

Lovibond® Water Testing

Tintometer® Group



Manual of Methods

MD 100 • MD 110 • MD 200

Alkalinity-m | Chlorine | Copper | Cyanuric Acid | Iron | pH

(EN) Manual of Methods

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(NL) Handboek Methoden

Zijde 328

(DE) Methodenhandbuch

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(FR) Méthodes Manuel

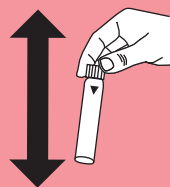
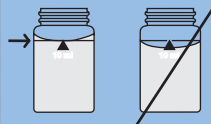
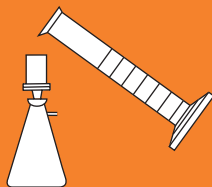
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(PT) Métodos Manual

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(ZH) 方法手册

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KS4.3 T / 20


Method name

Method number

Bar code for the detection of the methods

Measuring range

20

S:4.3

Display in the MD 100 / MD 110 / MD 200

Chemical Method

Instrument specific information

The test can be performed on the following devices. In addition, the required cuvette and the absorption range of the photometer are indicated.

Instrument Type	Cuvette	λ	Measuring Range
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	ø 24 mm	610 nm	0.1 - 4 mmol/l $K_{S4.3}$
SpectroDirect, XD 7000, XD 7500	ø 24 mm	615 nm	0.1 - 4 mmol/l $K_{S4.3}$

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Alka-M-Photometer	Tablet / 100	513210BT
Alka-M-Photometer	Tablet / 250	513211BT

Application List

- Waste Water Treatment
- Drinking Water Treatment
- Raw Water Treatment

Notes

1. The terms Alkalinity-m, m-Value, total alkalinity and Acid demand to $K_{S4.3}$ are identical.
2. For accurate results, exactly 10 ml of water sample must be used for the test.

Language codes ISO 639-1

Revision status

EN Handbook of Methods 01/20

Performing test procedure

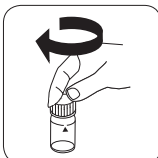
Implementation of the provision Acid capacity $K_{S_{4.3}}$ with Tablet

Select the method on the device

For this method, no ZERO measurements are to be carried out with the following devices: XD 7000, XD 7500



Fill 24 mm vial with **10 ml sample**.



Close vial(s).

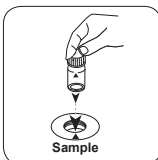


Place **sample vial** in the sample chamber. • Pay attention to the positioning.

• • •



Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. • Pay attention to the positioning.



Press the **TEST (XD: START)** button.

The result in Acid Capacity $K_{S_{4.3}}$ appears on the display.



Alkalinity-m T

M30

5 - 200 mg/L CaCO₃

tA

Acid / Indicator

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Alka-M-Photometer	Tablet / 100	513210BT
Alka-M-Photometer	Tablet / 250	513211BT

Notes

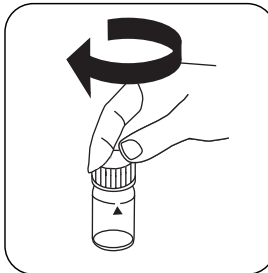
1. The terms Alkalinity-m, m-Value, total alkalinity and Acid demand to K_{S4.3} are identical.
2. For accurate results, exactly 10 ml of water sample must be used for the test.

Determination of Alkalinity, total = Alkalinity-m = m-Value with Tablet

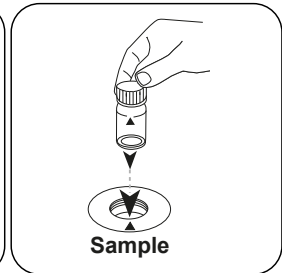
Select the method on the device.



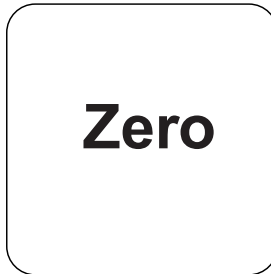
Fill 24 mm vial with **10 mL sample**.



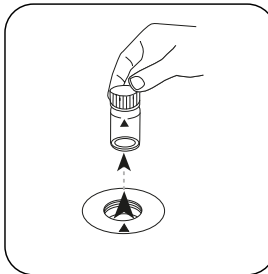
Close vial(s).



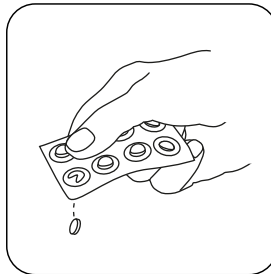
Place **sample vial** in the sample chamber. Pay attention to the positioning.



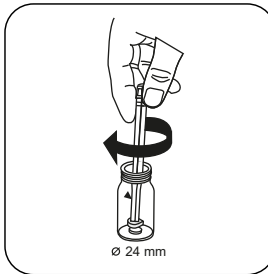
Press the **ZERO** button.



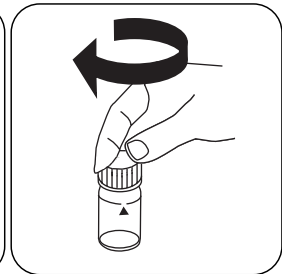
Remove the vial from the sample chamber.



Add **ALKA-M-PHOTOMETER** tablet.



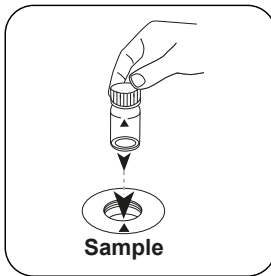
Crush tablet(s) by rotating slightly.



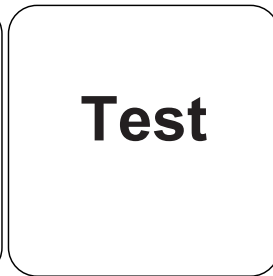
Close vial(s).



Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

The result in Alkalinity-m appears on the display.

EN

Analyses

The following table identifies the output values can be converted into other citation forms.

Unit	Cite form	Scale Factor
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

EN

Chemical Method

Acid / Indicator

Appendix

Derived from

EN ISO 9963-1

**Chlorine T****M100****0.01 - 6.0 mg/L Cl₂ ^{a)}****CL6****DPD****Material**

EN

Required material (partly optional):

Reagents	Packaging Unit	Part Number
DPD No.1	Tablet / 100	511050BT
DPD No. 1	Tablet / 250	511051BT
DPD No. 1	Tablet / 500	511052BT
DPD No. 3	Tablet / 100	511080BT
DPD No. 3	Tablet / 250	511081BT
DPD No. 3	Tablet / 500	511082BT
DPD No. 1 High Calcium ^{e)}	Tablet / 100	515740BT
DPD No. 1 High Calcium ^{e)}	Tablet / 250	515741BT
DPD No. 1 High Calcium ^{e)}	Tablet / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Tablet / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Tablet / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Tablet / 500	515732BT
DPD No. 4	Tablet / 100	511220BT
DPD No. 4	Tablet / 250	511221BT
DPD No. 4	Tablet / 500	511222BT
DPD No. 3 Evo	Tablet / 100	511420BT
DPD No. 3 Evo	Tablet / 250	511421BT
DPD No. 3 Evo	Tablet / 500	511422BT
DPD No. 4 Evo	Tablet / 100	511970BT
DPD No. 4 Evo	Tablet / 250	511971BT
DPD No. 4 Evo	Tablet / 500	511972BT

Available Standards

Title	Packaging Unit	Part Number
ValidCheck Chlorine 1,5 mg/l	1 pc.	48105510



Sampling

1. When preparing the sample, chlorine outgassing, e.g. through the pipette or shaking, must be avoided.
2. The analysis must take place immediately after taking the sample.

Preparation

1. Cleaning of vials:
As many household cleaners (e.g. dishwasher detergent) contain reducing substances, this can lead to lower results with the determination of chlorine. To avoid measurement errors, the glassware used should be free of chlorine consumption. To achieve this, all glassware should be placed in a sodium hypochlorite solution (0.1 g/L) for one hour and then rinsed thoroughly with deionised water.
2. For individual testing of free and total chlorine, the use of different sets of glassware is recommended (EN ISO 7393-2, 5.3)
3. The DPD colour development is carried out at a pH value of 6.2 to 6.5. The reagents therefore contain a buffer for the pH adjustment. Strong alkaline or acidic water samples must therefore be adjusted between pH 6 and pH 7 before the analysis (use 0.5 mol/L sulphuric acid or 1 mol/L sodium hydroxide).

Notes

1. Evo tablets can be used as an alternative to the corresponding standard tablet (e.g. DPD No.3 Evo instead of DPD No.3).

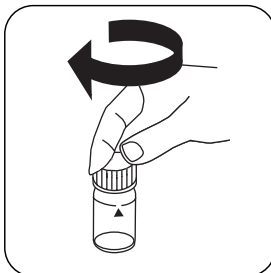


Determination of free chlorine with tablet

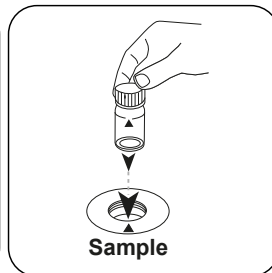
Select the method on the device.



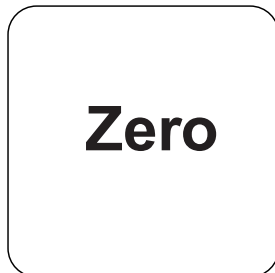
Fill 24 mm vial with **10 mL sample**.



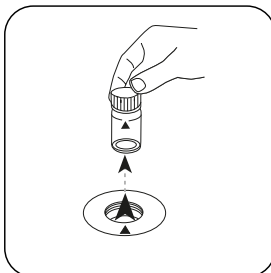
Close vial(s).



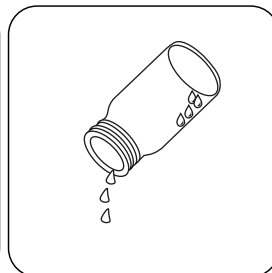
Place **sample vial** in the sample chamber. Pay attention to the positioning.



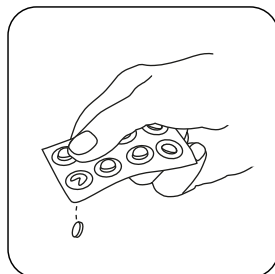
Press the **ZERO** button.



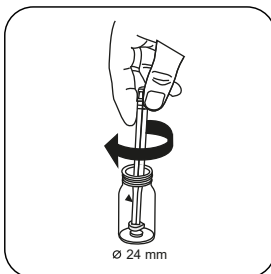
Remove the vial from the sample chamber.



Empty vial except for a few drops.



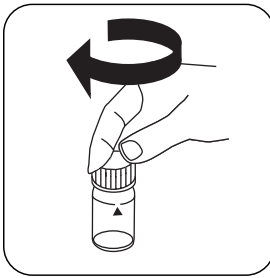
Add **DPD No. 1 tablet**.



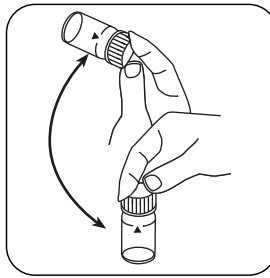
Crush tablet(s) by rotating slightly.



Fill up vial with **sample** to the **10 mL mark**.



Close vial(s).



Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.

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Test

Press the **TEST** (XD: **START**) button.

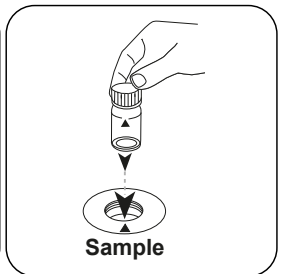
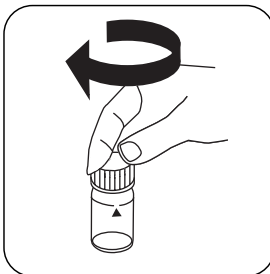
The result in mg/L free chlorine appears on the display.

Determination of total Chlorine with tablet

Select the method on the device.



Fill 24 mm vial with **10 mL sample**. Close vial(s).



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Zero

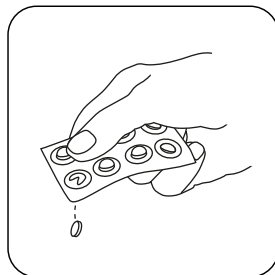
Press the **ZERO** button.



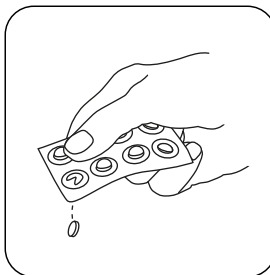
Remove the vial from the sample chamber.



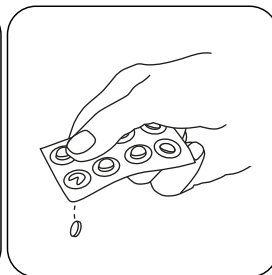
Empty vial except for a few drops.



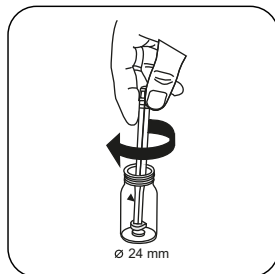
Add **DPD No. 1** tablet .



Add **DPD No. 3** tablet .



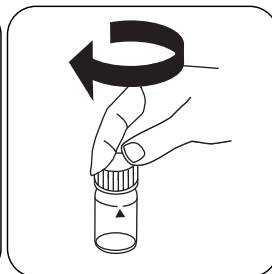
As an alternative to DPD No. 1 and No. 3 tablets, a DPD No. 4 tablet can be added.



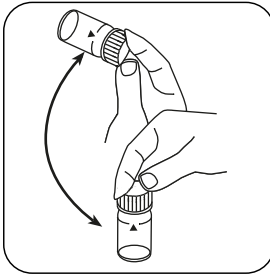
Crush tablet(s) by rotating slightly.



Fill up vial with **sample** to the **10 mL** mark.



Close vial(s).



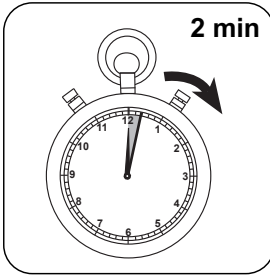
Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.



Wait for **2 minute(s) reaction time**.

Once the reaction period is finished, the measurement takes place automatically. The result in mg/L total Chlorine appears on the display.



Chemical Method

DPD

Appendix

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Interferences

Persistent Interferences

- All oxidising agents in the samples react like chlorine, which leads to higher results.

Removeable Interferences

- Interference from copper and iron (III) are eliminated by the addition of EDTA.
- The use of reagent tablets in samples with high calcium content* and/or high conductivity* can lead to turbidity of the sample and therefore incorrect measurements. In this case, the alternative reagent tablet DPD No.1 High Calcium and reagent tablet DPD No.3 High Calcium should be used.
*it is not possible to give exact values, because the development of turbidity depends on the composition and nature of the sample.
- Concentrations above 10 mg/L chlorine, in the event of using fluid reagents, can lead to results within the measuring range of up to 0 mg/L. In the event of a high concentration of chlorine, the sample must be diluted with chlorine-free water. 10 mL of the diluted sample should be mixed with the reagent and the measurement taken again (plausibility test).

Interference	from / [mg/L]
CrO ₄ ²⁻	0.01
MnO ₂	0.01

Method Validation

Limit of Detection	0.02 mg/L
Limit of Quantification	0.06 mg/L
End of Measuring Range	6 mg/L
Sensitivity	2.05 mg/L / Abs
Confidence Intervall	0.04 mg/L
Standard Deviation	0.019 mg/L
Variation Coefficient	0.87 %

Conformity

EN ISO 7393-2



^{a)} determination of free, combined and total | ^{b)} alternative reagent, used instead of DPD No.1/No.3 in case of turbidity in the water sample caused by high concentration of calcium and/or high conductivity

**Chlorine L****M101****0.02 - 4.0 mg/L Cl₂^{a)}****CL6****DPD****Material**

EN

Required material (partly optional):

Reagents	Packaging Unit	Part Number
DPD 1 Buffer Solution, Blue Bottle	15 mL	471010
DPD 1 Buffer Solution	100 mL	471011
DPD 1 Buffer Solution	1 pc.	471016
DPD 1 Reagent Solution, Green Bottle	15 mL	471020
DPD 1 Reagent Solution	100 mL	471021
DPD 1 Reagent Solution	1 pc.	471026
DPD 3 Solution, Red Bottle	15 mL	471030
DPD 3 Solution	100 mL	471031
DPD 3 Solution	1 pc.	471036
DPD Reagent Set	1 pc.	471056

Available Standards

Title	Packaging Unit	Part Number
ValidCheck Chlorine 1,5 mg/l	1 pc.	48105510

Sampling

1. When preparing the sample, Chlorine outgassing, e.g. through the pipette or shaking, must be avoided.
2. The analysis must take place immediately after taking the sample.



Preparation

1. Cleaning of vials:
As many household cleaners (e.g. dishwasher detergent) contain reducing substances, this can lead to lower results with the determination of Chlorine. To avoid measurement errors, the glassware used should be free of chlorine consumption. To achieve this, all glassware should be placed in a sodium hypochlorite solution (0.1 g/L) for one hour and then rinsed thoroughly with deionised water.
2. For individual testing of free and total Chlorine, the use of different sets of glassware is recommended (EN ISO 7393-2, 5.3)
3. The DPD colour development is carried out at a pH value of 6.2 to 6.5. The reagents therefore contain a buffer for the pH adjustment. Strong alkaline or acidic water samples must therefore be adjusted between pH 6 and pH 7 before the analysis (use 0.5 mol/l Sulphuric acid or 1 mol/l Sodium hydroxide).

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Notes

1. After use, ensure the cuvettes are once again closed with the respective same-coloured screw caps.
2. Reagent sets are to be stored in the cool at +6 °C to +10 °C.

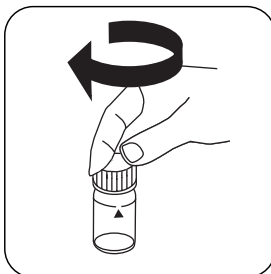


Determination of free chlorine with liquid reagent

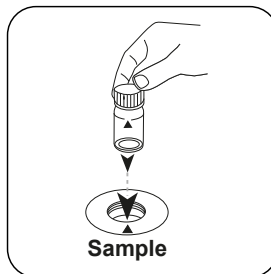
Select the method on the device.



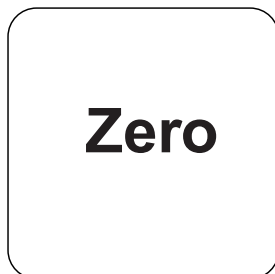
Fill 24 mm vial with **10 mL sample**.



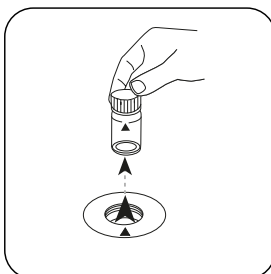
Close vial(s).



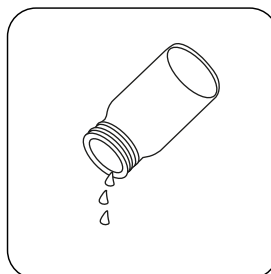
Place **sample vial** in the sample chamber. Pay attention to the positioning.



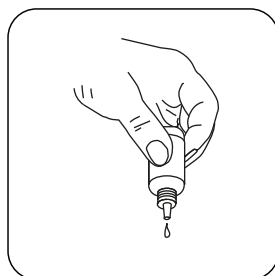
Press the **ZERO** button.



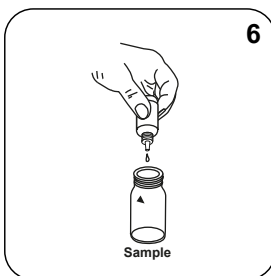
Remove the vial from the sample chamber.



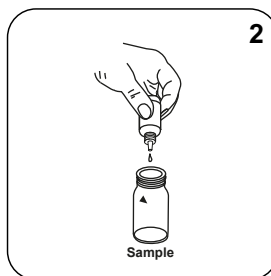
Empty vial.



Hold cuvettes vertically and add equal drops by pressing slowly.



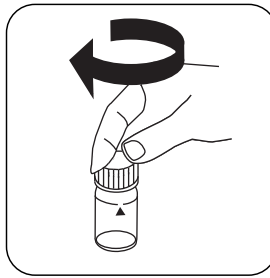
Add **6 drops DPD 1 Buffer Solution** to the **sample vial**.



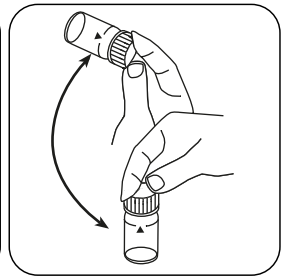
Add **2 drops DPD 1 Reagent Solution** to the **sample vial**.



Fill up vial with **sample** to the **10 mL** mark.

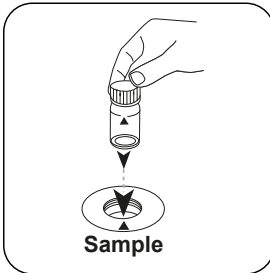


Close vial(s).

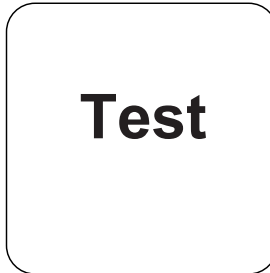


Invert several times to mix the contents.

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Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

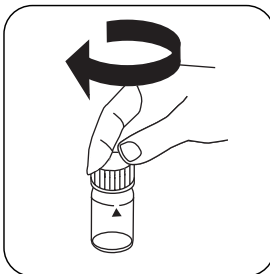
The result in mg/L free chlorine appears on the display.

Determination of totale Chlorine with liquid reagent

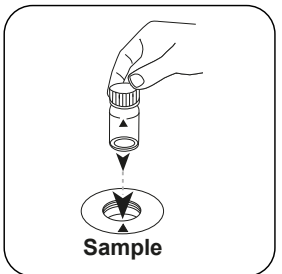
Select the method on the device.



Fill 24 mm vial with **10 mL** **sample**.



Close vial(s).

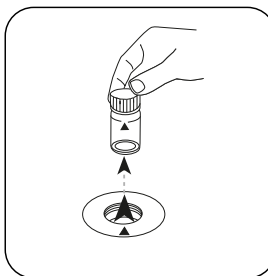


Place **sample vial** in the sample chamber. Pay attention to the positioning.

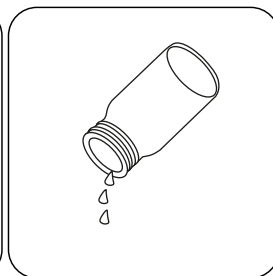


Zero

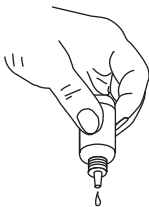
Press the **ZERO** button.



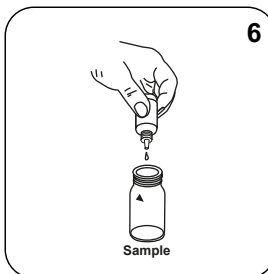
Remove the vial from the sample chamber.



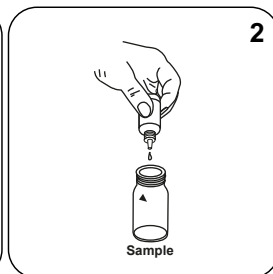
Empty vial.



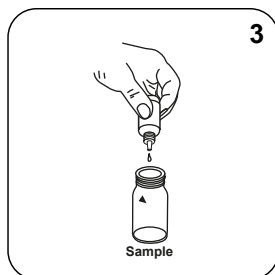
Hold cuvettes vertically and add equal drops by pressing slowly.



Add **6 drops DPD 1 Buffer Solution** to the sample vial.



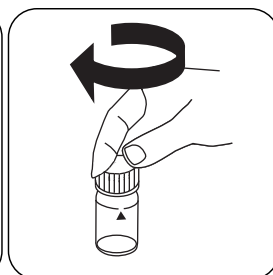
Add **2 drops DPD 1 Reagent Solution** to the sample vial.



Add **3 drops DPD 3 Solution** to the sample vial.



Fill up vial with **sample** to the **10 mL** mark.



Close vial(s).



Invert several times to mix the contents.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

EN



Wait for **2 minute(s) reaction time**.

Once the reaction period is finished, the measurement takes place automatically. The result in mg/L total Chlorine appears on the display.



Chemical Method

DPD

Appendix

EN

Interferences

Persistent Interferences

- All oxidising agents in the samples react like chlorine, which leads to higher results.

Removeable Interferences

- Interference from Copper and Iron (III) are eliminated by the addition of EDTA.
- Concentrations above 4 mg/L Chlorine, in the event of using fluid reagents, can lead to results within the measuring range of up to 0 mg/L. In this case, the sample must be diluted with chlorine-free water. 10 ml of the diluted sample should be mixed with the reagent and the measurement taken again (plausibility test).

Interference	from / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Conformity

EN ISO 7393-2

^{a)} determination of free, combined and total



Chlorine HR T

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

Material

EN

Required material (partly optional):

Reagents	Packaging Unit	Part Number
DPD No. 1 HR	Tablet / 100	511500BT
DPD No. 1 HR	Tablet / 250	511501BT
DPD No. 1 HR	Tablet / 500	511502BT
DPD No. 3 HR	Tablet / 100	511590BT
DPD No. 3 HR	Tablet / 250	511591BT
DPD No. 3 HR	Tablet / 500	511592BT
Set DPD No. 1 HR/No. 3 HR 100 Pc. #	100 each	517791BT
Set DPD No. 1 HR/No. 3 HR 250 Pc. #	250 each	517792BT
DPD No. 1 High Calcium ^{e)}	Tablet / 100	515740BT
DPD No. 1 High Calcium ^{e)}	Tablet / 250	515741BT
DPD No. 1 High Calcium ^{e)}	Tablet / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Tablet / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Tablet / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Tablet / 500	515732BT
DPD No.3 HR Evo	Tablet / 100	511920BT
DPD No. 3 HREvo	Tablet / 250	511921BT
DPD No. 3 HREvo	Tablet / 500	511922BT

Sampling

1. When preparing the sample, chlorine outgassing, e.g. through the pipette or shaking, must be avoided.
2. The analysis must take place immediately after taking the sample.



Preparation

1. Cleaning of vials:
As many household cleaners (e.g. dishwasher detergent) contain reducing substances, this can lead to lower results with the determination of chlorine. To avoid measurement errors, the glassware used should be free of chlorine consumption. To achieve this, all glassware should be placed in a sodium hypochlorite solution (0.1 g/L) for one hour and then rinsed thoroughly with deionised water.
2. For individual testing of free and total chlorine, the use of different sets of glassware is recommended (EN ISO 7393-2, 5.3)
3. The DPD colour development is carried out at a pH value of 6.2 to 6.5. The reagents therefore contain a buffer for the pH adjustment. Strong alkaline or acidic water samples must therefore be adjusted between pH 6 and pH 7 before the analysis (use 0.5 mol/L sulphuric acid or 1 mol/L sodium hydroxide).

EN

Notes

1. Evo tablets can be used as an alternative to the corresponding standard tablet (e.g. DPD No.3 Evo instead of DPD No.3).

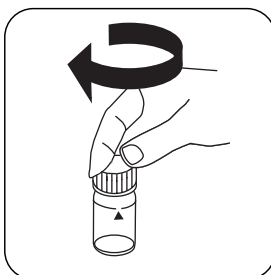


Determination of free chlorine HR with tablet

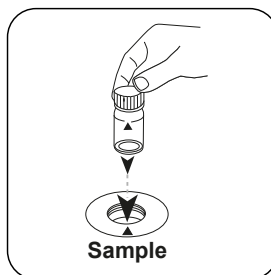
Select the method on the device.



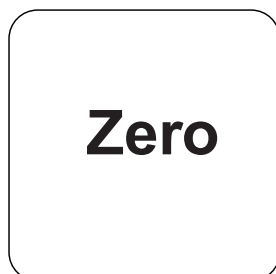
Fill 24 mm vial with **10 mL sample**.



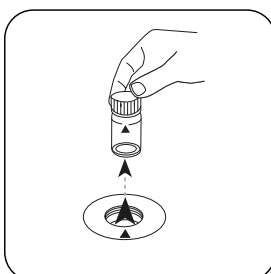
Close vial(s).



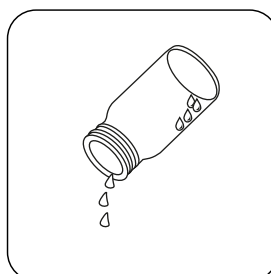
Place **sample vial** in the sample chamber. Pay attention to the positioning.



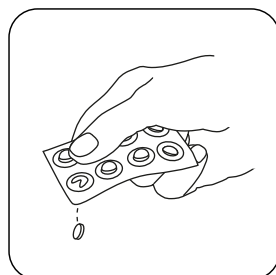
Press the **ZERO** button.



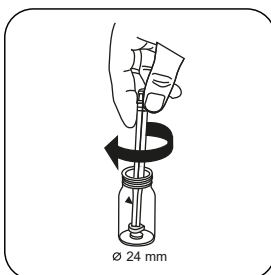
Remove the vial from the sample chamber.



Empty vial except for a few drops.



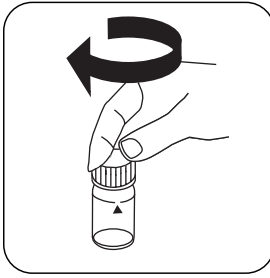
Add **DPD No. 1 HR tablet**.



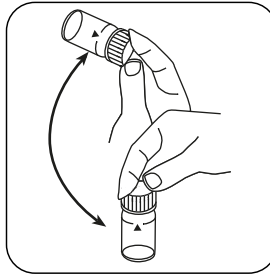
Crush tablet(s) by rotating slightly.



Fill up vial with **sample** to the **10 mL mark**.



Close vial(s).



Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.

EN

Test

Press the **TEST** (XD: **START**) button.

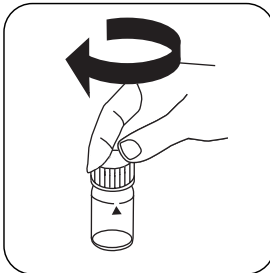
The result in mg/L free chlorine appears on the display.

Determination of total Chlorine HR with tablet

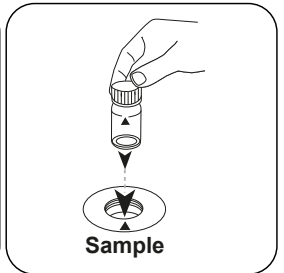
Select the method on the device.



Fill 24 mm vial with **10 mL sample**.



Close vial(s).

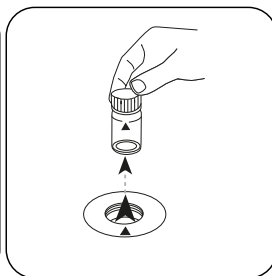


Place **sample vial** in the sample chamber. Pay attention to the positioning.



Zero

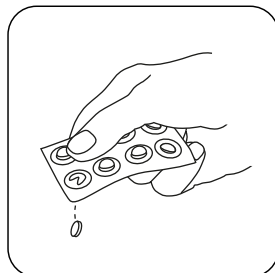
Press the **ZERO** button.



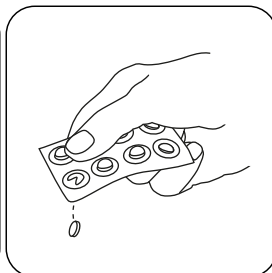
Remove the vial from the sample chamber.



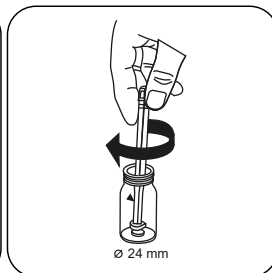
Empty vial except for a few drops.



Add **DPD No. 1 HR tablet**.



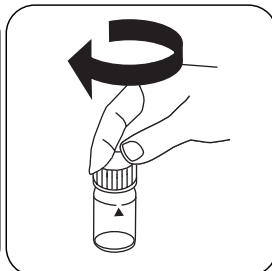
Add **DPD No. 3 HR tablet**.



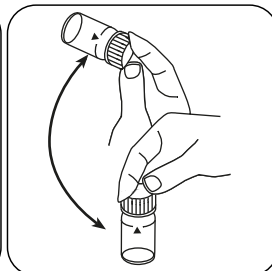
Crush tablet(s) by rotating slightly.



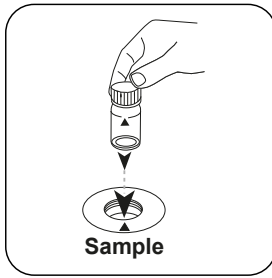
Fill up vial with **sample** to the **10 mL mark**.



Close vial(s).



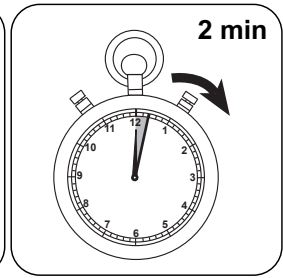
Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.



Wait for **2 minute(s)** reaction time.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/L total Chlorine appears on the display.



Chemical Method

DPD

Appendix

EN

Interferences

Persistent Interferences

- All oxidising agents in the samples react like chlorine, which leads to higher results.

Removeable Interferences

- Interference from Copper and Iron (III) are eliminated by the addition of EDTA.
- The use of reagent tablets in samples with high Calcium content* and/or high conductivity* can lead to turbidity of the sample and therefore incorrect measurements. In this case, the alternative reagent tablet DPD No. 1 High Calcium and reagent tablet DPD No. 3 High Calcium should be used.
*it is not possible to give exact values, because the development of turbidity depends on the composition and nature of the sample.

Conformity

EN ISO 7393-2

^{a)} determination of free, combined and total | ^{a)} alternative reagent, used instead of DPD No.1/No.3 in case of turbidity in the water sample caused by high concentration of calcium and/or high conductivity | * including stirring rod, 10 cm



Copper T

M150

0.05 - 5 mg/L Cu^{a)}

Cu

Biquinoline

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Copper No. 1	Tablet / 100	513550BT
Copper No. 1	Tablet / 250	513551BT
Copper No. 2	Tablet / 100	513560BT
Copper No. 2	Tablet / 250	513561BT
Set Copper No. 1/No. 2 100 Pc.#	100 each	517691BT
Set Copper No. 1/No. 2 250 Pc.#	250 each	517692BT
ValidCheck Copper 2 mg/l	1 pc.	48141525

Preparation

1. Strong alkaline or acidic water samples must be adjusted to pH 4 to 6 before analysis.

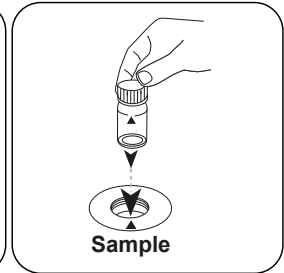
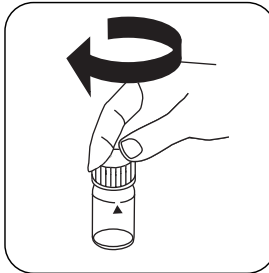
Determination of Copper, free with tablet

Select the method on the device.

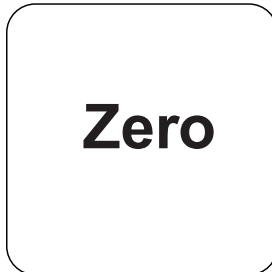
In addition, choose the test: free



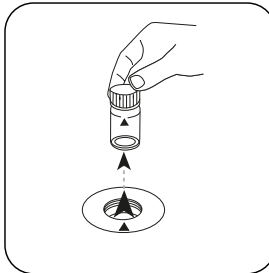
Fill 24 mm vial with **10 mL** **sample**.
Close vial(s).



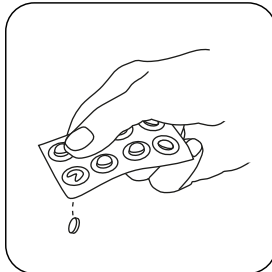
Place **sample vial** in the sample chamber. Pay attention to the positioning.



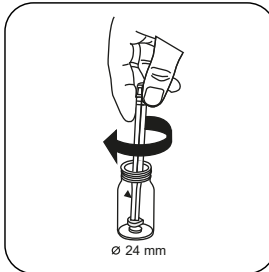
Press the **ZERO** button.



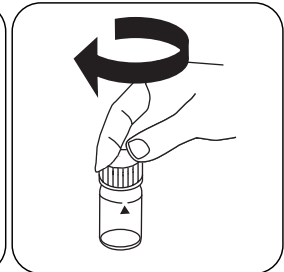
Remove the vial from the sample chamber.



Add **COPPER No. 1** tablet



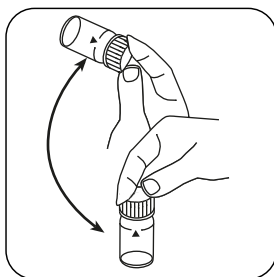
Crush tablet(s) by rotating slightly.



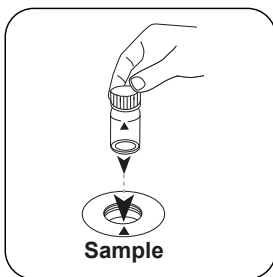
Close vial(s).



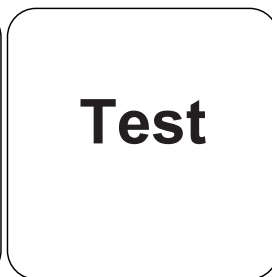
EN



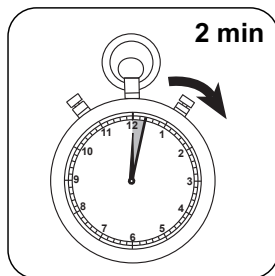
Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.



Wait for **2 minute(s) reaction time**.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/L free Copper appears on the display.

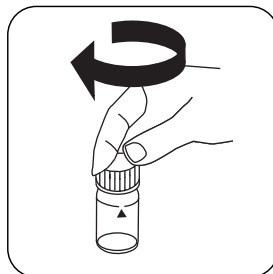
Determination of Copper, total with tablet

Select the method on the device.

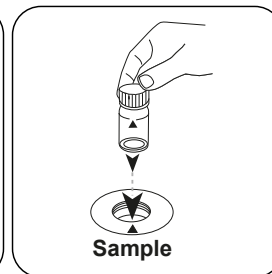
In addition, choose the test: total



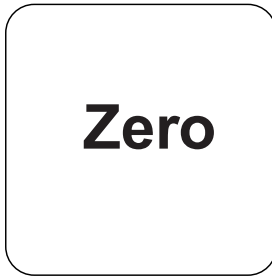
Fill 24 mm vial with **10 mL sample**.



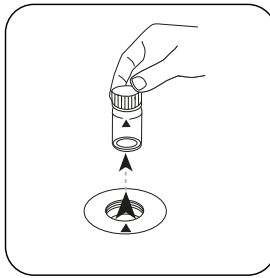
Close vial(s).



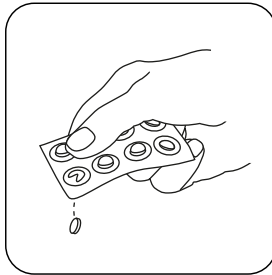
Place **sample vial** in the sample chamber. Pay attention to the positioning.



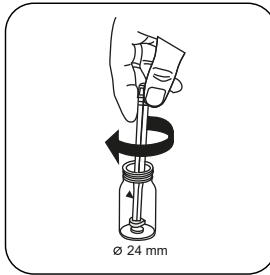
Press the **ZERO** button.



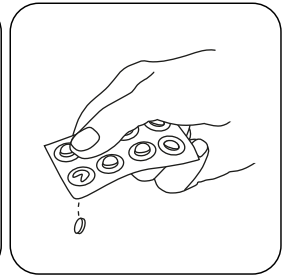
Remove the vial from the sample chamber.



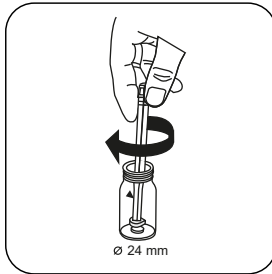
Add **COPPER No. 1 tablet**



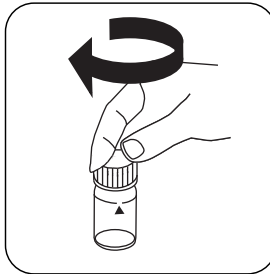
Crush tablet(s) by rotating slightly and dissolve.



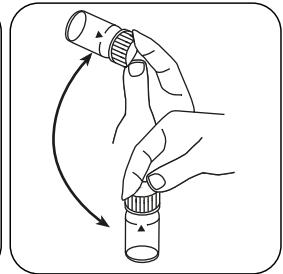
Add **COPPER No. 2 tablet**



Crush tablet(s) by rotating slightly.

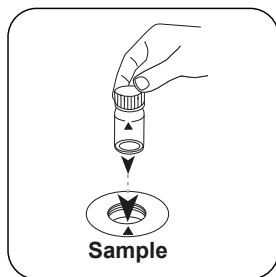


Close vial(s).

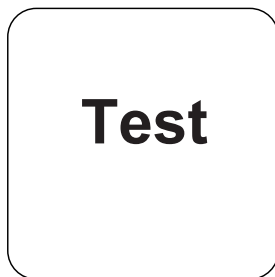


Dissolve tablet(s) by inverting.

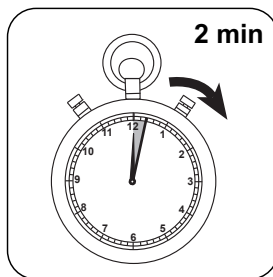
EN



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.



Wait for **2 minute(s)** reaction time.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/L total Copper appears on the display.

Chemical Method

Biquinoline

Appendix

Interferences

Persistent Interferences

1. Cyanide CN^- and Silver Ag^+ interfere with the test result.

Method Validation

Limit of Detection	0.05 mg/L
Limit of Quantification	0.15 mg/L
End of Measuring Range	5 mg/L
Sensitivity	3.8 mg/L / Abs
Confidence Intervall	0.026 mg/L
Standard Deviation	0.011 mg/L
Variation Coefficient	0.42 %

Bibliography

Photometrische Analyse, Lange/Vedjerek, Verlag Chemie 1980

^{a)} determination of free, combined and total | ^{*} including stirring rod, 10 cm



CyA T

M160

10 - 160 mg/L CyA

CyA

Melamine

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
CyA-Test	Tablet / 100	511370BT
CyA-Test	Tablet / 250	511371BT
Deionised Water	100 mL	461275
Deionised Water	250 mL	457022

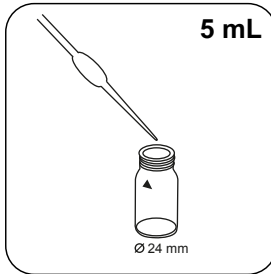
Notes

1. Cyanuric acid causes an extremely fine distributed turbidity with a milky appearance. Individual particles are not attributable to the presence of cyanuric acid.

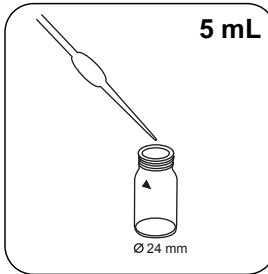


Determination of Cyanuric Acid Test with Tablet

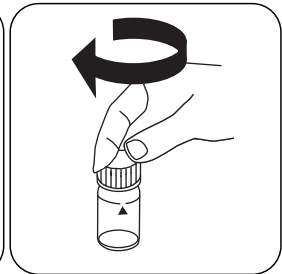
Select the method on the device.



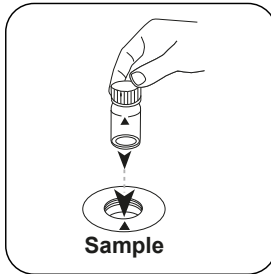
Fill 24 mm vial with **5 mL deionised water**.



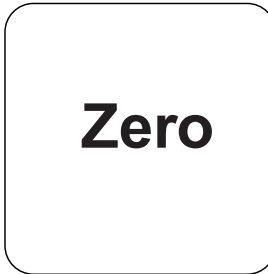
Put **5 mL sample** in the vial.



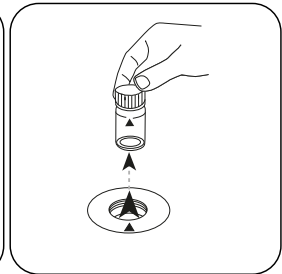
Close vial(s).



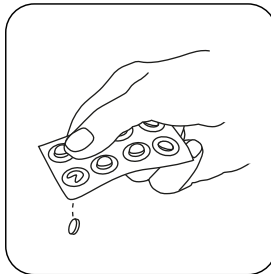
Place **sample vial** in the sample chamber. Pay attention to the positioning.



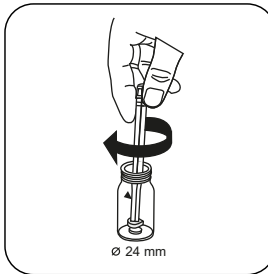
Press the **ZERO** button.



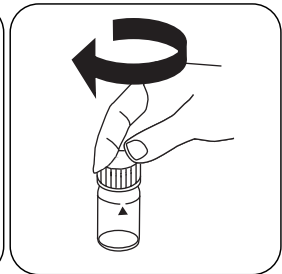
Remove the vial from the sample chamber.



Add **CyA-Test tablet**.



Crush tablet(s) by rotating slightly.



Close vial(s).



Invert several times to mix the contents (for at least 60 s until the tablet is completely dissolved).



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

The result in mg/L Cyanuric Acid appears on the display.

EN



Chemical Method

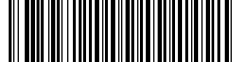
Melamine

Interferences

Persistent Interferences

1. Undissolved particles may lead to higher results. Therefore, it is important to dissolve the Tablet completely.

EN



Iron T

M220

0.02 - 1 mg/L Fe

FE

Ferrozine / Thioglycolate

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Iron II LR (Fe ²⁺)	Tablet / 100	515420BT
Iron II LR (Fe ²⁺)	Tablet / 250	515421BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablet / 100	515370BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablet / 250	515371BT

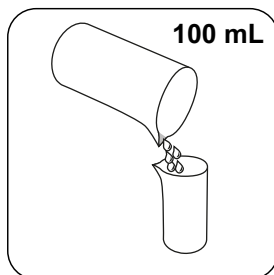
Preparation

1. Water that has been treated with organic compounds such as corrosion inhibitors, must be oxidised where necessary to break down the iron complex. 1 ml of concentrated Sulphuric acid ($\geq 95\%$) and 1 ml concentrated Nitric acid ($\geq 65\%$) is therefore added to 100 ml water sample and boiled down to approximately half the volume. After cooling down, the digestion procedure is continued.

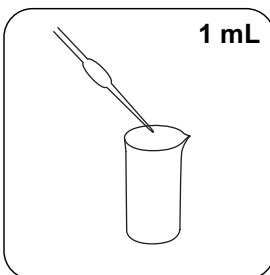
Notes

1. This method is for the determination of total dissolved Fe²⁺ and Fe³⁺.
2. For the determination of Fe²⁺, the IRON (II) LR Tablet, instead of the IRON LR Tablet is used.

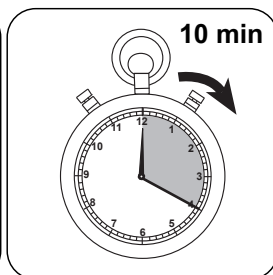
Digestion



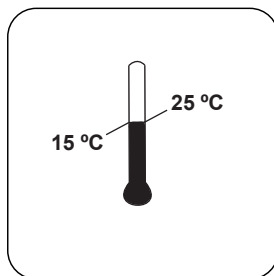
Fill a suitable sample vessel with **100 mL sample**.



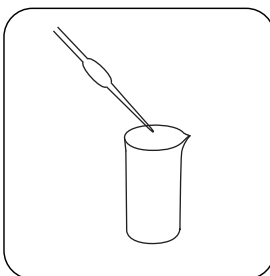
Add **1 mL concentrated sulfuric acid ($\geq 95\%$)**.



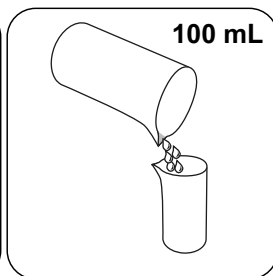
The sample is to be **heated for 10 minutes**, or for as long as it takes for everything to be completely dissolved.



Allow the sample to cool to room temperature.



Adjust **pH-value** of the sample with **ammonia solution (10-25 %)** to 3-5.



Fill the sample with **deionised water to 100 mL**.

This sample is used for the analysis of total solved and dissolved Iron.

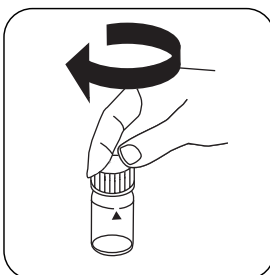
Determination of Iron (II,III), dissolved with Tablet

Select the method on the device.

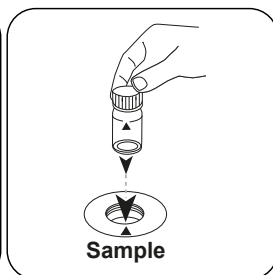
For testing of **dissolved and undissolved Iron**, carry out the described **digestion**.



Fill 24 mm vial with **10 mL sample**.



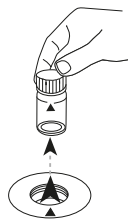
Close vial(s).



Place **sample vial** in the sample chamber. Pay attention to the positioning.

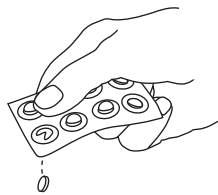


Zero

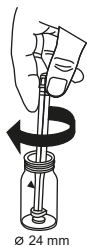


Press the **ZERO** button.

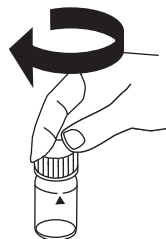
Remove the vial from the sample chamber.



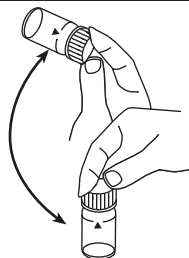
Add **IRON LR tablet**.



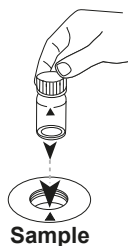
Crush tablet(s) by rotating slightly.



Close vial(s).



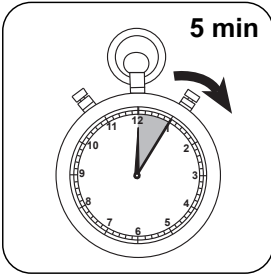
Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.

Test

Press the **TEST (XD: START)** button.



Wait for **5 minute(s)**
reaction time.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/L Iron appears on the display.

EN



Chemical Method

Ferrozine / Thioglycolate

Appendix

EN

Interferences

Removeable Interferences

1. The presence of copper increases the test result by 10 %. At a concentration of 10 mg/L copper in the sample, the measurement result is increased by 1 mg/L iron. The interference can be eliminated by the addition of thiourea

Method Validation

Limit of Detection	0.01 mg/L
Limit of Quantification	0.016 mg/L
End of Measuring Range	1 mg/L
Sensitivity	0.92 mg/L / Abs
Confidence Intervall	0.013 mg/L
Standard Deviation	0.005 mg/L
Variation Coefficient	1.23 %

Bibliography

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, p. 102



pH-value T

M330

6.5 - 8.4 pH

PH

Phenol Red

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Phenol Red Photometer	Tablet / 100	511770BT
Phenol Red Photometer	Tablet / 250	511771BT
Phenol Red Photometer	Tablet / 500	511772BT

Notes

1. For photometric determination of pH values only use PHENOL RED tablets in black printed foil pack and marked with PHOTOMETER.

Determination of pH-value with Tablet

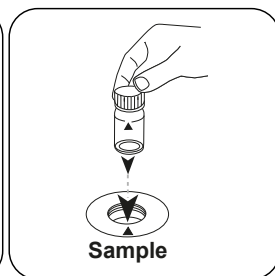
Select the method on the device.



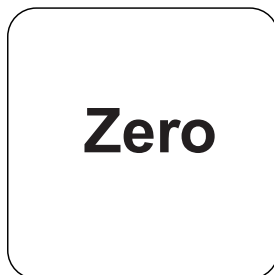
Fill 24 mm vial with **10 mL sample**.



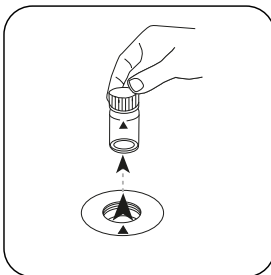
Close vial(s).



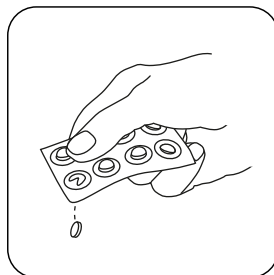
Place **sample vial** in the sample chamber. Pay attention to the positioning.



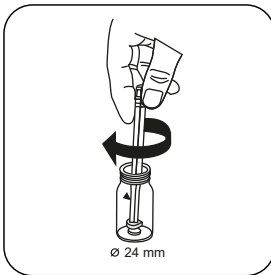
Press the **ZERO** button.



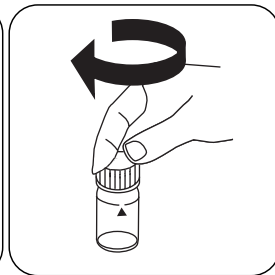
Remove the vial from the sample chamber.



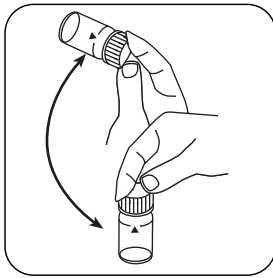
Add **PHENOL RED PHOTOMETER** tablet.



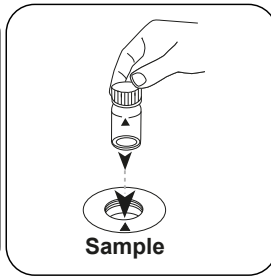
Crush tablet(s) by rotating slightly.



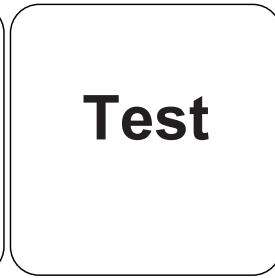
Close vial(s).



Dissolve tablet(s) by inverting.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

The result in pH value appears on the display.

Chemical Method

Phenol Red

Appendix

Interferences

EN

Persistent Interferences

1. Water samples with little Carbonate hardness* can lead to false pH values.
* $K_{S4,3} < 0.7 \text{ mmol/l} \triangleq \text{total alkalinity} < 35 \text{ mg/L CaCO}_3$.

Removeable Interferences

1. pH values below 6.5 and above 8.4 can produce results inside the measuring range. A plausibility test (pH-meter) is recommended.
2. Salt error
For salt concentrations below 2 g/L, no significant error, is expected due to the salt concentration of the reagent tablet. For higher salt concentrations the measurement values have to be adjusted as follows:

Salt content per sample in g/L	30 (seawater)	60	120	180
Correction	-0.15 ¹⁾	-0.21 ²⁾	-0.26 ²⁾	-0.29 ²⁾

¹⁾ according to Kolthoff (1922)

²⁾ according to Parson and Douglas (1926)

Bibliography

Colorimetric Chemical Analytical Methods, 9th Edition, London



pH value L

M331

6.5 - 8.4 pH

PH

Phenol Red

EN

Material

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Phenol Red Solution	15 mL	471040
Phenol Red Solution	100 mL	471041
Phenol Red Solution in 6-pack	1 pc.	471046

Preparation

1. Due to differing drop sizes results can show a discrepancy in accuracy by comparison with tablets.
This can be minimised by using a pipette (0.18 ml equivalent to 6 drops).

Notes

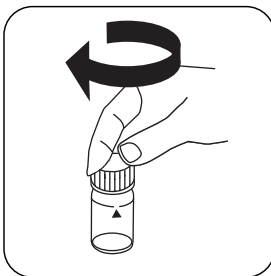
1. After use, ensure the cuvette is once again closed with the same-coloured screw caps.
2. Reagents are to be stored in the cool at +6 °C to +10 °C.

Determination of pH-value with liquid reagent

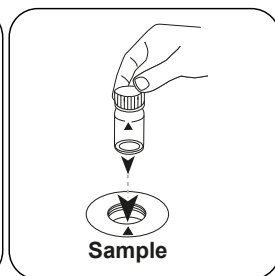
Select the method on the device.



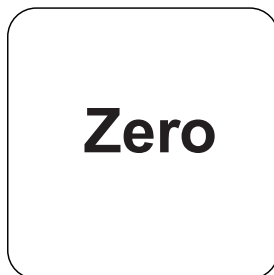
Fill 24 mm vial with **10 mL sample**.



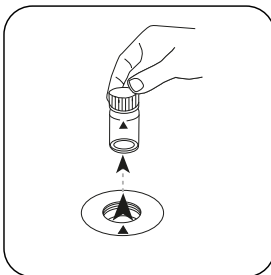
Close vial(s).



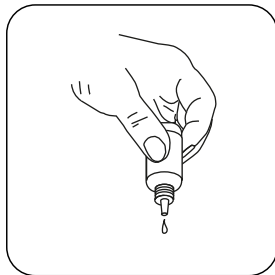
Place **sample vial** in the sample chamber. Pay attention to the positioning.



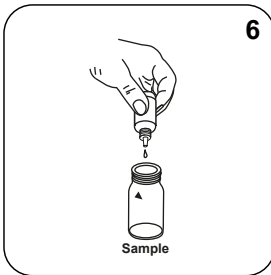
Press the **ZERO** button.



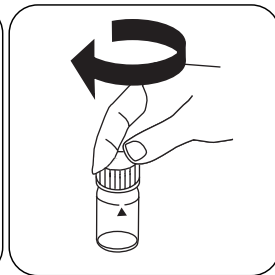
Remove the vial from the sample chamber.



Hold cuvettes vertically and add equal drops by pressing slowly.



Add **6 drops PHENOL Red-Lösung** to the **sample vial**.



Close vial(s).



Invert several times to mix the contents.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**) button.

The result in pH value appears on the display.

EN

Chemical Method

Phenol Red

Appendix

Interferences

EN

Removeable Interferences


1. Salt error Correction of test results (average values) for samples with salt contents of:

2.	Salt content of the sample	Correction
	30 g/L (seawater)	-0.15 ¹⁾
	60 g/L	-0.21 ²⁾
	120 g/L	-0.26 ²⁾
	180 g/L	-0.29 ²⁾
	¹⁾ according to Kolthoff (1922)	²⁾ according to Parson and Douglas (1926)

3. When testing chlorinated water the residual chlorine contents can influence the colour reaction of the liquid reagent. This can be avoided by adding a small crystal of Sodiumthiosulphate ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) to the sample solution before adding the PHENOL RED solution.

Bibliography

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20


Methoden Name

Methodennummer

Barcode zur Methodenerkennung

Messbereich

20

S:4.3

Chemische Methode

$K_{S_{4.3} T}$
0,1 - 4 mmol/l $K_{S_{4.3}}$
Säure / Indikator

Displayanzeige im MD 100 MD 110 / MD 200

Chemische Methode

Instrumentenspezifische Informationen

Der Test kann auf den folgenden Geräten durchgeführt werden. Zusätzlich sind die benötigte Küvette und der Absorptionsbereich der Photometer angegeben.

Geräte	Küvette	λ	Messbereich
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	ø 24 mm	610 nm	0,1 - 4 mmol/l $K_{S_{4.3}}$
SpectroDirect, XD 7000, XD 7500	ø 24 mm	615 nm	0,1 - 4 mmol/l $K_{S_{4.3}}$

Material

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Alka-M-Photometer	Tablette / 100	513210BT
Alka-M-Photometer	Tablette / 250	513211BT

Anwendungsbereich

- Abwasserbehandlung
- Trinkwasseraufbereitung
- Rohwasserbehandlung

Anmerkungen

1. Die Begriffe Alkalität-m, m-Wert, Gesamtalkalität und Säurekapazität $K_{S_{4.3}}$ sind identisch.
2. Die exakte Einhaltung des Probevolumens von 10 ml ist für die Genauigkeit des Analyseergebnisses entscheidend.

Sprachkürzel nach ISO 639-1

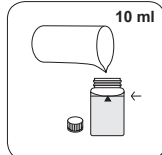
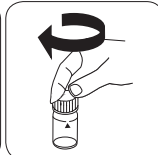
Revisionsstand

DE Methodenhandbuch 01/20

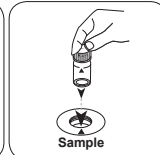
Durchführung der
Messung**Durchführung der Bestimmung Säurekapazität $K_{s4,3}$ mit Tablette**

Die Methode im Gerät auswählen.

Für diese Methode muss bei folgenden Geräten keine ZERO-Messung durchgeführt werden: XD 7000, XD 7500

24-mm-Küvette mit **10 ml Probe** füllen.

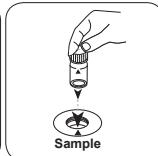
Küvette(n) verschließen.

Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.

• • •



Tablette(n) durch Umschwenken lösen.

Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.Taste **TEST** (XD: **START**) drücken.In der Anzeige erscheint das Ergebnis als Säurekapazität $K_{s4,3}$.



Alkalität-m T

M30

5 - 200 mg/L CaCO₃

tA

Säure / Indikator

DE

Material

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Alka-M-Photometer	Tablette / 100	513210BT
Alka-M-Photometer	Tablette / 250	513211BT

Anmerkungen

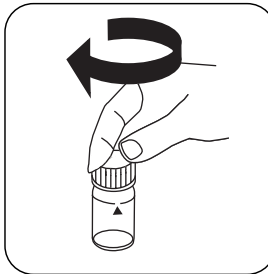
1. Die Begriffe Alkalität-m, m-Wert, Gesamtalkalität und Säurekapazität $K_{s4,3}$ sind identisch.
2. Die exakte Einhaltung des Probevolumens von 10 ml ist für die Genauigkeit des Analyseergebnisses entscheidend.

Durchführung der Bestimmung Alkalität, total= Alkalität-m = m-Wert mit Tablette

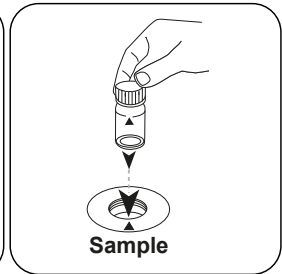
Die Methode im Gerät auswählen.



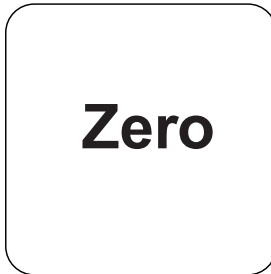
24-mm-Küvette mit **10 mL Probe** füllen.



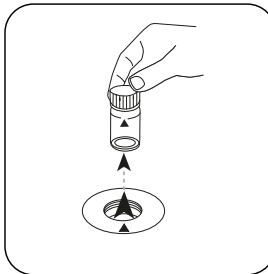
Küvette(n) verschließen.



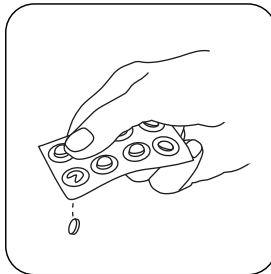
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



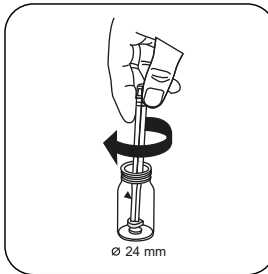
Taste **ZERO** drücken.



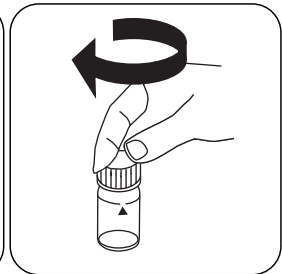
Küvette aus dem Messschacht nehmen.



Eine **ALKA-M-PHOTOMETER** Tablette zugeben.



Tablette(n) unter leichter Drehung zerdrücken.



Küvette(n) verschließen.

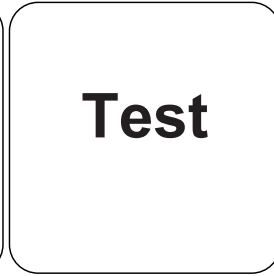
DE



Tablette(n) durch Umschwenken lösen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis als Alkalität-m.

DE

Auswertung

Die folgende Tabelle gibt an wie die ausgegebenen Werte in andere Zitierformen umgewandelt werden können.

Einheit	Zitierform	Umrechnungsfaktor
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

DE

Chemische Methode

Säure / Indikator

Appendix

Abgeleitet von
EN ISO 9963-1

**Chlor T****M100****0,01 - 6,0 mg/L Cl₂ ^{a)}****CL6****DPD****Material**

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
DPD No.1	Tablette / 100	511050BT
DPD No. 1	Tablette / 250	511051BT
DPD No. 1	Tablette / 500	511052BT
DPD No. 3	Tablette / 100	511080BT
DPD No. 3	Tablette / 250	511081BT
DPD No. 3	Tablette / 500	511082BT
DPD No. 1 High Calcium ^{e)}	Tablette / 100	515740BT
DPD No. 1 High Calcium ^{e)}	Tablette / 250	515741BT
DPD No. 1 High Calcium ^{e)}	Tablette / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Tablette / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Tablette / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Tablette / 500	515732BT
DPD No. 4	Tablette / 100	511220BT
DPD No. 4	Tablette / 250	511221BT
DPD No. 4	Tablette / 500	511222BT
DPD No. 3 Evo	Tablette / 100	511420BT
DPD No. 3 Evo	Tablette / 250	511421BT
DPD No. 3 Evo	Tablette / 500	511422BT
DPD No.4 Evo	Tablette / 100	511970BT
DPD No. 4 Evo	Tablette / 250	511971BT
DPD No. 4 Evo	Tablette / 500	511972BT

Verfügbare Standards

Titel	Verpackungseinheit	Bestell-Nr.
ValidCheck Chlor 1,5 mg/L	1 St.	48105510

Probenahme

1. Bei der Probenvorbereitung muss das Ausgasen von Chlor, z.B. durch Pipettieren und Schütteln, vermieden werden.
2. Die Analyse muss unmittelbar nach der Probenahme erfolgen.

Vorbereitung

1. Reinigung der Küvetten:
Da viele Haushaltsreiniger (z.B. Geschirrspülmittel) reduzierende Stoffe enthalten, kann es bei der Bestimmung von Chlor zu Minderbefunden kommen. Um diesen Messfehler auszuschließen, sollten die Glasgeräte chlorzehrungsfrei sein. Dazu werden die Glasgeräte für eine Stunde unter Natriumhypochloritlösung (0,1 g/L) aufbewahrt und danach gründlich mit VE-Wasser (Vollentsalztes Wasser) gespült.
2. Für die Einzelbestimmung von freiem Chlor und Gesamtchlor ist es sinnvoll, jeweils einen eigenen Satz Küvetten zu verwenden (siehe EN ISO 7393-2, Abs. 5.3).
3. Die DPD-Farbentwicklung erfolgt bei einem pH-Wert von 6,2 bis 6,5. Die Reagenzien enthalten daher einen Puffer zur pH-Wert Einstellung. Stark alkalische oder saure Wässer müssen jedoch vor der Analyse in einen pH-Bereich zwischen 6 und 7 gebracht werden (mit 0,5 mol/L Schwefelsäure bzw. 1 mol/L Natronlauge).

Anmerkungen

1. Evo-Tabletten können alternativ zu der entsprechenden Standard-Tablette verwendet werden (z.B. DPD Nr. 3 Evo anstatt DPD Nr. 3).



Durchführung der Bestimmung freies Chlor mit Tablette

Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



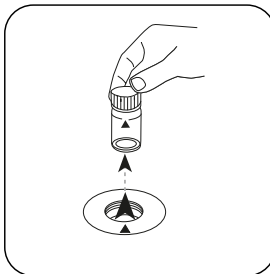
Küvette(n) verschließen.



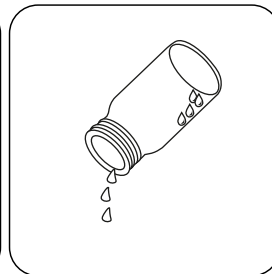
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



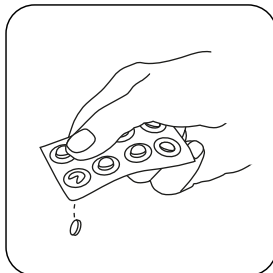
Taste **ZERO** drücken.



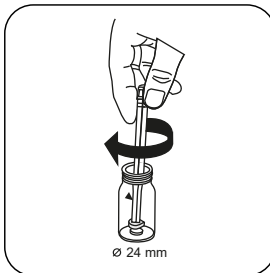
Küvette aus dem Messschacht nehmen.



Die Küvette bis auf einige Tropfen entleeren.



Eine **DPD No. 1 Tablette** zugeben.



Tablette(n) unter leichter Drehung zerdrücken.



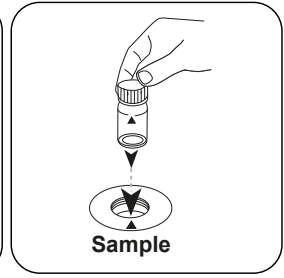
Küvette bis zur **10-mL-Marke** mit der **Probe** auffüllen.



Küvette(n) verschließen.



Tablette(n) durch Umschwenken lösen.



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.

DE

Test

Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis in mg/L freies Chlor.

Durchführung der Bestimmung gesamt Chlor mit Tablette

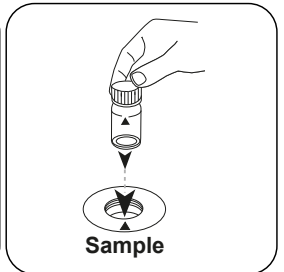
Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



Küvette(n) verschließen.

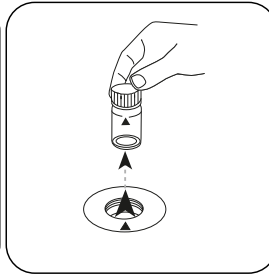


Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.

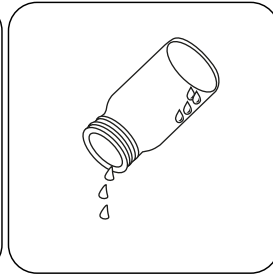


Zero

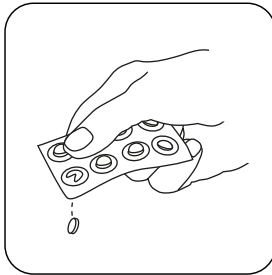
Taste **ZERO** drücken.



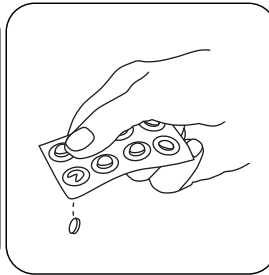
Küvette aus dem Messschacht nehmen.



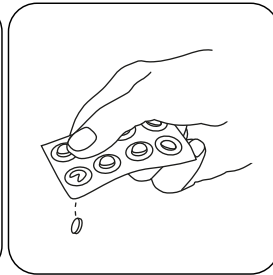
Die Küvette bis auf einige Tropfen entleeren.



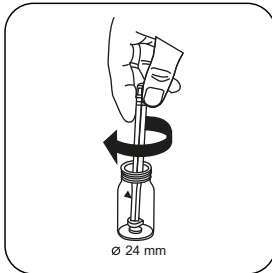
Eine **DPD No. 1** Tablette zugeben.



Eine **DPD No. 3** Tablette zugeben.



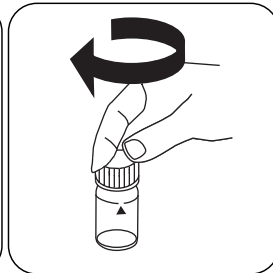
Alternativ zur DPD Nr. 1 und Nr. 3 Tablette kann eine DPD Nr. 4 Tablette zugegeben werden.



Tablette(n) unter leichter Drehung zerdrücken.



Küvette bis zur **10-mL-Mark**e mit der **Probe** auffüllen.



Küvette(n) verschließen.



Tablette(n) durch Umschwenken lösen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST (XD: START)** drücken.

DE



2 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.
In der Anzeige erscheint das Ergebnis in mg/L Gesamtchlor.



Chemische Methode

DPD

Appendix

DE

Störungen

Permanente Störungen

- Alle in den Proben vorhandenen Oxidationsmittel reagieren wie Chlor, was zu Mehrbefunden führt.

Ausschließbare Störungen

- Störungen durch Kupfer und Eisen(III) sind durch EDTA zu beseitigen.
- Bei Proben mit hohem Calciumgehalt* und/oder hoher Leitfähigkeit* kann es bei der Verwendung der Reagenztabletten zu einer Eintrübung der Probe und damit verbundener Fehlmessung kommen. In diesem Fall sind alternativ die Reagenztablette DPD No. 1 High Calcium und die Reagenztablette DPD No. 3 High Calcium zu verwenden.
*exakte Werte können nicht angegeben werden, da die Entstehung einer Trübung von Art und Zusammensetzung des Probenwassers abhängt.
- Konzentrationen über 10 mg/L Chlor, bei Verwendung von Tabletten, können zu Ergebnissen innerhalb des Messbereichs bis hin zu 0 mg/L führen. Bei einer zu hohen Chlorkonzentration muss die Probe mit chlorfreiem Wasser verdünnt werden. 10 mL der verdünnten Probe werden mit Reagenz versetzt und die Messung wiederholt (Plausibilitätstest).

Störung	Stört ab / [mg/L]
CrO_4^{2-}	0.01
MnO_2	0.01

Methodenvalidierung

Nachweisgrenze	0.02 mg/L
Bestimmungsgrenze	0.06 mg/L
Messbereichsende	6 mg/L
Empfindlichkeit	2.05 mg/L / Abs
Vertrauensbereich	0.04 mg/L
Verfahrensstandardabweichung	0.019 mg/L
Verfahrensvariationskoeffizient	0.87 %

Konform

EN ISO 7393-2



^{a)} Bestimmung von frei, gebunden, gesamt möglich | ^{a)} Hilfsreagenz, alternativ zur DPD No. 1 / No. 3 bei Eintrübungen der Probe durch hohen Calciumionengehalt und/oder hohe Leitfähigkeit

**Chlor L****M101****0,02 - 4,0 mg/L Cl₂^{a)}****CL6****DPD****Material**

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
DPD 1 Pufferlösung, blaue Flasche	15 mL	471010
DPD 1 Pufferlösung	100 mL	471011
DPD 1 Pufferlösung im 6-er Pack	1 St.	471016
DPD 1 Reagenzlösung, grüne Flasche	15 mL	471020
DPD 1 Reagenzlösung	100 mL	471021
DPD 1 Reagenzlösung im 6-er Pack	1 St.	471026
DPD 3 Lösung, rote Flasche	15 mL	471030
DPD 3 Lösung	100 mL	471031
DPD 3 Lösung im 6-er Pack	1 St.	471036
DPD Reagenzien Set	1 St.	471056

Verfügbare Standards

Titel	Verpackungseinheit	Bestell-Nr.
ValidCheck Chlor 1,5 mg/L	1 St.	48105510

Probenahme

1. Bei der Probenvorbereitung muss das Ausgasen von Chlor, z.B. durch Pipettieren und Schütteln, vermieden werden.
2. Die Analyse muss unmittelbar nach der Probenahme erfolgen.

Vorbereitung

1. Reinigung der Küvetten:
Da viele Haushaltsreiniger (z.B. Geschirrspülmittel) reduzierende Stoffe enthalten, kann es bei der Bestimmung von Chlor zu Minderbefunden kommen. Um diesen Messfehler auszuschließen, sollten die Glasgeräte chlorzehrungsfrei sein. Dazu werden die Glasgeräte für eine Stunde unter Natriumhypochloritlösung (0,1 g/L) aufbewahrt und danach gründlich mit VE-Wasser (Vollentsalztes Wasser) gespült.
2. Für die Einzelbestimmung von freiem Chlor und Gesamtchlor ist es sinnvoll, jeweils einen eigenen Satz Küvetten zu verwenden (siehe EN ISO 7393-2, Abs. 5.3).
3. Die DPD-Farmentwicklung erfolgt bei einem pH-Wert von 6,2 bis 6,5. Die Reagenzien enthalten daher einen Puffer zur pH-Wert Einstellung. Stark alkalische oder saure Wässer müssen jedoch vor der Analyse in einen pH-Bereich zwischen 6 und 7 gebracht werden (mit 0,5 mol/l Schwefelsäure bzw. 1 mol/l Natronlauge).

DE

Anmerkungen

1. Nach Gebrauch sind die Tropfflaschen mit der jeweils gleichfarbigen Schraubkappe sofort wieder zu verschließen.
2. Den Reagenssatz bei +6 °C bis +10 °C kühl lagern.



Durchführung der Bestimmung freies Chlor mit Flüssigreagenz

Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



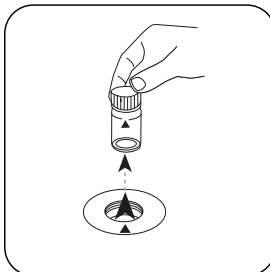
Küvette(n) verschließen.



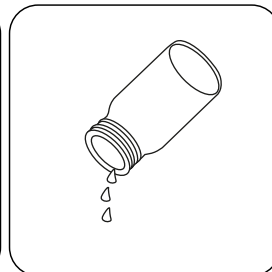
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



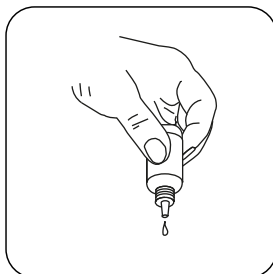
Taste **ZERO** drücken.



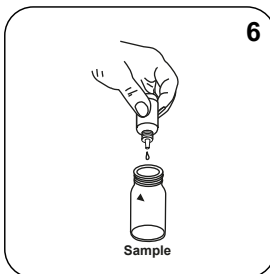
Küvette aus dem Messschacht nehmen.



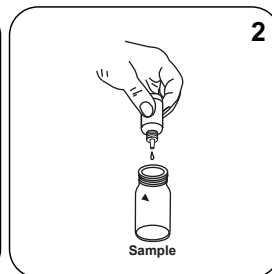
Küvette entleeren.



Die Tropfflaschen senkrecht halten und durch langsames Drücken gleich große Tropfen zugeben.



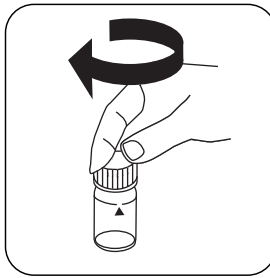
6 Tropfen DPD 1 Puffer-Lösung in die **Probeküvette** geben.



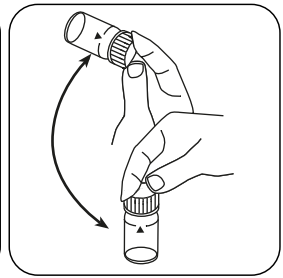
2 Tropfen DPD 1 Reagenz-Lösung in die **Probeküvette** geben.



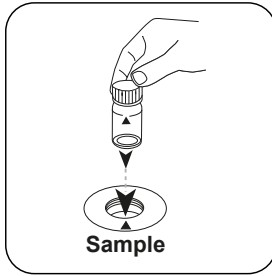
Küvette bis zur **10-mL-Mark**e mit der **Probe** auffüllen.



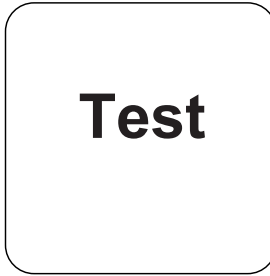
Küvette(n) verschließen.



Inhalt durch Umschwenken mischen.



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

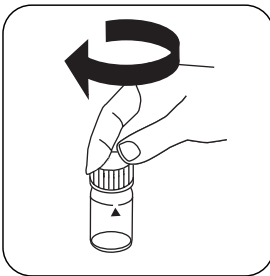
In der Anzeige erscheint das Ergebnis in mg/L freies Chlor.

Durchführung der Bestimmung gesamtes Chlor mit Flüssigreagenz

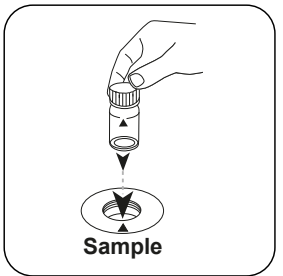
Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



Küvette(n) verschließen.



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



Zero

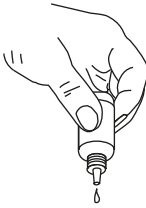
Taste **ZERO** drücken.



Küvette aus dem Messschacht nehmen.



Küvette entleeren.



Die Tropfflaschen senkrecht halten und durch langsames Drücken gleich große Tropfen zugeben.



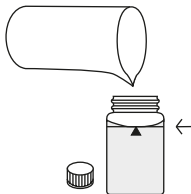
6 Tropfen DPD
1 Puffer-Lösung in die **Probenküvette** geben.



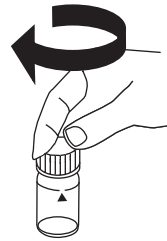
2 Tropfen DPD
1 Reagenz-Lösung in die **Probenküvette** geben.



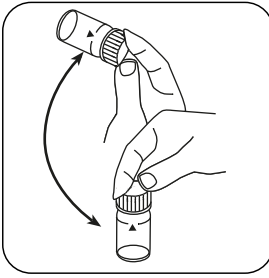
3 Tropfen DPD 3 Lösung in die **Probenküvette** geben.



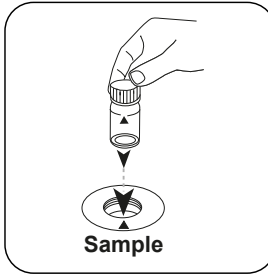
Küvette bis zur **10-mL-Marke** mit der **Probe** auffüllen.



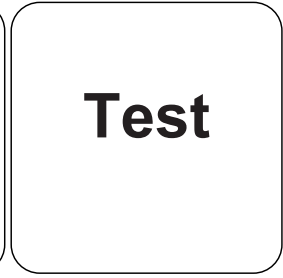
Küvette(n) verschließen.



Inhalt durch Umschwenken mischen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

DE



2 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.
In der Anzeige erscheint das Ergebnis in mg/L Gesamtchlor.



Chemische Methode

DPD

Appendix

DE

Störungen

Permanente Störungen

- Alle in den Proben vorhandenen Oxidationsmittel reagieren wie Chlor, was zu Mehrbefunden führt.

Ausschließbare Störungen

- Störungen durch Kupfer und Eisen(III) sind durch EDTA zu beseitigen.
- Konzentrationen über 4 mg/L Chlor, bei Verwendung von Flüssigreagenzien, können zu Ergebnissen innerhalb des Messbereichs bis hin zu 0 mg/L führen. In diesem Fall muss die Probe mit chlorfreiem Wasser verdünnt werden. 10 ml der verdünnten Probe werden mit Reagenz versetzt und die Messung wiederholt (Plausibilitätstest).

Störung	Stört ab / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Konform

EN ISO 7393-2

^{a)} Bestimmung von frei, gebunden, gesamt möglich

**Chlor HR T****M103****0,1 - 10 mg/L Cl₂^{a)}****CL10****DPD****Material**

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
DPD No. 1 HR	Tablette / 100	511500BT
DPD No. 1 HR	Tablette / 250	511501BT
DPD No. 1 HR	Tablette / 500	511502BT
DPD No. 3 HR	Tablette / 100	511590BT
DPD No. 3 HR	Tablette / 250	511591BT
DPD No. 3 HR	Tablette / 500	511592BT
Set DPD No. 1 HR/No. 3 HR #	je 100	517791BT
Set DPD No. 1 HR/No. 3 HR #	je 250	517792BT
DPD No. 1 High Calcium ^{e)}	Tablette / 100	515740BT
DPD No. 1 High Calcium ^{e)}	Tablette / 250	515741BT
DPD No. 1 High Calcium ^{e)}	Tablette / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Tablette / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Tablette / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Tablette / 500	515732BT
DPD No.3 HREvo	Tablette / 100	511920BT
DPD No. 3 HREvo	Tablette / 250	511921BT
DPD No. 3 HREvo	Tablette / 500	511922BT

Probenahme

1. Bei der Probenvorbereitung muss das Ausgasen von Chlor, z.B. durch Pipettieren und Schütteln, vermieden werden.
2. Die Analyse muss unmittelbar nach der Probenahme erfolgen.



Vorbereitung

1. Reinigung der Küvetten:
Da viele Haushaltsreiniger (z.B. Geschirrspülmittel) reduzierende Stoffe enthalten, kann es bei der Bestimmung von Chlor zu Minderbefunden kommen. Um diesen Messfehler auszuschließen, sollten die Glasgeräte chlorzehrungsfrei sein. Dazu werden die Glasgeräte für eine Stunde unter Natriumhypochloritlösung (0,1 g/L) aufbewahrt und danach gründlich mit VE-Wasser (Vollentsalztes Wasser) gespült.
2. Für die Einzelbestimmung von freiem Chlor und Gesamtchlor ist es sinnvoll, jeweils einen eigenen Satz Küvetten zu verwenden (siehe EN ISO 7393-2, Abs. 5.3).
3. Die DPD-Farmentwicklung erfolgt bei einem pH-Wert von 6,2 bis 6,5. Die Reagenzien enthalten daher einen Puffer zur pH-Wert Einstellung. Stark alkalische oder saure Wässer müssen jedoch vor der Analyse in einen pH-Bereich zwischen 6 und 7 gebracht werden (mit 0,5 mol/L Schwefelsäure bzw. 1 mol/L Natronlauge).

DE

Anmerkungen

1. Evo-Tabletten können alternativ zu der entsprechenden Standard-Tablette verwendet werden (z.B. DPD Nr. 3 Evo anstatt DPD Nr. 3).



Durchführung der Bestimmung freies Chlor HR mit Tablette

Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



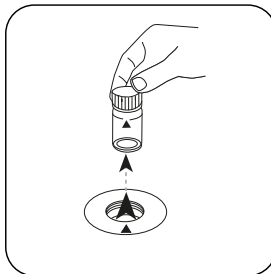
Küvette(n) verschließen.



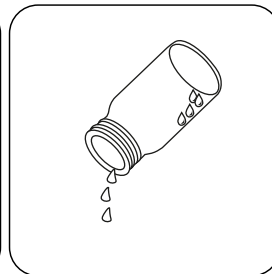
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



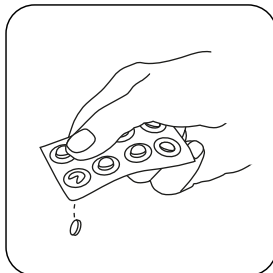
Taste **ZERO** drücken.



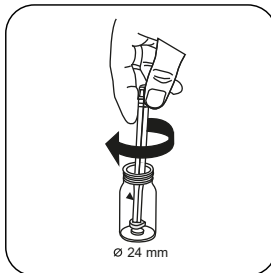
Küvette aus dem Messschacht nehmen.



Die Küvette bis auf einige Tropfen entleeren.



Eine **DPD No. 1 HR Tablette** zugeben.



Tablette(n) unter leichter Drehung zerdrücken.



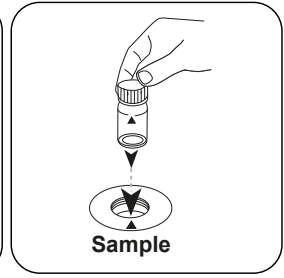
Küvette bis zur **10-mL-Marke** mit der **Probe** auffüllen.



Küvette(n) verschließen.



Tablette(n) durch Umschwenken lösen.



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.

DE

Test

Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis in mg/L freies Chlor.

Durchführung der Bestimmung gesamtes Chlor HR mit Tablette

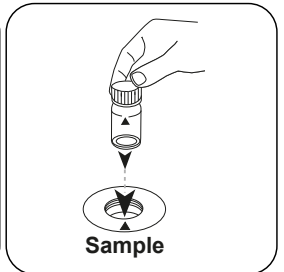
Die Methode im Gerät auswählen.



24-mm-Küvette mit **10 mL Probe** füllen.



Küvette(n) verschließen.



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



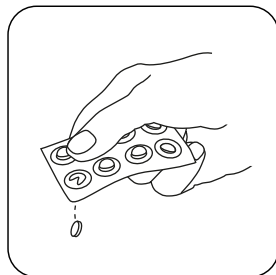
Zero

DE

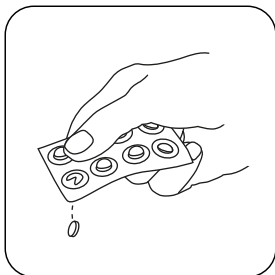
Taste **ZERO** drücken.

Küvette aus dem
Messschacht nehmen.

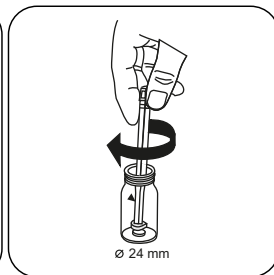
Die Küvette bis auf einige
Tropfen entleeren.



Eine **DPD No. 1 HR**
Tablette zugeben.



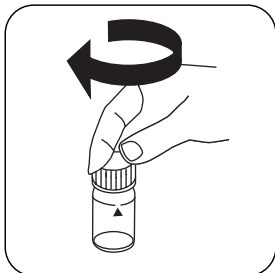
Eine **DPD No. 3 HR**
Tablette zugeben.



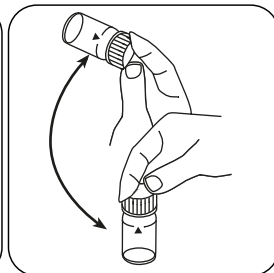
Tablette(n) unter leichter
Drehung zerdrücken.



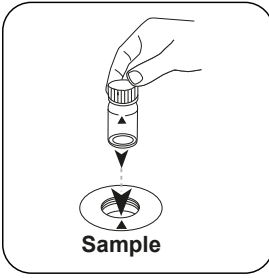
Küvette bis zur **10-mL-**
Marke mit der **Probe**
auffüllen.



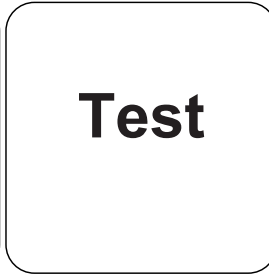
Küvette(n) verschließen.



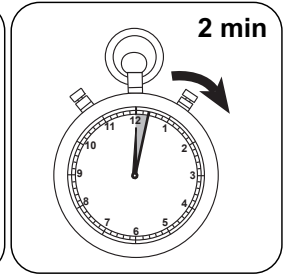
Tablette(n) durch
Umschwenken lösen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.



2 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.

In der Anzeige erscheint das Ergebnis in mg/L Gesamtchlor.

DE



Chemische Methode

DPD

Appendix

DE

Störungen

Permanente Störungen

- Alle in den Proben vorhandenen Oxidationsmittel reagieren wie Chlor, was zu Mehrbefunden führt.

Ausschließbare Störungen

- Störungen durch Kupfer und Eisen(III) sind durch EDTA zu beseitigen.
- Bei Proben mit hohem Calciumgehalt* und/oder hoher Leitfähigkeit* kann es bei der Verwendung der Reagenztabletten zu einer Eintrübung der Probe und damit verbundener Fehlmessung kommen. In diesem Fall sind alternativ die Reagenztablette DPD No. 1 High Calcium und die Reagenztablette DPD No. 3 High Calcium zu verwenden.

*exakte Werte können nicht angegeben werden, da die Entstehung einer Trübung von Art und Zusammensetzung des Probenwassers abhängt.

Konform

EN ISO 7393-2

^{a)} Bestimmung von frei, gebunden, gesamt möglich | ^{b)} Hilfsreagenz, alternativ zur DPD No. 1 / No. 3 bei Eintrübungen der Probe durch hohen Calciumionengehalt und/oder hohe Leitfähigkeit | * inklusive Rührstab



Kupfer T

M150

0,05 - 5 mg/L Cu^{a)}

Cu

Biquinolin

DE

Material

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Copper No. 1	Tablette / 100	513550BT
Copper No. 1	Tablette / 250	513551BT
Copper No. 2	Tablette / 100	513560BT
Copper No. 2	Tablette / 250	513561BT
Set Copper No. 1/No. 2 [#]	je 100	517691BT
Set Copper No. 1/No. 2 [#]	je 250	517692BT
ValidCheck Kupfer 2 mg/L	1 St.	48141525

Vorbereitung

1. Stark alkalische oder saure Wässer sollten vor der Analyse auf einen pH-Wert von 4 bis 6 eingestellt werden.

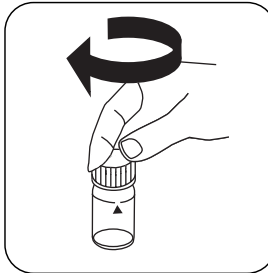
Durchführung der Bestimmung Kupfer, frei mit Tablette

Die Methode im Gerät auswählen.

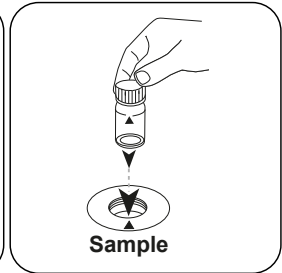
Wählen Sie zudem die Bestimmung: frei



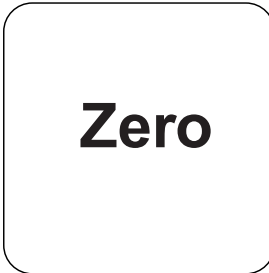
24-mm-Küvette mit **10 mL Probe** füllen.



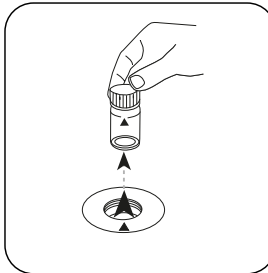
Küvette(n) verschließen.



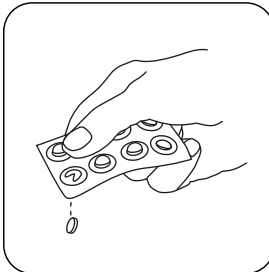
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



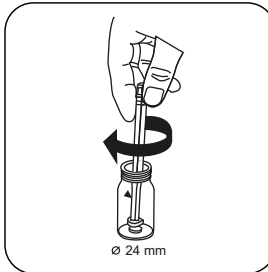
Taste **ZERO** drücken.



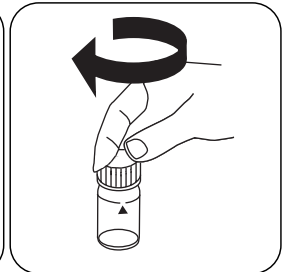
Küvette aus dem Messschacht nehmen.



Eine **COPPER No. 1** Tablette zugeben.



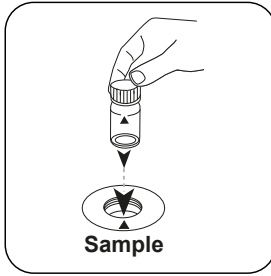
Tablette(n) unter leichter Drehung zerdrücken.



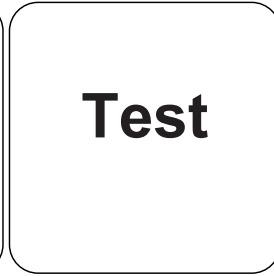
Küvette(n) verschließen.



Tablette(n) durch Umschwenken lösen.

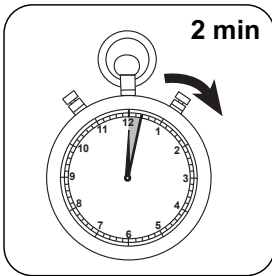


Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

DE



2 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.

In der Anzeige erscheint das Ergebnis in mg/L freies Kupfer.

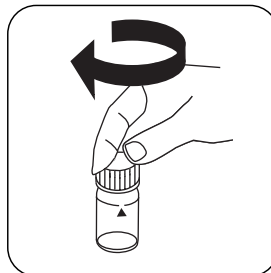
Durchführung der Bestimmung Kupfer, gesamt mit Tablette

Die Methode im Gerät auswählen.

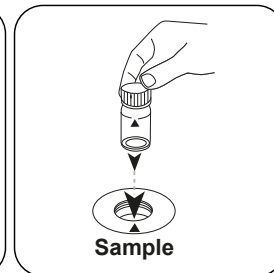
Wählen Sie zudem die Bestimmung: gesamt



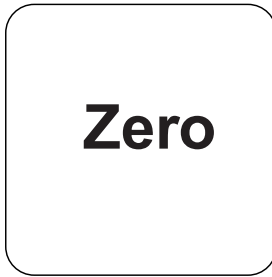
24-mm-Küvette mit **10 mL Probe** füllen.



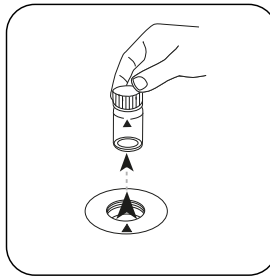
Küvette(n) verschließen.



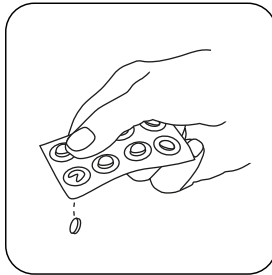
Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



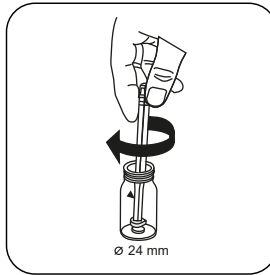
Taste **ZERO** drücken.



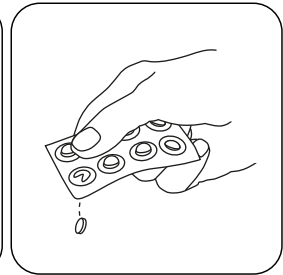
Küvette aus dem
Messschacht nehmen.



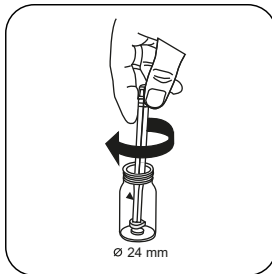
Eine **COPPER No.**
1 Tablette zugeben.



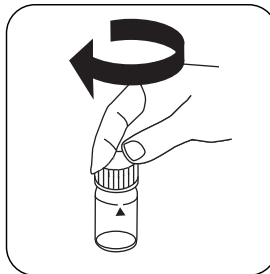
Die Tablette(n) unter
leichter Drehung
zerdrücken und lösen.



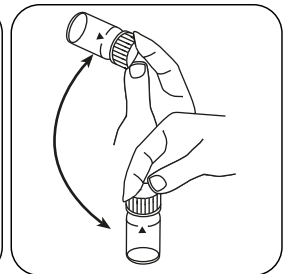
Eine **COPPER No.**
2 Tablette zugeben.



Tablette(n) unter leichter
Drehung zerdrücken.

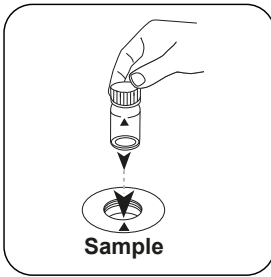


Küvette(n) verschließen.

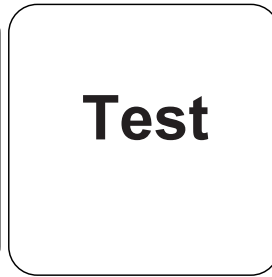


Tablette(n) durch
Umschwenken lösen.

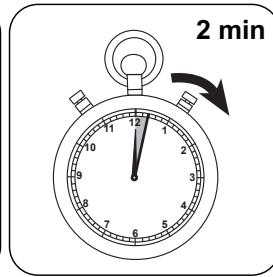
DE



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.



2 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.

In der Anzeige erscheint das Ergebnis in mg/L gesamtes Kupfer.

DE

Chemische Methode

Biquinolin

Appendix

Störungen

Permanente Störungen

1. Cyanide CN^- und Silber Ag^+ stören die Bestimmung.

Methodenvalidierung

Nachweisgrenze	0.05 mg/L
Bestimmungsgrenze	0.15 mg/L
Messbereichsende	5 mg/L
Empfindlichkeit	3.8 mg/L / Abs
Vertrauensbereich	0.026 mg/L
Verfahrensstandardabweichung	0.011 mg/L
Verfahrensvariationskoeffizient	0.42 %

Literaturverweise

Photometrische Analyse, Lange/Vedjelek, Verlag Chemie 1980

^{a)} Bestimmung von frei, gebunden, gesamt möglich | * inklusive Rührstab



CyA T

M160

10 - 160 mg/L CyA

CyA

Melamin

Material

DE

Benötigtes Material (zum Teil optional):

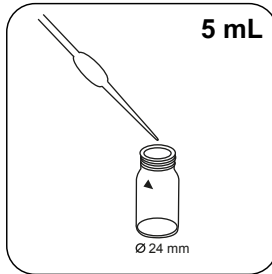
Reagenzien	Form/Menge	Bestell-Nr.
CyA-Test	Tablette / 100	511370BT
CyA-Test	Tablette / 250	511371BT
VE-Wasser	100 mL	461275
VE-Wasser	250 mL	457022

Anmerkungen

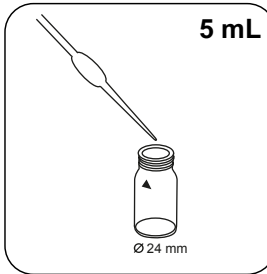
1. Cyanursäure verursacht eine sehr fein verteilte Trübung mit milchigem Aussehen. Einzelne Partikel sind nicht auf das Vorhandensein von Cyanursäure zurückzuführen.

Durchführung der Bestimmung Cyanursäure-Test mit Tablette

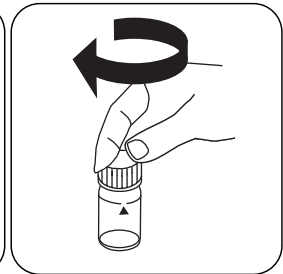
Die Methode im Gerät auswählen.



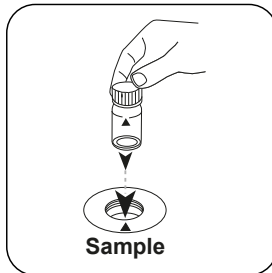
24 mm-Küvette mit **5 mL VE-Wasser** füllen.



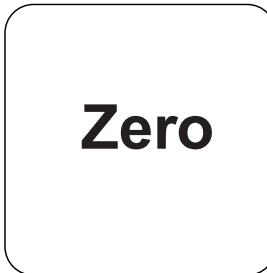
5 mL Probe in die Küvette geben.



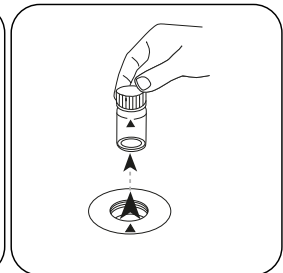
Küvette(n) verschließen.



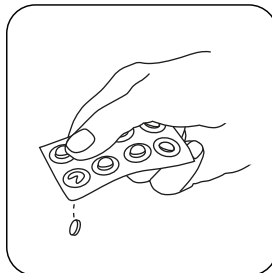
Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



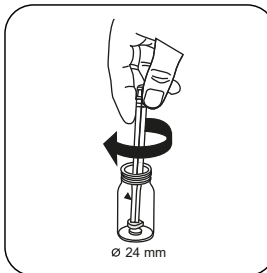
Taste **ZERO** drücken.



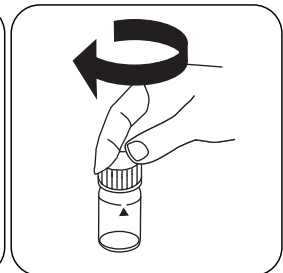
Küvette aus dem Messschacht nehmen.



Eine **CyA-Test Tablette** zugeben.



Tablette(n) unter leichter Drehung zerdrücken.



Küvette(n) verschließen.



Inhalt durch Umschwenken mischen (mindestens 60 s lang, bis die Tablette vollständig aufgelöst ist).



Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.




Test

Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis in mg/L Cyanursäure.

DE



Chemische Methode

Melamin

Störungen

Permanente Störungen

1. Nicht aufgelöste Partikel können zu Mehrbefunden führen. Daher ist es wichtig die Tabletten vollständig aufzulösen.

DE



Eisen T

M220

0,02 - 1 mg/L Fe

FE

Ferrozine / Thioglycolat

Material

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Iron II LR (Fe ²⁺)	Tablette / 100	515420BT
Iron II LR (Fe ²⁺)	Tablette / 250	515421BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablette / 100	515370BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Tablette / 250	515371BT

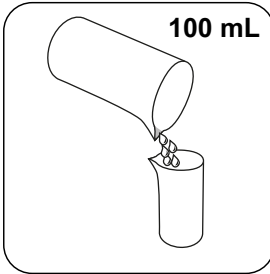
Vorbereitung

1. Wässer, die mit organischen Verbindungen als Korrosionsschutz usw. behandelt worden sind, müssen ggf. oxidiert werden, um die Eisenkomplexe zu zerstören. Dazu wird eine 100 ml Probe mit 1 ml konzentrierter Schwefelsäure (≥ 95 %) und 1 ml konzentrierter Salpetersäure (≥ 65 %) versetzt und auf die Hälfte eingedampft. Nach dem Abkühlen wird der Aufschluss durchgeführt.

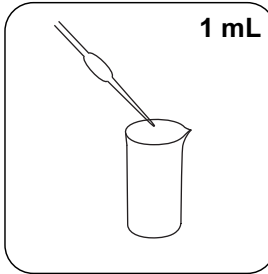
Anmerkungen

1. Bei dieser Methode erfolgt die Bestimmung von gesamt gelösten Fe²⁺ und Fe³⁺.
2. Zur Bestimmung von Fe²⁺ wird die IRON (II) LR Tablette, anstelle der IRON LR Tablette verwendet.

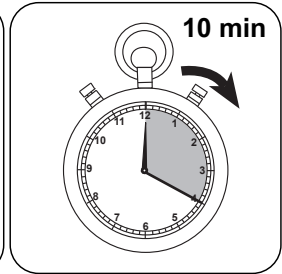
Aufschluss



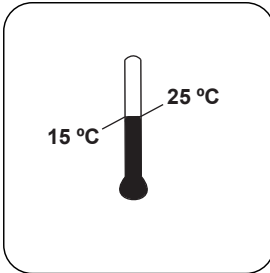
Ein geeignetes
Probengefäß mit **100 mL**
Probe füllen.



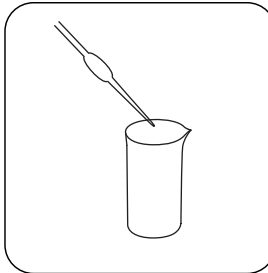
1 mL konzentrierte
Schwefelsäure (≥ 95 %)
zugeben.



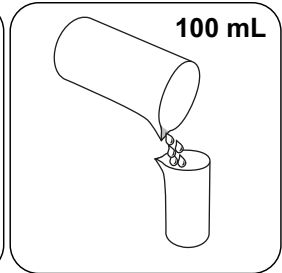
Die Probe für **10 Minuten**
erhitzen, oder so lange, bis
sich alles vollständig gelöst
hat.



Die Probe auf
Raumtemperatur
abkühlen lassen.



Den **pH-Wert** der Probe
mit **Ammoniaklösung**
(10-25 %) auf 3-5
einstellen.



Die Probe mit **VE-Wasser**
auf **100 mL** auffüllen.

Diese Probe für die Analyse von gesamten gelösten und ungelösten Eisen verwenden.
Durchführung der Bestimmung Eisen(II,III), gelöst mit Tablette

Die Methode im Gerät auswählen.

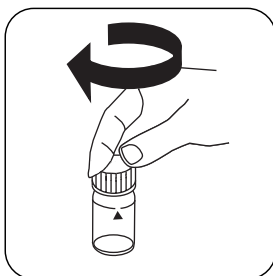
Für die Bestimmung von **gelöstem und ungelöstem Eisen** den beschriebenen
Aufschluss durchführen.



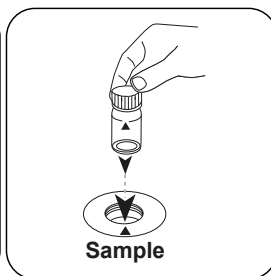
DE



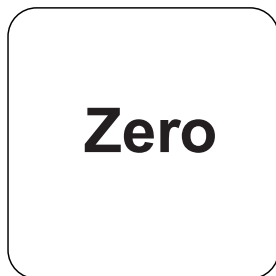
24-mm-Küvette mit **10 mL**
Probe füllen.



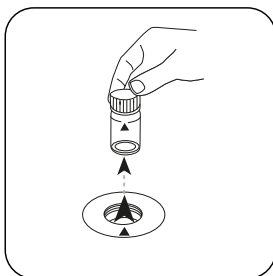
Küvette(n) verschließen.



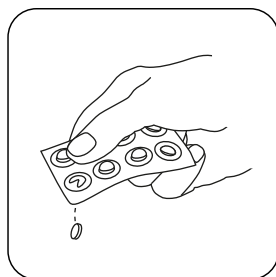
Die **Probeküvette** in
den Messschacht stellen.
Positionierung beachten.



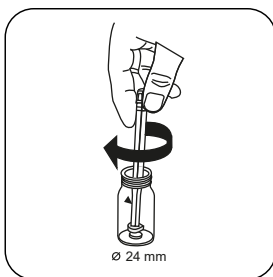
Taste **ZERO** drücken.



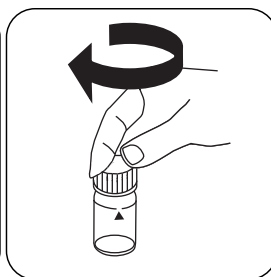
Küvette aus dem
Messschacht nehmen.



Eine **IRON LR** Tablette
zugeben.



Tablette(n) unter leichter
Drehung zerdrücken.



Küvette(n) verschließen.



Tablette(n) durch Umschwenken lösen.

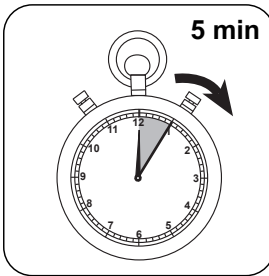


Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST (XD: START)** drücken.

DE



5 Minute(n) Reaktionszeit abwarten.

Nach Ablauf der Reaktionszeit erfolgt automatisch die Messung.
In der Anzeige erscheint das Ergebnis in mg/L Eisen.



Chemische Methode

Ferrozine / Thioglycolat

Appendix

DE

Störungen

Ausschließbare Störungen

1. Die Anwesenheit von Kupfer erhöht das Messergebnis um 10 %. Bei einer Konzentration von 10 mg/L Kupfer in der Probe wird das Messergebnis um 1 mg/L Eisen erhöht.
Die Störung kann durch die Zugabe von Thioharnstoff beseitigt werden

Methodenvalidierung

Nachweisgrenze	0.01 mg/L
Bestimmungsgrenze	0.016 mg/L
Messbereichsende	1 mg/L
Empfindlichkeit	0.92 mg/L / Abs
Vertrauensbereich	0.013 mg/L
Verfahrensstandardabweichung	0.005 mg/L
Verfahrensvariationskoeffizient	1.23 %

Literaturverweise

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



pH-Wert T

M330

6,5 - 8,4 pH

PH

Phenolrot

Material

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Phenol Red Photometer	Tablette / 100	511770BT
Phenol Red Photometer	Tablette / 250	511771BT
Phenol Red Photometer	Tablette / 500	511772BT

Anmerkungen

1. Für die photometrische pH-Wert Bestimmung sind nur PHENOL RED-Tabletten mit schwarzem Folienaufdruck zu verwenden, die mit dem Begriff PHOTOMETER gekennzeichnet sind.

Durchführung der Bestimmung pH-Wert mit Tablette

Die Methode im Gerät auswählen.



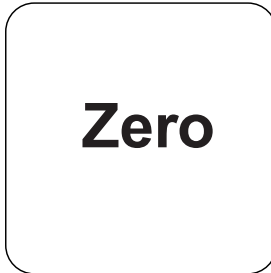
24-mm-Küvette mit **10 mL Probe** füllen.



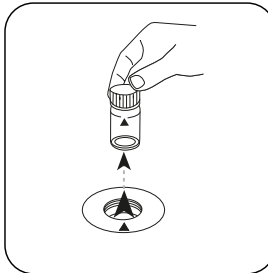
Küvette(n) verschließen.



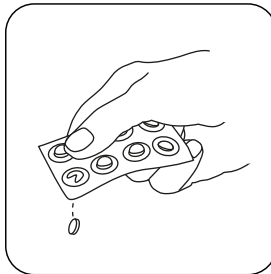
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



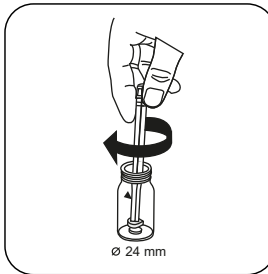
Taste **ZERO** drücken.



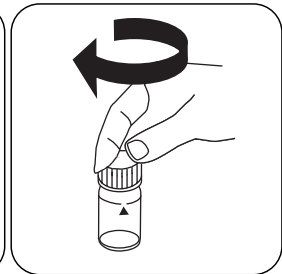
Küvette aus dem Messschacht nehmen.



Eine **PHENOL RED PHOTOMETER** Tablette zugeben.



Tablette(n) unter leichter Drehung zerdrücken.



Küvette(n) verschließen.

DE



Tablette(n) durch Umschwenken lösen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis als pH-Wert.

DE

Chemische Methode

Phenolrot

Appendix

Störungen

DE

Permanente Störungen

1. Wasserproben mit geringer Carbonathärte* können falsche pH-Werte ergeben.
* $K_{S4,3} < 0,7 \text{ mmol/l} \triangleq \text{Gesamthärte} < 35 \text{ mg/L CaCO}_3$.

Ausschließbare Störungen

1. pH-Werte unter 6,5 und über 8,4 können zu Ergebnissen innerhalb des Messbereiches führen. Es wird ein Plausibilitätstest (pH-Meter) empfohlen.
2. Salzfehler:
Bei Salzgehalten bis 2 g/L ist kein nennenswerter Salzfehler aufgrund des Salzgehaltes der Reagenztablette zu erwarten. Bei höheren Salzgehalten sind die Messwerte wie folgt zu korrigieren:

Salzgehalt der Probe in g/L	30 (Meerwasser)	60	120	180
Korrektur	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾ nach Kolthoff (1922)

²⁾ nach Parson und Douglas (1926)

Literaturverweise

Colorimetric Chemical Analytical Methods, 9th Edition, London



pH-Wert L

M331

6,5 - 8,4 pH

PH

Phenolrot

Material

DE

Benötigtes Material (zum Teil optional):

Reagenzien	Form/Menge	Bestell-Nr.
Phenolrot Lösung	15 mL	471040
Phenolrot Lösung	100 mL	471041
Phenolrot Lösung im -6er Pack	1 St.	471046

Vorbereitung

1. Auf Grund unterschiedlicher Tropfengröße kann das Messergebnis größere Abweichungen als bei Verwendung von Tabletten aufweisen.
Bei Verwendung einer Pipette (0,18 ml entsprechen 6 Tropfen) kann diese Abweichung minimiert werden.

Anmerkungen

1. Nach Gebrauch ist die Trofflasche mit der gleichfarbigen Schraubkappe sofort wieder zu verschließen.
2. Das Reagenz bei +6 °C bis +10 °C kühl lagern.



Durchführung der Bestimmung pH-Wert mit Flüssigreagenz

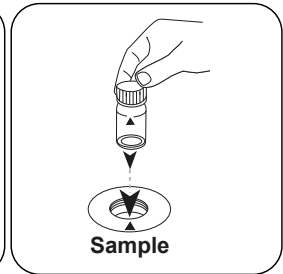
Die Methode im Gerät auswählen.



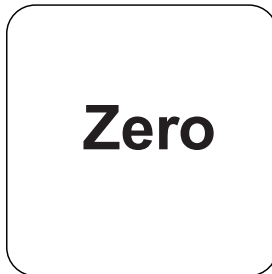
24-mm-Küvette mit **10 mL Probe** füllen.



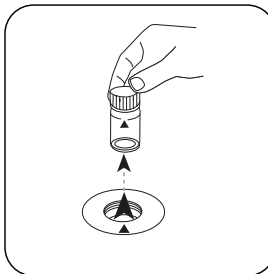
Küvette(n) verschließen.



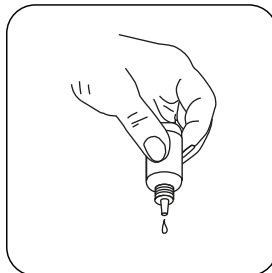
Die **Probeküvette** in den Messschacht stellen. Positionierung beachten.



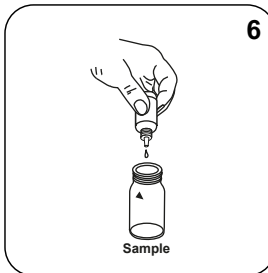
Taste **ZERO** drücken.



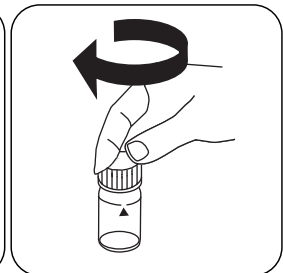
Küvette aus dem Messschacht nehmen.



Die Tropfflaschen senkrecht halten und durch langsames Drücken gleich große Tropfen zugeben.



6 Tropfen PHENOL Red-Lösung in die **Probeküvette** geben.



Küvette(n) verschließen.

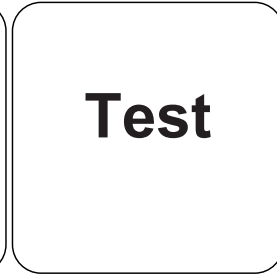
DE



Inhalt durch Umschwenken mischen.



Die **Probenküvette** in den Messschacht stellen. Positionierung beachten.



Taste **TEST** (XD: **START**) drücken.

In der Anzeige erscheint das Ergebnis als pH-Wert.

DE

Chemische Methode

Phenolrot

Appendix

Störungen

DE

Ausschließbare Störungen


1. Salzfehler: Korrektur des Messwertes (durchschnittliche Werte) für Proben mit einem Salzgehalt von:

2.	Salzgehalt der Probe	Korrektur
	30 g/L (Meerwasser)	-0,15 ¹⁾
	60 g/L	-0,21 ²⁾
	120 g/L	-0,26 ²⁾
	180 g/L	-0,29 ²⁾
	¹⁾ nach Kolthoff (1922)	²⁾ nach Parson und Douglas (1926)

3. Bei der Untersuchung von gechlortem Wasser kann der vorhandene Restchlorgehalt die Farbreaktion des Flüssigreagenzes beeinflussen. Dies wird verhindert, indem ein kleiner Kristall Natriumthiosulfat ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) in die Probelösung gegeben wird, bevor die PHENOL RED-Lösung zugesetzt wird.

Literaturverweise

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20


Nombre del método

Número de método

Código de barras para reconocer el método

Rango de medición

20

S:4.3

Indicación en la pantalla de MD 100 / MD 110 / MD 200

Método químico

Información específica del instrumento

La prueba puede realizarse en los siguientes dispositivos. Además, se muestran la cubeta requerida y el rango de absorción del fotómetro.

Dispositivos	Cubeta	λ	Rango de medición
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	\varnothing 24 mm	610 nm	0.1 - 4 mmol/l $K_{S4.3}$
SpectroDirect, XD 7000, XD 7500	\varnothing 24 mm	615 nm	0.1 - 4 mmol/l $K_{S4.3}$

Material

Material requerido (parcialmente opcional):

Título	Unidad de embalaje	Referencia No
Fotómetro alca-M	Tabletas / 100	513210BT
Fotómetro alca-M	Tabletas / 250	513211BT

Lista de aplicaciones

- Tratamiento de aguas residuales
- Tratamiento de aguas potables
- Tratamiento de aguas de aporte

Notas

1. Las definiciones de alcalinidad-m, valor-m y capacidad ácida $K_{S4.3}$ son idénticas.
2. Añadir un volumen de muestra de exactamente 10 ml, ya que este volumen influye de forma decisiva en la exactitud del resultado.

Códigos de idioma ISO 639-1

Estado de revisión

ES Manual de Métodos 01/20

ES

Realización de la determinación

Ejecución de la determinación Capacidad ácida $K_{24.3}$ con tableta

Seleccionar el método en el aparato.

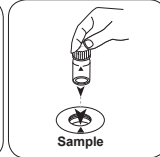
Para este método no es necesario realizar medición CERO en los aparatos siguientes: XD 7000, XD 7500



Llenar la cubeta de 24 mm con 10 ml de muestra .

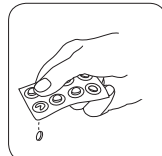


Cerrar la(s) cubeta(s).



Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

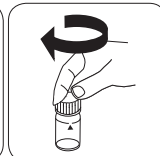
• • •



Añadir **tableta ALKA-M-PHOTOMETER**.



Triturar la(s) tableta(s) girando ligeramente.



Cerrar la(s) cubeta(s).



Alcalinidad-m T

M30

5 - 200 mg/L CaCO₃

tA

Ácido / Indicador

ES

Material

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Fotómetro alca-M	Tabletas / 100	513210BT
Fotómetro alca-M	Tabletas / 250	513211BT

Notas

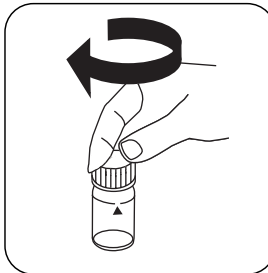
1. Las definiciones de alcalinidad-m, valor-m y capacidad ácida $K_{s4.3}$ son idénticas.
2. Añadir un volumen de muestra de exactamente 10 ml, ya que este volumen influye de forma decisiva en la exactitud del resultado.

Ejecución de la determinación Alcalinidad, total= alcalinidad-m = valor-m con tableta

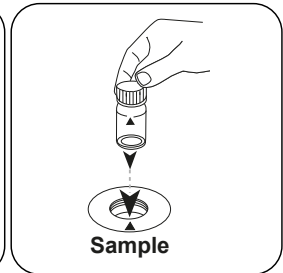
Seleccionar el método en el aparato.



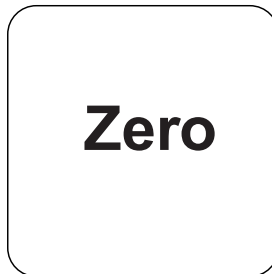
Llenar la cubeta de 24 mm con **10 mL de muestra** .



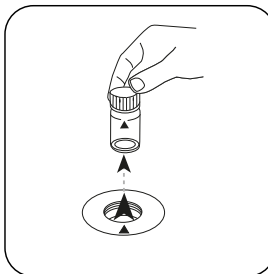
Cerrar la(s) cubeta(s).



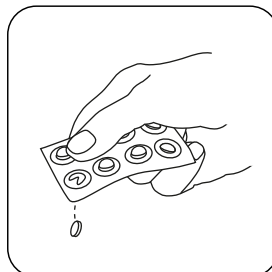
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



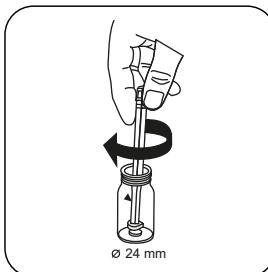
Pulsar la tecla **ZERO**.



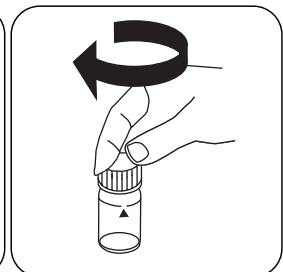
Extraer la cubeta del compartimiento de medición.



Añadir tableta **ALKA-M-PHOTOMETER**.



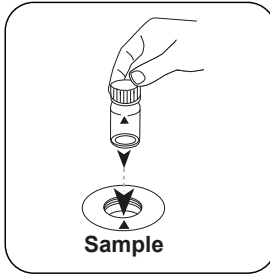
Triturar la(s) tableta(s) girando ligeramente.



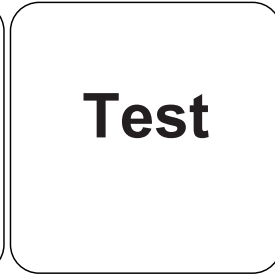
Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.



Poner la **cubeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

A continuación se visualizará el resultado como Alcalinidad-m.

ES

Evaluación

La siguiente tabla muestra cómo los valores de salida se pueden convertir a otros formularios de citas.

Unidad	Conversión	Factor de conversión
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4.3}	0.02

ES

Método químico

Ácido / Indicador

Apéndice

Derivado de

EN ISO 9963-1



Cloro T

M100

0.01 - 6.0 mg/L Cl₂ ^{a)}

CL6

DPD

Material

ES

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
DPD n°1	Tabletas / 100	511050BT
DPD n° 1	Tabletas / 250	511051BT
DPD n° 1	Tabletas / 500	511052BT
DPD n° 3	Tabletas / 100	511080BT
DPD n° 3	Tabletas / 250	511081BT
DPD n° 3	Tabletas / 500	511082BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 100	515740BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 250	515741BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 500	515742BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 100	515730BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 250	515731BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 500	515732BT
DPD n° 4	Tabletas / 100	511220BT
DPD n° 4	Tabletas / 250	511221BT
DPD n° 4	Tabletas / 500	511222BT
DPD n° 3 Evo	Tabletas / 100	511420BT
DPD n° 3 Evo	Tabletas / 250	511421BT
DPD n° 3 Evo	Tabletas / 500	511422BT
DPD n°4 Evo	Tabletas / 100	511970BT
DPD n° 4 Evo	Tabletas / 250	511971BT
DPD n° 4 Evo	Tabletas / 500	511972BT

Standards disponibles

Título	Unidad de embalaje	No. de referencia
ValidCheck cloro 1,5 mg/l	1 Cantidad	48105510



Muestreo

1. Evitar durante la preparación de la muestra la desgasificación de cloro, p. ej., al pipetar o agitar.
2. La determinación se ha de realizar inmediatamente después de la toma de la muestra.

Preparación

1. Limpieza de las cubetas:
Muchos productos de limpieza (p. ej., detergentes de lavavajillas) poseen componentes reductores, que pueden reducir los resultados en la determinación del cloro. Para evitar estas alteraciones, los aparatos de vidrio deben estar exentos de componentes corrosivos al cloro. Para ello, deberá sumergir los aparatos de vidrio durante una hora en una solución de hipoclorito sódico (0,1 g/L), enjuagándolos minuciosamente a continuación con agua desionizada.
2. Para la determinación individual de cloro libre y cloro total se recomienda utilizar siempre los mismos sets de cubetas respectivamente (véase EN ISO 7393-2, párrafo 5.3).
3. El desarrollo coloreo por DPD se efectúa entre un valor de pH de 6,2 - 6,5. Por ello poseen las tabletas un tampón para la graduación del valor de pH. Sin embargo, las muestras acuosas muy ácidas o muy básicas se deberán neutralizar a un valor de pH entre 6 y 7 antes de realizar el análisis (con 0,5 mol/L de ácido sulfúrico o 1 mol/L de hidróxido sódico).

Notas

1. Las tabletas Evo pueden utilizarse como alternativa a la tableta estándar correspondiente (por ejemplo, DPD nº 3 Evo en lugar de DPD nº 3).

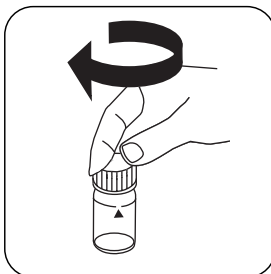


Ejecución de la determinación Cloro libre con tableta

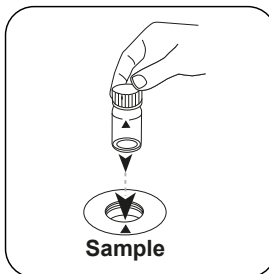
Seleccionar el método en el aparato.



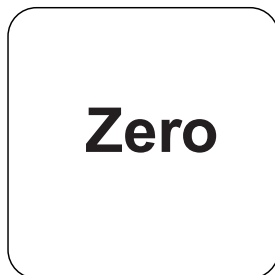
Llenar la cubeta de 24 mm con **10 mL de muestra** .



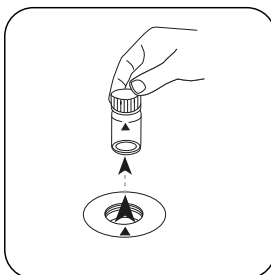
Cerrar la(s) cubeta(s).



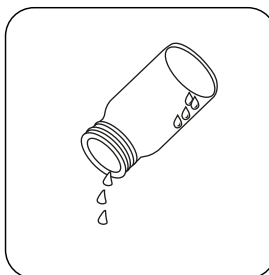
Poner la **cubeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



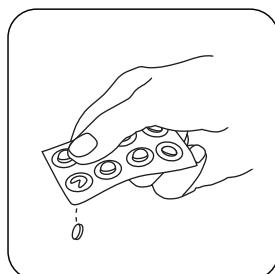
Pulsar la tecla **ZERO**.



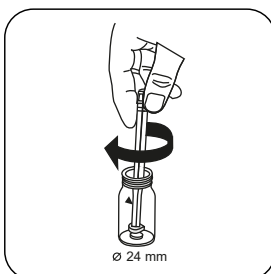
Extraer la cubeta del compartimento de medición.



Vaciar la cubeta excepto algunas gotas.



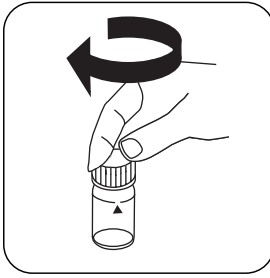
Añadir **tableta DPD No. 1**.



Triturar la(s) tableta(s) girando ligeramente.



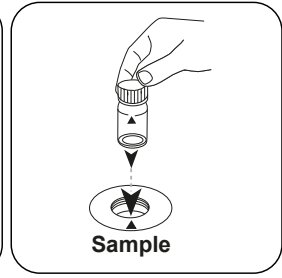
Llenar la cubeta con la **muestra hasta la marca de 10 mL** .



Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.



Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

ES

Test

Pulsar la tecla **TEST** (XD: **START**).

A continuación se visualizará el resultado en mg/L Cloro libre.

Ejecución de la determinación Cloro total con tableta

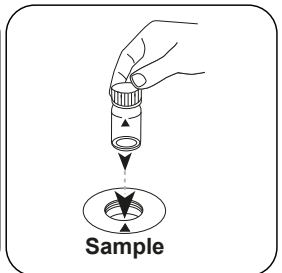
Seleccionar el método en el aparato.



Llenar la cubeta de 24 mm con **10 mL de muestra** .



Cerrar la(s) cubeta(s).

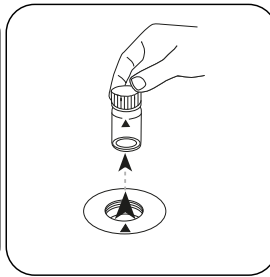


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

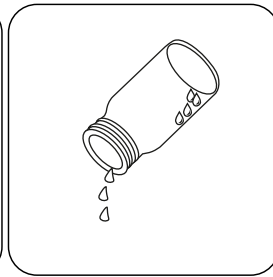


Zero

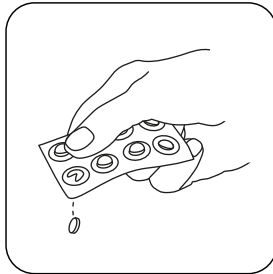
Pulsar la tecla **ZERO**.



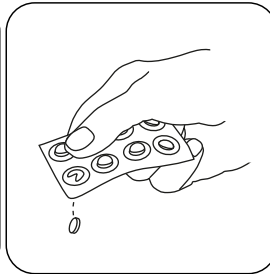
Extraer la cubeta del compartimiento de medición.



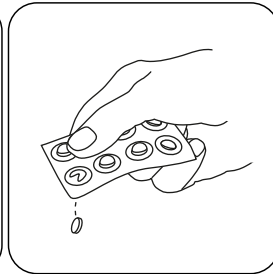
Vaciar la cubeta excepto algunas gotas.



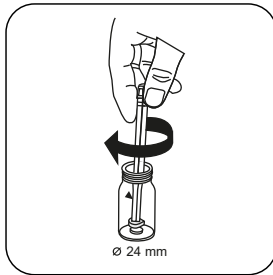
Añadir **tableta DPD No. 1**.



Añadir **tableta DPD No. 3**.



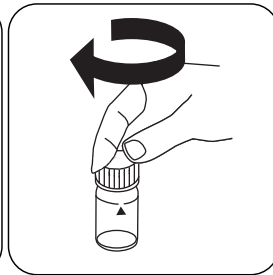
Alternativa a la tableta DPD No 1 y No 3, se puede agregar una tableta DPD No. 4.



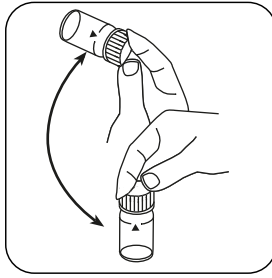
Triturar la(s) tableta(s) girando ligeramente.



Llenar la cubeta con la **muestra hasta la marca de 10 mL**.



Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.

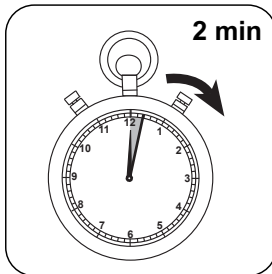


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

ES



Esperar **2 minutos como periodo de reacción.**

Finalizado el periodo de reacción se realizará la determinación automáticamente.

A continuación se visualizará el resultado en mg/L Cloro total.



Método químico

DPD

Apéndice

ES

Interferencia

Interferencias persistentes

- Todos los elementos oxidantes existentes en la muestra reaccionan como el cloro, lo que produce un resultado más elevado.

Interferencias extraíbles

- Las perturbaciones debido a cobre y hierro (III) deben suprimirse mediante EDTA.
- En las muestras con una elevada concentración de iones de calcio* y/o alta conductividad*, se puede producir un enturbiamiento de la muestra con el uso de las tabletas de reactivo, alterando el resultado. En este caso, utilizar alternativamente la tableta reactiva DPD n° 1 High Calcium y la tableta reactiva DPD n° 3 High Calcium. *no se pueden dar valores exactos, ya que la aparición de enturbiamiento dependerá del tipo y composición de la muestra.
- Las concentraciones de cloro mayores a 10 mg/L, cuando se usan tabletas pueden conducir a resultados de dentro del campo de medición hasta 0 mg/L. Con una concentración de cloro alta, se deberá diluir la muestra con agua sin cloro. Se mezclan 10 mL de muestra diluida con reactivo y se repite la medición (prueba de plausibilidad).

Interferencia	de / [mg/L]
CrO ₄ ²⁻	0.01
MnO ₂	0.01

Validación del método

Límite de detección	0.02 mg/L
Límite de determinación	0.06 mg/L
Límite del rango de medición	6 mg/L
Sensibilidad	2.05 mg/L / Abs
Intervalo de confianza	0.04 mg/L
Desviación estándar	0.019 mg/L
Coefficiente de variación	0.87 %

Conforme a

EN ISO 7393-2



^{a)} Posible determinación de libre, combinado, total | ^{a)} Reactivo auxiliar, alternativo a DPD No.1/3 en enturbiamientos de la prueba debido a concentraciones elevadas de calcio y/o elevada conductividad

ES



Cloro L

M101

0.02 - 4.0 mg/L Cl₂ ^{a)}

CL6

DPD

ES

Material

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Solución de tampón DPD 1, frasco azul	15 mL	471010
Solución de tampón DPD 1	100 mL	471011
Solución de tampón DPD 1 en pack de 6	1 Cantidad	471016
Solución reactiva DPD 1, frasco verde	15 mL	471020
Solución de reactivo DPD 1	100 mL	471021
Solución reactiva DPD 1 en pack de 6	1 Cantidad	471026
Solución DPD 3, frasco rojo	15 mL	471030
Solución DPD 3	100 mL	471031
Solución DPD 3 en pack de 6	1 Cantidad	471036
Juego de reactivos para DPD	1 Cantidad	471056

Standards disponibles

Título	Unidad de embalaje	No. de referencia
ValidCheck cloro 1,5 mg/l	1 Cantidad	48105510

Muestreo

1. Evitar durante la preparación de la muestra la desgasificación de cloro, p. ej., al pipetar o agitar.
2. La determinación se ha de realizar inmediatamente después de la toma de la muestra.

Preparación

1. Limpieza de las cubetas:
Muchos productos de limpieza (p. ej., detergentes de lavavajillas) poseen componentes reductores, que pueden reducir los resultados en la determinación del cloro. Para evitar estas alteraciones, los aparatos de vidrio deben estar exentos de componentes corrosivos al cloro. Para ello, deberá sumergir los aparatos de vidrio durante una hora en una solución de hipoclorito sódico (0,1 g/L), enjuagándolos minuciosamente a continuación con agua desionizada.
2. Para la determinación individual de cloro libre y cloro total se recomienda utilizar siempre los mismos sets de cubetas respectivamente (véase EN ISO 7393-2, párrafo 5.3).
3. El desarrollo coloreo por DPD se efectúa entre un valor de pH de 6,2 - 6,5. Por ello poseen las tabletas un tampón para la graduación del valor de pH. Sin embargo, las muestras acuosas muy ácidas o muy básicas se deberán neutralizar a un valor de pH entre 6 y 7 antes de realizar el análisis (con 0,5 mol/l de ácido sulfúrico o 1 mol/l de hidróxido sódico).

Notas

1. Después de usarlas, las botellas cuentagotas deben cerrarse de nuevo inmediatamente con la tapa roscada del mismo color, respectivamente.
2. Guardar el set reactivo a una temperatura entre +6 °C y +10 °C.



Ejecución de la determinación Cloro libre con reactivos líquidos

Seleccionar el método en el aparato.



10 mL
Llenar la cubeta de 24 mm con **10 mL de muestra**.



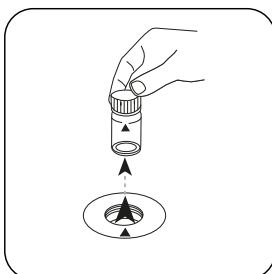
Cerrar la(s) cubeta(s).



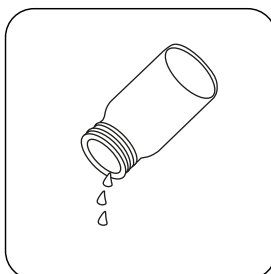
Poner la **cubeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



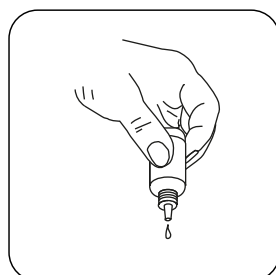
Pulsar la tecla **ZERO**.



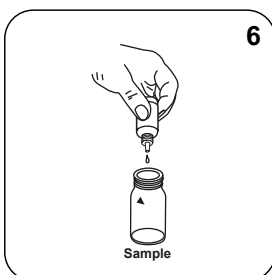
Extraer la cubeta del compartimento de medición.



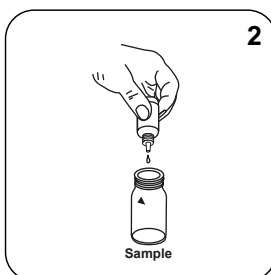
Vaciar la cubeta.



Mantener la botella cuentagotas vertical y añadir gotas del mismo tamaño presionando lentamente.



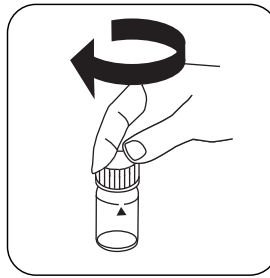
Añadir **6 gotas de DPD 1 Buffer Solution** en la cubeta con la muestra.



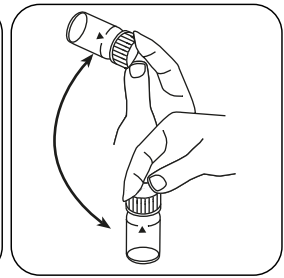
Añadir **2 gotas de DPD 1 Reagent Solution** en la cubeta con la muestra.



Llenar la cubeta con la **muestra** hasta la **marca de 10 mL** .

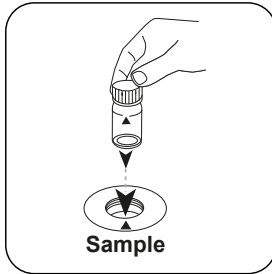


Cerrar la(s) cubeta(s).

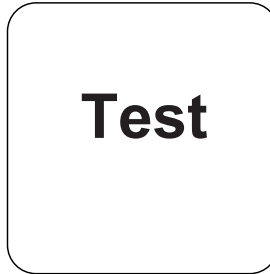


Mezclar el contenido girando.

ES



Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

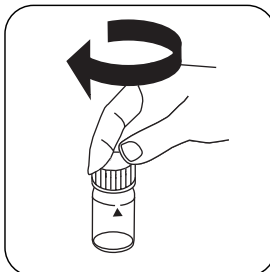
A continuación se visualizará el resultado en mg/L Cloro libre.

Ejecución de la determinación Cloro total con reactivos líquidos

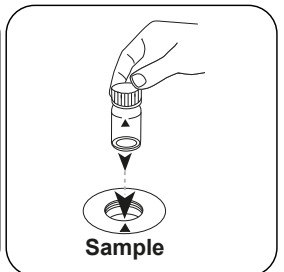
Seleccionar el método en el aparato.



Llenar la cubeta de 24 mm con **10 mL de muestra** .



Cerrar la(s) cubeta(s).

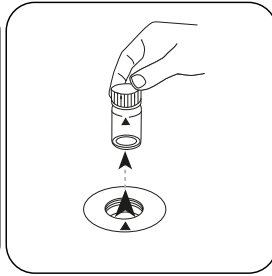


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

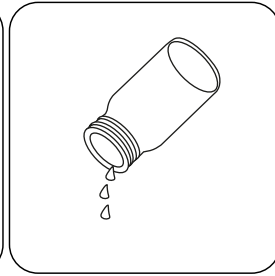


Zero

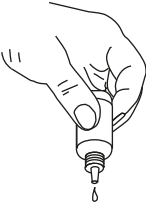
Pulsar la tecla **ZERO**.



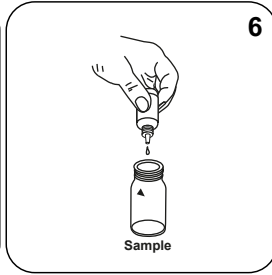
Extraer la cubeta del compartimiento de medición.



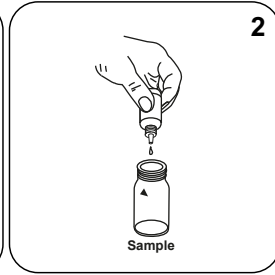
Vaciar la cubeta.



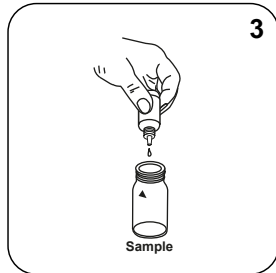
Mantener la botella cuentagotas vertical y añadir gotas del mismo tamaño presionando lentamente.



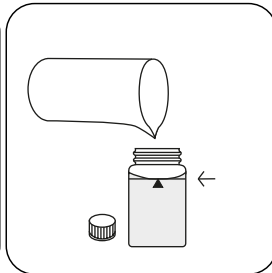
Añadir **6 gotas de DPD 1 Buffer Solution** en la cubeta con la muestra.



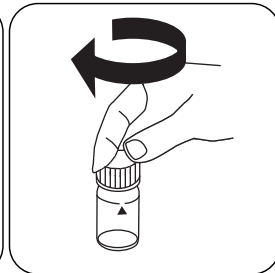
Añadir **2 gotas de DPD 1 Reagent Solution** en la cubeta con la muestra.



Añadir **3 gotas de DPD 3 Solution** en la cubeta con la muestra.



Llenar la cubeta con la **muestra hasta la marca de 10 mL**.



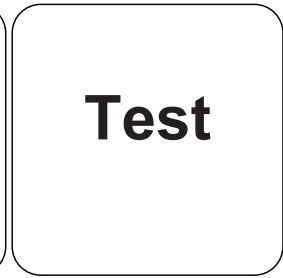
Cerrar la(s) cubeta(s).



Mezclar el contenido girando.

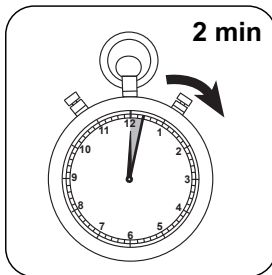


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

ES



Esperar **2 minutos** como **periodo de reacción** .

Finalizado el periodo de reacción se realizará la determinación automáticamente.

A continuación se visualizará el resultado en mg/L Cloro total.



Método químico

DPD

Apéndice

ES

Interferencia

Interferencias persistentes

- Todos los elementos oxidantes existentes en la muestra reaccionan como el cloro, lo que produce un resultado más elevado.

Interferencias extraíbles

- Las perturbaciones debido a cobre y hierro (III) deben suprimirse mediante EDTA.
- Las concentraciones de cloro mayores a 4 mg/L, cuando se usan reactivos líquidos pueden conducir a resultados de dentro del campo de medición hasta 0 mg/L. En este caso, se deberá diluir la muestra con agua sin cloro. Se mezclan 10 ml de muestra diluida con reactivo y se repite la medición (prueba de plausibilidad).

Interferencia	de / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Conforme a

EN ISO 7393-2

^{a)} Posible determinación de libre, combinado, total



Cloro HR T

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

Material

ES

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
DPD n° 1 HR	Tabletas / 100	511500BT
DPD n° 1 HR	Tabletas / 250	511501BT
DPD n° 1 HR	Tabletas / 500	511502BT
DPD n° 3 HR	Tabletas / 100	511590BT
DPD n° 3 HR	Tabletas / 250	511591BT
DPD n° 3 HR	Tabletas / 500	511592BT
Juego DPD n° 1 HR/n° 3 HR #	100 cada	517791BT
Juego DPD n° 1 HR/n° 3 HR #	250 cada	517792BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 100	515740BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 250	515741BT
DPD n° 1 High Calcium ^{e)}	Tabletas / 500	515742BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 100	515730BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 250	515731BT
DPD n° 3 High Calcium ^{e)}	Tabletas / 500	515732BT
DPD n°3 HR Evo	Tabletas / 100	511920BT
DPD n° 3 HR Evo	Tabletas / 250	511921BT
DPD n° 3 HR Evo	Tabletas / 500	511922BT

Muestreo

1. Evitar durante la preparación de la muestra la desgasificación de cloro, p. ej., al pipetar o agitar.
2. La determinación se ha de realizar inmediatamente después de la toma de la muestra.



Preparación

1. Limpieza de las cubetas:
Muchos productos de limpieza (p. ej., detergentes de lavavajillas) poseen componentes reductores, que pueden reducir los resultados en la determinación del cloro. Para evitar estas alteraciones, los aparatos de vidrio deben estar exentos de componentes corrosivos al cloro. Para ello, deberá sumergir los aparatos de vidrio durante una hora en una solución de hipoclorito sódico (0,1 g/L), enjuagándolos minuciosamente a continuación con agua desionizada.
2. Para la determinación individual de cloro libre y cloro total se recomienda utilizar siempre los mismos sets de cubetas respectivamente (véase EN ISO 7393-2, párrafo 5.3).
3. El desarrollo coloreo por DPD se efectúa entre un valor de pH de 6,2 - 6,5. Por ello poseen las tabletas un tampón para la graduación del valor de pH. Sin embargo, las muestras acuosas muy ácidas o muy básicas se deberán neutralizar a un valor de pH entre 6 y 7 antes de realizar el análisis (con 0,5 mol/L de ácido sulfúrico o 1 mol/L de hidróxido sódico).

Notas

1. Las tabletas Evo pueden utilizarse como alternativa a la tableta estándar correspondiente (por ejemplo, DPD nº 3 Evo en lugar de DPD nº 3).



Ejecución de la determinación Cloro HR libre con tableta

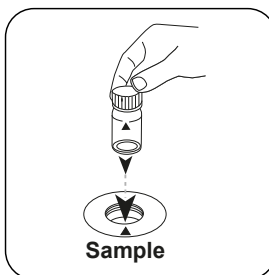
Seleccionar el método en el aparato.



Llenar la cubeta de 24 mm con **10 mL de muestra** .



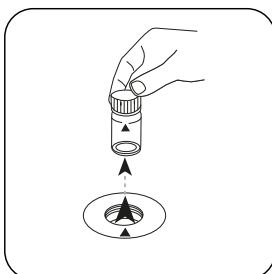
Cerrar la(s) cubeta(s).



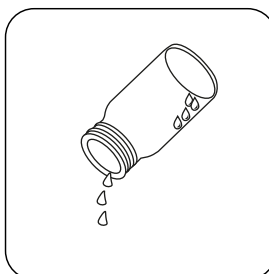
Poner la **cubeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



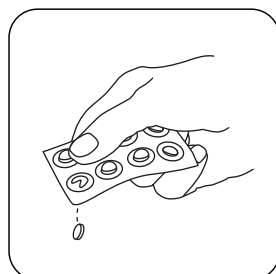
Pulsar la tecla **ZERO**.



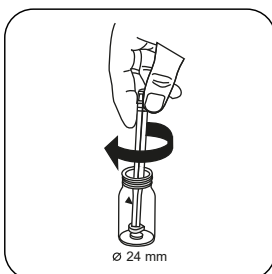
Extraer la cubeta del compartimento de medición.



Vaciar la cubeta excepto algunas gotas.



Añadir **tableta DPD No. 1 HR** .



Triturar la(s) tableta(s) girando ligeramente.



Llenar la cubeta con la **muestra hasta la marca de 10 mL** .



Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.



Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

ES

Test

Pulsar la tecla **TEST** (XD: **START**).

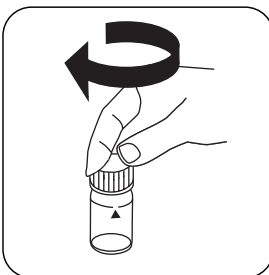
A continuación se visualizará el resultado en mg/L Cloro libre.

Ejecución de la determinación Cloro HR total con tableta

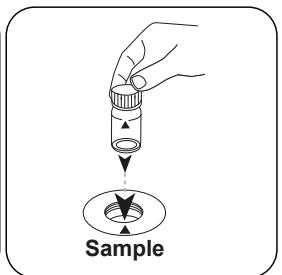
Seleccionar el método en el aparato.



Llenar la cubeta de 24 mm con **10 mL de muestra** .



Cerrar la(s) cubeta(s).

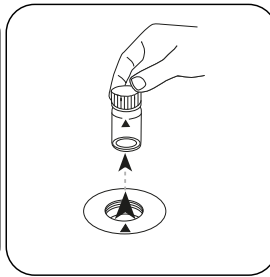


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!

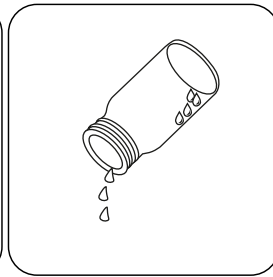


Zero

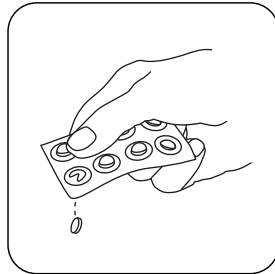
Pulsar la tecla **ZERO**.



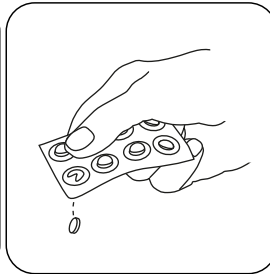
Extraer la cubeta del compartimiento de medición.



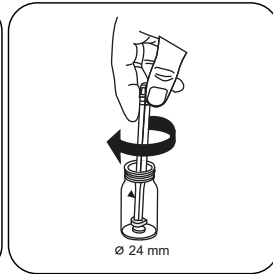
Vaciar la cubeta excepto algunas gotas.



Añadir **tableta DPD No. 1 HR**.



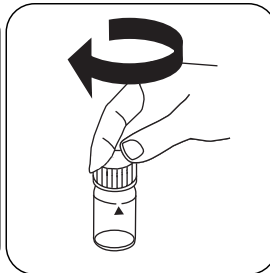
Añadir **tableta DPD No. 3 HR**.



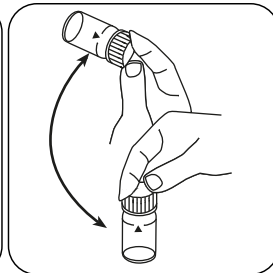
Triturar la(s) tableta(s) girando ligeramente.



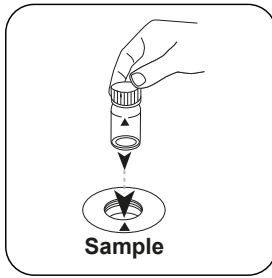
Llenar la cubeta con la muestra hasta la **marca de 10 mL**.



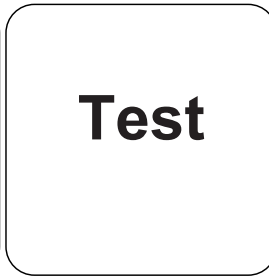
Cerrar la(s) cubeta(s).



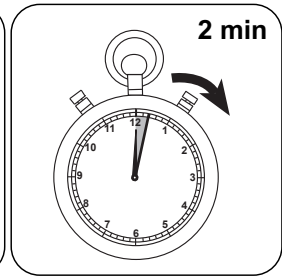
Disolver la(s) tableta(s) girando.



Poner la **cupeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).



Esperar **2 minutos como periodo de reacción**.

Finalizado el periodo de reacción se realizará la determinación automáticamente.
A continuación se visualizará el resultado en mg/L Cloro total.



Método químico

DPD

Apéndice

ES

Interferencia

Interferencias persistentes

- Todos los elementos oxidantes existentes en la muestra reaccionan como el cloro, lo que produce un resultado más elevado.

Interferencias extraíbles

- Las perturbaciones debido a cobre y hierro (III) deben suprimirse mediante EDTA.
- En las muestras con una elevada concentración de iones de calcio* y/o alta conductividad*, se puede producir un enturbiamiento de la muestra con el uso de las tabletas de reactivo, alterando el resultado. En este caso, utilizar alternativamente la tableta reactiva DPD nº 1 High Calcium y la tableta reactiva DPD nº 3 High Calcium. *no se pueden dar valores exactos, ya que la aparición de enturbiamiento dependerá del tipo y composición de la muestra.

Conforme a

EN ISO 7393-2

^{a)} Posible determinación de libre, combinado, total | ^{b)} Reactivo auxiliar, alternativo a DPD No.1/3 en enturbiamientos de la prueba debido a concentraciones elevadas de calcio y/o elevada conductividad



Cobre T

M150

0.05 - 5 mg/L Cu^{a)}

Cu

Biquinolina

ES

Material

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Cobre n° 1	Tabletas / 100	513550BT
Cobre n° 1	Tabletas / 250	513551BT
Cobre n° 2	Tabletas / 100	513560BT
Cobre n° 2	Tabletas / 250	513561BT
Juego cobre n° 1/n° 2 ^a	100 cada	517691BT
Juego cobre n° 1/n° 2 ^a	250 cada	517692BT
ValidCheck cobre 2 mg/l	1 Cantidad	48141525

Preparación

1. Las muestras acuosas muy alcalinas o muy ácidas se deberán neutralizar a un valor de pH de 4 a 6.



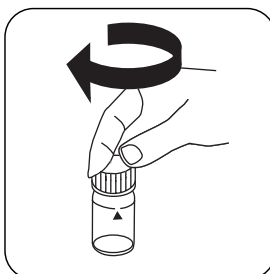
Ejecución de la determinación Cobre libre con tableta

Seleccionar el método en el aparato.

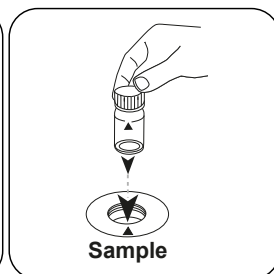
Seleccione además la determinación: libre



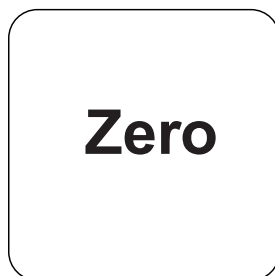
Llenar la cubeta de 24 mm con **10 mL de muestra** .



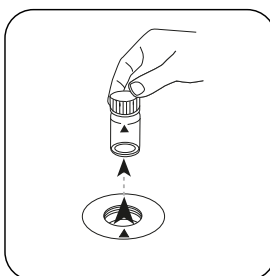
Cerrar la(s) cubeta(s).



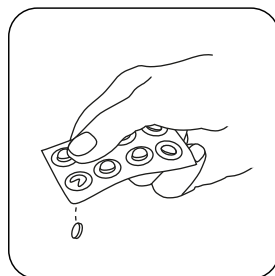
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



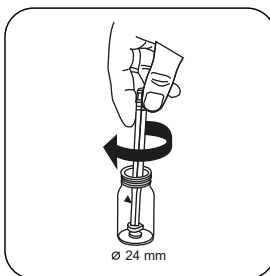
Pulsar la tecla **ZERO**.



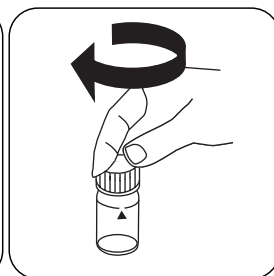
Extraer la cubeta del compartimiento de medición.



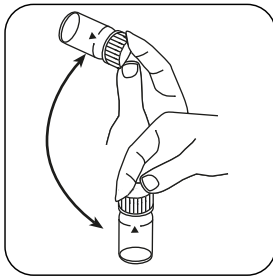
Añadir **tableta COPPER No. 1**.



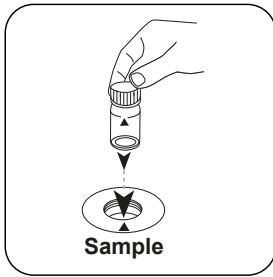
Triturar la(s) tableta(s) girando ligeramente.



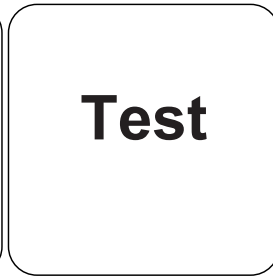
Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.

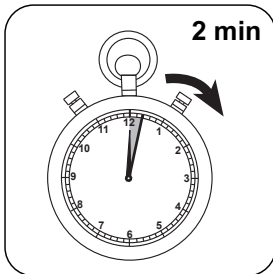


Poner la **cupeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

ES



Esperar **2 minutos como periodo de reacción**.

Finalizado el periodo de reacción se realizará la determinación automáticamente.

A continuación se visualizará el resultado en mg/L Cobre libre.

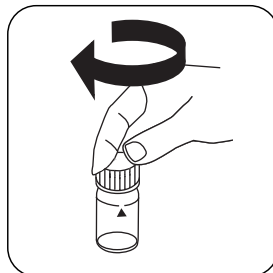
Ejecución de la determinación Cobre total con tableta

Seleccionar el método en el aparato.

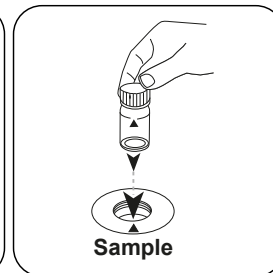
Seleccione además la determinación: total



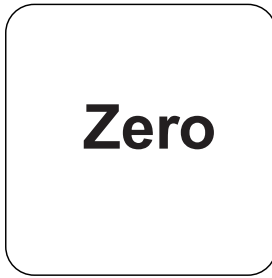
Llenar la cupeta de 24 mm con **10 mL de muestra**.



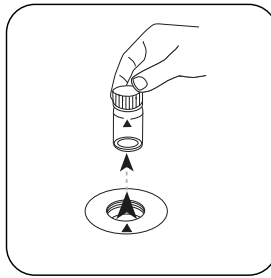
Cerrar la(s) cupeta(s).



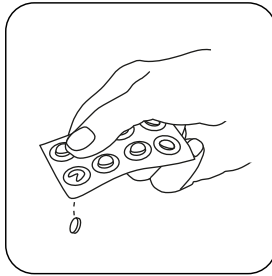
Poner la **cupeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



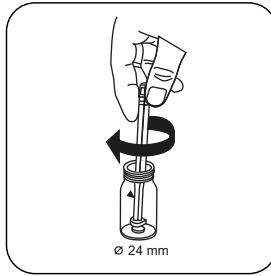
Pulsar la tecla **ZERO**.



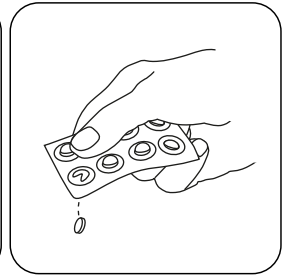
Extraer la cubeta del compartimiento de medición.



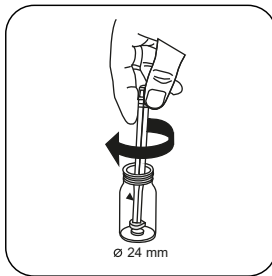
Añadir **tableta COPPER No. 1**.



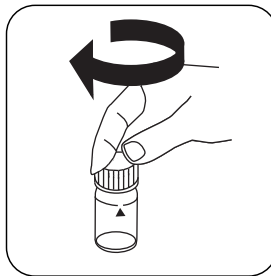
Triturar la(s) tableta(s) girando ligeramente y disolver.



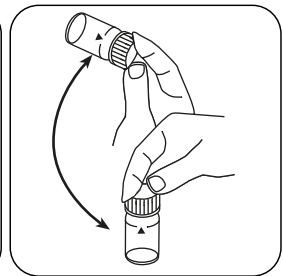
Añadir **tableta COPPER No. 2**.



Triturar la(s) tableta(s) girando ligeramente.



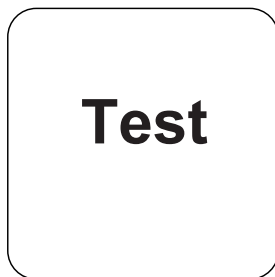
Cerrar la(s) cubeta(s).



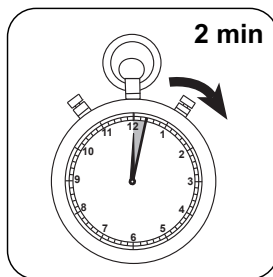
Disolver la(s) tableta(s) girando.



Poner la **cupeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).



Esperar **2 minutos como periodo de reacción**.

Finalizado el periodo de reacción se realizará la determinación automáticamente.

A continuación se visualizará el resultado en mg/L Cobre total.

ES

Método químico

Biquinolina

Apéndice

Interferencia

Interferencias persistentes

1. Cianuro CN⁻ y Plata Ag⁺ perturban la determinación.

Validación del método

Límite de detección	0.05 mg/L
Límite de determinación	0.15 mg/L
Límite del rango de medición	5 mg/L
Sensibilidad	3.8 mg/L / Abs
Intervalo de confianza	0.026 mg/L
Desviación estándar	0.011 mg/L
Coefficiente de variación	0.42 %

Bibliografía

Photometrische Analyse, Lange/Vedjerek, Verlag Chemie 1980

^{a)} Posible determinación de libre, combinado, total



CyA T

M160

10 - 160 mg/L CyA

CyA

Melamina

ES

Material

Material requerido (parcialmente opcional):

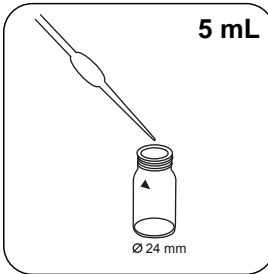
Reactivos	Unidad de embalaje	No. de referencia
CyA-Test	Tabletas / 100	511370BT
CyA-Test	Tabletas / 250	511371BT
Agua desionizada	100 mL	461275
Agua desionizada	250 mL	457022

Notas

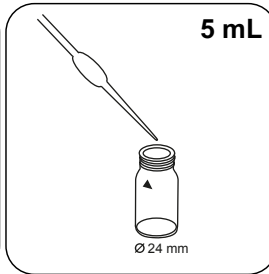
1. El ácido cianúrico provoca un enturbiamiento muy fino de la solución, produciendo un aspecto lechoso. Si hay partículas individuales en la muestra no se deberán a la presencia de ácido cianúrico.

Ejecución de la determinación Prueba de ácido cianúrico con tableta

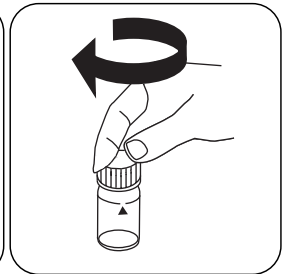
Seleccionar el método en el aparato.



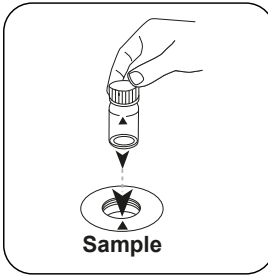
Llenar la cubeta de 24 mm con **5 mL de agua desionizada**.



Añadir **5 mL de muestra** en la cubeta.



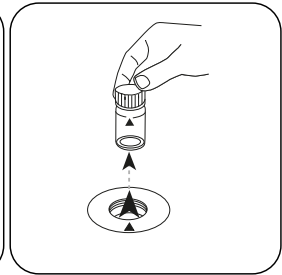
Cerrar la(s) cubeta(s).



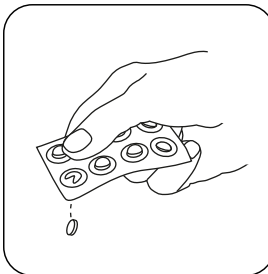
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



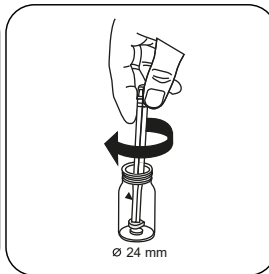
Pulsar la tecla **ZERO**.



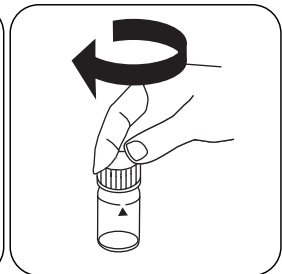
Extraer la cubeta del compartimiento de medición.



Añadir **tableta CyA-Test**.



Triturar la(s) tableta(s) girando ligeramente.



Cerrar la(s) cubeta(s).



Mezclar el contenido girando (durante al menos 60 s hasta la completa disolución de la tableta).




Poner la **cupeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

A continuación se visualizará el resultado en mg/L ácido cianúrico .



Método químico

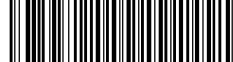
Melamina

Interferencia

Interferencias persistentes

1. Las partículas no disueltas pueden producir resultados mayores. Por ello, es importante que las tabletas se disuelvan completamente.

ES



Hierro T

M220

0.02 - 1 mg/L Fe

FE

Ferrocina / Tioglicolato

ES

Material

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Hierro II LR (Fe ²⁺)	Tabletas / 100	515420BT
Hierro II LR (Fe ²⁺)	Tabletas / 250	515421BT
Hierro LR (Fe ²⁺ und Fe ³⁺)	Tabletas / 100	515370BT
Hierro LR (Fe ²⁺ und Fe ³⁺)	Tabletas / 250	515371BT

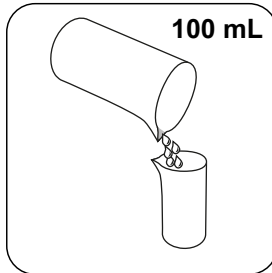
Preparación

1. Las aguas que han sido tratadas con compuestos orgánicos como protección contra la corrosión, etc., pueden oxidarse para destruir los complejos de hierro. Para ello se disuelve una muestra de 100 ml con 1 ml de ácido sulfúrico concentrado y 1 ml de ácido nítrico concentrado y se evapora a la mitad. Después de enfriarse se realiza la disgregación.

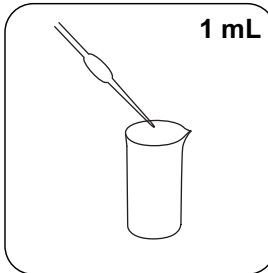
Notas

1. Con este método se realiza la determinación del Fe²⁺ y Fe³⁺ total disuelto.
2. Para determinar Fe²⁺ se utiliza la tableta IRON (II) LR, en lugar de la tableta IRON LR.

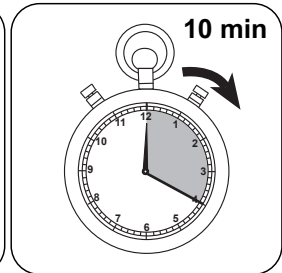
Disgregación



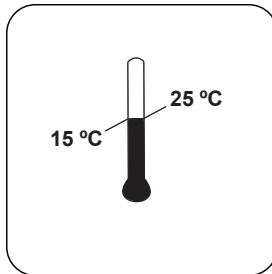
Llenar un recipiente de muestra apropiado con **100 mL de muestra**.



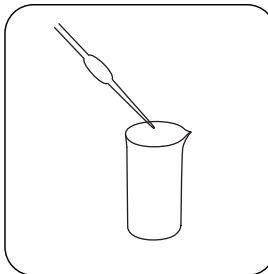
Añadir **1 mL de ácido sulfúrico concentrado**.



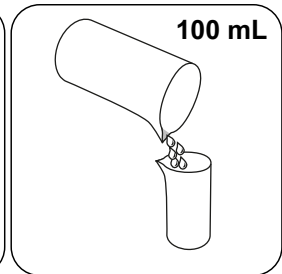
Calentar la muestra durante 10 minutos, o hasta que se haya disuelto totalmente.



Dejar enfriar la muestra a **temperatura ambiente**.



Ajustar el **valor de pH** de la muestra con **solución amoniacal a 3-5**.



Rellenar la muestra con **agua desionizada hasta 100 mL**.

Utilizar esta muestra para el análisis de total de hierro disuelto y no disuelto.

Ejecución de la determinación Hierro (II,III), disuelto con tableta

Seleccionar el método en el aparato.

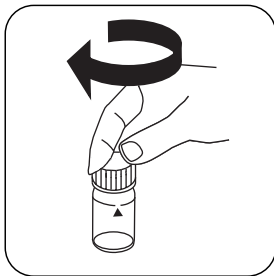
Para la determinación de **Hierro disuelto y sin disolver** realizar la disgregación descrita.



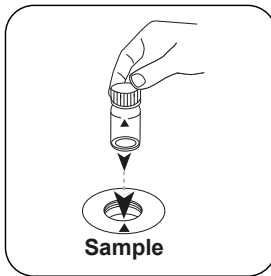
ES



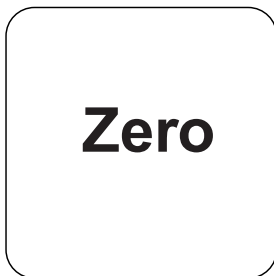
Llenar la cubeta de 24 mm con **10 mL de muestra** .



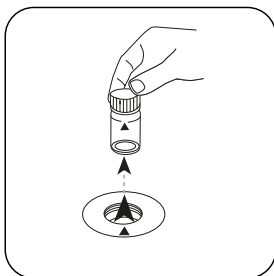
Cerrar la(s) cubeta(s).



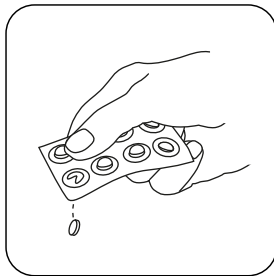
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



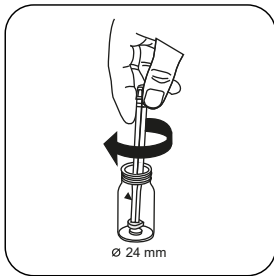
Pulsar la tecla **ZERO**.



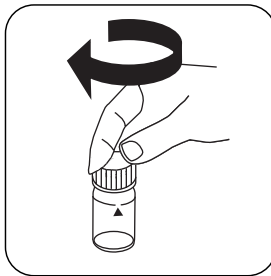
Extraer la cubeta del compartimiento de medición.



Añadir **tableta IRON LR**.



Triturar la(s) tableta(s) girando ligeramente.



Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.

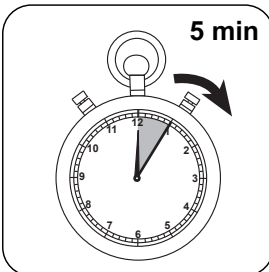


Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

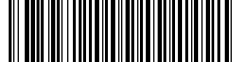
ES



Esperar **5 minutos como período de reacción**.

Finalizado el periodo de reacción se realizará la determinación automáticamente.

A continuación se visualizará el resultado en mg/L Hierro.



Método químico

Ferrocina / Tioglicolato

Apéndice

ES

Interferencia

Interferencias extraíbles

1. La presencia de cobre aumenta el resultado de medición un 10 %. Con una concentración de 10 mg/L de cobre en la muestra, el resultado de la medición se aumenta en 1 mg/L de hierro.
La perturbación puede eliminarse añadiendo tiourea.

Validación del método

Límite de detección	0.01 mg/L
Límite de determinación	0.016 mg/L
Límite del rango de medición	1 mg/L
Sensibilidad	0.92 mg/L / Abs
Intervalo de confianza	0.013 mg/L
Desviación estándar	0.005 mg/L
Coefficiente de variación	1.23 %

Bibliografía

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



Valor de pH T

M330

6.5 - 8.4 pH

PH

Rojo de fenol

Material

ES

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Rojo de fenol fotómetro	Tabletas / 100	511770BT
Rojo de fenol fotómetro	Tabletas / 250	511771BT
Rojo de fenol fotómetro	Tabletas / 500	511772BT

Notas

1. Para la determinación fotométrica del valor de pH solo deben usarse tabletas PHENOL RED selladas en una lámina negra con la palabra adicional PHOTOMETER.

Ejecución de la determinación Valor de pH con tableta

Seleccionar el método en el aparato.



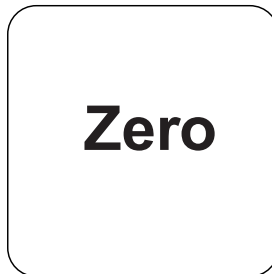
Llenar la cubeta de 24 mm con **10 mL de muestra** .



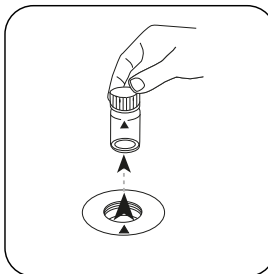
Cerrar la(s) cubeta(s).



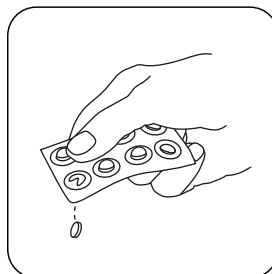
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



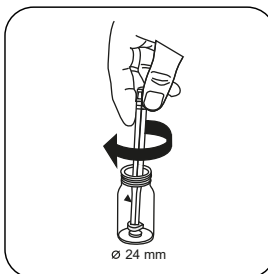
Pulsar la tecla **ZERO**.



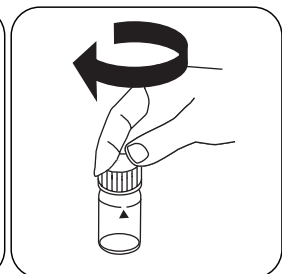
Extraer la cubeta del compartimiento de medición.



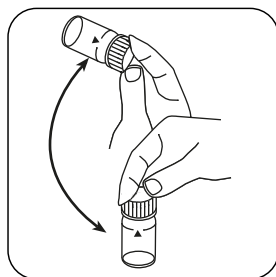
Añadir tableta **PHENOL RED PHOTOMETER**.



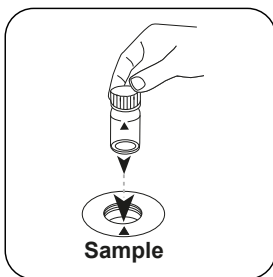
Triturar la(s) tableta(s) girando ligeramente.



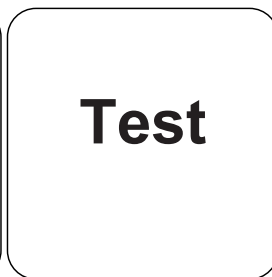
Cerrar la(s) cubeta(s).



Disolver la(s) tableta(s) girando.



Poner la **cupeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

A continuación se visualizará el resultado como valor de pH.

ES

Método químico

Rojo de fenol

Apéndice

Interferencia

ES

Interferencias persistentes

1. Las muestras de agua con baja dureza de carbonato* pueden entregar valores de pH falsos.

* $K_{S4.3} < 0,7 \text{ mmol/l} \triangleq \text{Alcalinidad total} < 35 \text{ mg/L CaCO}_3$.

Interferencias extraíbles

1. Los valores de pH inferiores a 6,5 y superiores a 8,4 pueden conducir a resultados dentro del campo de medición. Se recomienda realizar una prueba de plausibilidad (medidor de pH).
2. Error de sal:
Con concentraciones de sal de hasta 2 g/L no cabe esperar un error de sal destacable, debido a la concentración de sal de la tableta de reactivo. Cuando las concentraciones de sal son mayores, los valores de medición deben corregirse del modo siguiente:

Concentración salina de la muestra en g/L	30 (agua de mar)	60	120	180
Corrección	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾según Kolthoff (1922)

²⁾según Parson y Douglas (1926)

Bibliografía

Colorimetric Chemical Analytical Methods, 9th Edition, London



Valor de pH L

M331

6.5 - 8.4 pH

PH

Rojo de fenol

Material

ES

Material requerido (parcialmente opcional):

Reactivos	Unidad de embalaje	No. de referencia
Solución de rojo de fenol	15 mL	471040
Solución de rojo de fenol	100 mL	471041
Solución rojo de fenol en pack de 6	1 Cantidad	471046

Preparación

1. El tamaño de las gotas, al contrario de las tabletas, puede aumentar las desviaciones del resultado. Mediante el uso de una pipeta (0,18 ml corresponden a 6 gotas) se pueden minimizar estas desviaciones.

Notas

1. Después de usarla, la botella cuentagotas debe cerrarse de nuevo inmediatamente con la tapa roscada del mismo color.
2. Guardar el reactivo a una temperatura entre +6 °C y +10 °C.

Ejecución de la determinación Valor de pH con reactivos líquidos

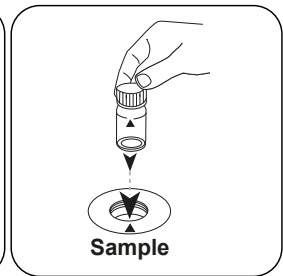
Seleccionar el método en el aparato.



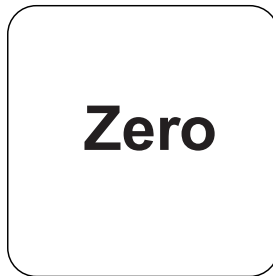
Llenar la cubeta de 24 mm con **10 mL de muestra** .



Cerrar la(s) cubeta(s).



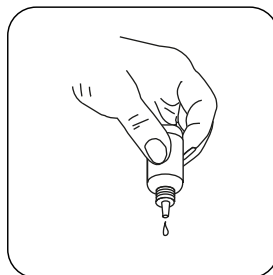
Poner la **cubeta de muestra** en el compartimiento de medición. ¡Debe tenerse en cuenta el posicionamiento!



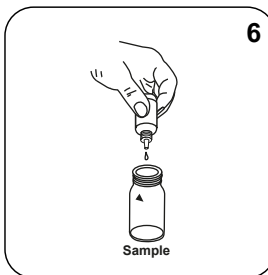
Pulsar la tecla **ZERO**.



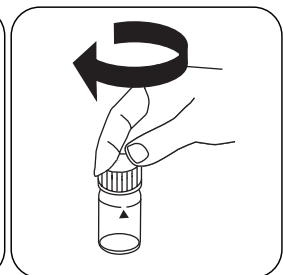
Extraer la cubeta del compartimiento de medición.



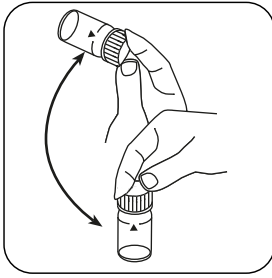
Mantener la botella cuentagotas vertical y añadir gotas del mismo tamaño presionando lentamente.



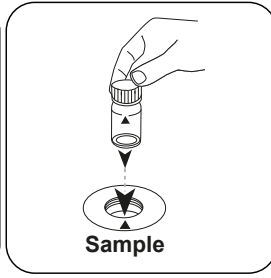
Añadir **6 gotas de PHENOL Red-Lösung** en la cubeta con la muestra.



Cerrar la(s) cubeta(s).



Mezclar el contenido girando.



Poner la **cupeta de muestra** en el compartimento de medición. ¡Debe tenerse en cuenta el posicionamiento!



Pulsar la tecla **TEST** (XD: **START**).

A continuación se visualizará el resultado como valor de pH.

Método químico

Rojo de fenol

Apéndice

Interferencia

ES

Interferencias extraíbles

1. Error de sal: Corrección de valor analizado (valores medios) para muestras con una concentración salina de:


2.	Concentración salina de la muestra	Corrección
	30 g/L (agua de mar)	-0,15 ¹⁾
	60 g/L	-0,21 ²⁾
	120 g/L	-0,26 ²⁾
	180 g/L	-0,29 ²⁾
	¹⁾ según Kolthoff (1922)	²⁾ según Parson y Douglas (1926)

3. En la determinación de muestras acuosas cloradas pueden influir los restos de cloro en la reacción coloreada del reactivo líquido. Esto puede evitarse añadiendo a la muestra un pequeño cristal de tiosulfato sódico ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$), antes de incorporar el reactivo PHENOL RED.

Bibliografía

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20



Nom de la méthode → KS4.3 T

Numéro de méthode → 20

Code à barres pour reconnaître la méthode → [Barcode]

Plage de mesure → 0.1 - 4 mmol/l $K_{S4.3}$

Méthode chimique → Acide / Indicateur

Affichage dans le MD 100 / MD 110 / MD 200 → S:4.3

Informations spécifiques à l'instrument

Le test peut être effectué sur les appareils suivants. De plus, la cuvette requise et la plage d'absorption du photomètre sont indiquées.

Appareils	Cuvette	λ	Gamme de mesure
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	ø 24 mm	610 nm	0.1 - 4 mmol/l $K_{S4.3}$
SpectroDirect, XD 7000, XD 7500	ø 24 mm	615 nm	0.1 - 4 mmol/l $K_{S4.3}$

Matériel

Matériel requis (partiellement optionnel):

Titre	Pack contenant	Code
Alka-M-Photometer	Pastilles / 100	513210BT
Alka-M-Photometer	Pastilles / 250	513211BT

Liste d'applications

- Traitement des eaux usées
- Traitement de l'eau potable
- Traitement de l'eau brute

Indication

1. Les termes Alcalinité-m, Valeur m, Alcalinité totale et Capacité acide $K_{S4.3}$ sont identiques.
2. L'observation exacte du volume d'échantillon de 10 ml est décisive pour l'exactitude du résultat de l'analyse.

Codes de langue ISO 639-1 → FR

État de révision → 01/20

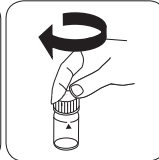
FR Méthodes Manuel 01/20

Procédure du test

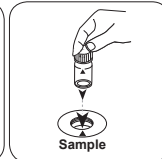
Réalisation de la quantification Capacité acide $K_{s4.3}$ avec pastille

Sélectionnez la méthode sur l'appareil.

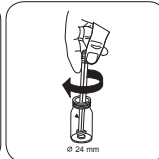
Cette méthode ne nécessite aucune mesure du zéro sur les appareils suivants : XD 7000, XD 7500

Remplissez une cuvette de 24 mm de **10 ml d'échantillon**.

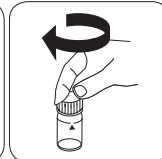
Fermez la(les) cuvette(s).

Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

• • •

Ajoutez une **pastille de ALKA-M-PHOTOMETER**.

Écrasez la(les) pastille(s) en la(les) tournant un peu.



Fermez la(les) cuvette(s).



Alcalinité-m T

M30

5 - 200 mg/L CaCO₃

tA

Acide / Indicateur

FR

Matériel

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
Photomètre Alca-M	Pastilles / 100	513210BT
Photomètre Alca-M	Pastilles / 250	513211BT

Indication

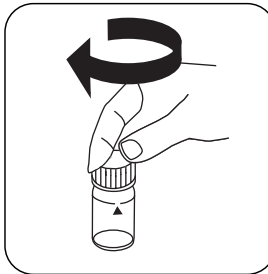
1. Les termes Alcalinité-m, Valeur m, Alcalinité totale et Capacité acide $K_{s4,3}$ sont identiques.
2. L'observation exacte du volume d'échantillon de 10 ml est décisive pour l'exactitude du résultat de l'analyse.

Réalisation de la quantification Alcalinité, total= Alcalinité-m = valeur-m avec pastille

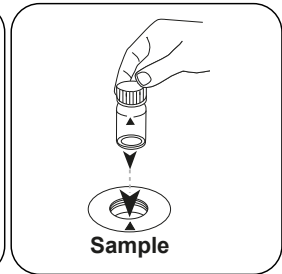
Sélectionnez la méthode sur l'appareil.



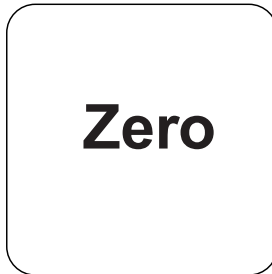
Remplissez une cuvette de 24 mm de **10 mL** d'échantillon.



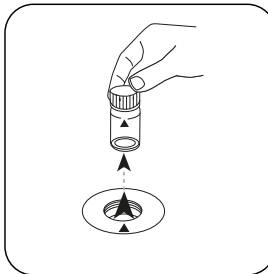
Fermez la(les) cuvette(s).



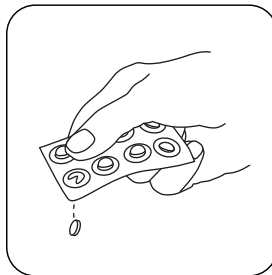
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



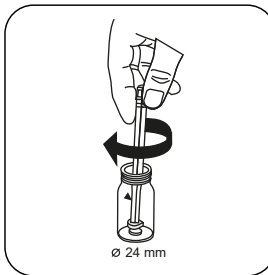
Appuyez sur la touche **ZERO**.



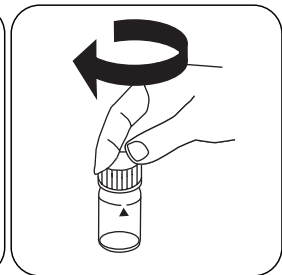
Retirez la cuvette de la chambre de mesure.



Ajoutez une **pastille de ALKA-M-PHOTOMETER**.

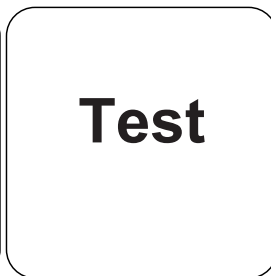
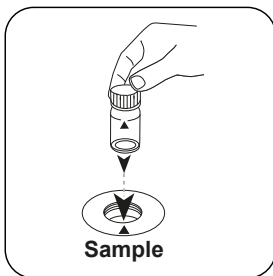
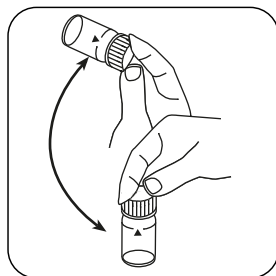


Écrasez la(les) pastille(s) en la(les) tournant un peu.



Fermez la(les) cuvette(s).

FR



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Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.

Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en alcalinité-m.

Analyses

Le tableau suivant identifie les valeurs de sortie qui peuvent être converties en d'autres formes de citation.

Unité	Formes de citation	Facteur de conversion
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

FR

Méthode chimique

Acide / Indicateur

Appendice

Dérivé de

EN ISO 9963-1



Chlore T

M100

0.01 - 6.0 mg/L Cl₂ ^{a)}

CL6

DPD

Matériel

FR

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
DPD N° 1	Pastilles / 100	511050BT
DPD N° 1	Pastilles / 250	511051BT
DPD N° 1	Pastilles / 500	511052BT
DPD N° 3	Pastilles / 100	511080BT
DPD N° 3	Pastilles / 250	511081BT
DPD N° 3	Pastilles / 500	511082BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 100	515740BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 250	515741BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 500	515742BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 100	515730BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 250	515731BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 500	515732BT
DPD N° 4	Pastilles / 100	511220BT
DPD N° 4	Pastilles / 250	511221BT
DPD N° 4	Pastilles / 500	511222BT
DPD N° 3 Evo	Pastilles / 100	511420BT
DPD N° 3 Evo	Pastilles / 250	511421BT
DPD N° 3 Evo	Pastilles / 500	511422BT
DPD N° 4 Evo	Pastilles / 100	511970BT
DPD N° 4 Evo	Pastilles / 250	511971BT
DPD N° 4 Evo	Pastilles / 500	511972BT

Standards disponibles

Titre	Pack contenant	Code
ValidCheck Chlore 1,5 mg/l	1 Pièces	48105510



Échantillonnage

1. Lors de la préparation de l'échantillon, il faudra éviter le dégazage du chrome, par ex. par pipetage ou agitation.
2. L'analyse devra avoir lieu immédiatement après le prélèvement de l'échantillon.

Préparation

1. Nettoyage des cuvettes :
Beaucoup de produits de nettoyage domestiques (par ex. liquide vaisselle) contenant des agents réducteurs, il est possible que lors de la quantification du chlore, les résultats soient plus bas. Pour exclure ces erreurs, les instruments en verre utilisés devraient être insensibles aux effets du chlore. Pour ce faire, il convient de laisser les instruments en verre pendant une heure dans une solution d'hypochlorite de sodium (0,1 g/L) et de bien les rincer ensuite à l'eau déminéralisée (eau entièrement dessalée).
2. Pour la quantification individuelle du chlore libre et du chlore total, il est recommandé d'utiliser à chaque fois un nouveau lot de cuvettes (voir EN ISO 7393-2, § 5.3).
3. La coloration due au DPD a lieu à un pH compris entre 6,2 et 6,5. C'est pourquoi, les réactifs contiennent un tampon pour l'ajustage du pH. Avant l'analyse, les eaux fortement alcalines ou acides devraient être cependant ajustées sur un pH compris entre 6 et 7 (avec 0,5 mol/L d'acide sulfurique ou 1 mol/L de soude caustique).

Indication

1. Les pastilles Evo peuvent être utilisées en remplacement de la pastille standard correspondante (par exemple, DPD n° 3 Evo au lieu de DPD n° 3).

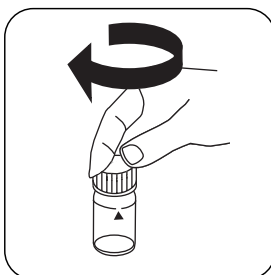


Réalisation de la quantification Chlore libre avec pastilles

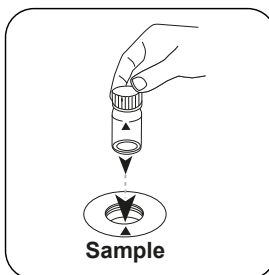
Sélectionnez la méthode sur l'appareil.



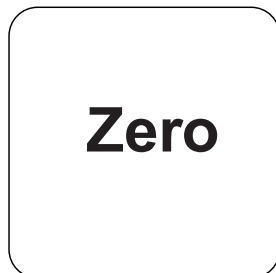
Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



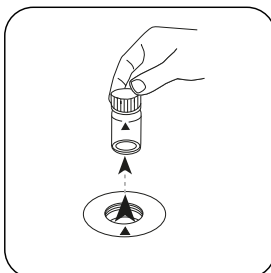
Fermez la(les) cuvette(s).



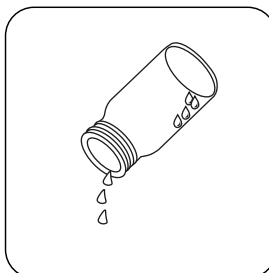
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



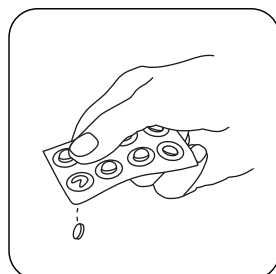
Appuyez sur la touche **ZERO**.



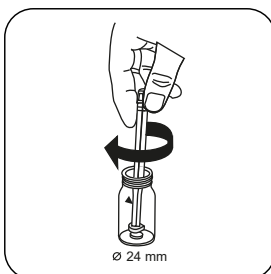
Retirez la cuvette de la chambre de mesure.



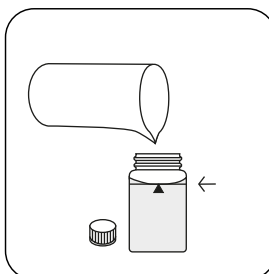
Videz pratiquement la cuvette en y laissant quelques gouttes.



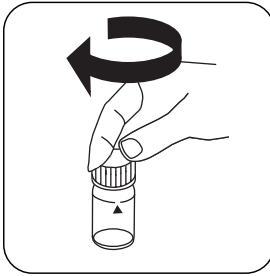
Ajoutez une **pastille de DPD No. 1**.



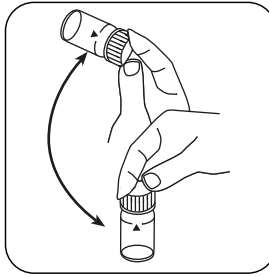
Écrasez la(les) pastille(s) en la(les) tournant un peu.



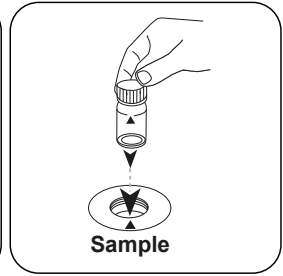
Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.



Fermez la(les) cuvette(s).



Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.



Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

FR

Test

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en mg/L chlore libre.

Réalisation de la quantification Chlore total avec pastilles

Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



Fermez la(les) cuvette(s).



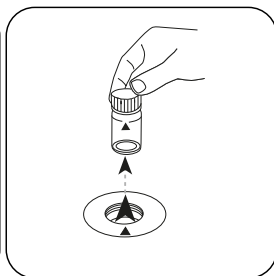
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



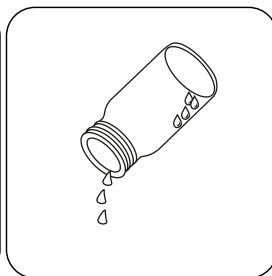
Zero

FR

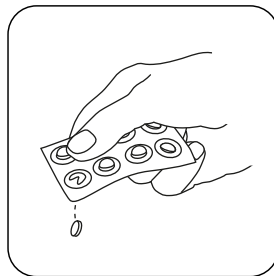
Appuyez sur la touche **ZERO**.



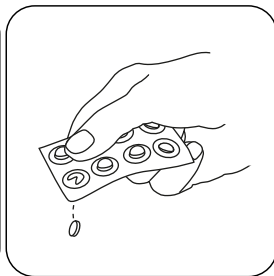
Retirez la cuvette de la chambre de mesure.



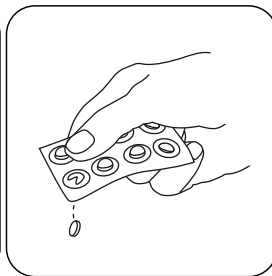
Videz pratiquement la cuvette en y laissant quelques gouttes.



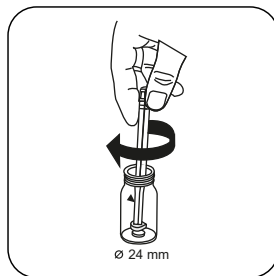
Ajoutez une **pastille de DPD No. 1**.



Ajoutez une **pastille de DPD No. 3**.



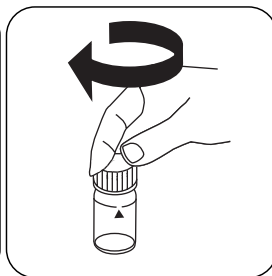
En alternative aux comprimés DPD n° 1 et n° 3, un comprimé DPD n° 4 peut être ajouté.



Écrasez la(les) pastille(s) en la(les) tournant un peu.



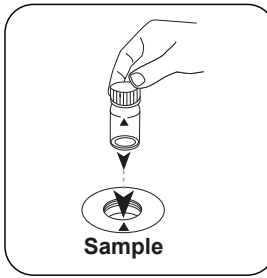
Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.



Fermez la(les) cuvette(s).



Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.



Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST** (XD: **START**).

FR



Attendez la fin du **temps de réaction de 2 minute(s)**.

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L chlore total.



Méthode chimique

DPD

Appendice

FR

Interférences

Interférences persistantes

- Les agents oxydants contenus dans les échantillons réagissent tous comme le chlore, ce qui entraîne des résultats plus élevés.

Interférences exclues

- Les perturbations causées par le cuivre et le fer (III) seront éliminées par EDTA.
- Dans le cas des échantillons à haute concentration en calcium* et/ou conductibilité élevée*, l'utilisation des pastilles de réactif peut causer des turbidités et donc fausser les résultats. Utilisez alors la pastille de réactif DPD N° 1 High Calcium et la pastille de réactif DPD N° 3 High Calcium.
*Nous ne pouvons fournir de valeurs exactes, l'apparition d'une turbidité dépendant du type et de la composition de l'eau d'échantillonnage.
- Les concentrations de chlore supérieures à 10 mg/L peuvent donner des résultats dans la plage de mesure allant jusqu'à 0 mg/L en utilisant des pastilles. En cas de concentration trop élevée de chlore, diluez l'échantillon à l'eau déchlorée. Le réactif est ajouté à 10 mL d'échantillon dilué. Ensuite, la mesure est répétée (test de plausibilité).

Interférences	de / [mg/L]
CrO ₄ ²⁻	0.01
MnO ₂	0.01

Méthode Validation

Limite de détection	0.02 mg/L
Limite de détermination	0.06 mg/L
Fin de la gamme de mesure	6 mg/L
Sensibilité	2.05 mg/L / Abs
Intervalle de confiance	0.04 mg/L
Déviatoin standard	0.019 mg/L
Coefficient de variation	0.87 %

Conformité

EN ISO 7393-2



^aDétermination du libre, combiné et total | ^aautre réactif, utilisé à la place de DPD No.1/3 en cas de turbidité dans l'échantillon d'eau due à une concentration élevée de calcium et/ou une conductivité élevée

**Chlore L****M101****0.02 - 4.0 mg/L Cl₂^{a)}****CL6****DPD****Matériel**

FR

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
DPD 1 solution tampon, flacon bleu	15 mL	471010
DPD 1 solution tampon	100 mL	471011
Solution tampon DPD 1 dans un lot de 6	1 Pièces	471016
DPD 1 solution de réactif, flacon vert	15 mL	471020
DPD 1 solution de réactif	100 mL	471021
Solution de réactif DPD 1 dans un lot de 6	1 Pièces	471026
DPD 3 solution, flacon rouge	15 mL	471030
DPD 3 solution	100 mL	471031
Solution DPD 3 dans un lot de 6	1 Pièces	471036
Kit de réactifs DPD	1 Pièces	471056

Standards disponibles

Titre	Pack contenant	Code
ValidCheck Chlore 1,5 mg/l	1 Pièces	48105510

Échantillonnage

1. Lors de la préparation de l'échantillon, il faudra éviter le dégazage du chrome, par ex. par pipetage ou agitation.
2. L'analyse devra avoir lieu immédiatement après le prélèvement de l'échantillon.



Préparation

1. Nettoyage des cuvettes :
Beaucoup de produits de nettoyage domestiques (par ex. liquide vaisselle) contenant des agents réducteurs, il est possible que lors de la quantification du chlore, les résultats soient plus bas. Pour exclure ces erreurs, les instruments en verre utilisés devraient être insensibles aux effets du chlore. Pour ce faire, il convient de laisser les instruments en verre pendant une heure dans une solution d'hypochlorite de sodium (0,1 g/L) et de bien les rincer ensuite à l'eau déminéralisée (eau entièrement dessalée).
2. Pour la quantification individuelle du chlore libre et du chlore total, il est recommandé d'utiliser à chaque fois un nouveau lot de cuvettes (voir EN ISO 7393-2, § 5.3).
3. La coloration due au DPD a lieu à un pH compris entre 6,2 et 6,5. C'est pourquoi, les réactifs contiennent un tampon pour l'ajustage du pH. Avant l'analyse, les eaux fortement alcalines ou acides devraient être cependant ajustées sur un pH compris entre 6 et 7 (avec 0,5 mol/l d'acide sulfurique ou 1 mol/l de soude caustique).

Indication

1. Après emploi, refermez immédiatement les flacons compte-goutte en utilisant le capot de même couleur.
2. Conservez le lot de réactif à une température de +6 °C à +10 °C.

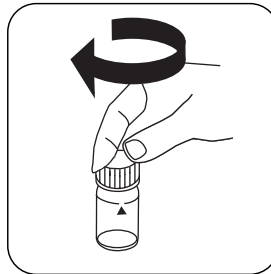


Réalisation de la quantification Chlore libre avec réactifs liquides

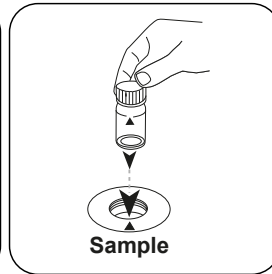
Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



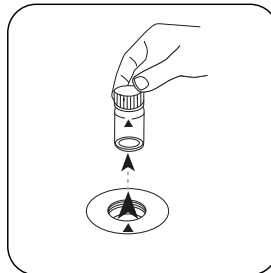
Fermez la(les) cuvette(s).



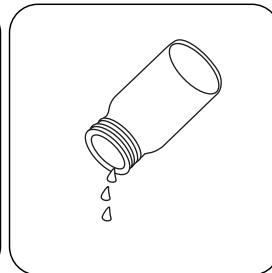
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



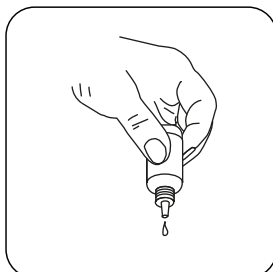
Appuyez sur la touche **ZERO**.



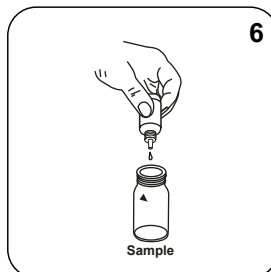
Retirez la cuvette de la chambre de mesure.



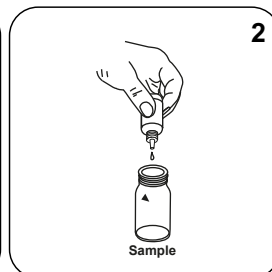
Videz la cuvette.



Tenez les flacons compte-goutte à la verticale et ajoutez des gouttes uniformes en appuyant lentement.



Ajoutez **6 gouttes de DPD 1 Buffer Solution** dans la cuvette réservée à l'échantillon.



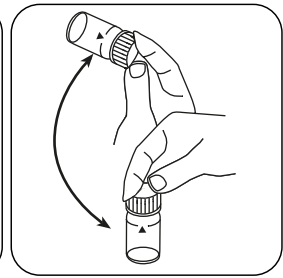
Ajoutez **2 gouttes de DPD 1 Reagent Solution** dans la cuvette réservée à l'échantillon.



Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.

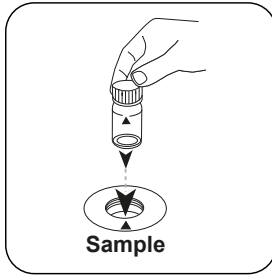


Fermez la(les) cuvette(s).

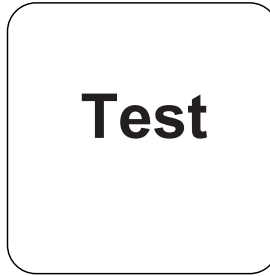


Mélangez le contenu en mettant le tube plusieurs fois à l'envers puis à l'endroit.

FR



Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST (XD: START)**.

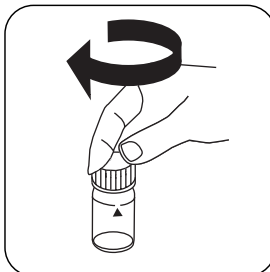
Le résultat s'affiche à l'écran en mg/L chlore libre.

Réalisation de la quantification Chlore total avec réactifs liquides

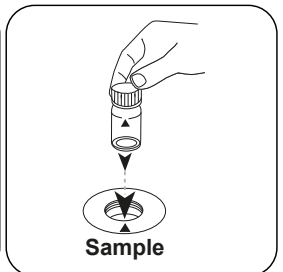
Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL** d'**échantillon**.



Fermez la(les) cuvette(s).

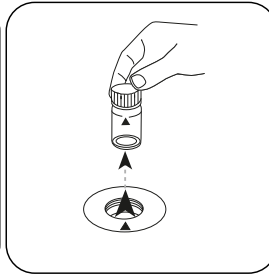


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

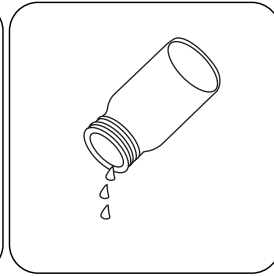


Zero

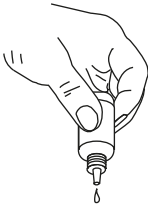
Appuyez sur la touche **ZERO**.



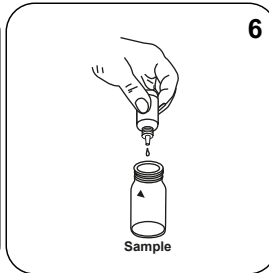
Retirez la cuvette de la chambre de mesure.



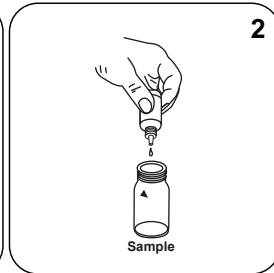
Videz la cuvette.



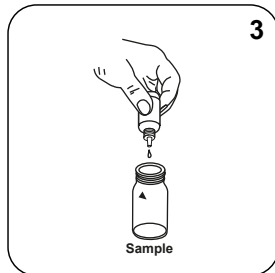
Tenez les flacons compte-goutte à la verticale et ajoutez des gouttes uniformes en appuyant lentement.



Ajoutez **6 gouttes de DPD 1 Buffer Solution** dans la cuvette réservée à l'échantillon.



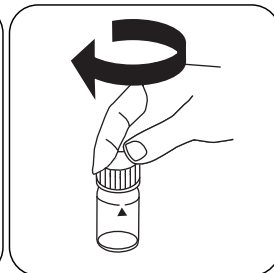
Ajoutez **2 gouttes de DPD 1 Reagent Solution** dans la cuvette réservée à l'échantillon.



Ajoutez **3 gouttes de DPD 3 Solution** dans la cuvette réservée à l'échantillon.



Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.



Fermez la(les) cuvette(s).



Mélangez le contenu en mettant le tube plusieurs fois à l'envers puis à l'endroit.

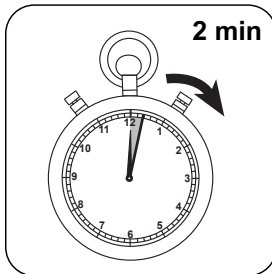


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST** (XD: **START**).

FR



Attendez la fin du **temps de réaction de 2 minute(s)**.

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L chlore total.



Méthode chimique

DPD

Appendice

FR

Interférences

Interférences persistantes

- Les agents oxydants contenus dans les échantillons réagissent tous comme le chlore, ce qui entraîne des résultats plus élevés.

Interférences exclues

- Les perturbations causées par le cuivre et le fer (III) seront éliminées par EDTA.
- Les concentrations de chlore supérieures à 4 mg/L peuvent donner des résultats dans la plage de mesure allant jusqu'à 0 mg/L en utilisant des réactifs liquides. Dans ce cas, diluez l'échantillon à l'eau déchlorée. Le réactif est ajouté à 10 ml d'échantillon dilué. Ensuite, la mesure est répétée (test de plausibilité).

Interférences	de / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Conformité

EN ISO 7393-2

^{a)}Détermination du libre, combiné et total



Chlore HR T

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

Matériel

FR

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
DPD N° 1 HR	Pastilles / 100	511500BT
DPD N° 1 HR	Pastilles / 250	511501BT
DPD N° 1 HR	Pastilles / 500	511502BT
DPD N° 3 HR	Pastilles / 100	511590BT
DPD N° 3 HR	Pastilles / 250	511591BT
DPD N° 3 HR	Pastilles / 500	511592BT
Kit DPD N° 1 HR/N° 3 HR #	100 chacun	517791BT
Kit DPD N° 1 HR/N° 3 HR #	250 chacun	517792BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 100	515740BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 250	515741BT
DPD N° 1 High Calcium ^{e)}	Pastilles / 500	515742BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 100	515730BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 250	515731BT
DPD N° 3 High Calcium ^{e)}	Pastilles / 500	515732BT
DPD N°3 HR Evo	Pastilles / 100	511920BT
DPD N° 3 HR Evo	Pastilles / 250	511921BT
DPD N° 3 HR Evo	Pastilles / 500	511922BT

Échantillonnage

1. Lors de la préparation de l'échantillon, il faudra éviter le dégazage du chrome, par ex. par pipetage ou agitation.
2. L'analyse devra avoir lieu immédiatement après le prélèvement de l'échantillon.



Préparation

1. Nettoyage des cuvettes :
Beaucoup de produits de nettoyage domestiques (par ex. liquide vaisselle) contenant des agents réducteurs, il est possible que lors de la quantification du chlore, les résultats soient plus bas. Pour exclure ces erreurs, les instruments en verre utilisés devraient être insensibles aux effets du chlore. Pour ce faire, il convient de laisser les instruments en verre pendant une heure dans une solution d'hypochlorite de sodium (0,1 g/L) et de bien les rincer ensuite à l'eau déminéralisée (eau entièrement dessalée).
2. Pour la quantification individuelle du chlore libre et du chlore total, il est recommandé d'utiliser à chaque fois un nouveau lot de cuvettes (voir EN ISO 7393-2, § 5.3).
3. La coloration due au DPD a lieu à un pH compris entre 6,2 et 6,5. C'est pourquoi, les réactifs contiennent un tampon pour l'ajustage du pH. Avant l'analyse, les eaux fortement alcalines ou acides devraient être cependant ajustées sur un pH compris entre 6 et 7 (avec 0,5 mol/L d'acide sulfurique ou 1 mol/L de soude caustique).

Indication

1. Les pastilles Evo peuvent être utilisées en remplacement de la pastille standard correspondante (par exemple, DPD n° 3 Evo au lieu de DPD n° 3).

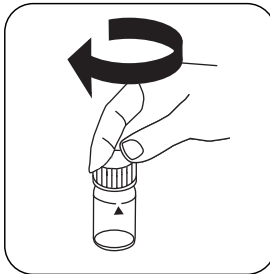


Réalisation de la quantification Chlore HR libre avec pastilles

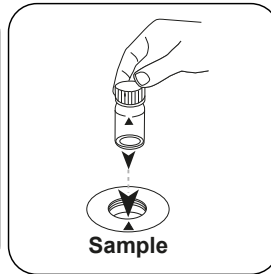
Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



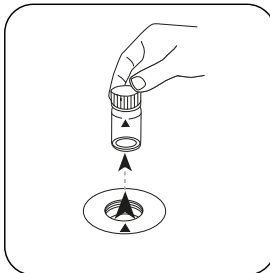
Fermez la(les) cuvette(s).



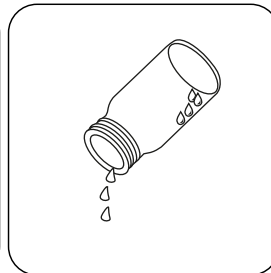
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **ZERO**.



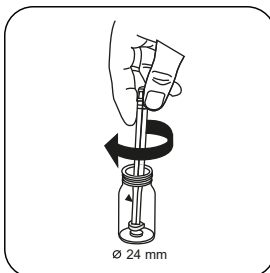
Retirez la cuvette de la chambre de mesure.



Videz pratiquement la cuvette en y laissant quelques gouttes.



Ajoutez une **pastille de DPD No. 1 HR**.



Écrasez la(les) pastille(s) en la(les) tournant un peu.



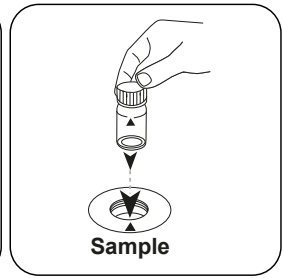
Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.



Fermez la(les) cuvette(s).



Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.



Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

FR

Test

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en mg/L chlore libre.

Réalisation de la quantification Chlore HR total avec pastilles

Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



Fermez la(les) cuvette(s).

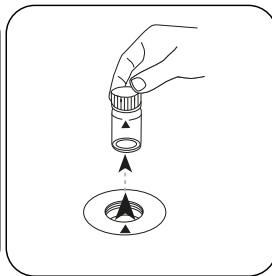


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

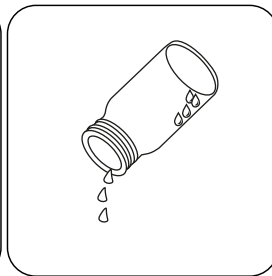


Zero

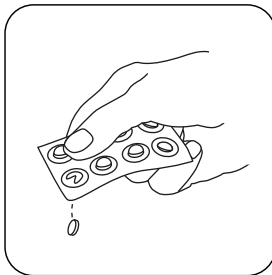
Appuyez sur la touche **ZERO**.



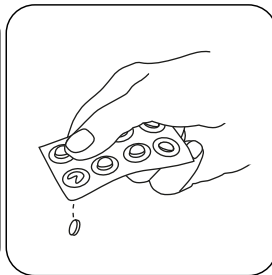
Retirez la cuvette de la chambre de mesure.



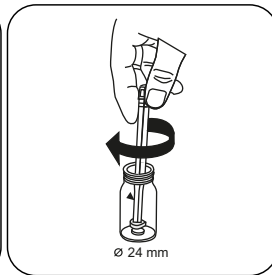
Videz pratiquement la cuvette en y laissant quelques gouttes.



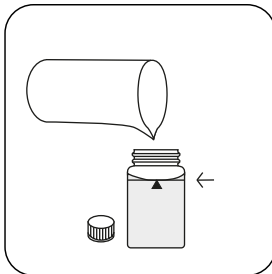
Ajoutez une **pastille de DPD No. 1 HR**.



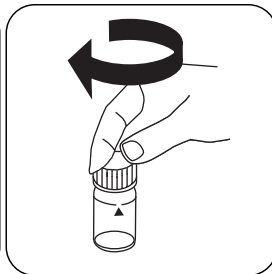
Ajoutez une **pastille de DPD No. 3 HR**.



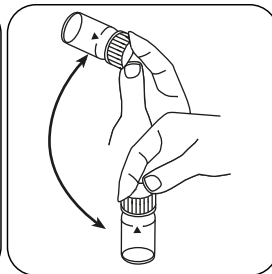
Écrasez la(les) pastille(s) en la(les) tournant un peu.



Remplissez la cuvette jusqu'au **repère de 10 mL** en y versant l'**échantillon**.



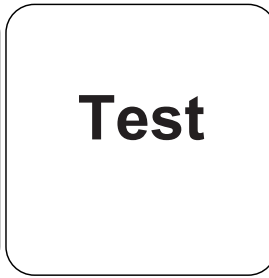
Fermez la(les) cuvette(s).



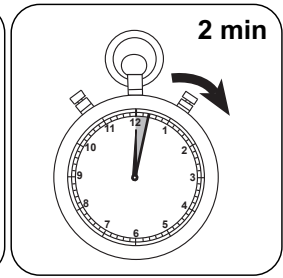
Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.



Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST** (XD: **START**).



Attendez la fin du **temps de réaction de 2 minute(s)**.

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L chlore total.



Méthode chimique

DPD

Appendice

FR

Interférences

Interférences persistantes

- Les agents oxydants contenus dans les échantillons réagissent tous comme le chlore, ce qui entraîne des résultats plus élevés.

Interférences exclues

- Les perturbations causées par le cuivre et le fer (III) seront éliminées par EDTA.
- Dans le cas des échantillons à haute concentration en calcium* et/ou conductibilité élevée*, l'utilisation des pastilles de réactif peut causer des turbidités et donc fausser les résultats. Utilisez alors la pastille de réactif DPD N° 1 High Calcium et la pastille de réactif DPD N° 3 High Calcium.
*Nous ne pouvons fournir de valeurs exactes, l'apparition d'une turbidité dépendant du type et de la composition de l'eau d'échantillonnage.

Conformité

EN ISO 7393-2

^aDétermination du libre, combiné et total | ^aautre réactif, utilisé à la place de DPD No.1/3 en cas de turbidité dans l'échantillon d'eau due à une concentration élevée de calcium et/ou une conductivité élevée | ^b agitateur inclus



Cuivre T

M150

0.05 - 5 mg/L Cu^{a)}

Cu

Biquinoline

FR

Matériel

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
Cuivre N° 1	Pastilles / 100	513550BT
Cuivre N° 1	Pastilles / 250	513551BT
Cuivre N° 2	Pastilles / 100	513560BT
Cuivre N° 2	Pastilles / 250	513561BT
Kit cuivre N° 1/N° 2 [#]	100 chacun	517691BT
Kit cuivre N° 1/N° 2 [#]	250 chacun	517692BT
ValidCheck Cuivre 2 mg/l	1 Pièces	48141525

Préparation

1. Avant l'analyse, les eaux fortement alcalines ou acides devraient être ajustées sur un pH 4 à 6.



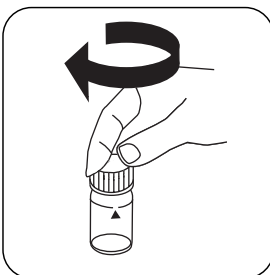
Réalisation de la quantification Cuivre, libre avec pastille

Sélectionnez la méthode sur l'appareil.

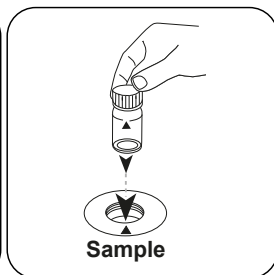
Sélectionnez également la quantification : libre



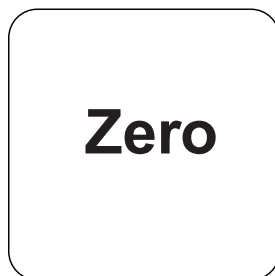
Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



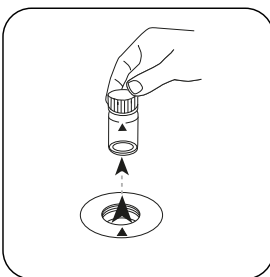
Fermez la(les) cuvette(s).



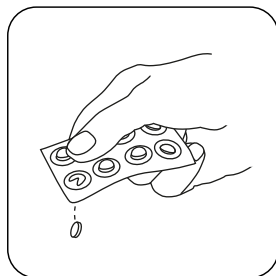
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



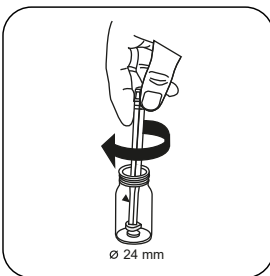
Appuyez sur la touche **ZERO**.



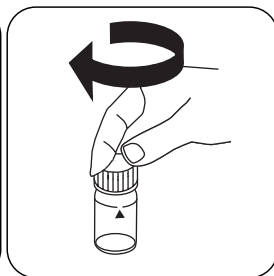
Retirez la cuvette de la chambre de mesure.



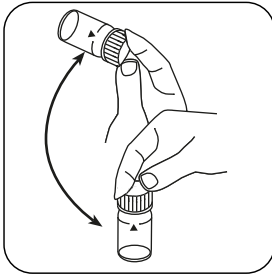
Ajoutez une **pastille de COPPER No. 1**.



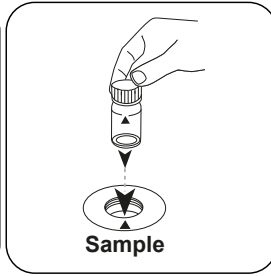
Écrasez la(les) pastille(s) en la(les) tournant un peu.



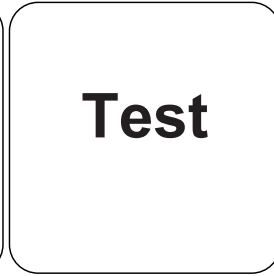
Fermez la(les) cuvette(s).



Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.

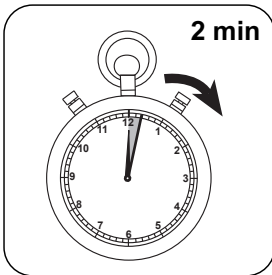


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST** (XD: **START**).

FR



Attendez la fin du **temps de réaction de 2 minute(s)**.

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L Cuivre, libre.

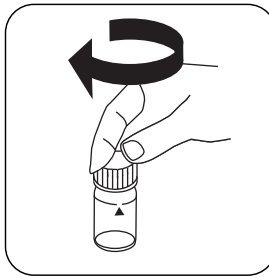
Réalisation de la quantification Cuivre, total avec pastille

Sélectionnez la méthode sur l'appareil.

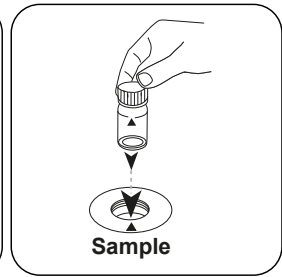
Sélectionnez également la quantification : total



Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.

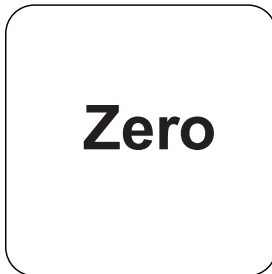


Fermez la(les) cuvette(s).

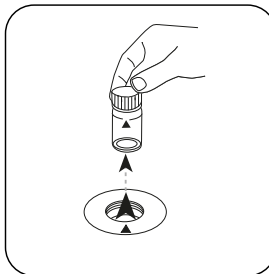


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

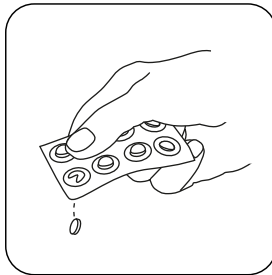
FR



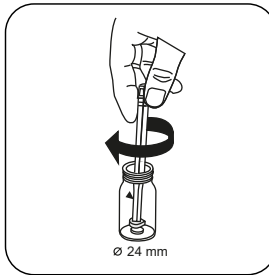
Appuyez sur la touche **ZERO**.



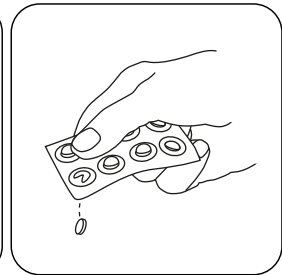
Retirez la cuvette de la chambre de mesure.



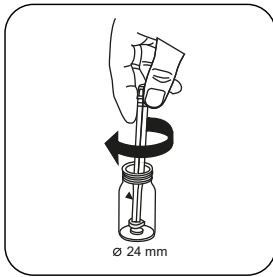
Ajoutez une **pastille de COPPER No. 1**.



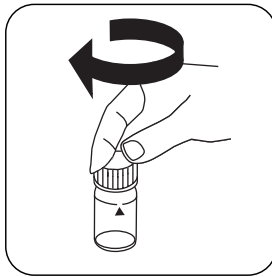
Écrasez et dissolvez la(les) pastille(s) en la(les) tournant un peu.



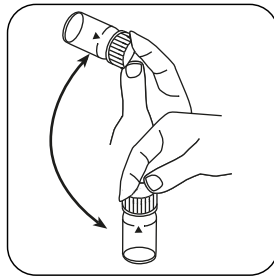
Ajoutez une **pastille de COPPER No. 2**.



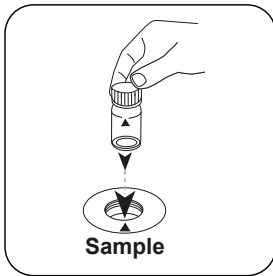
Écrasez la(les) pastille(s)
en la(les) tournant un peu.



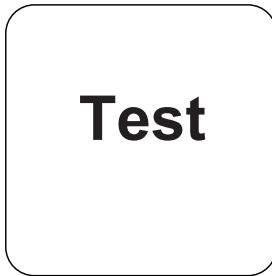
Fermez la(les) cuvette(s).



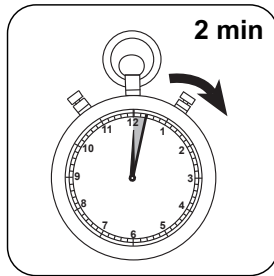
Dissolvez la(les) pastille(s)
en mettant le tube plusieurs
fois à l'envers.



Placez la **cuvette réservée**
à l'échantillon dans la
chambre de mesure.
Attention à la positionner
correctement.



Appuyez sur la touche
TEST (XD: START).



Attendez la fin du **temps de**
réaction de 2 minute(s) .

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L Cuivre, total.

Méthode chimique

Biquinoline

Appendice

Interférences

Interférences persistantes

1. Cyanure CN⁻ et Argent Ag⁺ perturbent la quantification.

Méthode Validation

Limite de détection	0.05 mg/L
Limite de détermination	0.15 mg/L
Fin de la gamme de mesure	5 mg/L
Sensibilité	3.8 mg/L / Abs
Intervalle de confiance	0.026 mg/L
Déviation standard	0.011 mg/L
Coefficient de variation	0.42 %

Bibliographie

Photometrische Analyse, Lange/Vedjerek, Verlag Chemie 1980

^aDétermination du libre, combiné et total | ^b* agitateur inclus

**CyA T****M160****10 - 160 mg/L CyA****CyA****Mélatamine**

FR

Matériel

Matériel requis (partiellement optionnel):

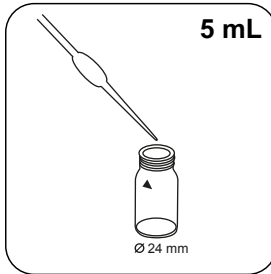
Réactifs	Pack contenant	Code
Test CyA	Pastilles / 100	511370BT
Test CyA	Pastilles / 250	511371BT
Eau purifiée	100 mL	461275
Eau purifiée	250 mL	457022

Indication

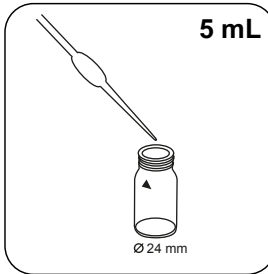
1. L'acide cyanurique cause une turbidité très fine répartie d'aspect laiteux. Certaines particules ne s'expliquent pas par la présence d'acide cyanurique.

Réalisation de la quantification Test à l'acide cyanurique avec pastille

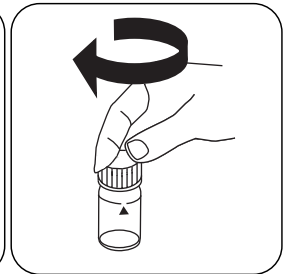
Sélectionnez la méthode sur l'appareil.



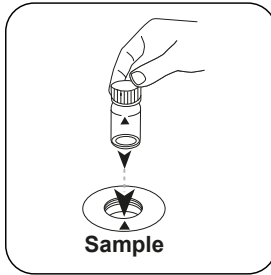
Remplissez une cuvette de 24 mm de **5 mL d'eau déminéralisée**.



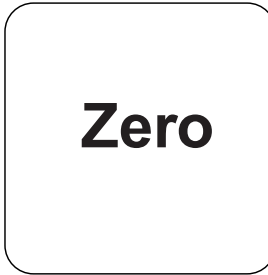
Versez **5 mL d'échantillon** dans la cuvette.



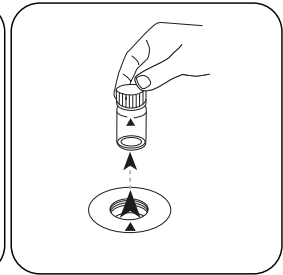
Fermez la(les) cuvette(s).



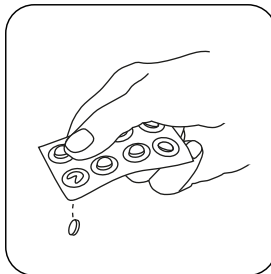
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



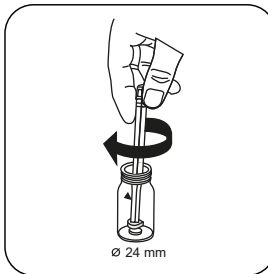
Appuyez sur la touche **ZERO**.



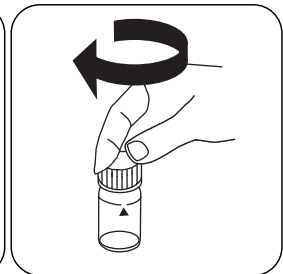
Retirez la cuvette de la chambre de mesure.



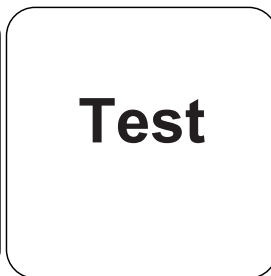
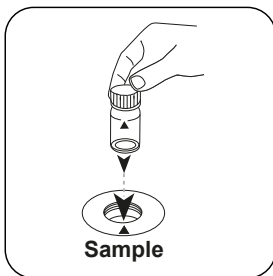
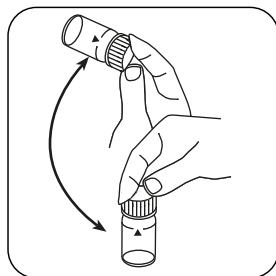
Ajoutez une **pastille de CyA-Test**.



Écrasez la(les) pastille(s) en la(les) tournant un peu.



Fermez la(les) cuvette(s).



FR

Retourner plusieurs fois pour mélanger le contenu (pendant au moins 60 s, jusqu'à ce que la pastille soit complètement dissoute) .

Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en mg/L Acide cyanurique.



Méthode chimique

Mélatamine

Interférences

Interférences persistantes

1. Les particules non dissoutes peuvent entraîner des résultats plus élevés. Il est donc important de dissoudre entièrement les pastilles.

FR



Fer T

M220

0.02 - 1 mg/L Fe

FE

Ferrozine / Thioglycolate

FR

Matériel

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
Fer II LR (Fe ²⁺)	Pastilles / 100	515420BT
Fer II LR (Fe ²⁺)	Pastilles / 250	515421BT
Fer LR (Fe ²⁺ und Fe ³⁺)	Pastilles / 100	515370BT
Fer LR (Fe ²⁺ und Fe ³⁺)	Pastilles / 250	515371BT

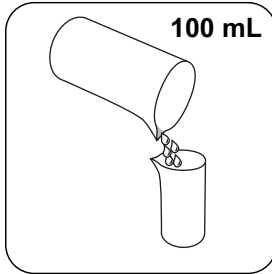
Préparation

- Il faudra éventuellement oxyder les eaux auparavant traitées avec des composés organiques pour les protéger de la corrosion etc. afin de détruire les complexes du fer. À un échantillon de 100 ml, on ajoutera 1 ml d'acide sulfurique concentré et 1 ml d'acide nitrique concentré pour réduire le tout de moitié par évaporation. Le fractionnement est effectué après le refroidissement.

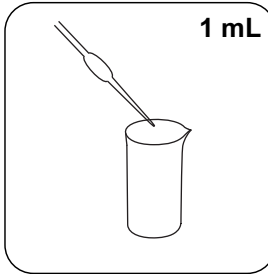
Indication

- Cette méthode permet de quantifier le Fe²⁺ total dissous et Fe³⁺.
- Pour la quantification du Fe²⁺, la pastille IRON LR est remplacée par la pastille IRON (II) LR.

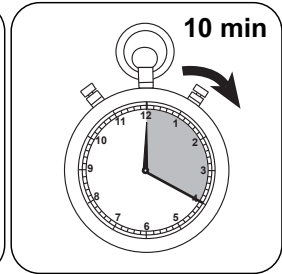
Fractionnement



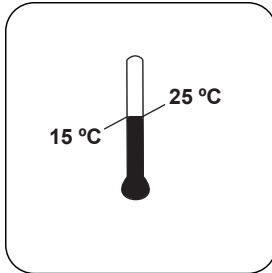
Versez **100 mL d'échantillon** dans un tube pour échantillon adéquat.



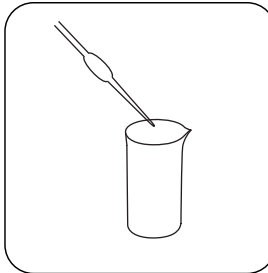
Ajoutez **1 mL de d'acide sulfurique concentré**.



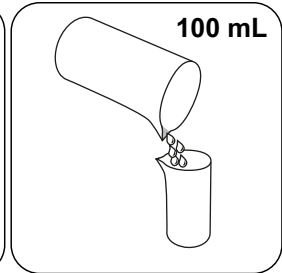
Réchauffez l'échantillon pendant **10 minutes**, ou jusqu'à ce que tout soit entièrement dissous.



Laissez refroidir l'échantillon à **température ambiante**.



Réglez le **pH** de l'échantillon avec **solution ammoniacquée** pour obtenir 3-5.



Complétez l'échantillon en ajoutant **d'eau déminéralisée** pour obtenir **100 mL**.

Utilisez cet échantillon pour analyser Fer total résolu et dissous.

Réalisation de la quantification Fer (II,III), dissous avec pastille

Sélectionnez la méthode sur l'appareil.

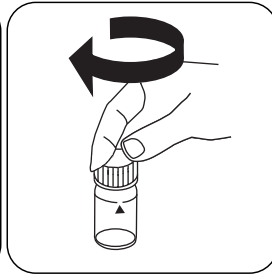
Pour la quantification de **Fer dissous et non dissous**, procédez au fractionnement décrit.



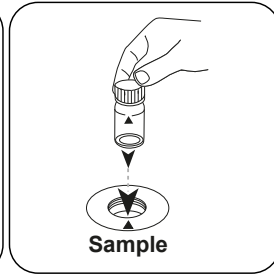
FR



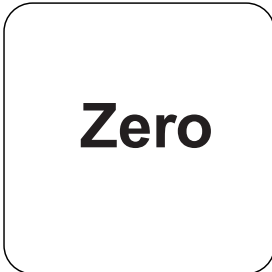
Remplissez une cuvette de 24 mm de **10 mL d'échantillon**.



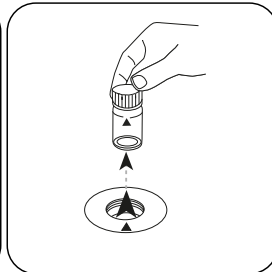
Fermez la(les) cuvette(s).



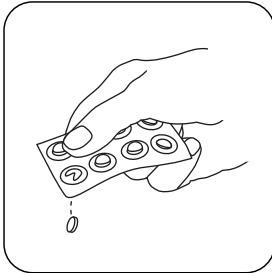
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



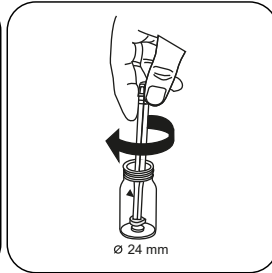
Appuyez sur la touche **ZERO**.



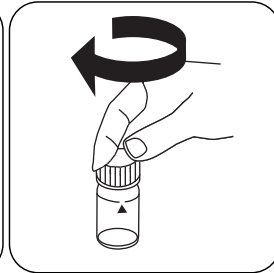
Retirez la cuvette de la chambre de mesure.



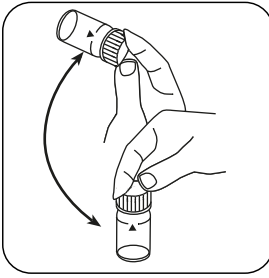
Ajoutez une **pastille de IRON LR**.



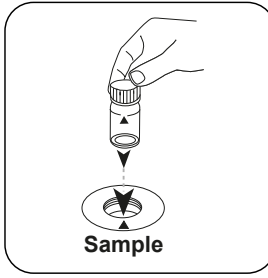
Écrasez la(les) pastille(s) en la(les) tournant un peu.



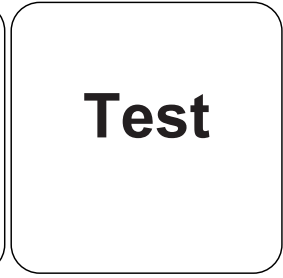
Fermez la(les) cuvette(s).



Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.

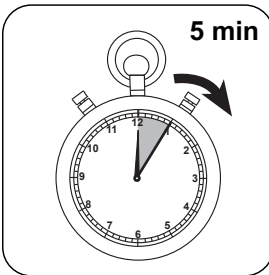


Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



Appuyez sur la touche **TEST** (XD: **START**).

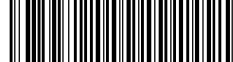
FR



Attendez la fin du **temps de réaction de 5 minute(s)**.

À l'issue du temps de réaction, la mesure est effectuée automatiquement.

Le résultat s'affiche à l'écran en mg/L fer.



Méthode chimique

Ferrozine / Thioglycolate

Appendice

FR

Interférences

Interférences exclues

1. La présence de cuivre augmente le résultat de 10 %. À une concentration de cuivre égale à 10 mg/L dans l'échantillon, le résultat augmente de 1 mg/L de fer. La perturbation peut être éliminée par un apport de thiocarbamide.

Méthode Validation

Limite de détection	0.01 mg/L
Limite de détermination	0.016 mg/L
Fin de la gamme de mesure	1 mg/L
Sensibilité	0.92 mg/L / Abs
Intervalle de confiance	0.013 mg/L
Déviatation standard	0.005 mg/L
Coefficient de variation	1.23 %

Bibliographie

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



Valeur du pH T

M330

6.5 - 8.4 pH

PH

Rouge de phénol

FR

Matériel

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
Rouge de phénol Photomètre	Pastilles / 100	511770BT
Rouge de phénol Photomètre	Pastilles / 250	511771BT
Rouge de phénol Photomètre	Pastilles / 500	511772BT

Indication

1. Pour la quantification photométrique du pH, n'utilisez que des pastilles PHENOL RED avec étiquette noire, sur lesquelles le terme PHOTOMER est apposé.

Réalisation de la quantification Valeur du pH avec pastille

Sélectionnez la méthode sur l'appareil.



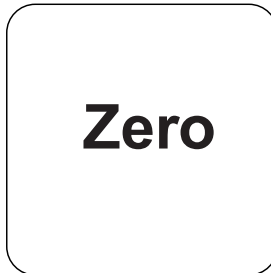
Remplissez une cuvette de 24 mm de **10 mL** d'échantillon.



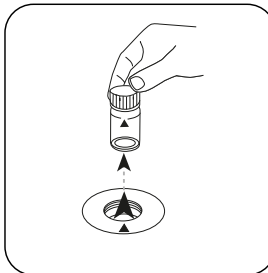
Fermez la(les) cuvette(s).



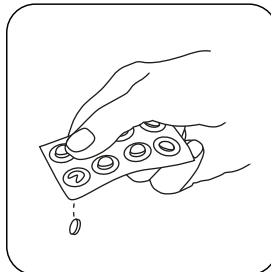
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



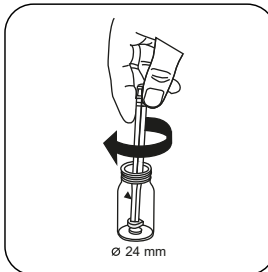
Appuyez sur la touche **ZERO**.



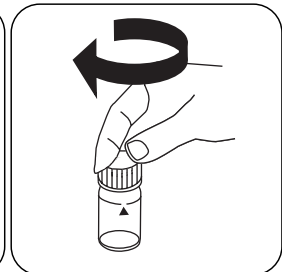
Retirez la cuvette de la chambre de mesure.



Ajoutez une **pastille de PHENOL RED PHOTOMETER**.

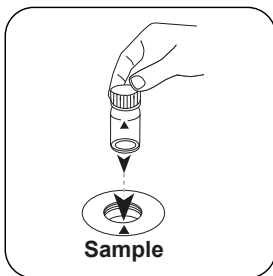


Écrasez la(les) pastille(s) en la(les) tournant un peu.



Fermez la(les) cuvette(s).

FR



FR

Dissolvez la(les) pastille(s) en mettant le tube plusieurs fois à l'envers.

Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en valeur du pH.

Méthode chimique

Rouge de phénol

Appendice

Interférences

FR

Interférences persistantes

1. Les échantillons d'eau avec faible dureté carbonatée* peuvent fausser les pH.
* $K_{S_{4,3}} < 0,7$ mmol/l \triangleq alcalinité totale < 35 mg/L CaCO_3 .

Interférences exclues

1. Les pH inférieurs à 6,5 et supérieurs à 8,4 peuvent provoquer des résultats dans la plage de mesure. Il est recommandé d'effectuer un test de plausibilité (appareil de mesure du pH).
2. Erreur de sel :
À des concentrations du sel jusqu'à 2 g/L, il ne faut s'attendre à aucune erreur digne de ce nom en raison de la concentration en sel de la pastille de réactif. À des concentrations supérieures, les valeurs mesurées seront corrigées comme suit :

Concentration en sel de l'échantillon en g/L	30 (eau de mer)	60	120	180
Correction	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾selon Kolthoff (1922)

²⁾selon Parson et Douglas (1926)

Bibliographie

Colorimetric Chemical Analytical Methods, 9th Edition, London



Valeur du pH L

M331

6.5 - 8.4 pH

PH

Rouge de phénol

FR

Matériel

Matériel requis (partiellement optionnel):

Réactifs	Pack contenant	Code
Solution de phénol rouge	15 mL	471040
Solution de phénol rouge	100 mL	471041
Solution de phénol rouge dans un lot de 6	1 Pièces	471046

Préparation

- En raison des différentes tailles de gouttes, le résultat peut présenter des écarts supérieurs à ceux des pastilles.
Cet écart peut être réduit à un minimum en utilisant une pipette (0,18 ml correspondent à 6 gouttes).

Indication

- Après emploi, refermez immédiatement le flacon compte-goutte en utilisant le capot de même couleur.
- Conservez le réactif à une température de +6 °C à +10 °C.

Réalisation de la quantification Valeur du pH avec réactif liquide

Sélectionnez la méthode sur l'appareil.



Remplissez une cuvette de 24 mm de **10 mL** d'échantillon.



Fermez la(les) cuvette(s).



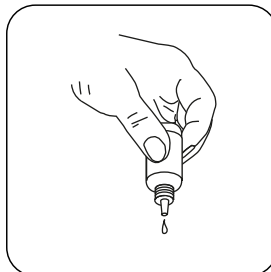
Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.



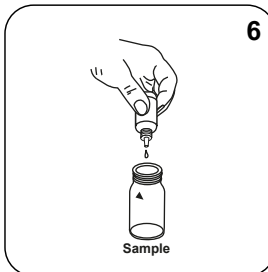
Appuyez sur la touche **ZERO**.



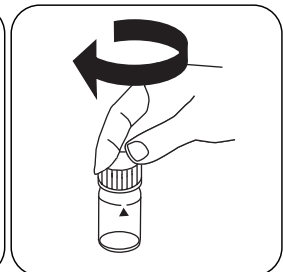
Retirez la cuvette de la chambre de mesure.



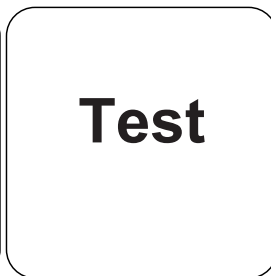
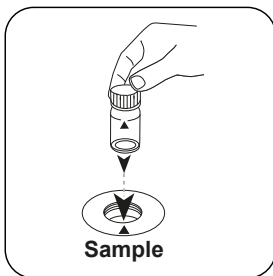
Tenez les flacons compte-goutte à la verticale et ajoutez des gouttes uniformes en appuyant lentement.



Ajoutez **6 gouttes de PHENOL Red-Lösung** dans la cuvette réservée à l'échantillon.



Fermez la(les) cuvette(s).



FR

Mélangez le contenu en mettant le tube plusieurs fois à l'envers puis à l'endroit.

Placez la **cuvette réservée à l'échantillon** dans la chambre de mesure. Attention à la positionner correctement.

Appuyez sur la touche **TEST** (XD: **START**).

Le résultat s'affiche à l'écran en valeur du pH.

Méthode chimique

Rouge de phénol

Appendice

Interférences

FR

Interférences exclues

1. Erreur de sel : Correction de la mesure du sel (valeurs moyennes) pour les échantillons présentant une concentration en sel de :


2.	Concentration en sel de l'échantillon	Correction
	30 g/L (eau de mer)	-0,15 ¹⁾
	60 g/L	-0,21 ²⁾
	120 g/L	-0,26 ²⁾
	180 g/L	-0,29 ²⁾
	¹⁾ selon Kolthoff (1922)	²⁾ selon Parson et Douglas (1926)

3. Lors de l'analyse de l'eau chlorée, la concentration résiduelle en chlore peut influencer la coloration du réactif liquide. Ceci est empêché en introduisant un petit cristal de hiosulfate de sodium ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) dans la solution d'échantillonnage avant d'ajouter la solution PHENOL RED.

Bibliographie

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20



Denominazione metodo

Numero metodo

Codice a barre per riconoscere il metodo

Range di misura

$K_{S_{4.3}} T$
0.1 - 4 mmol/l $K_{S_{4.3}}$

20
S:4.3

Indicazione sul display del MD 100 / MD 110 / MD 200

Metodo chimico

Acido/indicatore

Informazioni specifiche dello strumento

Il test può essere eseguito sui seguenti dispositivi. Inoltre, sono indicate la cuvetta richiesta e il range di assorbimento del fotometro.

Dispositivi	Cuvetta	λ	Campo di misura
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	ø 24 mm	610 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$
SpectroDirect, XD 7000, XD 7500	ø 24 mm	615 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$

Materiale

Materiale richiesto (in parte facoltativo):

Titolo	Unità di imballaggio	N. ordine
Alka-M-Photometer	Pastiglia / 100	513210BT
Alka-M-Photometer	Pastiglia / 250	513211BT

Campo di applicazione

- Trattamento acqua di scarico
- Trattamento acqua potabile
- Trattamento acqua non depurata

Note

1. I termini alcalinità M, valore M, alcalinità totale e capacità acida $K_{S_{4.3}}$ sono equivalenti.
2. Per l'accuratezza del risultato dell'analisi è fondamentale che il volume del campione misuri esattamente 10 ml.

ISO 639-1 codici linguistici

Stato di revisione

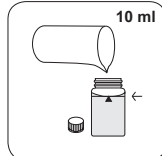
IT Manuale dei Metodi 01/20

Svolgimento della misurazione

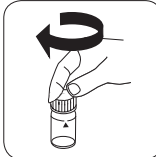
Esecuzione della rilevazione Capacità acida $K_{s4.3}$ con pastiglia

Selezionare il metodo nel dispositivo.

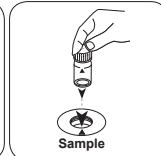
Con i seguenti dispositivi, per questo metodo non è necessario eseguire una misurazione ZERO: XD 7000, XD 7500



Riempire una cuvetta da 24 mm con **10 ml di campione**.

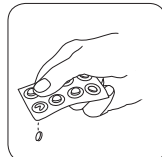


Chiudere la/e cuvetta/e.

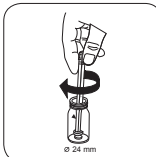


Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

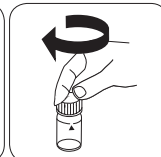
• • •



Aggiungere una **pastiglia ALKA-M-PHOTOMETER**.



Frantumare la/e pastiglia/e con una leggera rotazione.



Chiudere la/e cuvetta/e.



Alcalinità M T

M30

5 - 200 mg/L CaCO₃

tA

Acido/indicatore

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Alka-M-Photometer	Pastiglia / 100	513210BT
Alka-M-Photometer	Pastiglia / 250	513211BT

Note

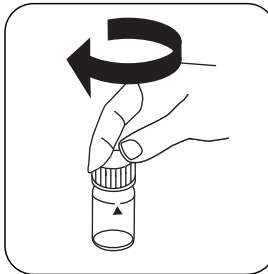
1. I termini alcalinità M, valore M, alcalinità totale e capacità acida $K_{S4,3}$ sono equivalenti.
2. Per l'accuratezza del risultato dell'analisi è fondamentale che il volume del campione misuri esattamente 10 ml.

Esecuzione della rilevazione Alcalinit , totale = alcalinit  M = valore M con pastiglia

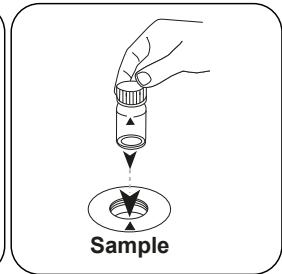
Selezionare il metodo nel dispositivo.



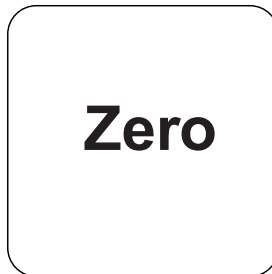
Riempire una cuvetta da
24 mm con **10 mL di
campione**.



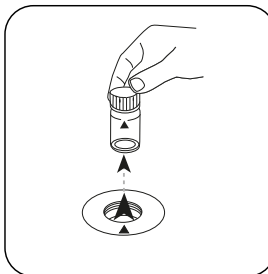
Chiudere la/e cuvetta/e.



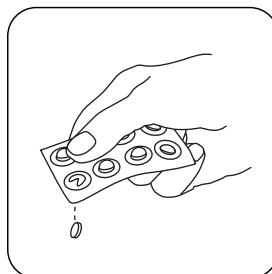
Posizionare la **cuvetta
del campione** nel vano di
misurazione. Fare attenzione
al posizionamento.



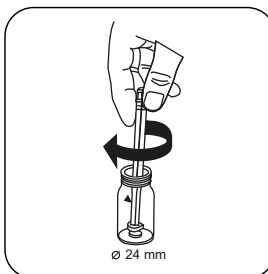
Premere il tasto **ZERO**.



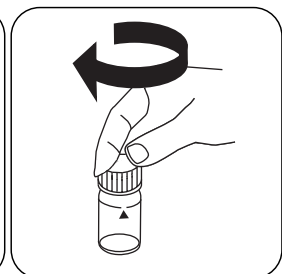
Prelevare la cuvetta dal
vano di misurazione.



Aggiungere **una pastiglia
ALKA-M-PHOTOMETER**.



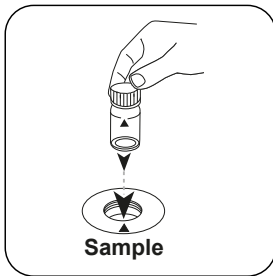
Frantumare la/e pastiglia/e
con una leggera rotazione.



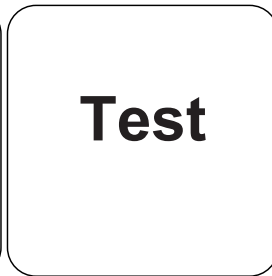
Chiudere la/e cuvetta/e.



Far sciogliere la/e
pastiglia/e agitando.



Posizionare la **cuvetta
del campione** nel
vano di misurazione.
Fare attenzione al
posizionamento.



Premere il tasto **TEST** (XD:
START).

Sul display compare il risultato come alcalinità-m.

Valutazione

La seguente tabella identifica i valori di output che possono essere convertiti in altre forme di citazione.

Unità di misura	Forma di citazione	Fattore di conversione
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

IT

Metodo chimico

Acido/indicatore

Appendice

Derivato di

EN ISO 9963-1

**Cloro T****M100****0.01 - 6.0 mg/L Cl₂ ^{a)}****CL6****DPD**

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
DPD No.1	Pastiglia / 100	511050BT
DPD No. 1	Pastiglia / 250	511051BT
DPD No. 1	Pastiglia / 500	511052BT
DPD No. 3	Pastiglia / 100	511080BT
DPD No. 3	Pastiglia / 250	511081BT
DPD No. 3	Pastiglia / 500	511082BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 100	515740BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 250	515741BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 500	515732BT
DPD No. 4	Pastiglia / 100	511220BT
DPD No. 4	Pastiglia / 250	511221BT
DPD No. 4	Pastiglia / 500	511222BT
DPD No. 3 Evo	Pastiglia / 100	511420BT
DPD No. 3 Evo	Pastiglia / 250	511421BT
DPD No. 3 Evo	Pastiglia / 500	511422BT
DPD No.4 Evo	Pastiglia / 100	511970BT
DPD No. 4 Evo	Pastiglia / 250	511971BT
DPD No. 4 Evo	Pastiglia / 500	511972BT

Standards disponibles

Titolo	Unità di imballaggio	N. ordine
ValidCheck Cloro 1,5 mg/l	1 pz.	48105510



Prelievo del campione

1. Nella preparazione del campione occorre evitare la degassificazione del cloro, ad es. utilizzando pipette e agitando.
2. L'analisi deve essere eseguita subito dopo il prelievo del campione.

Preparazione

1. Pulizia delle cuvette:
Poiché molti detergenti ad uso domestico (ad es. detersivo per piatti) contengono sostanze riducenti, nella rilevazione del cloro si potrebbero ottenere risultati troppo bassi. Per escludere tali errori di misura è necessario che i dispositivi in vetro siano esenti dal consumo di cloro. I dispositivi in vetro inoltre vengono conservati in una soluzione di ipoclorito di sodio (0,1 g/L) per un'ora e successivamente vengono risciacquati abbondantemente con acqua demineralizzata.
2. Per la singola rilevazione del cloro libero e del cloro totale è opportuno utilizzare un apposito kit di cuvette per ciascuna procedura (vedere EN ISO 7393-2, par. 5.3).
3. Lo sviluppo della colorazione del DPD avviene con un valore di pH compreso tra 6,2 e 6,5. I reagenti contengono pertanto un tampone per la regolazione del valore di pH. Le acque fortemente alcaline o acide tuttavia devono essere portate prima dell'analisi entro un range di pH compreso tra 6 e 7 (con 0,5 mol/L di acido solforico o 1 mol/L di liscivia).

Note

1. Le compresse Evo possono essere utilizzate come alternativa alla corrispondente compressa standard (ad esempio DPD No. 3 Evo invece di DPD No. 3).



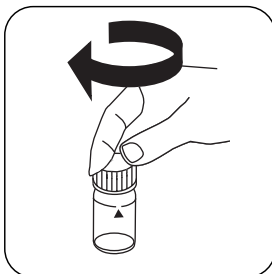
Esecuzione della rilevazione Cloro, libero con compressa

Selezionare il metodo nel dispositivo.

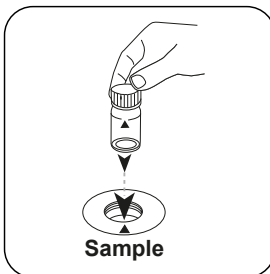
IT



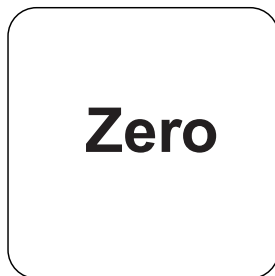
Riempire una cuvetta da 24 mm con **10 mL di campione**.



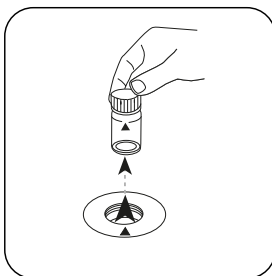
Chiudere la/e cuvetta/e.



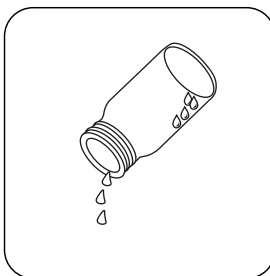
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



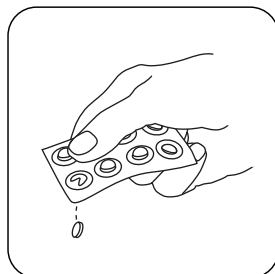
Premere il tasto **ZERO**.



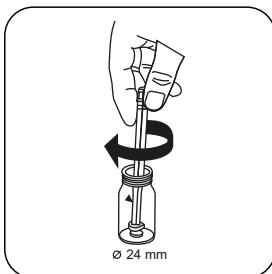
Prelevare la cuvetta dal vano di misurazione.



Svuotare la cuvetta finché non rimangono alcune gocce.



Aggiungere **una pastiglia DPD No. 1**.



Frantumare la/e pastiglia/e con una leggera rotazione.



Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.



Chiudere la/e cuvetta/e.



Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

IT

Test

Premere il tasto **TEST** (XD: **START**).

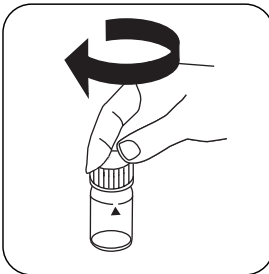
Sul display compare il risultato in mg/L di Cloro libero.

Esecuzione della rilevazione Cloro, totale con compressa

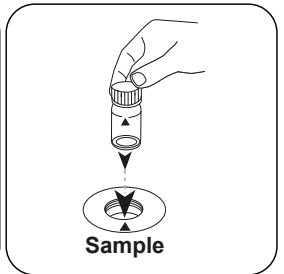
Selezionare il metodo nel dispositivo.



Riempire una cuvetta da 24 mm con **10 mL di campione**.



Chiudere la/e cuvetta/e.

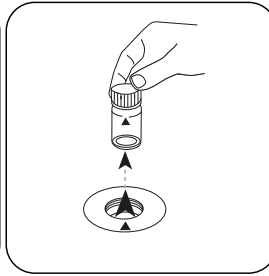


Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

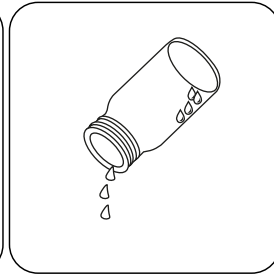


Zero

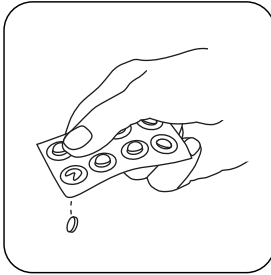
Premere il tasto **ZERO**.



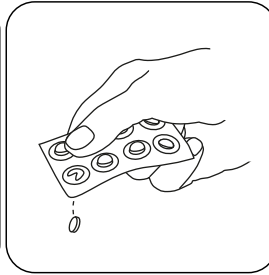
Prelevare la cuvetta dal vano di misurazione.



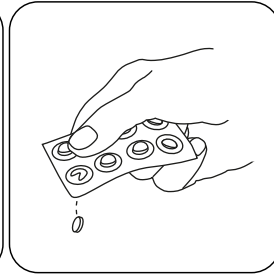
Svuotare la cuvetta finché non rimangono alcune gocce.



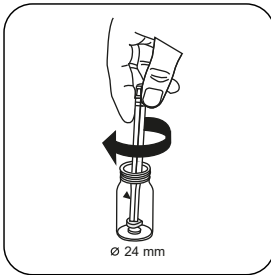
Aggiungere **una pastiglia DPD No. 1**.



Aggiungere **una pastiglia DPD No. 3**.



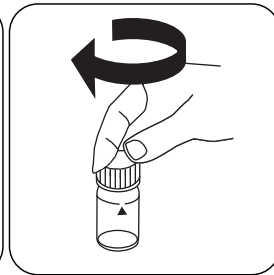
In alternativa al DPD No. 1 e No. 3 tablet, un DPD No. 4 tablet può essere aggiunto.



Frantumare la/e pastiglia/e con una leggera rotazione.



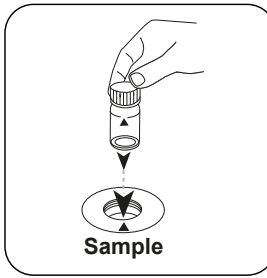
Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.



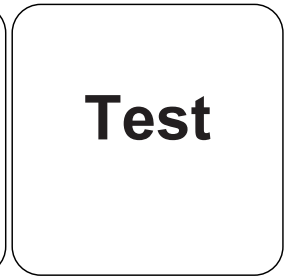
Chiudere la/e cuvetta/e.



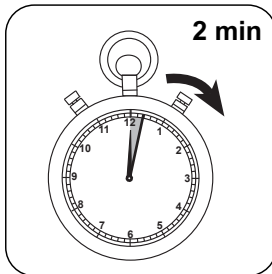
Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).



Attendere un **tempo di reazione di 2 minuti/i** .

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione. Sul display compare il risultato in mg/L di Cloro totale.



Metodo chimico

DPD

Appendice

IT

Interferenze

Interferenze permanenti

- Tutti gli ossidanti presenti nei campioni reagiscono come il cloro dando risultati troppo elevati.

Interferenze escludibili

- Le interferenze da parte di rame e ferro(III) devono essere eliminate con EDTA.
- In caso di campioni con un elevato tenore di calcio* e/o un'elevata conducibilità*, utilizzando le pastiglie di reagenti potrebbe verificarsi un intorbidimento del campione con conseguenti errori di misurazione. In questo caso si possono utilizzare in alternativa la pastiglia di reagente DPD No. 1 High Calcium e la pastiglia di reagente DPD No. 3 High Calcium.
*Non è possibile indicare i valori esatti in quanto l'intorbidimento dipende dal tipo e dalla composizione dell'acqua campione.
- Se si utilizzano pastiglie, le concentrazioni di cloro maggiori di 10 mg/L possono dare risultati entro il range di misura fino a 0 mg/L. Se la concentrazione di cloro è troppo elevata, il campione deve essere diluito con acqua priva di cloro. 10 mL del campione diluito vengono addizionati con il reagente e la misurazione viene ripetuta (test di plausibilità).

Interferenze	da / [mg/L]
CrO ₄ ²⁻	0.01
MnO ₂	0.01

Validazione metodo

Limite di rilevabilità	0.02 mg/L
Limite di quantificazione	0.06 mg/L
Estremità campo di misura	6 mg/L
Sensibilità	2.05 mg/L / Abs
Intervallo di confidenza	0.04 mg/L
Deviazione standard della procedura	0.019 mg/L
Coefficiente di variazione della procedura	0.87 %



Conforme

EN ISO 7393-2

*Determinazione di libero, vincolato, totale possibile | *Reagente ausiliario, in alternativa a DPD n. 1 / no 3 in caso di torbidità del campione a causa di alto contenuto di ioni di calcio e / o alta conduttività

IT

**Cloro L****M101****0.02 - 4.0 mg/L Cl₂^{a)}****CL6****DPD**

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
DPD 1 soluzione tampone, bottiglia blu	15 mL	471010
Soluzione tampone DPD 1	100 mL	471011
DPD 1 Soluzione tampone in confezione da 6	1 pz.	471016
DPD 1 soluzione reagente, bottiglia verde	15 mL	471020
Soluzione reagente DPD 1	100 mL	471021
DPD 1 Soluzione reagente in confezione da 6	1 pz.	471026
DPD 3 soluzione, bottiglia rossa	15 mL	471030
Soluzione DPD 3	100 mL	471031
DPD 3 Soluzione in confezione da 6	1 pz.	471036
Set di reagenti DPD	1 pz.	471056

Standards disponibles

Titolo	Unità di imballaggio	N. ordine
ValidCheck Cloro 1,5 mg/l	1 pz.	48105510

Prelievo del campione

1. Nella preparazione del campione occorre evitare la degassificazione del cloro, ad es. utilizzando pipette e agitando.
2. L'analisi deve essere eseguita subito dopo il prelievo del campione.



Preparazione

1. Pulizia delle cuvette:
Poiché molti detersivi ad uso domestico (ad es. detersivo per piatti) contengono sostanze riducenti, nella rilevazione del cloro si potrebbero ottenere risultati troppo bassi. Per escludere tali errori di misura è necessario che i dispositivi in vetro siano esenti dal consumo di cloro. I dispositivi in vetro inoltre vengono conservati in una soluzione di ipoclorito di sodio (0,1 g/L) per un'ora e successivamente vengono risciacquati abbondantemente con acqua demineralizzata.
2. Per la singola rilevazione del cloro libero e del cloro totale è opportuno utilizzare un apposito kit di cuvette per ciascuna procedura (vedere EN ISO 7393-2, par. 5.3).
3. Lo sviluppo della colorazione del DPD avviene con un valore di pH compreso tra 6,2 e 6,5. I reagenti contengono pertanto un tampone per la regolazione del valore di pH. Le acque fortemente alcaline o acide tuttavia devono essere portate prima dell'analisi entro un range di pH compreso tra 6 e 7 (con 0,5 mol/l di acido solforico o 1 mol/l di liscivia).

Note

1. Dopo l'uso bisogna richiudere immediatamente le bottiglie contagocce con i rispettivi tappi dello stesso colore.
2. Conservare al fresco il kit di reagenti a una temperatura compresa tra +6 °C e +10 °C.

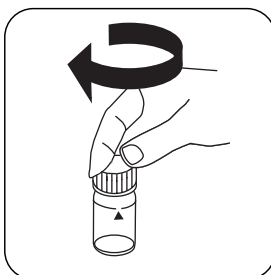


Esecuzione della rilevazione Cloro, libero con reagente liquido

Selezionare il metodo nel dispositivo.



Riempire una cuvetta da 24 mm con **10 mL di campione**.



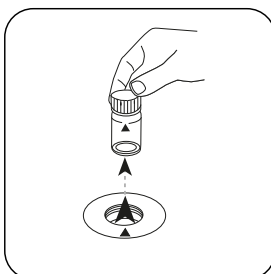
Chiudere la/e cuvetta/e.



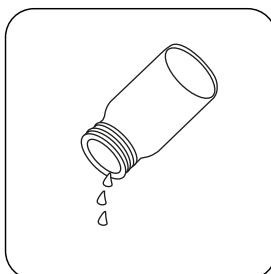
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



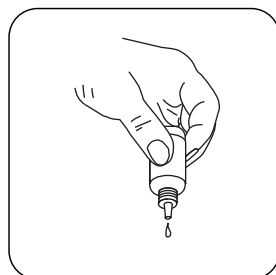
Premere il tasto **ZERO**.



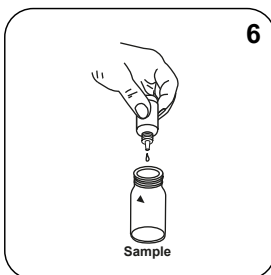
Prelevare la cuvetta dal vano di misurazione.



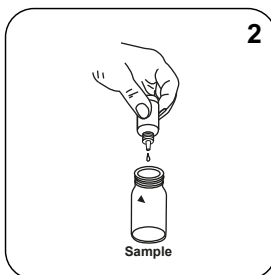
Svuotare la cuvetta.



Tenere le botticelle contagocce in posizione verticale e introdurre, premendo lentamente, gocce della stessa dimensione nella cuvetta.



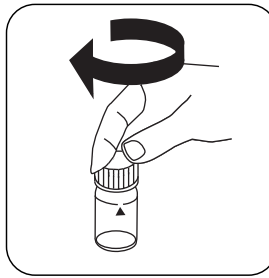
Introdurre **6 gocce di DPD 1 Buffer Solution** nella cuvetta del campione.



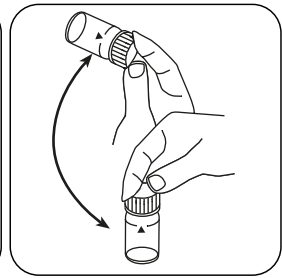
Introdurre **2 gocce di DPD 1 Reagent Solution** nella cuvetta del campione.



Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.

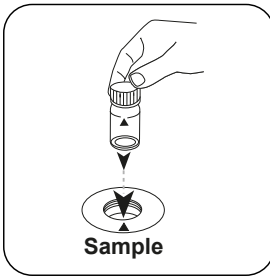


Chiudere la/e cuvetta/e.

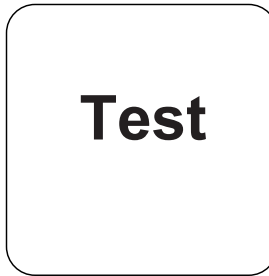


Miscelare il contenuto capovolgendo.

IT



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).

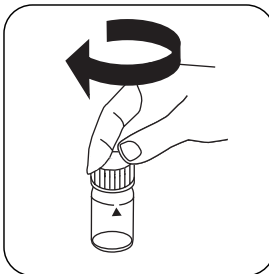
Sul display compare il risultato in mg/L di Cloro libero.

Esecuzione della rilevazione Cloro, totale con reagente liquido

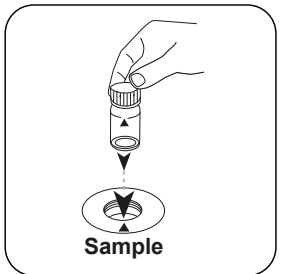
Selezionare il metodo nel dispositivo.



Riempire una cuvetta da 24 mm con **10 mL di campione**.



Chiudere la/e cuvetta/e.

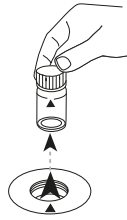


Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Zero

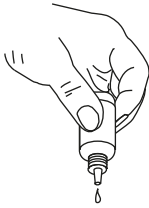
Premere il tasto **ZERO**.



Prelevare la cuvetta dal vano di misurazione.



Svuotare la cuvetta.



Tenere le boccette contagocce in posizione verticale e introdurre, premendo lentamente, gocce della stessa dimensione nella cuvetta.



6

Introdurre **6 gocce di DPD 1 Buffer Solution** nella cuvetta del campione.



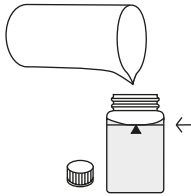
2

Introdurre **2 gocce di DPD 1 Reagent Solution** nella cuvetta del campione.

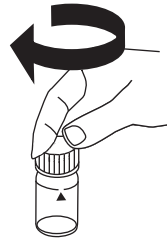


3

Introdurre **3 gocce di DPD 3 Solution** nella cuvetta del campione.



Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.



Chiudere la/e cuvetta/e.



Miscelare il contenuto capovolgendo.

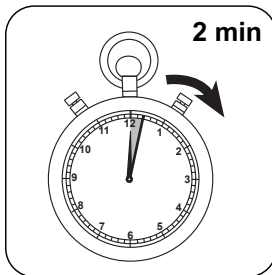


Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).

IT



Attendere un **tempo di reazione di 2 minuti/i** .

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione. Sul display compare il risultato in mg/L di Cloro totale.



Metodo chimico

DPD

Appendice

IT

Interferenze

Interferenze permanenti

- Tutti gli ossidanti presenti nei campioni reagiscono come il cloro dando risultati troppo elevati.

Interferenze escludibili

- Le interferenze da parte di rame e ferro(III) devono essere eliminate con EDTA.
- Se si utilizzano reagenti liquidi, le concentrazioni di cloro maggiori di 4 mg/L possono dare risultati entro il range di misura fino a 0 mg/L. In questo caso il campione deve essere diluito con acqua priva di cloro. 10 ml del campione diluito vengono addizionati con il reagente e la misurazione viene ripetuta (test di plausibilità).

Interferenze	da / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Conforme

EN ISO 7393-2

^{a)}Determinazione di libero, vincolato, totale possibile

**Cloro HR T****M103****0.1 - 10 mg/L Cl₂^{a)}****CL10****DPD**

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
DPD No. 1 HR	Pastiglia / 100	511500BT
DPD No. 1 HR	Pastiglia / 250	511501BT
DPD No. 1 HR	Pastiglia / 500	511502BT
DPD No. 3 HR	Pastiglia / 100	511590BT
DPD No. 3 HR	Pastiglia / 250	511591BT
DPD No. 3 HR	Pastiglia / 500	511592BT
Set DPD No. 1 HR/No. 3 HR #	ciascuna 100	517791BT
Set DPD No. 1 HR/No. 3 HR #	ciascuna 250	517792BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 100	515740BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 250	515741BT
DPD No. 1 Alto Calcio ^{e)}	Pastiglia / 500	515742BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 100	515730BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 250	515731BT
DPD No. 3 High Calcium ^{e)}	Pastiglia / 500	515732BT
DPD No.3 HR Evo	Pastiglia / 100	511920BT
DPD No. 3 HR Evo	Pastiglia / 250	511921BT
DPD No. 3 HR Evo	Pastiglia / 500	511922BT

Prelievo del campione

1. Nella preparazione del campione occorre evitare la degassificazione del cloro, ad es. utilizzando pipette e agitando.
2. L'analisi deve essere eseguita subito dopo il prelievo del campione.



Preparazione

1. Pulizia delle cuvette:
Poiché molti detersivi ad uso domestico (ad es. detersivo per piatti) contengono sostanze riducenti, nella rilevazione del cloro si potrebbero ottenere risultati troppo bassi. Per escludere tali errori di misura è necessario che i dispositivi in vetro siano esenti dal consumo di cloro. I dispositivi in vetro inoltre vengono conservati in una soluzione di ipoclorito di sodio (0,1 g/L) per un'ora e successivamente vengono risciacquati abbondantemente con acqua demineralizzata.
2. Per la singola rilevazione del cloro libero e del cloro totale è opportuno utilizzare un apposito kit di cuvette per ciascuna procedura (vedere EN ISO 7393-2, par. 5.3).
3. Lo sviluppo della colorazione del DPD avviene con un valore di pH compreso tra 6,2 e 6,5. I reagenti contengono pertanto un tampone per la regolazione del valore di pH. Le acque fortemente alcaline o acide tuttavia devono essere portate prima dell'analisi entro un range di pH compreso tra 6 e 7 (con 0,5 mol/L di acido solforico o 1 mol/L di liscivia).

Note

1. Le compresse Evo possono essere utilizzate come alternativa alla corrispondente compressa standard (ad esempio DPD No. 3 Evo invece di DPD No. 3).



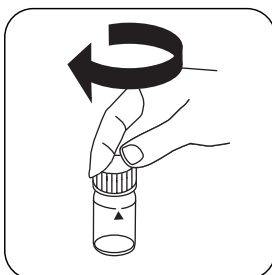
Esecuzione della rilevazione Cloro HR, libero con compressa

Selezionare il metodo nel dispositivo.

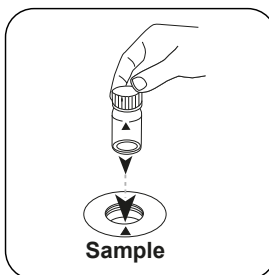
IT



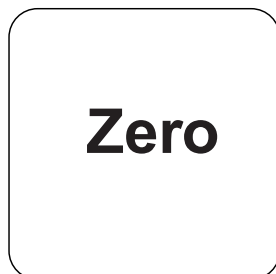
Riempire una cuvetta da 24 mm con **10 mL di campione**.



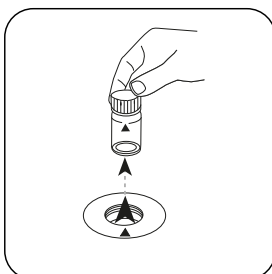
Chiudere la/e cuvetta/e.



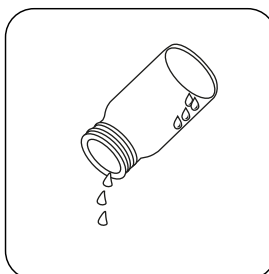
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



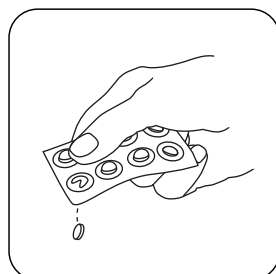
Premere il tasto **ZERO**.



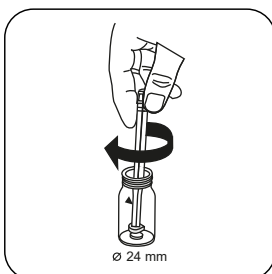
Prelevare la cuvetta dal vano di misurazione.



Svuotare la cuvetta finché non rimangono alcune gocce.



Aggiungere **una pastiglia DPD No. 1 HR**.



Frantumare la/e pastiglia/e con una leggera rotazione.



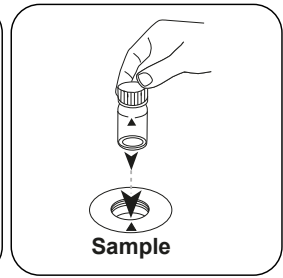
Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.



Chiudere la/e cuvetta/e.



Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

IT

Test

Premere il tasto **TEST** (XD: **START**).

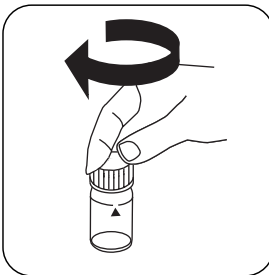
Sul display compare il risultato in mg/L di Cloro libero.

Esecuzione della rilevazione Cloro HR, totale con compressa

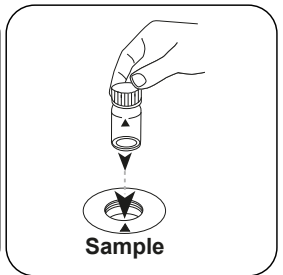
Selezionare il metodo nel dispositivo.



Riempire una cuvetta da 24 mm con **10 mL di campione**.



Chiudere la/e cuvetta/e.

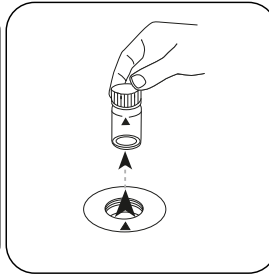


Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

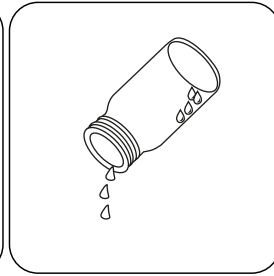


Zero

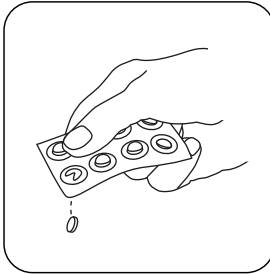
Premere il tasto **ZERO**.



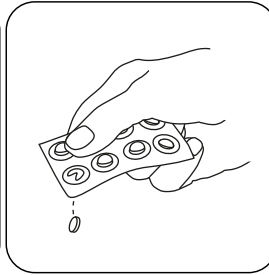
Prelevare la cuvetta dal vano di misurazione.



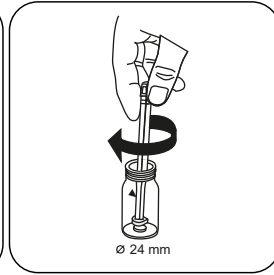
Svuotare la cuvetta finché non rimangono alcune gocce.



Aggiungere **una pastiglia DPD No. 1 HR**.



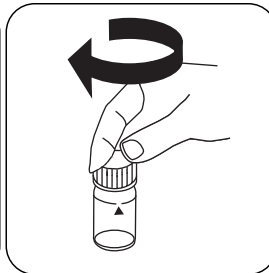
Aggiungere **una pastiglia DPD No. 3 HR**.



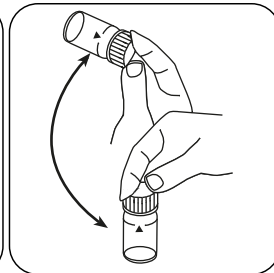
Frantumare la/e pastiglia/e con una leggera rotazione.



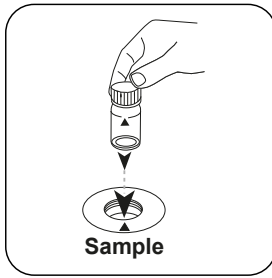
Immettere il **campione** nella cuvetta fino a raggiungere la **tacca dei 10 mL**.



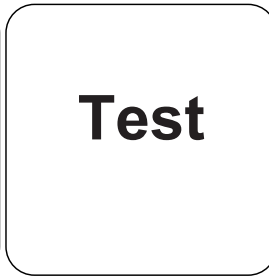
Chiudere la/e cuvetta/e.



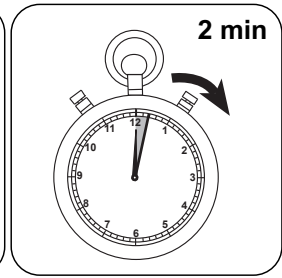
Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).



Attendere un **tempo di reazione di 2 minuti**.

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione. Sul display compare il risultato in mg/L di Cloro totale.



Metodo chimico

DPD

Appendice

IT

Interferenze

Interferenze permanenti

- Tutti gli ossidanti presenti nei campioni reagiscono come il cloro dando risultati troppo elevati.

Interferenze escludibili

- Le interferenze da parte di rame e ferro(III) devono essere eliminate con EDTA.
- In caso di campioni con un elevato tenore di calcio* e/o un'elevata conducibilità*, utilizzando le pastiglie di reagente potrebbe verificarsi un intorbidimento del campione con conseguenti errori di misurazione. In questo caso si possono utilizzare in alternativa la pastiglia di reagente DPD No. 1 High Calcium e la pastiglia di reagente DPD No. 3 High Calcium.

*Non è possibile indicare i valori esatti in quanto l'intorbidimento dipende dal tipo e dalla composizione dell'acqua campione.

Conforme

EN ISO 7393-2

^{*)}Determinazione di libero, vincolato, totale possibile | ^{**)}Reagente ausiliario, in alternativa a DPD n. 1 / no 3 in caso di torbidità del campione a causa di alto contenuto di ioni di calcio e / o alta conduttività | ^{***)}Bacchetta compresa



Rame T

M150

0.05 - 5 mg/L Cu^{a)}

Cu

Bichinolina

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Rame No. 1	Pastiglia / 100	513550BT
Rame No. 1	Pastiglia / 250	513551BT
Rame No. 2	Pastiglia / 100	513560BT
Rame No. 2	Pastiglia / 250	513561BT
Set Rame No. 1/no. 2 ^a	ciascuna 100	517691BT
Set Rame No. 1/no. 2 ^a	ciascuna 250	517692BT
ValidCheck Rame 2 mg/l	1 pz.	48141525

Preparazione

1. Le acque fortemente alcaline o acide dovrebbero essere regolate prima dell'analisi su un valore di pH da 4 a 6.

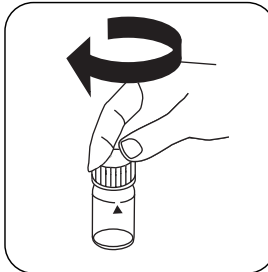
Esecuzione della rilevazione Rame, libero con pastiglia

Selezionare il metodo nel dispositivo.

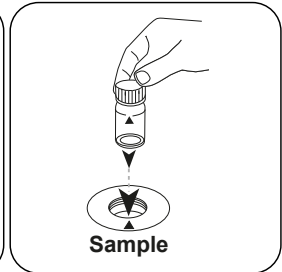
Selezionare inoltre la determinazione: libero



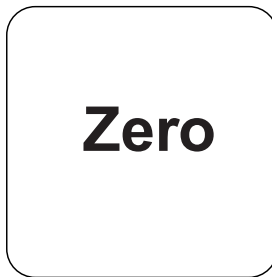
Riempire una cuvetta da 24 mm con **10 mL di campione**.



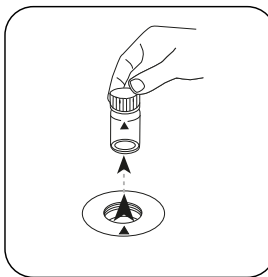
Chiudere la/e cuvetta/e.



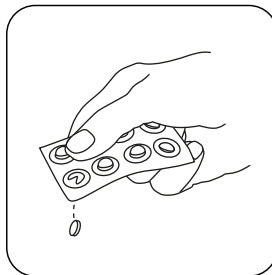
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



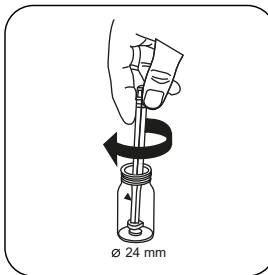
Premere il tasto **ZERO**.



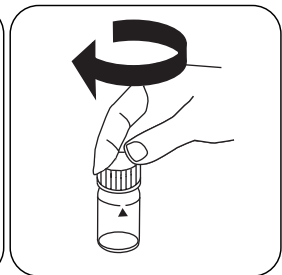
Prelevare la cuvetta dal vano di misurazione.



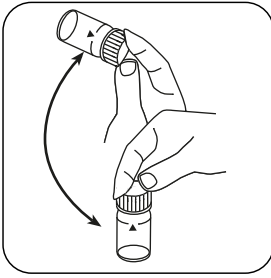
Aggiungere **una pastiglia COPPER No. 1**.



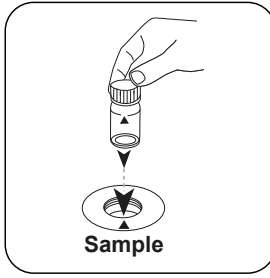
Frantumare la/e pastiglia/e con una leggera rotazione.



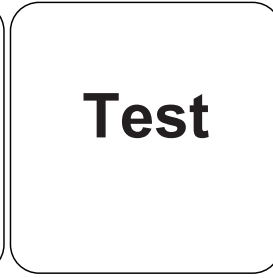
Chiudere la/e cuvetta/e.



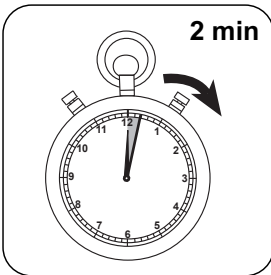
Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).



Attendere un **tempo di reazione di 2 minuto/i**.

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione.

Sul display compare il risultato in mg/L di Rame libero.

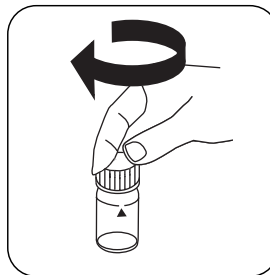
Esecuzione della rilevazione Rame, totale con pastiglia

Selezionare il metodo nel dispositivo.

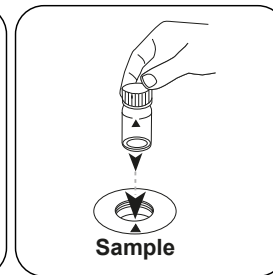
Selezionare inoltre la determinazione: totale



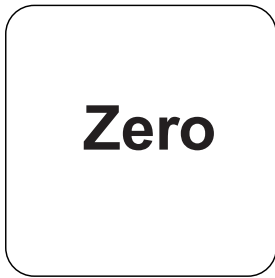
Riempire una cuvetta da 24 mm con **10 mL di campione**.



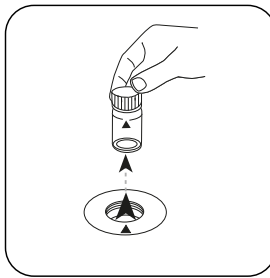
Chiudere la/e cuvetta/e.



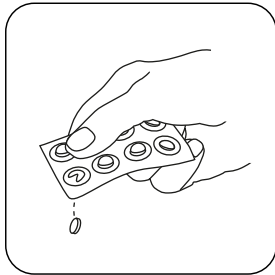
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



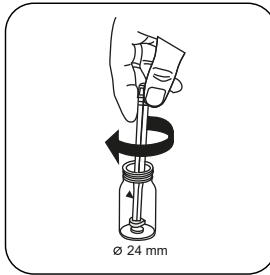
Premere il tasto **ZERO**.



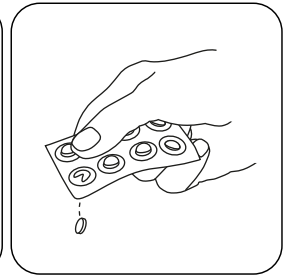
Prelevare la cuvetta dal vano di misurazione.



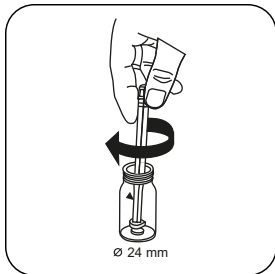
Aggiungere **una pastiglia COPPER No. 1**.



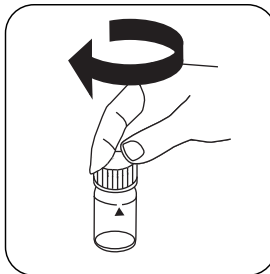
Frantumare e far sciogliere la/e pastiglia/e con una leggera rotazione.



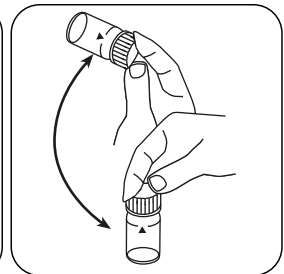
Aggiungere **una pastiglia COPPER No. 2**.



Frantumare la/e pastiglia/e con una leggera rotazione.

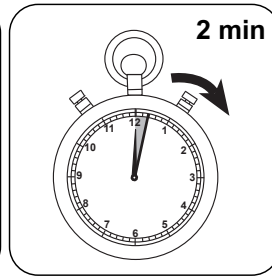
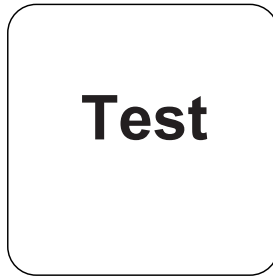
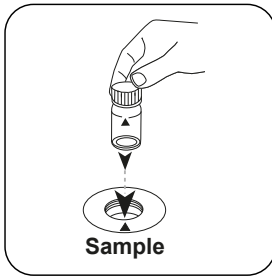


Chiedere la/e cuvetta/e.



Far sciogliere la/e pastiglia/e agitando.

IT



IT

Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

Premere il tasto **TEST** (XD: **START**).

Attendere un **tempo di reazione di 2 minuto/i**.

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione. Sul display compare il risultato in mg/L di Rame totale.

Metodo chimico

Bichinolina

Appendice

Interferenze

Interferenze permanenti

1. Cianuro CN⁻ e Argento Ag⁺ interferiscono con la rilevazione.

Validazione metodo

Limite di rilevabilità	0.05 mg/L
Limite di quantificazione	0.15 mg/L
Estremità campo di misura	5 mg/L
Sensibilità	3.8 mg/L / Abs
Intervallo di confidenza	0.026 mg/L
Deviazione standard della procedura	0.011 mg/L
Coefficiente di variazione della procedura	0.42 %

Riferimenti bibliografici

Photometrische Analyse, Lange/Vedjelek, Verlag Chemie 1980

^aDeterminazione di libero, vincolato, totale possibile | ^bBacchetta compresa



CYA T

M160

10 - 160 mg/L CyA

CyA

Melammina

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Test CyA	Pastiglia / 100	511370BT
Test CyA	Pastiglia / 250	511371BT
Acqua demineralizzata	100 mL	461275
Acqua demineralizzata	250 mL	457022

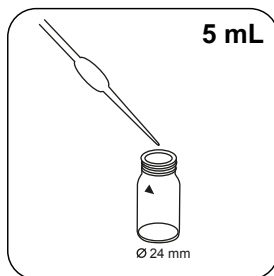
Note

1. L'acido cianurico provoca un intorbidimento distribuito molto finemente dall'aspetto lattiginoso. Singole particelle non sono imputabili alla presenza di acido cianurico.

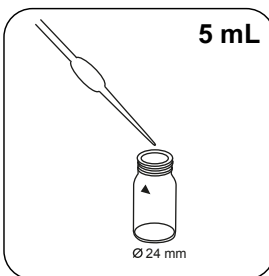


Esecuzione della rilevazione Test acido cianurico con pastiglia

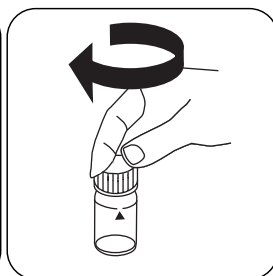
Selezionare il metodo nel dispositivo.



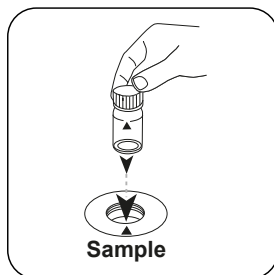
Riempire una cuvetta da 24 mm con **5 mL di acqua demineralizzata**.



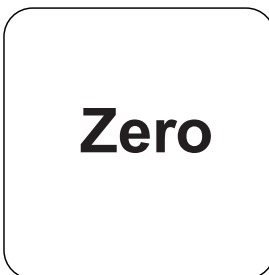
Immettere **5 mL di campione** nella cuvetta.



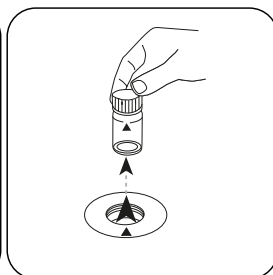
Chiudere la/e cuvetta/e.



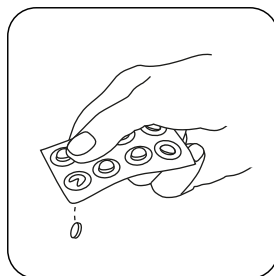
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



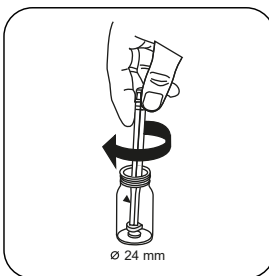
Premere il tasto **ZERO**.



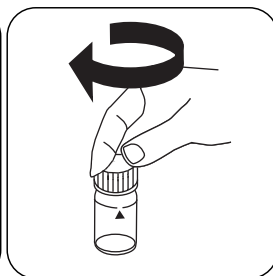
Prelevare la cuvetta dal vano di misurazione.



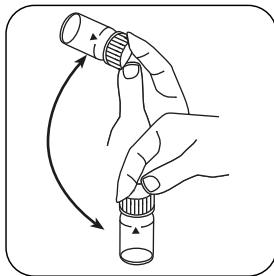
Aggiungere **una pastiglia CyA-Test**.



Frantumare la/e pastiglia/e con una leggera rotazione.

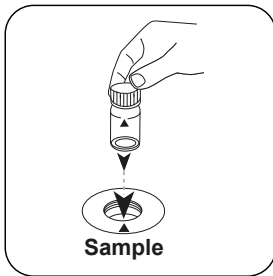


Chiudere la/e cuvetta/e.

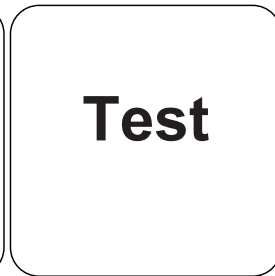


Miscelare il contenuto capovolgendo (per almeno 60 s fino al completo scioglimento della pastiglia).

Sul display compare il risultato in mg/L di acido cianurico .



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).



Metodo chimico

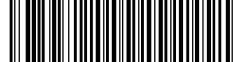
Melamina

Interferenze

Interferenze permanenti

1. Le particelle non disciolte possono portare a risultati troppo elevati. Pertanto è importante sciogliere completamente le pastiglie.

IT



Ferro T

M220

0.02 - 1 mg/L Fe

FE

Ferrozine / acido tioglicolico

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Ferro II LR (Fe ²⁺)	Pastiglia / 100	515420BT
Ferro II LR (Fe ²⁺)	Pastiglia / 250	515421BT
Ferro LR (Fe ²⁺ und Fe ³⁺)	Pastiglia / 100	515370BT
Ferro LR (Fe ²⁺ und Fe ³⁺)	Pastiglia / 250	515371BT

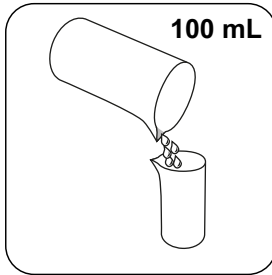
Preparazione

1. Le acque che sono state trattate con composti organici che proteggono dalla corrosione devono essere eventualmente ossidate per disgregare i complessi di ferro. A tale scopo si aggiunge un campione da 100 ml con 1 ml di acido solforico concentrato e 1 ml di acido nitrico concentrato e lo si fa evaporare fino alla metà. Dopo il raffreddamento viene eseguita la digestione.

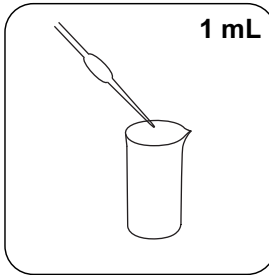
Note

1. Con questo metodo viene rilevato il totale del Fe²⁺ e del Fe³⁺ disciolti.
2. Per rilevare il Fe²⁺ si utilizza, invece della pastiglia IRON LR, la pastiglia IRON (II) LR.

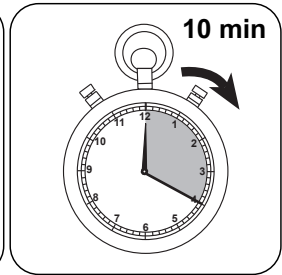
Digestione



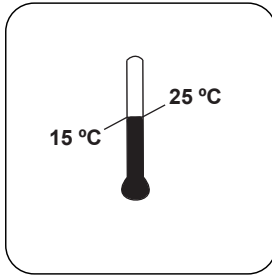
Riempire un recipiente per campioni adeguato con **100 mL di campione**.



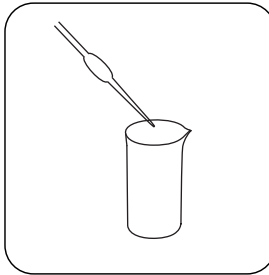
Aggiungere **1 mL di acido solforico concentrato**.



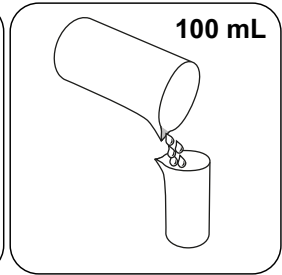
Riscaldare il campione per 10 minuti o finché non si sarà sciolto completamente.



Lasciar raffreddare il campione a **temperatura ambiente**.



Regolare il **valore di pH** del campione con **soluzione di ammoniaca su 3-5**.



Aggiungere al campione **acqua demineralizzata fino a raggiungere i 100 mL**.

Utilizzare questo campione per l'analisi di Ferro soluto e disciolto totale.

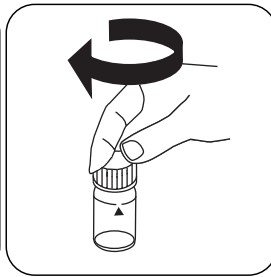
Esecuzione della rilevazione Ferro(II,III), disciolto con pastiglia

Selezionare il metodo nel dispositivo.

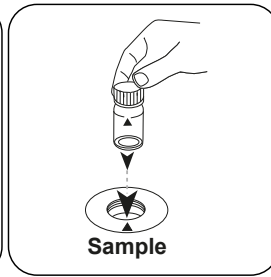
Per la determinazione di **Ferro disciolto e non disciolto** eseguire la **digestione** descritta.



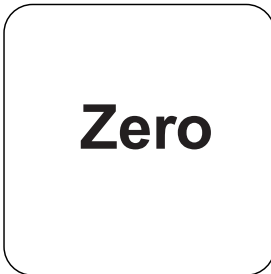
Riempire una cuvetta da 24 mm con **10 mL di campione**.



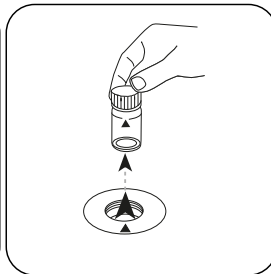
Chiudere la/e cuvetta/e.



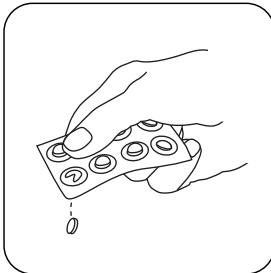
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



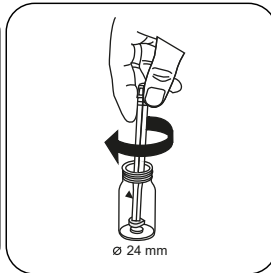
Premere il tasto **ZERO**.



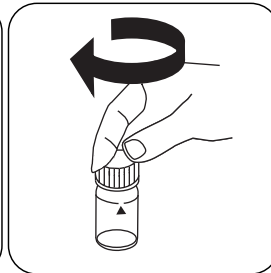
Prelevare la cuvetta dal vano di misurazione.



Aggiungere **una pastiglia IRON LR**.



Frantumare la/e pastiglia/e con una leggera rotazione.



Chiudere la/e cuvetta/e.



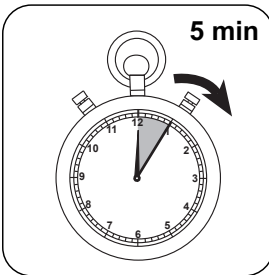
Far sciogliere la/e pastiglia/e agitando.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.

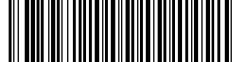


Premere il tasto **TEST** (XD: **START**).



Attendere un **tempo di reazione di 5 minuti/i** .

Allo scadere del tempo di reazione viene effettuata automaticamente la misurazione. Sul display compare il risultato in mg/L di Ferro.



Metodo chimico

Ferrozine / acido tioglicolico

Appendice

IT

Interferenze

Interferenze escludibili

1. La presenza di rame aumenta il risultato della misurazione del 10 %. Con una concentrazione di 10 mg/L di rame nel campione il risultato della misurazione viene aumentato di 1 mg/L di ferro.
L'interferenza può essere eliminata con l'aggiunta di tiourea.

Validazione metodo

Limite di rilevabilità	0.01 mg/L
Limite di quantificazione	0.016 mg/L
Estremità campo di misura	1 mg/L
Sensibilità	0.92 mg/L / Abs
Intervallo di confidenza	0.013 mg/L
Deviazione standard della procedura	0.005 mg/L
Coefficiente di variazione della procedura	1.23 %

Riferimenti bibliografici

Photometrische Analyse, Lange/Vjedelek, Verlag Chemie 1980, pag. 102



Valore pH T

M330

6.5 - 8.4 pH

PH

Rosso fenolo

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Fotometro rosso fenolo	Pastiglia / 100	511770BT
Fotometro rosso fenolo	Pastiglia / 250	511771BT
Fotometro rosso fenolo	Pastiglia / 500	511772BT

Note

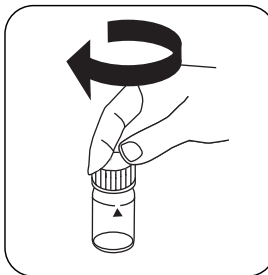
1. Per la rilevazione fotometrica del valore di pH si devono utilizzare soltanto pastiglie PHENOL RED con etichetta nera contrassegnate con il termine PHOTOMETER.

Esecuzione della rilevazione Valore pH con pastiglia

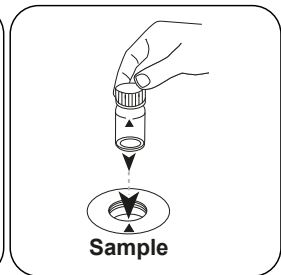
Selezionare il metodo nel dispositivo.



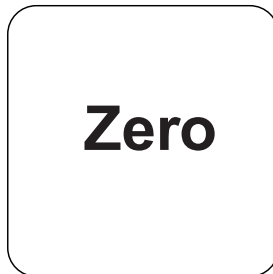
Riempire una cuvetta da 24 mm con **10 mL di campione**.



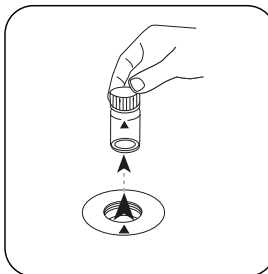
Chiudere la/e cuvetta/e.



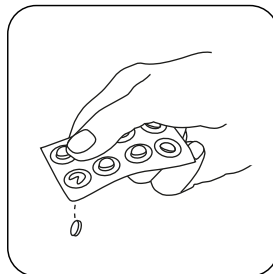
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



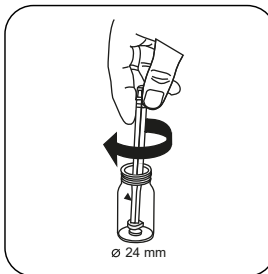
Premere il tasto **ZERO**.



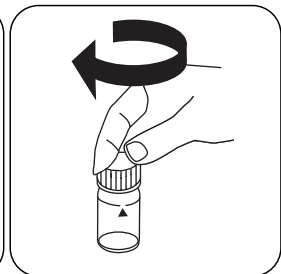
Prelevare la cuvetta dal vano di misurazione.



Aggiungere una pastiglia **PHENOL RED PHOTOMETER**.



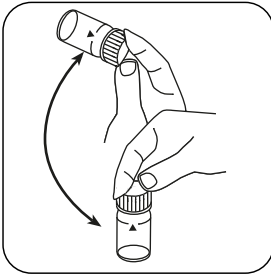
Frantumare la/e pastiglia/e con una leggera rotazione.



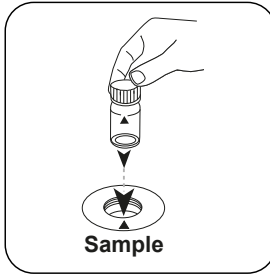
Chiudere la/e cuvetta/e.



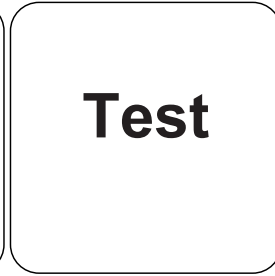
IT



Far sciogliere la/e
pastiglia/e agitando.



Posizionare la **cuvetta
del campione** nel
vano di misurazione.
Fare attenzione al
posizionamento.



Premere il tasto **TEST** (XD:
START).

Sul display compare il risultato come valore pH.

Metodo chimico

Rosso fenolo

Appendice

Interferenze

Interferenze permanenti

1. I campioni di acqua con una bassa durezza carbonatica* possono far ottenere valori di pH errati.

* $K_{S4,3} < 0,7 \text{ mmol/l} \triangleq \text{alcalinità totale} < 35 \text{ mg/L CaCO}_3$.

Interferenze escludibili

1. I valori di pH minori di 6,5 e maggiori di 8,4 possono dare risultati entro il range di misura. Si consiglia un test di plausibilità (misuratore di pH).

2. Errore salino:

Con una salinità fino a 2 g/L non è previsto alcun errore salino significativo dovuto alla salinità della pastiglia di reagente. Con salinità maggiori è necessario correggere i valori di misura nel modo seguente:

Salinità del campione in g/L	30 (acqua di mare)	60	120	180
Correzione	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾ secondo Kolthoff (1922)

²⁾ secondo Parson e Douglas (1926)

Riferimenti bibliografici

Colorimetric Chemical Analytical Methods, 9th Edition, London



Valore pH L

M331

6.5 - 8.4 pH

PH

Rosso fenolo

IT

Materiale

Materiale richiesto (in parte facoltativo):

Reagenti	Unità di imballaggio	N. ordine
Soluzione di rosso fenolo	15 mL	471040
Soluzione di rosso fenolo	100 mL	471041
Soluzione di rosso fenolo in confezione da 6	1 pz.	471046

Preparazione

1. Per via della dimensione variabile delle gocce, il risultato della misurazione può presentare divergenze maggiori di quanto avvenga con l'uso delle pastiglie. Utilizzando una pipetta (0,18 ml corrispondono a 6 gocce) si può ridurre al minimo questa divergenza.

Note

1. Dopo l'uso bisogna richiudere immediatamente la boccetta contagocce con il relativo tappo dello stesso colore.
2. Conservare al fresco il reagente a una temperatura compresa tra +6 °C e +10 °C.

Esecuzione della rilevazione Valore pH con reagente liquido

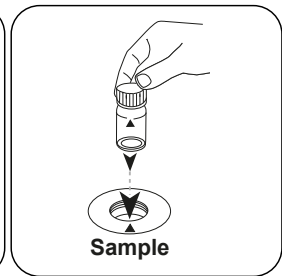
Selezionare il metodo nel dispositivo.



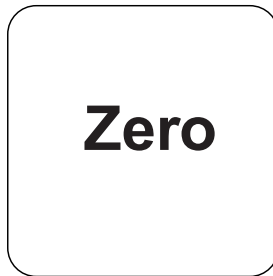
Riempire una cuvetta da 24 mm con **10 mL di campione**.



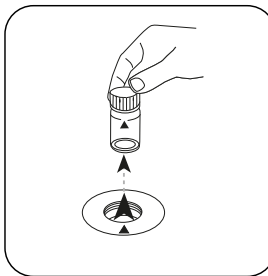
Chiudere la/e cuvetta/e.



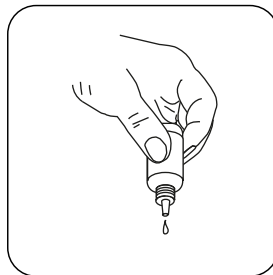
Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



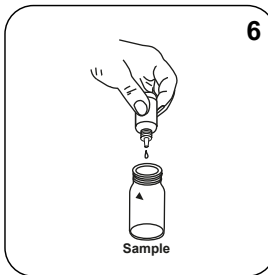
Premere il tasto **ZERO**.



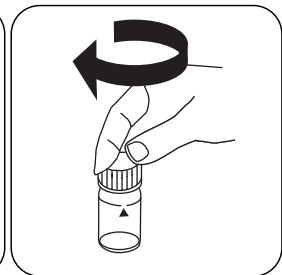
Prelevare la cuvetta dal vano di misurazione.



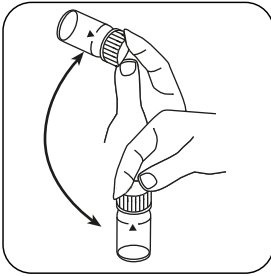
Tenere le boccette contagocce in posizione verticale e introdurre, premendo lentamente, gocce della stessa dimensione nella cuvetta.



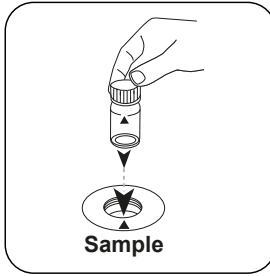
Introdurre **6 gocce di PHENOL Red-Lösung** nella cuvetta del campione.



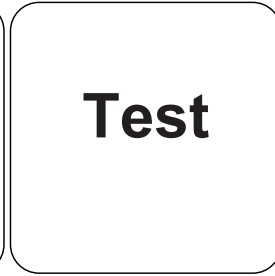
Chiudere la/e cuvetta/e.



Miscelare il contenuto capovolgendo.



Posizionare la **cuvetta del campione** nel vano di misurazione. Fare attenzione al posizionamento.



Premere il tasto **TEST** (XD: **START**).

Sul display compare il risultato come valore pH.

Metodo chimico

Rosso fenolo

Appendice

Interferenze

IT

Interferenze escludibili

1. Errore salino: Correzione del valore di misura (valori medi) per i campioni con una salinità di:


2.	Salinità del campione	Correzione
	30 g/L (acqua di mare)	-0,15 ¹⁾
	60 g/L	-0,21 ²⁾
	120 g/L	-0,26 ²⁾
	180 g/L	-0,29 ²⁾
	¹⁾ secondo Kolthoff (1922)	²⁾ secondo Parson e Douglas (1926)

3. Nell'analisi di acqua clorurata, il tenore di cloro residuo può influenzare la reazione cromatica del reagente liquido. Tale interferenza viene evitata immettendo un piccolo cristallo di tiosolfato di sodio ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) nella soluzione campione prima di aggiungere la soluzione PHENOL RED.

Riferimenti bibliografici

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20



Nome do método

Número do método

Código de barras para a detecção dos métodos

Área de medição

$K_{S_{4.3}} T$
0.1 - 4 mmol/l $K_{S_{4.3}}$
Ácido / Indicador

20
S:4.3

Indicado no display: MD 100 / MD 110 / MD 200

Método Químico

Informação específica do instrumento

O teste pode ser realizado nos seguintes dispositivos. Além disso, a cubeta necessária e a faixa de absorção do fotômetro são indicadas.

Dispositivos	Cubeta	λ	Faixa de Medição
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	\varnothing 24 mm	610 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$
SpectroDirect, XD 7000, XD 7500	\varnothing 24 mm	615 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$

Material

Material necessário (parcialmente opcional):

Título	Unidade de Embalagem	Artigo No
Alka-M-Photometer	Pastilhas / 100	513210BT
Alka-M-Photometer	Pastilhas / 250	513211BT

Lista de Aplicações

- Tratamento de Esgotos
- Tratamento de Água Potável
- Tratamento de Água Bruta

Notas

1. Os termos alcalinidade-m, m-valor, alcalinidade total e capacidade de acidez $K_{S_{4.3}}$ são idênticos.
2. O cumprimento exato do volume da amostra de 10 ml é decisivo para a precisão do resultado de análise.

Códigos de idioma ISO 639-1

Nível de revisão

PT Métodos Manual 01/20

Efetuar a medição

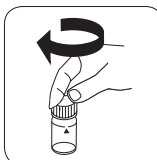
Realização da determinação Capacidade de acidez $K_{s4.3}$ com pastilha

Escolher o método no equipamento.

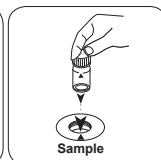
Para este método não tem de ser efetuada uma medição ZERO nos seguintes equipamentos: XD 7000, XD 7500



Encher a célula de 24 mm com 10 ml de amostra .

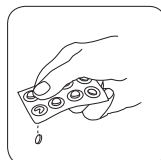


Fechar a(s) célula(s).

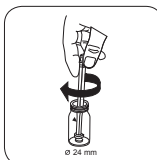


Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.

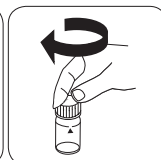
• • •



Pastilha ALKA-M-PHOTO-METER.



Esmagar a(s) pastilha(s) rodando ligeiramente.



Fechar a(s) célula(s).

PT Métodos Manual 01/20

PT



Alcalinidade-m T

M30

5 - 200 mg/L CaCO₃

tA

Ácido / Indicador

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
Alca-M-fotómetro	Pastilhas / 100	513210BT
Alca-M-fotómetro	Pastilhas / 250	513211BT

Notas

- Os termos alcalinidade-m, m-valor, alcalinidade total e capacidade de acidez $K_{s4,3}$ são idênticos.
- O cumprimento exato do volume da amostra de 10 ml é decisivo para a precisão do resultado de análise.

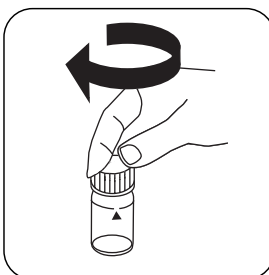


Realização da determinação Alcalinidade, total= alcalinidade-m= m-valor com pastilha

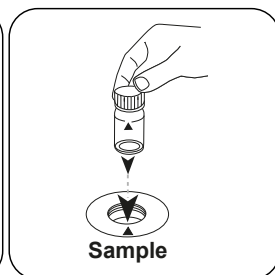
Escolher o método no equipamento.



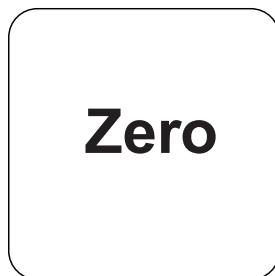
Encher a célula de 24 mm com **10 mL de amostra**.



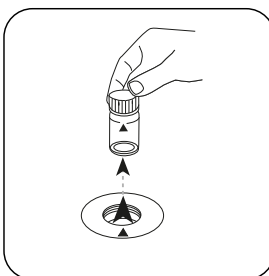
Fechar a(s) célula(s).



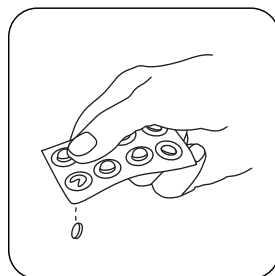
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



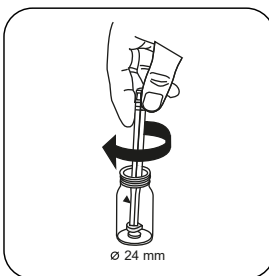
Premir a tecla **ZERO**.



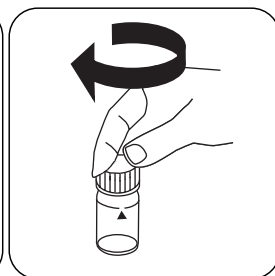
Retirar a célula do compartimento de medição.



Pastilha ALKA-M-PHOTOMETER.



Esmagar a(s) pastilha(s) rodando ligeiramente.



Fechar a(s) célula(s).

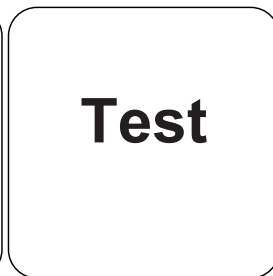
PT



Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

No visor aparece o resultado como Alcalinidade-m.

PT

Análises

A tabela a seguir identifica os valores de saída que podem ser convertidos em outras formas de citação.

Unidade	Forma de citação	Fator de conversão
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

PT

Método Químico

Ácido / Indicador

Apêndice

Derivado de

EN ISO 9963-1

**Cloro T****M100****0.01 - 6.0 mg/L Cl₂^{a)}****CL6****DPD**

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
DPD Nº. 1	Pastilhas / 100	511050BT
DPD Nº. 1	Pastilhas / 250	511051BT
DPD Nº. 1	Pastilhas / 500	511052BT
DPD Nº. 3	Pastilhas / 100	511080BT
DPD Nº. 3	Pastilhas / 250	511081BT
DPD Nº. 3	Pastilhas / 500	511082BT
DPD Nº. 1 Alto Cálcio ^{e)}	Pastilhas / 100	515740BT
DPD Nº. 1 Alto Cálcio ^{e)}	Pastilhas / 250	515741BT
DPD Nº. 1 Alto Cálcio ^{e)}	Pastilhas / 500	515742BT
DPD Nº. 3 Alto Cálcio ^{e)}	Pastilhas / 100	515730BT
DPD Nº. 3 Alto Cálcio ^{e)}	Pastilhas / 250	515731BT
DPD Nº. 3 Alto Cálcio ^{e)}	Pastilhas / 500	515732BT
DPD Nº. 4	Pastilhas / 100	511220BT
DPD Nº. 4	Pastilhas / 250	511221BT
DPD Nº. 4	Pastilhas / 500	511222BT
DPD Nº. 3 Evo	Pastilhas / 100	511420BT
DPD Nº. 3 Evo	Pastilhas / 250	511421BT
DPD Nº. 3 Evo	Pastilhas / 500	511422BT
DPD Nº. 4 Evo	Pastilhas / 100	511970BT
DPD Nº. 4 Evo	Pastilhas / 250	511971BT
DPD Nº. 4 Evo	Pastilhas / 500	511972BT

Padrões disponíveis

Título	Unidade de Embalagem	Código do Produto
ValidCheck Cloro 1,5 mg/l	1 pc.	48105510

Amostragem

1. Na preparação da amostra é preciso evitar a libertação de gases de cloro, p. ex. através da pipetagem e agitação.
2. A análise tem de ser efetuada logo após a recolha da amostra.

Preparação

1. Limpeza das células:
Uma vez que muitos produtos de limpeza domésticos (p. ex. lava-louça) contêm substâncias redutoras, na determinação de cloro pode haver demasiadas reduções. Para excluir este erro de medição, os equipamentos de vidro não deviam ter a capacidade de absorção de cloro. Para esse efeito, os equipamentos de vidro são guardados por uma hora sob solução de hipoclorito de sódio (0,1 g/L) e depois devem ser bem enxaguados com água desmineralizada.
2. Para a determinação individual de cloro livre e cloro total é conveniente usar respetivamente um conjunto próprio de células (ver EN ISO 7393-2, alínea 5.3).
3. A formação de cores DPD ocorre com um valor pH entre 6,2 e 6,5. Os reagentes contêm, por isso, um tampão para ajustar o valor pH. As águas fortemente alcalinas ou ácidas devem, porém, antes da análise, ser ajustadas para um valor pH entre 6 e 7 (com 0,5 mol/L de ácido sulfúrico ou 1 mol/L soda cáustica).

Notas

1. Os pastilhas Evo podem ser utilizadas como alternativa à pastilha padrão correspondente (por exemplo, DPD N° 3 Evo em vez da DPD N° 3).

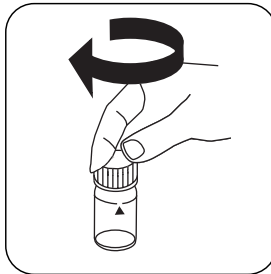


Realização da determinação Cloro livre com pastilha

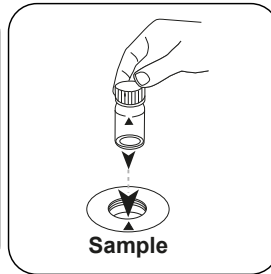
Escolher o método no equipamento.



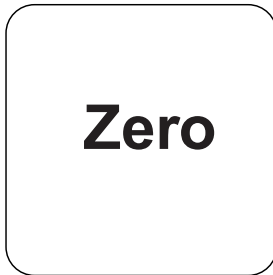
Encher a célula de 24 mm com **10 mL de amostra**.



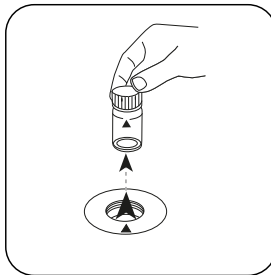
Fechar a(s) célula(s).



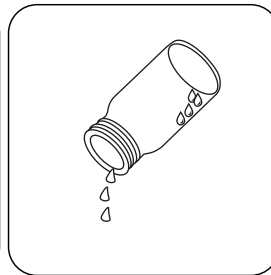
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



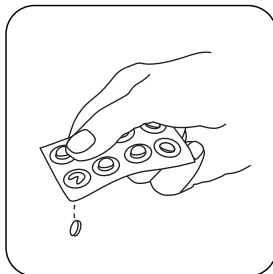
Premir a tecla **ZERO**.



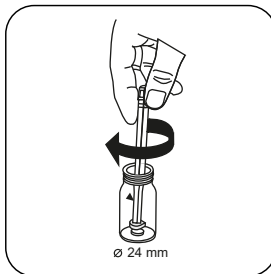
Retirar a célula do compartimento de medição.



Esvaziar a célula até ficarem apenas algumas gotas.



Pastilha DPD No. 1.



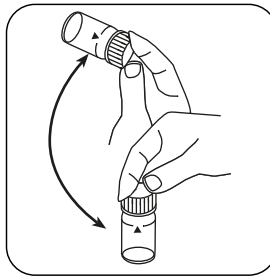
Esmagar a(s) pastilha(s) rodando ligeiramente.



Encher a célula até à **marca de 10 mL** com a amostra.



Fechar a(s) célula(s).



Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.

PT

Test

Premir a tecla **TEST** (XD: **START**).

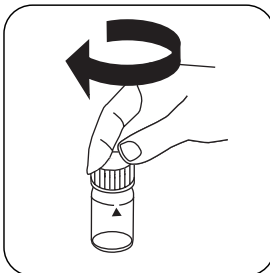
No visor aparece o resultado em mg/L Cloro livre.

Realização da determinação Cloro total com pastilha

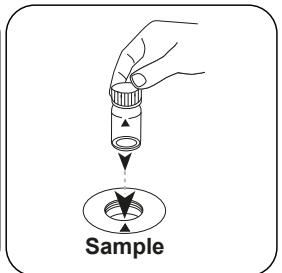
Escolher o método no equipamento.



Encher a célula de 24 mm com **10 mL de amostra**.



Fechar a(s) célula(s).



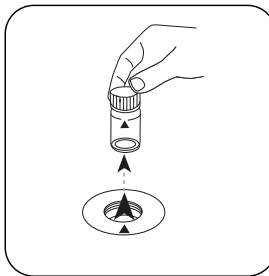
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



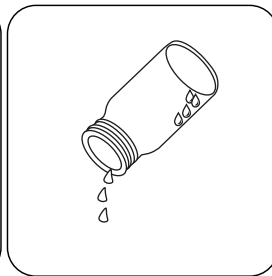
Zero

PT

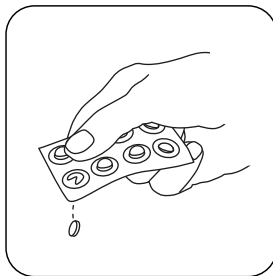
Pressionar a tecla **ZERO**.



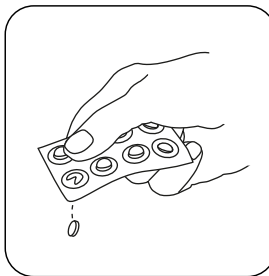
Retirar a célula do compartimento de medição.



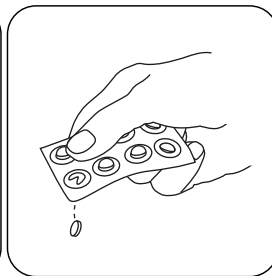
Esvaziar a célula até ficarem apenas algumas gotas.



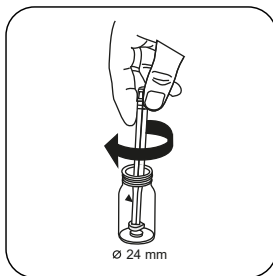
Pastilha DPD No. 1.



Pastilha DPD No. 3.



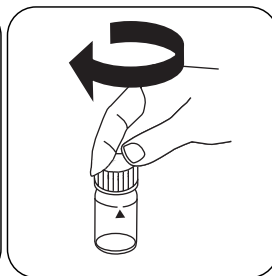
Como alternativa aos comprimidos DPD No. 1 e No. 3, pode ser adicionado 1 comprimido DPD No. 4.



Esmagar a(s) pastilha(s) rodando ligeiramente.



Encher a célula até à **marca de 10 mL** com a amostra .



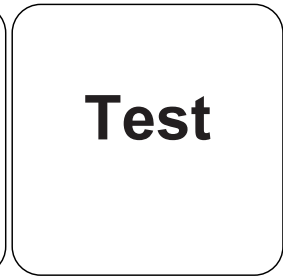
Fechar a(s) célula(s).



Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).



Aguardar **2 minuto(s) de tempo de reação**.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Cloro total.



Método Químico

DPD

Apêndice

PT

Texto de Interferências

Interferências Persistentes

- Todos os oxidantes presentes nas amostras reagem como o cloro, o que leva a resultados demasiado altos.

Interferências Removíveis

- As interferências por cobre e ferro(III) devem ser eliminadas por EDTA.
- Nas amostras com elevado teor de cálcio* e/ou elevada condutividade* pode ocorrer, se forem usadas as pastilhas de reagente, uma turvação da amostra e, por conseguinte, a medição pode ficar errada. Neste caso, deve usar em alternativa a pastilha de reagente DPD No. 1 High Calcium e a pastilha de reagente DPD No. 3 High Calcium.
*não podem ser indicados valores exatos, uma vez que a formação de uma turvação depende do tipo e da composição da água da amostra.
- Concentrações de cloro superiores a 10 mg/L, se forem usadas pastilhas, podem causar resultados dentro da área de medição até 0 mg/L. No caso de uma concentração demasiado alta de cloro, deve diluir a amostra com água sem cloro. 10 mL da amostra diluída é colocada em reagente e a medição é repetida (teste de plausibilidade).

Interferências	a partir de / [mg/L]
CrO_4^{2-}	0.01
MnO_2	0.01

Validação de método

Limite de Detecção	0.02 mg/L
Limite de Determinação	0.06 mg/L
Fim da Faixa de Medição	6 mg/L
Sensibilidade	2.05 mg/L / Abs
Faixa de Confiança	0.04 mg/L
Desvio Padrão	0.019 mg/L
Coefficiente de Variação	0.87 %

Conformidade

EN ISO 7393-2



^aDeterminação do possível livre, vinculado, total | ^aReagente auxiliar, alternativamente ao DPD no. 1 / não 3 quando a amostra é nublada devido ao alto teor de íons de cálcio e / ou alta condutividade

**Cloro L****M101****0.02 - 4.0 mg/L Cl₂^{a)}****CL6****DPD**

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
DPD 1 solução tampão, frasco azul	15 mL	471010
Solução tampão DPD 1	100 mL	471011
DPD 1 solução tampão em embalagem de 6	1 pc.	471016
Solução de reagente DPD 1, frasco verde	15 mL	471020
Solução de reagente DPD 1	100 mL	471021
Solução de reagente DPD 1 numa embalagem de 6 unidades	1 pc.	471026
DPD 3 Solução, frasco vermelho	15 mL	471030
Solução DPD 3	100 mL	471031
Solução DPD 3 numa embalagem de 6 unidades	1 pc.	471036
Kit de reagentes DPD	1 pc.	471056

Padrões disponíveis

Título	Unidade de Embalagem	Código do Produto
ValidCheck Cloro 1,5 mg/l	1 pc.	48105510

Amostragem

1. Na preparação da amostra é preciso evitar a libertação de gases de cloro, p. ex. através da pipetagem e agitação.
2. A análise tem de ser efetuada logo após a recolha da amostra.

Preparação

1. Limpeza das células:
Uma vez que muitos produtos de limpeza domésticos (p. ex. lava-louça) contêm substâncias redutoras, na determinação de cloro pode haver demasiadas reduções. Para excluir este erro de medição, os equipamentos de vidro não devem ter a capacidade de absorção de cloro. Para esse efeito, os equipamentos de vidro são guardados por uma hora sob solução de hipoclorito de sódio (0,1 g/L) e depois devem ser bem enxaguados com água desmineralizada.
2. Para a determinação individual de cloro livre e cloro total é conveniente usar respetivamente um conjunto próprio de células (ver EN ISO 7393-2, alínea 5.3).
3. A formação de cores DPD ocorre com um valor pH entre 6,2 e 6,5. Os reagentes contêm, por isso, um tampão para ajustar o valor pH. As águas fortemente alcalinas ou ácidas devem, porém, antes da análise, ser ajustadas para um valor pH entre 6 e 7 (com 0,5 mol/l de ácido sulfúrico ou 1 mol/l soda cáustica).

Notas

1. Depois de usados, os frascos conta-gotas devem ser novamente fechados com a respetiva tampa de enroscar à cor.
2. Guardar o conjunto de reagentes em local fresco entre +6 °C e +10 °C.

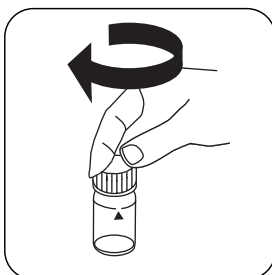


Realização da determinação Cloro livre com reagente líquido

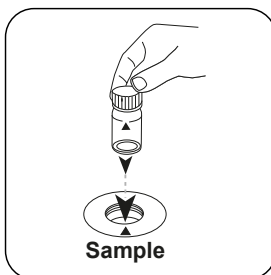
Escolher o método no equipamento.



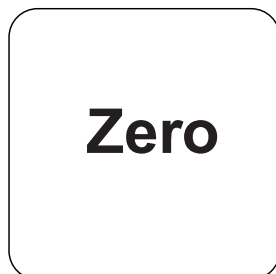
Encher a célula de 24 mm com **10 mL de amostra**.



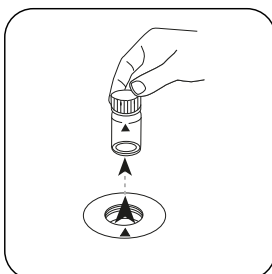
Fechar a(s) célula(s).



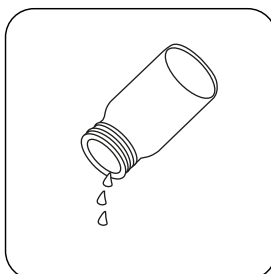
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



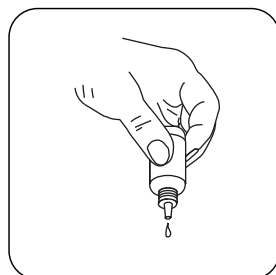
Premir a tecla **ZERO**.



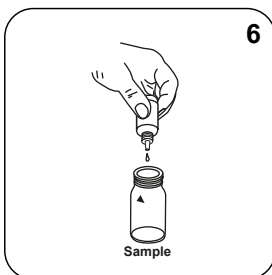
Retirar a célula do compartimento de medição.



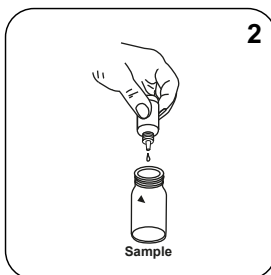
Esvaziar a célula.



Manter os frascos conta gotas na vertical e pressionar lentamente para adicionar gotas de igual dimensão.



Adicionar **6 gotas DPD 1 Buffer Solution** à célula de amostra.



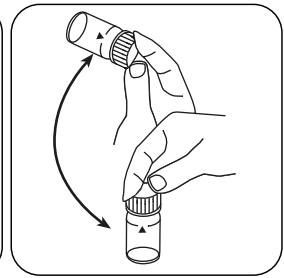
Adicionar **2 gotas DPD 1 Reagent Solution** à célula de amostra.



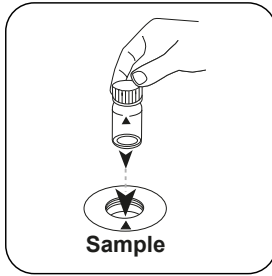
Encher a célula até à **marca de 10 mL** com a amostra .



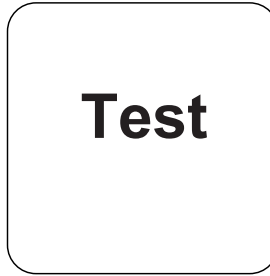
Fechar a(s) célula(s).



Misturar o conteúdo girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

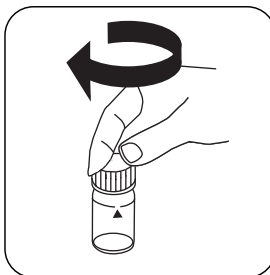
No visor aparece o resultado em mg/L Cloro livre.

Realização da determinação Cloro total com reagente líquido

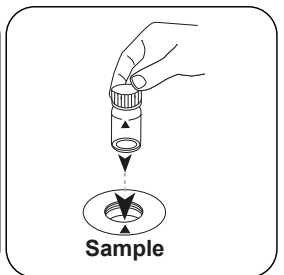
Escolher o método no equipamento.



Encher a célula de 24 mm com **10 mL de amostra** .



Fechar a(s) célula(s).

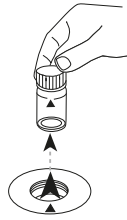


Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Zero

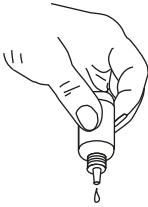
PT
 Premir a tecla **ZERO**.



Retirar a célula do compartimento de medição.



Esvaziar a célula.



Manter os frascos conta gotas na vertical e pressionar lentamente para adicionar gotas de igual dimensão.



6

Adicionar **6 gotas DPD 1 Buffer Solution** à célula de amostra.



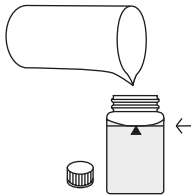
2

Adicionar **2 gotas DPD 1 Reagent Solution** à célula de amostra.

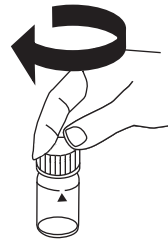


3

Adicionar **3 gotas DPD 3 Solution** à célula de amostra.



Encher a célula até à **marca de 10 mL** com a amostra .



Fechar a(s) célula(s).



Misturar o conteúdo girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).



Aguardar **2 minuto(s) de tempo de reação**.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Cloro total.



Método Químico

DPD

Apêndice

PT

Texto de Interferências

Interferências Persistentes

- Todos os oxidantes presentes nas amostras reagem como o cloro, o que leva a resultados demasiado altos.

Interferências Removíveis

- As interferências por cobre e ferro(III) devem ser eliminadas por EDTA.
- Concentrações de cloro superiores a 4 mg/L, se forem usados reagentes líquidos, podem causar resultados dentro da área de medição até 0 mg/L. Neste caso, deve diluir a amostra com água sem cloro. 10 ml da amostra diluída é colocada em reagente e a medição é repetida (teste de plausibilidade).

Interferências	a partir de / [mg/L]
CrO_4^{2-}	0,01
MnO_2	0,01

Conformidade

EN ISO 7393-2

^{a)}Determinação do possível livre, vinculado, total



Cloro HR T

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

Material

PT

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
DPD N.º. 1 HR	Pastilhas / 100	511500BT
DPD N.º. 1 HR	Pastilhas / 250	511501BT
DPD N.º. 1 HR	Pastilhas / 500	511502BT
DPD N.º. 3 HR	Pastilhas / 100	511590BT
DPD N.º. 3 HR	Pastilhas / 250	511591BT
DPD N.º. 3 HR	Pastilhas / 500	511592BT
Definir N.º DPD 1 HR/No. 3 HR #	cada 100	517791BT
Definir N.º DPD 1 HR/No. 3 HR #	cada 250	517792BT
DPD N.º. 1 Alto Cálcio ^{e)}	Pastilhas / 100	515740BT
DPD N.º. 1 Alto Cálcio ^{e)}	Pastilhas / 250	515741BT
DPD N.º. 1 Alto Cálcio ^{e)}	Pastilhas / 500	515742BT
DPD N.º. 3 Alto Cálcio ^{e)}	Pastilhas / 100	515730BT
DPD N.º. 3 Alto Cálcio ^{e)}	Pastilhas / 250	515731BT
DPD N.º. 3 Alto Cálcio ^{e)}	Pastilhas / 500	515732BT
DPD N.º. 3 HR Evo	Pastilhas / 100	511920BT
DPD N.º. 3 HR Evo	Pastilhas / 250	511921BT
DPD N.º. 3 HR Evo	Pastilhas / 500	511922BT

Amostragem

1. Na preparação da amostra é preciso evitar a libertação de gases de cloro, p. ex. através da pipetagem e agitação.
2. A análise tem de ser efetuada logo após a recolha da amostra.



Preparação

1. Limpeza das células:
Uma vez que muitos produtos de limpeza domésticos (p. ex. lava-louça) contêm substâncias redutoras, na determinação de cloro pode haver demasiadas reduções. Para excluir este erro de medição, os equipamentos de vidro não deviam ter a capacidade de absorção de cloro. Para esse efeito, os equipamentos de vidro são guardados por uma hora sob solução de hipoclorito de sódio (0,1 g/L) e depois devem ser bem enxaguados com água desmineralizada.
2. Para a determinação individual de cloro livre e cloro total é conveniente usar respetivamente um conjunto próprio de células (ver EN ISO 7393-2, alínea 5.3).
3. A formação de cores DPD ocorre com um valor pH entre 6,2 e 6,5. Os reagentes contêm, por isso, um tampão para ajustar o valor pH. As águas fortemente alcalinas ou ácidas devem, porém, antes da análise, ser ajustadas para um valor pH entre 6 e 7 (com 0,5 mol/L de ácido sulfúrico ou 1 mol/L soda cáustica).

Notas

1. Os pastilhas Evo podem ser utilizadas como alternativa à pastilha padrão correspondente (por exemplo, DPD N° 3 Evo em vez da DPD N° 3).

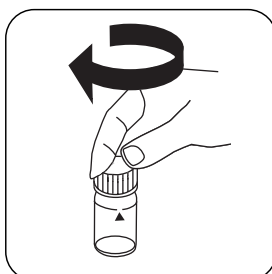


Realização da determinação Cloro HR livre com pastilha

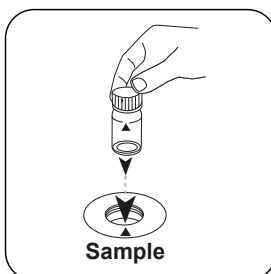
Escolher o método no equipamento.



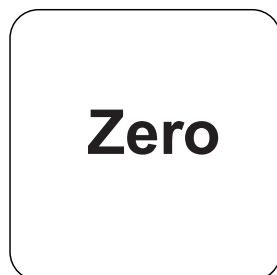
Encher a célula de 24 mm com **10 mL de amostra**.



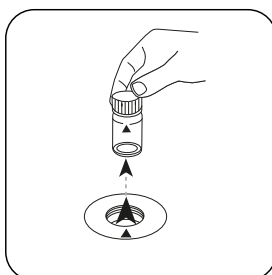
Fechar a(s) célula(s).



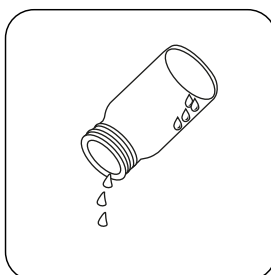
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



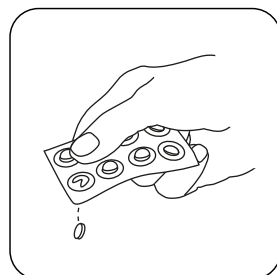
Premir a tecla **ZERO**.



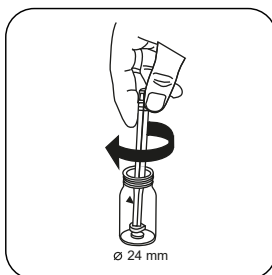
Retirar a célula do compartimento de medição.



Esvaziar a célula até ficarem apenas algumas gotas.



Pastilha DPD No. 1 HR.



Esmagar a(s) pastilha(s) rodando ligeiramente.



Encher a célula até à **marca de 10 mL** com a amostra.



Fechar a(s) célula(s).



Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.

PT

Test

Premir a tecla **TEST** (XD: **START**).

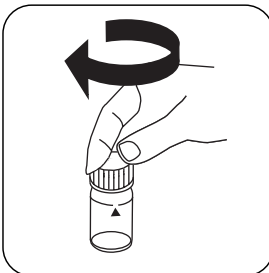
No visor aparece o resultado em mg/L Cloro livre.

Realização da determinação Cloro HR total com pastilha

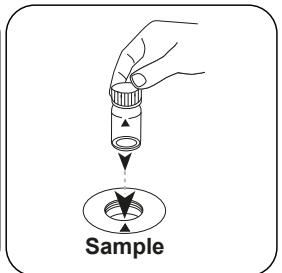
Escolher o método no equipamento.



Encher a célula de 24 mm com **10 mL de amostra**.



Fechar a(s) célula(s).



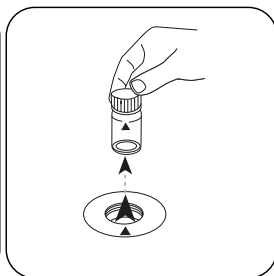
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



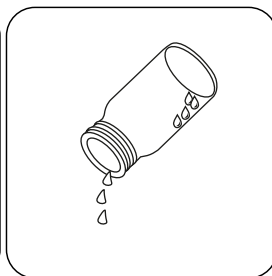
Zero

PT

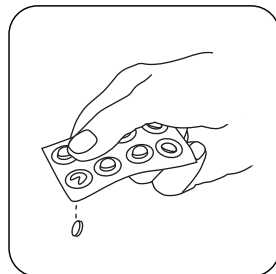
Pressionar a tecla **ZERO**.



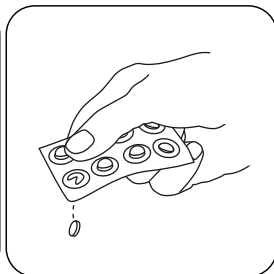
Retirar a célula do compartimento de medição.



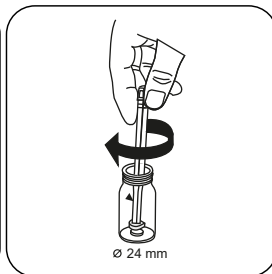
Esvaziar a célula até ficarem apenas algumas gotas.



Pastilha DPD No. 1 HR .



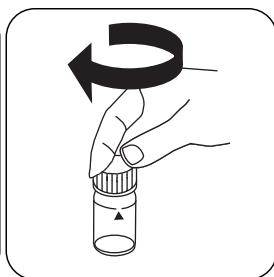
Pastilha DPD No. 3 HR .



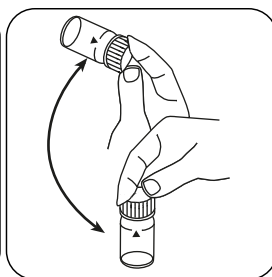
Esmagar a(s) pastilha(s) rodando ligeiramente.



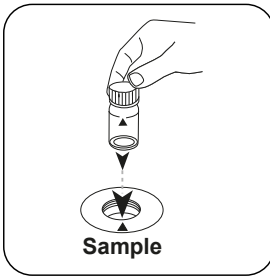
Encher a célula até à marca de 10 mL com a amostra .



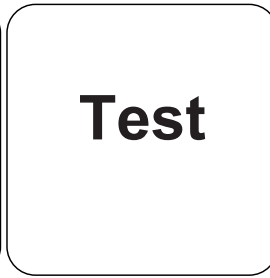
Fechar a(s) célula(s).



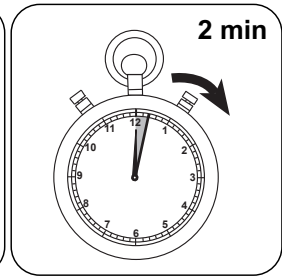
Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).



Aguardar **2 minuto(s) de tempo de reação**.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Cloro total.



Método Químico

DPD

Apêndice

PT

Texto de Interferências

Interferências Persistentes

- Todos os oxidantes presentes nas amostras reagem como o cloro, o que leva a resultados demasiado altos.

Interferências Removíveis

- As interferências por cobre e ferro(III) devem ser eliminadas por EDTA.
- Nas amostras com elevado teor de cálcio* e/ou elevada condutividade* pode ocorrer, se forem usadas as pastilhas de reagente, uma turvação da amostra e, por conseguinte, a medição pode ficar errada. Neste caso, deve usar em alternativa a pastilha de reagente DPD No. 1 High Calcium e a pastilha de reagente DPD No. 3 High Calcium.

*não podem ser indicados valores exatos, uma vez que a formação de uma turvação depende do tipo e da composição da água da amostra.

Conformidade

EN ISO 7393-2

^oDeterminação do possível livre, vinculado, total | ^oReagente auxiliar, alternativamente ao DPD no. 1 / não 3 quando a amostra é nublada devido ao alto teor de íons de cálcio e / ou alta condutividade | ^oincluindo vareta de agitação

**Cobre T****M150****0.05 - 5 mg/L Cu^a****Cu****Biquinoline**

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
Cobre Não. 1	Pastilhas / 100	513550BT
Cobre Não. 1	Pastilhas / 250	513551BT
Cobre Não. 2	Pastilhas / 100	513560BT
Cobre Não. 2	Pastilhas / 250	513561BT
Definir número de cobre 1/Não. 2 [#]	cada 100	517691BT
Definir número de cobre 1/Não. 2 [#]	cada 250	517692BT
ValidCheck Cobre 2 mg/l	1 pc.	48141525

Preparação

1. As águas fortemente alcalinas ou ácidas deviam, antes da análise, ser ajustadas para um valor pH de 4 a 6.



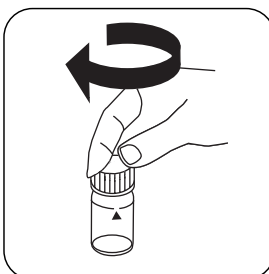
Realização da determinação Cobre, livre com pastilha

Escolher o método no equipamento.

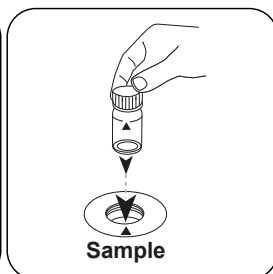
Escolha ainda a determinação: livre



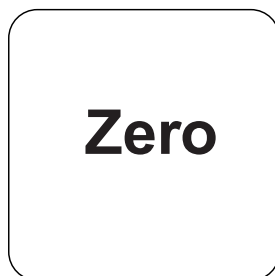
Encher a célula de 24 mm com **10 mL de amostra**.



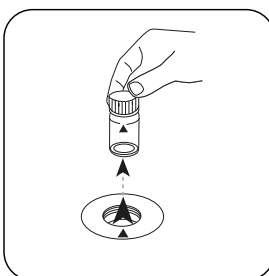
Fechar a(s) célula(s).



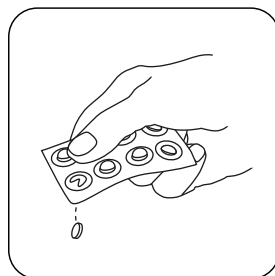
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



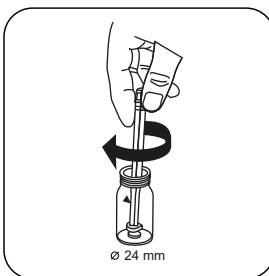
Premir a tecla **ZERO**.



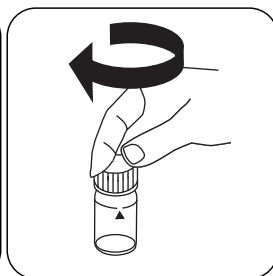
Retirar a célula do compartimento de medição.



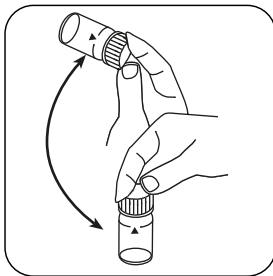
Pastilha COPPER No. 1.



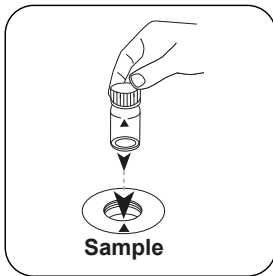
Esmagar a(s) pastilha(s) rodando ligeiramente.



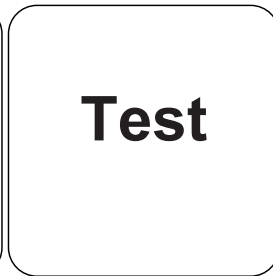
Fechar a(s) célula(s).



Dissolver a(s) pastilha(s) girando.

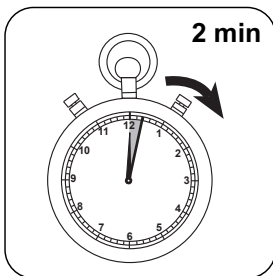


Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

PT



Aguardar **2 minuto(s)** de tempo de reação.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Cobre livre.

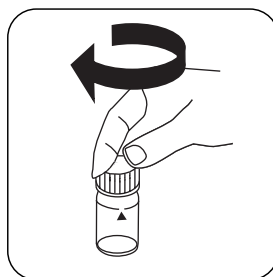
Realização da determinação Cobre, total com pastilha

Escolher o método no equipamento.

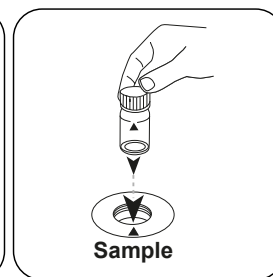
Escolha ainda a determinação: total



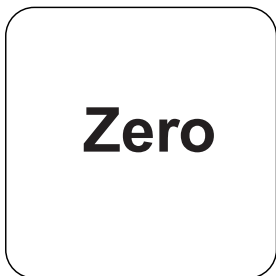
Encher a célula de 24 mm com **10 mL de amostra**.



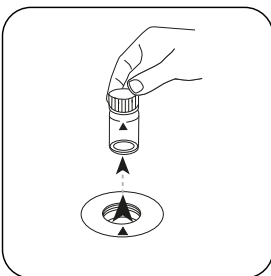
Fechar a(s) célula(s).



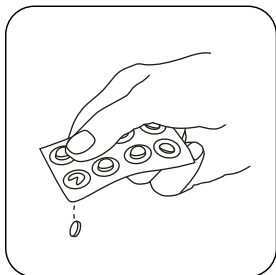
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



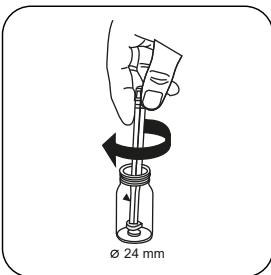
Premir a tecla **ZERO**.



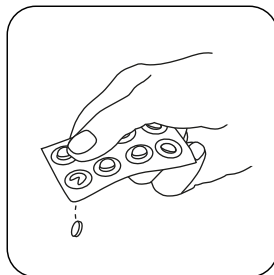
Retirar a célula do compartimento de medição.



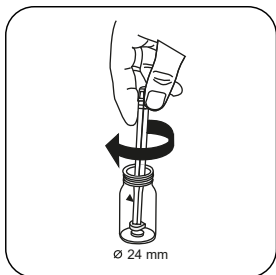
Pastilha COPPER No. 1.



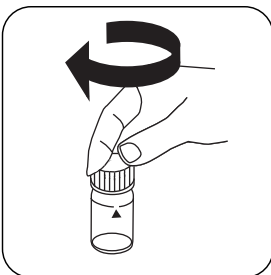
Esmagar a(s) pastilha(s) rodando ligeiramente e dissolver.



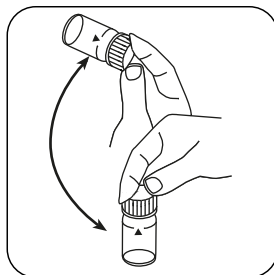
Pastilha COPPER No. 2.



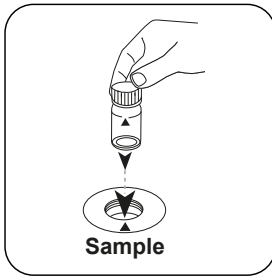
Esmagar a(s) pastilha(s) rodando ligeiramente.



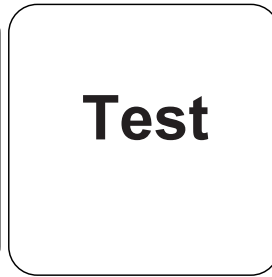
Fechar a(s) célula(s).



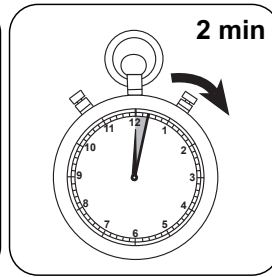
Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).



Aguardar **2 minuto(s) de tempo de reação**.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Cobre total.

Método Químico

Biquinoline

Apêndice

Texto de Interferências

Interferências Persistentes

1. Cianeto CN⁻ e Prata Ag⁺ interferem a determinação.

Validação de método

Limite de Detecção	0.05 mg/L
Limite de Determinação	0.15 mg/L
Fim da Faixa de Medição	5 mg/L
Sensibilidade	3.8 mg/L / Abs
Faixa de Confiança	0.026 mg/L
Desvio Padrão	0.011 mg/L
Coefficiente de Variação	0.42 %

Bibliografia

Análise fotométrica, Lange/Vjedelek, Verlag Chemie 1980

^aDeterminação do possível livre, vinculado, total | ^bIncluindo vareta de agitação



CyA T

M160

10 - 160 mg/L CyA

CyA

Melamine

PT

Material

Material necessário (parcialmente opcional):

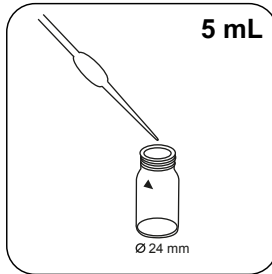
Reagentes	Unidade de Embalagem	Código do Produto
Teste CyA	Pastilhas / 100	511370BT
Teste CyA	Pastilhas / 250	511371BT
água desmineralizada	100 mL	461275
água desmineralizada	250 mL	457022

Notas

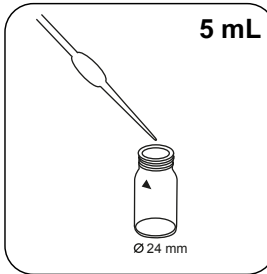
1. O ácido cianúrico causa uma turvação muito finamente distribuída com aspeto leitoso. A presença de algumas partículas não remete para a presença de ácido cianúrico.

Realização da determinação Teste de ácido cianúrico com pastilha

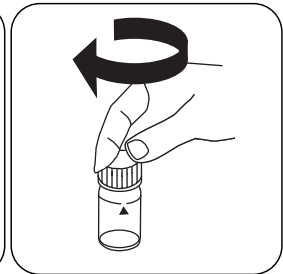
Escolher o método no equipamento.



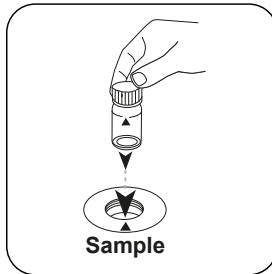
Encher a célula de 24 mm com **5 mL de água desmineralizada**.



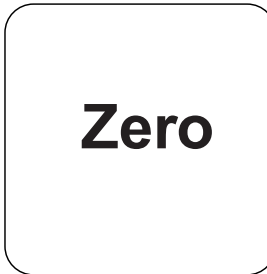
Adicionar **5 mL de amostra** à célula.



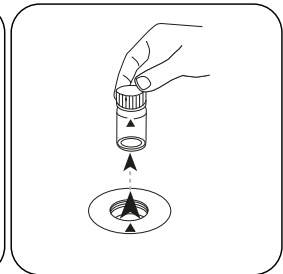
Fechar a(s) célula(s).



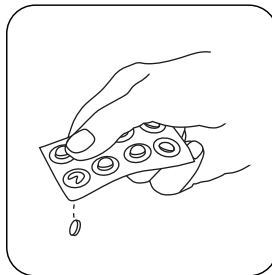
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



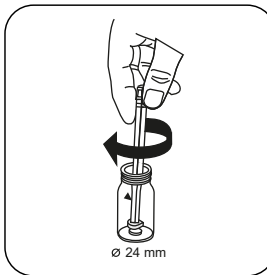
Premir a tecla **ZERO**.



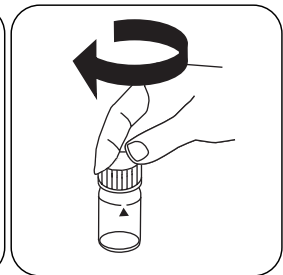
Retirar a célula do compartimento de medição.



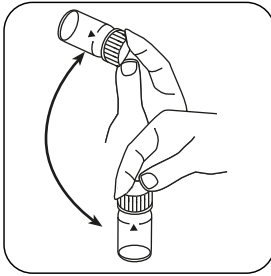
Pastilha CyA-Test.



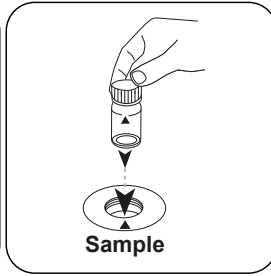
Esmagar a(s) pastilha(s) rodando ligeiramente.



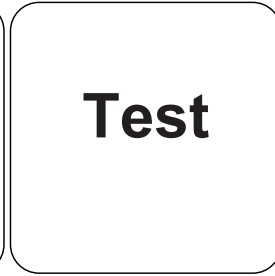
Fechar a(s) célula(s).



Misturar o conteúdo girando (durante pelo menos 60 s até que o pastilha esteja completamente dissolvido).



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

No visor aparece o resultado em mg/L ácido cianúrico.



Método Químico

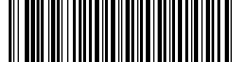
Melamine

Texto de Interferências

Interferências Persistentes

1. Partículas não dissolvidas podem causar resultados demasiado altos. Por isso, é importante dissolver totalmente as pastilhas.

PT



Ferro T

M220

0.02 - 1 mg/L Fe

FE

Ferrozine / Thioglycolate

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
Iron II LR (Fe ²⁺)	Pastilhas / 100	515420BT
Iron II LR (Fe ²⁺)	Pastilhas / 250	515421BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Pastilhas / 100	515370BT
Iron LR (Fe ²⁺ und Fe ³⁺)	Pastilhas / 250	515371BT

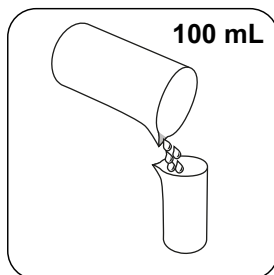
Preparação

1. As águas que foram tratadas com compostos orgânicos como proteção anticorrosiva, etc. têm de ser eventualmente oxidadas para destruir os complexos de ferro. Para isso, transfere-se uma amostra de 100 ml com 1 ml de ácido sulfúrico concentrado e 1 ml de ácido nítrico concentrado e evaporada para metade. Depois de arrefecer, passa-se à digestão.

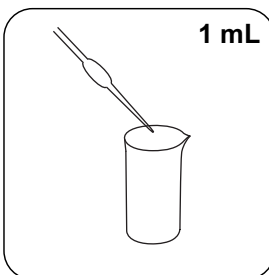
Notas

1. Neste método ocorre a determinação de Fe²⁺ e Fe³⁺ totalmente dissolvido.
2. Para determinar Fe²⁺ usa-se a pastilha IRON (II) LR, em vez da pastilha IRON LR.

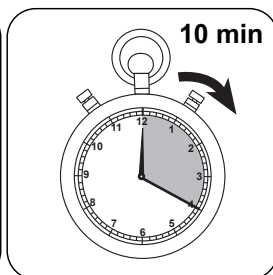
Digestão



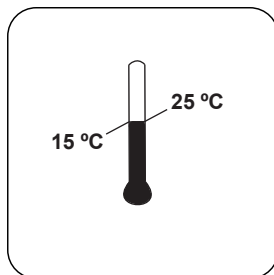
Encher um recipiente de amostra adequado com **100 mL de amostra** .



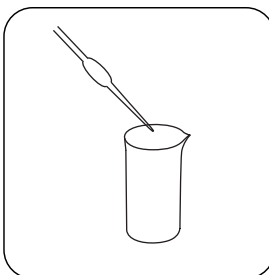
Adicionar **1 mL ácido sulfúrico concentrado** .



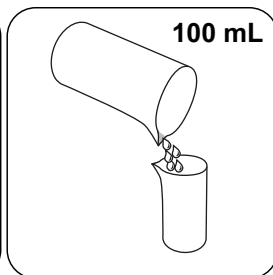
A amostra deve **aquecer durante 10 minutos**, ou até tudo se ter totalmente dissolvido.



Deixar a amostra arrefecer até à **temperatura ambiente** .



Ajustar o **valor pH** da amostra com **solução amoniacal para 3-5**.



Encher a amostra com **água desmineralizada até 100 mL** .

Usar esta amostra para a análise de total de ferro solvido e dissolvido.

Realização da determinação Ferro(II,III), dissolvido com pastilha

Escolher o método no equipamento.

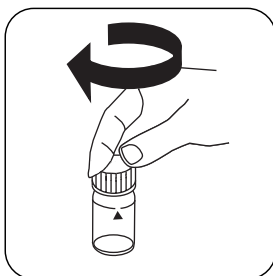
Para a determinação de **Ferro dissolvido e não dissolvido** deve realizar a **digestão** descrita.



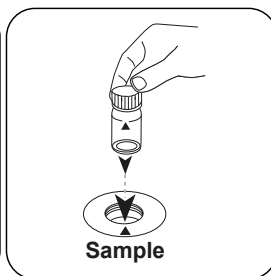
PT



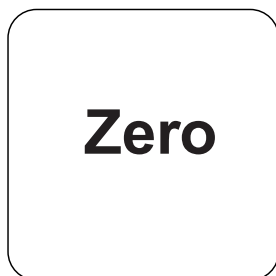
Encher a célula de 24 mm com **10 mL de amostra**.



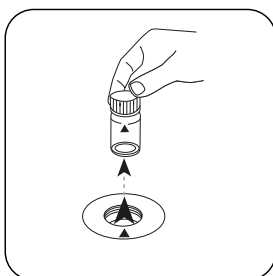
Fechar a(s) célula(s).



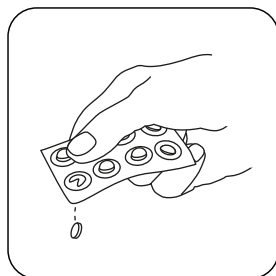
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



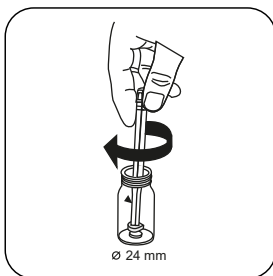
Premir a tecla **ZERO**.



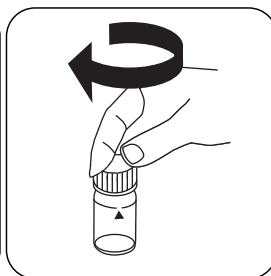
Retirar a célula do compartimento de medição.



Pastilha IRON LR.



Esmagar a(s) pastilha(s) rodando ligeiramente.



Fechar a(s) célula(s).



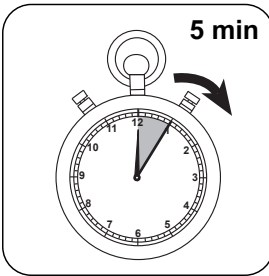
Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



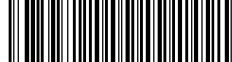
Premir a tecla **TEST** (XD: **START**).



Aguardar **5 minuto(s) de tempo de reação**.

Decorrido o tempo de reação, a medição é efetuada automaticamente.

No visor aparece o resultado em mg/L Ferro.



Método Químico

Ferrozine / Thioglycolate

Apêndice

PT

Texto de Interferências

Interferências Removíveis

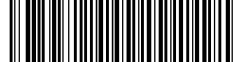
1. A presença de cobre aumenta o resultado de medição em 10 %. Numa concentração de 10 mg/L de cobre na amostra, o resultado de medição aumenta em 1 mg/L de ferro.
A interferência pode ser eliminada com a adição de tiourea

Validação de método

Limite de Detecção	0.01 mg/L
Limite de Determinação	0.016 mg/L
Fim da Faixa de Medição	1 mg/L
Sensibilidade	0.92 mg/L / Abs
Faixa de Confiança	0.013 mg/L
Desvio Padrão	0.005 mg/L
Coefficiente de Variação	1.23 %

Bibliografia

Análise fotométrica, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



Valor pH T

M330

6.5 - 8.4 pH

PH

Phenol Red

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
Fotómetro Fenol Vermelho	Pastilhas / 100	511770BT
Fotómetro Fenol Vermelho	Pastilhas / 250	511771BT
Fotómetro Fenol Vermelho	Pastilhas / 500	511772BT

Notas

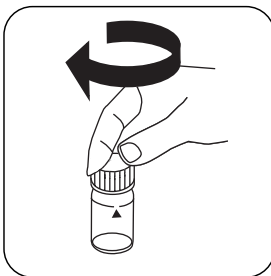
1. Para a determinação fotométrica do valor pH deve usar somente pastilhas PHENOL RED com impressão de película preta, que estão identificadas com o termo PHOTOMETER.

Realização da determinação Valor pH com pastilha

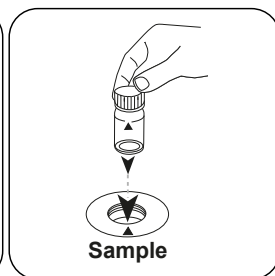
Escolher o método no equipamento.



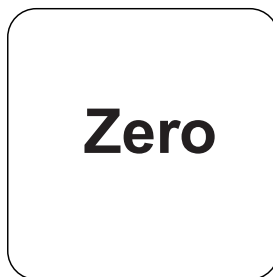
Encher a célula de 24 mm com **10 mL de amostra**.



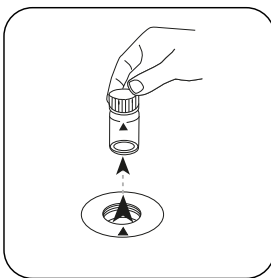
Fechar a(s) célula(s).



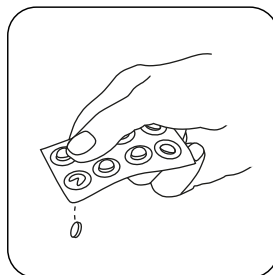
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



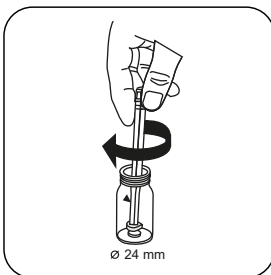
Premir a tecla **ZERO**.



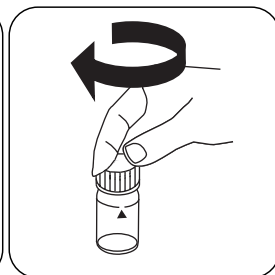
Retirar a célula do compartimento de medição.



Pastilha PHENOL RED PHOTOMETER.



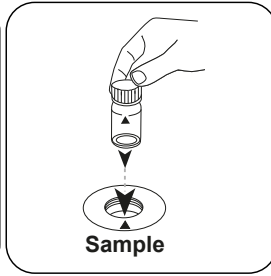
Esmagar a(s) pastilha(s) rodando ligeiramente.



Fechar a(s) célula(s).



Dissolver a(s) pastilha(s) girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

No visor aparece o resultado como valor pH.

PT

Método Químico

Phenol Red

Apêndice

Texto de Interferências

PT

Interferências Persistentes

1. As amostras de água com baixa dureza de carbonato* podem obter valores pH incorretos.

* $K_{S4,3} < 0,7 \text{ mmol/l} \triangleq \text{Alcalinidade total} < 35 \text{ mg/L CaCO}_3$.

Interferências Removíveis

1. Os valores pH inferiores a 6,5 e superiores a 8,4 podem causar resultados dentro da área de medição. Recomenda-se um teste de plausibilidade (medidor de pH).
2. Erro de sal:
No caso de teores de sal até 2 g/L não é expectável nenhum erro de sal significativo devido ao teor de sal da pastilha de reagente. No caso de teores de sal superiores, deve corrigir os valores de medição do seguinte modo:

Teor de sal da amostra emg/L	30 (água do mar)	60	120	180
Correção	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾segundo Kolthoff (1922)

²⁾segundo Parson e Douglas (1926)

Bibliografia

Colorimetric Chemical Analytical Methods, 9th Edition, London



Valor pH L

M331

6.5 - 8.4 pH

PH

Phenol Red

PT

Material

Material necessário (parcialmente opcional):

Reagentes	Unidade de Embalagem	Código do Produto
Solução de vermelho fenol	15 mL	471040
Solução de vermelho fenol	100 mL	471041
Solução de vermelho fenol em embalagem de -6	1 pc.	471046

Preparação

1. Devido aos diferentes tamanhos de gotas, o resultado de medição pode apresentar desvios maiores do que ao utilizar pastilhas.
Se utilizar uma pipeta (0,18 ml corresponde a 6 gotas) pode reduzir este desvio.

Notas

1. Depois de usado, o frasco conta-gotas deve ser novamente fechado com a respetiva tampa de enroscar à cor.
2. Guardar o reagente em local fresco entre +6 °C e +10 °C.

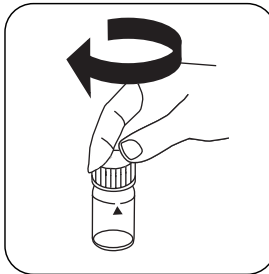


Realização da determinação Valor pH com reagente líquido

Escolher o método no equipamento.



Encher a célula de 24 mm com **10 mL de amostra**.



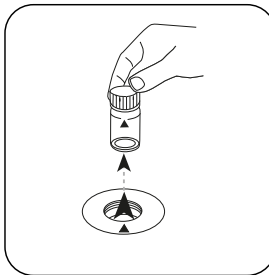
Fechar a(s) célula(s).



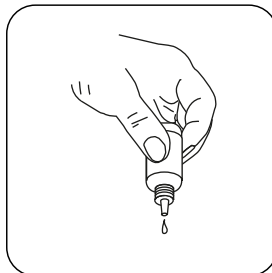
Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



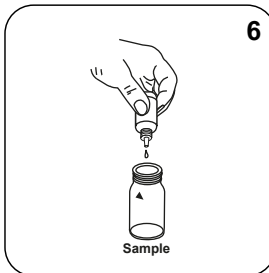
Premir a tecla **ZERO**.



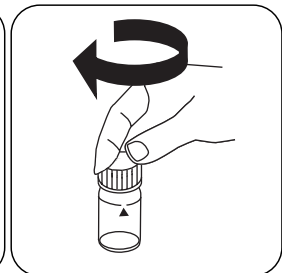
Retirar a célula do compartimento de medição.



Manter os frascos conta gotas na vertical e pressionar lentamente para adicionar gotas de igual dimensão.



Adicionar **6 gotas PHENOL Red-Lösung** à célula de amostra.



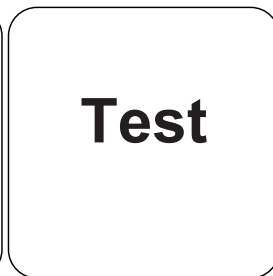
Fechar a(s) célula(s).



Misturar o conteúdo girando.



Colocar a **célula de amostra** no compartimento de medição. Observar o posicionamento.



Premir a tecla **TEST** (XD: **START**).

No visor aparece o resultado como valor pH.

Método Químico

Phenol Red

Apêndice

Texto de Interferências

PT

Interferências Removíveis

1. Erro de sal: Correção do valor de medição (valores médios) para amostras com um teor de sal de:


2.	Teor de sal da amostra	Correção
	30 g/L (água do mar)	-0,15 ¹⁾
	60 g/L	-0,21 ²⁾
	120 g/L	-0,26 ²⁾
	180 g/L	-0,29 ²⁾
	¹⁾ segundo Kolthoff (1922)	²⁾ segundo Parson e Douglas (1926)

3. Na análise de água clorada, o teor de cloro residual existente pode influenciar a reação de cor do reagente líquido. Isto é evitado, na medida em que se insere um pequeno cristal de tiosulfato de sódio ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) na solução de amostra antes de ser adicionada a solução PHENOL RED.

Bibliografia

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20



Naam van de methode

Nummer methode

Streepjescode ter identificatie van de methode

Meetbereik

$K_{S_{4.3}} T$ M20
0.1 - 4 mmol/l $K_{S_{4.3}}$ S:4.3
Zuur / Indicator

Chemische methode

Uitlezing in MD
100 MD 110 / MD 200

Instrument specifieke informatie

De test kan op de volgende apparaten worden uitgevoerd. Bovendien worden de vereiste cuvette en het absorptiebereik van de fotometer aangegeven.

Toestellen	Cuvet	λ	Meetbereik
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	\varnothing 24 mm	610 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$
SpectroDirect, XD 7000, XD 7500	\varnothing 24 mm	615 nm	0.1 - 4 mmol/l $K_{S_{4.3}}$

Reagentia

Benodigd materiaal (deels optioneel):

Titel	Verpakkingseenheid	Bestelnr.
Alka-M-Photometer	Tablet / 100	513210BT
Alka-M-Photometer	Tablet / 250	513211BT

Toepassingsbereik

- Afvalwaterzuivering
- Behandeling drinkwater
- Zuivering vervuild water

Aantekeningen

1. De termen alkaliteit-m, m-waarde, totale alkaliteit en zuurcapaciteit_{S_{4.3}} zijn identiek.
2. De exacte naleving van het monstervolume van 10 ml is bepalend voor de nauwkeurigheid van het analysesresultaat.

Beknopte naam conform de norm ISO 639-1

Herziene versie

NL Handboek van Methoden 01/20

Uitvoering van de meting

Uitvoering van de bepaling Zuurcapaciteit $K_{s4,3}$ met tablet

De methode in het apparaat selecteren.

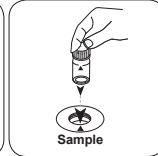
Voor deze methode moet bij de volgende apparaten geen nulmeting worden uitgevoerd:
XD 7000, XD 7500



Spoelbakje van 24 mm met **10 ml staal** vullen.

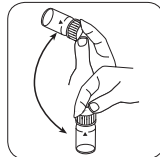


De spoelbakjes afsluiten.



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.

• • •



Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST (XD: START)** indrukken.

De display toont het resultaat als Zuurcapaciteit $K_{s4,3}$.

Test



Alkaliteit-m T

M30

5 - 200 mg/L CaCO₃

tA

Zuur / Indicator

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
Alka-M-fotometer	Tablet / 100	513210BT
Alka-M-fotometer	Tablet / 250	513211BT

Aantekeningen

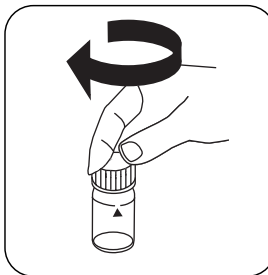
1. De termen alkaliteit-m, m-waarde, totale alkaliteit en zuurcapaciteit_{KS4.3} zijn identiek.
2. De exacte naleving van het monstervolume van 10 ml is bepalend voor de nauwkeurigheid van het analyseresultaat.

Uitvoering van de bepaling Alkaliteit, totaal= alkaliteit-m= m-waarde met tablet

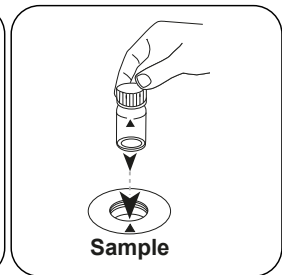
De methode in het apparaat selecteren.



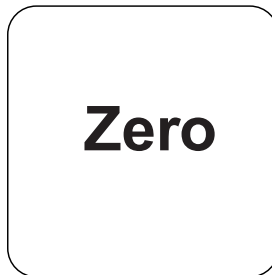
Spoelbakje van 24 mm met **10 mL staal** vullen.



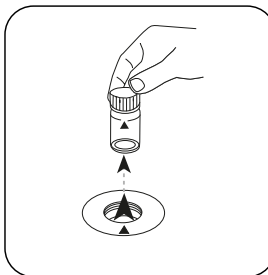
De spoelbakjes afsluiten.



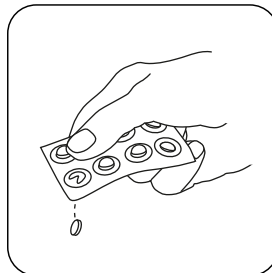
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



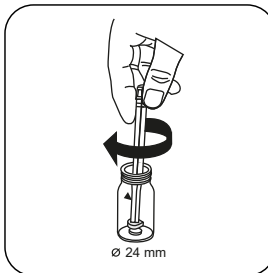
De toets **NUL** indrukken.



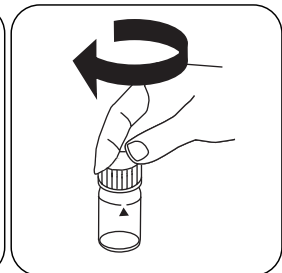
Het spoelbakje uit de meetschacht nemen.



Een **ALKA-M-FOTOMETER** tablet toevoegen.



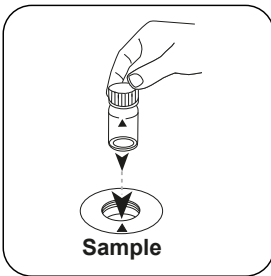
De tabletten onder lichte rotatie verpletteren.



De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

De display toont het resultaat als Alkaliteit-m.

NL

Evaluatie

De volgende tabel geeft aan dat de uitvoerwaarden kunnen worden geconverteerd naar andere citatievormen.

Eenheid	Dagvaardingsformulier	Omrekeningsfactor
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4,3}	0.02

NL

Chemische methode

Zuur / Indicator

Aanhangsel

Afgeleid van

EN ISO 9963-1

**Chloor T****M100****0.01 - 6.0 mg/L Cl₂^{a)}****CL6****DPD**

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
DPD Nr.1	Tablet / 100	511050BT
DPD Nr. 1	Tablet / 250	511051BT
DPD Nr. 1	Tablet / 500	511052BT
DPD Nr. 3	Tablet / 100	511080BT
DPD Nr. 3	Tablet / 250	511081BT
DPD Nr. 3	Tablet / 500	511082BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 100	515740BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 250	515741BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 500	515742BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 100	515730BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 250	515731BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 500	515732BT
DPD Nr. 4	Tablet / 100	511220BT
DPD Nr. 4	Tablet / 250	511221BT
DPD Nr. 4	Tablet / 500	511222BT
DPD No. 3 Evo	Tablet / 100	511420BT
DPD No. 3 Evo	Tablet / 250	511421BT
DPD No. 3 Evo	Tablet / 500	511422BT
DPD Nr.4 Evo	Tablet / 100	511970BT
DPD Nr. 4 Evo	Tablet / 250	511971BT
DPD Nr. 4 Evo	Tablet / 500	511972BT

Beschikbare standaarden

Omschrijving	Verpakkingseenheid	Bestelnr.
ValidCheck Chloor 1,5 mg/l	1 St.	48105510



Bemonstering

1. Tijdens de monstervoorbereiding moet worden vermeden dat het chloor wordt uitgestoten, bijvoorbeeld door pipetteren en schudden.
2. De analyse moet onmiddellijk na de bemonstering worden uitgevoerd.

Vorbereiding

1. Het schoonmaken van de spoelbakjes:
Aangezien veel huishoudelijke reinigingsmiddelen (bijv. afwasmiddelen) minder schadelijke stoffen bevatten, kan de bepaling van chloor leiden tot minder goede resultaten. Om deze meefout uit te sluiten, moeten de glasapparaten chloorvrij zijn. Hiertoe wordt het glaswerk gedurende één uur onder natriumhypochlorietoplossing (0,1 g/L) bewaard en vervolgens grondig gespoeld met gedeïoniseerd water.
2. Voor de individuele bepaling van vrij chloor en totaal chloor is het zinvol om een aparte set spoelbakjes te gebruiken (zie EN ISO 7393-2, paragraaf 5.3).
3. De DPD-kleurontwikkeling vindt plaats bij een pH-waarde van 6,2 tot 6,5. De reagentia bevatten daarom een buffer voor de aanpassing van de pH-waarde. Sterk alkalisch of zuur water moet echter vóór de analyse in een pH-gebied tussen 6 en 7 (met 0,5 mol/L-zwavelzuur of 1 mol/L-natriumhydroxideoplossing) worden geplaatst.

Aantekeningen

1. Evo-tabletten kunnen worden gebruikt als alternatief voor de overeenkomstige standaardtabletten (bv. DPD nr. 3 Evo in plaats van DPD nr. 3).

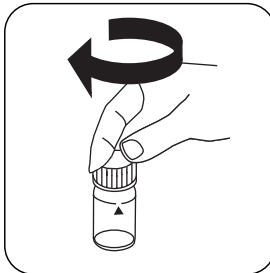


Uitvoering van de bepaling vrij chloor met tablet

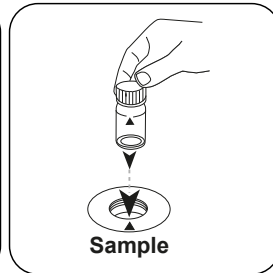
De methode in het apparaat selecteren.



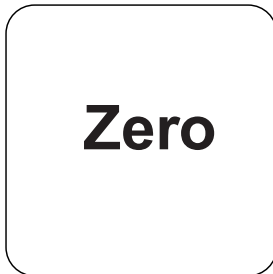
Spoelbakje van 24 mm met **10 mL staal** vullen.



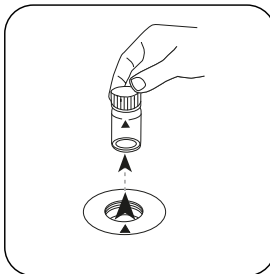
De spoelbakjes afsluiten.



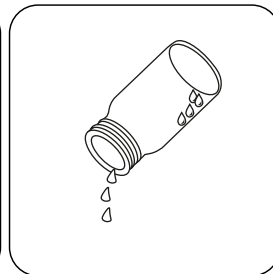
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



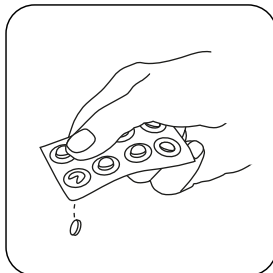
De toets **NUL** indrukken.



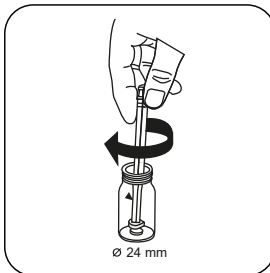
Het spoelbakje uit de meetschacht nemen.



Het spoelbakje tot op enkele druppels ledigen.



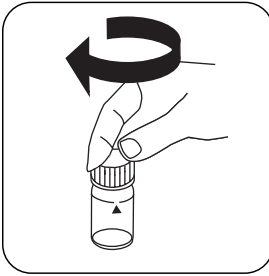
Een **DPD Nr. 1 tablet** toevoegen.



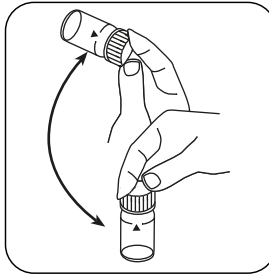
De tabletten onder lichte rotatie verpletteren.



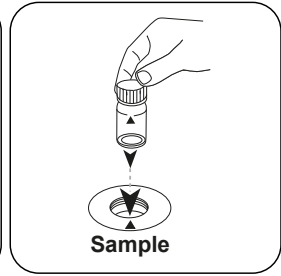
Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.



De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien



Het **staal**spoelbakje in de meetschacht plaatsen. Op de positionering letteren.

NL

Test

De toets **TEST** (XD: **START**) indrukken.

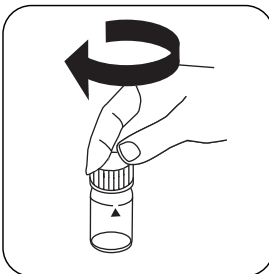
De display toont het resultaat in mg/L vrij chloor.

Uitvoering van de bepaling totaal chloor met tablet

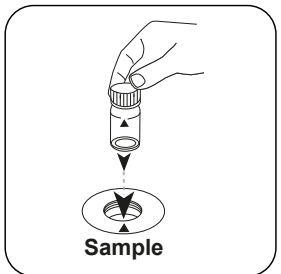
De methode in het apparaat selecteren.



Spoelbakje van 24 mm met **10 mL** staal vullen.



De spoelbakjes afsluiten.

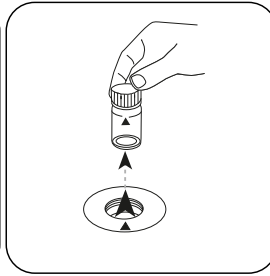


Het **staal**spoelbakje in de meetschacht plaatsen. Op de positionering letteren.

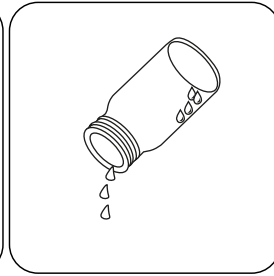


Zero

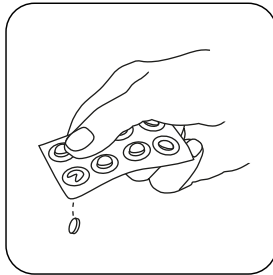
De toets **NUL** indrukken.



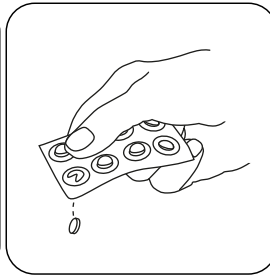
Het spoelbakje uit de meetschacht nemen.



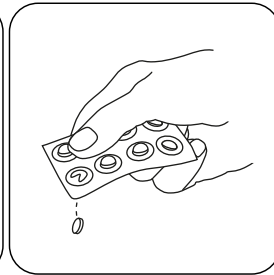
Het spoelbakje tot op enkele druppels ledigen.



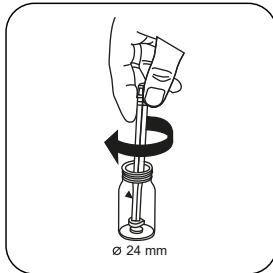
Een DPD Nr. 1 tablet toevoegen.



Een DPD Nr. 3 tablet toevoegen.



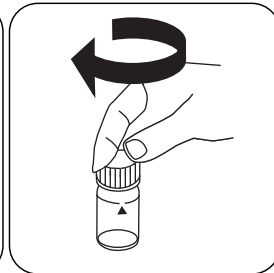
Als alternatief voor DPD nr. 1 en nr. 3 tabletten kan 1 DPD nr. 4 tablet worden toegevoegd.



De tabletten onder lichte rotatie verpletteren.



Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.



De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaats. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

NL



De reactietijd van **2 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L Totaal chloor.



Chemische methode

DPD

Aanhangsel

NL

Verstoringen

Permanente verstoringen

- Alle oxidatiemiddelen in de monsters reageren als chloor, wat tot extra resultaten leidt.

Uit te sluiten verstoringen

- Storingen veroorzaakt door koper en ijzer(III) worden door EDTA geëlimineerd.
- Bij monsters met een hoog calciumgehalte* en/of een hoge geleidbaarheid* kan het gebruik van reagenstabletten leiden tot vertroebeling van het monster en de daarmee samenhangende onjuiste meting. In dit geval zijn de reagenstabletten DPD-nr. 1 High Calcium en het reagenstablet DPD-nr. 3 High Calcium te gebruiken.
*exacte waarden kunnen niet worden gegeven omdat de troebelheidsvorming afhankelijk is van de aard en samenstelling van het monsterwater.
- Concentraties van meer dan 10 mg/L chloor, bij gebruik van tabletten, kunnen leiden tot resultaten binnen het meetbereik tot 0 mg/L. Als de chloorconcentratie te hoog is, moet het monster worden verdund met chloorvrij water. Voeg reagens toe aan 10 mL van het verdunde monster en herhaal de meting (plausibiliteitstest).

Verstoringen	verstoort vanaf
CrO ₄ ²⁻	0.01
MnO ₂	0.01

Validatie van de methodes

Aantoonbaarheidsgrens	0.02 mg/L
Bepaalbaarheidsgrens	0.06 mg/L
Einde meetbereik	6 mg/L
Gevoeligheid	2.05 mg/L / Abs
Betrouwbaarheidsgrenzen	0.04 mg/L
Standaardafwijking procedure	0.019 mg/L
Variatiecoëfficiënt procedure	0.87 %

Conform

EN ISO 7393-2



^{a)} bepaling van de vrije, gebonden, totaal mogelijke | ^{a)} hulpreagens, alternatief voor DPD-nr. 1 / nr. 3 in geval van troebelheid van het monster als gevolg van een hoog calciumionengehalte en/of een hoge geleidbaarheid



Chloor L

M101

0.02 - 4.0 mg/L Cl₂^{a)}

CL6

DPD

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
DPD 1 bufferoplossing, blauw flesje	15 mL	471010
DPD 1-bufferoplossing	100 mL	471011
DPD 1 bufferoplossing in verpakking van 6 stuks	1 St.	471016
DPD 1 reagensoplossing, groen flesje	15 mL	471020
DPD 1-reagensoplossing	100 mL	471021
DPD 1 reagensoplossing in verpakking van 6 stuks	1 St.	471026
DPD 3 oplossing, rood flesje	15 mL	471030
DPD 3 oplossing	100 mL	471031
DPD 3 oplossing in verpakking van 6 stuks	1 St.	471036
DPD reagentia set	1 St.	471056

Beschikbare standaarden

Omschrijving	Verpakkingseenheid	Bestelnr.
ValidCheck Chloor 1,5 mg/l	1 St.	48105510

Bemonstering

1. Tijdens de monstervoorbereiding moet worden vermeden dat het chloor wordt uitgestoten, bijvoorbeeld door pipetteren en schudden.
2. De analyse moet onmiddellijk na de bemonstering worden uitgevoerd.

Vorbereiding

1. Het schoonmaken van de spoelbakjes:
Aangezien veel huishoudelijke reinigingsmiddelen (bijv. afwasmiddelen) minder schadelijke stoffen bevatten, kan de bepaling van chloor leiden tot minder goede resultaten. Om deze meefout uit te sluiten, moeten de glasapparaten chloorvrij zijn. Hiertoe wordt het glaswerk gedurende één uur onder natriumhypochlorietoplossing (0,1 g/L) bewaard en vervolgens grondig gespoeld met gedeïoniseerd water.
2. Voor de individuele bepaling van vrij chloor en totaal chloor is het zinvol om een aparte set spoelbakjes te gebruiken (zie EN ISO 7393-2, paragraaf 5.3).
3. De DPD-kleurontwikkeling vindt plaats bij een pH-waarde van 6,2 tot 6,5. De reagentia bevatten daarom een buffer voor de aanpassing van de pH-waarde. Sterk alkalisch of zuur water moet echter vóór de analyse in een pH-gebied tussen 6 en 7 (met 0,5 mol/l-zwavelzuur of 1 mol/l-natriumhydroxideoplossing) worden geplaatst.

Aantekeningen

1. Na gebruik moeten de druppelflacons onmiddellijk worden gesloten met de schroefdop van dezelfde kleur.
2. Bewaar het reagens ingesteld op +6 °C tot +10 °C op een koele plaats.

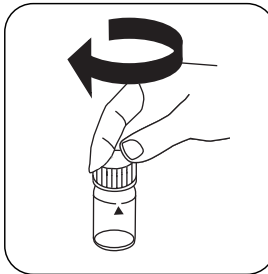


Uitvoering van de bepaling vrij chloor met vloeibaar reagens

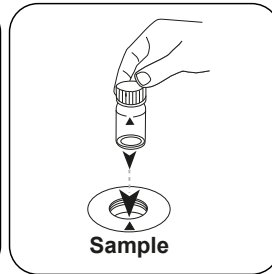
De methode in het apparaat selecteren.



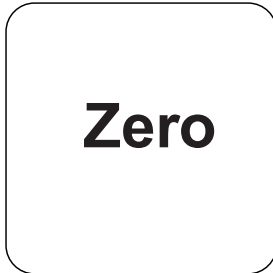
Spoelbakje van 24 mm met 10 mL staal vullen.



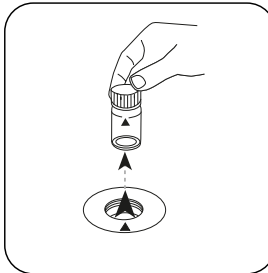
De spoelbakjes afsluiten.



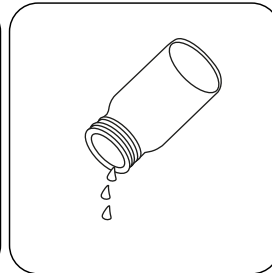
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



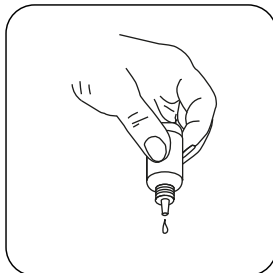
De toets **NUL** indrukken.



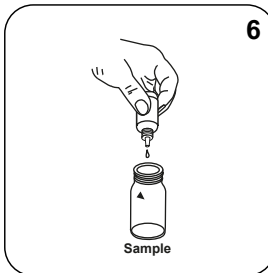
Het spoelbakje uit de meetschacht nemen.



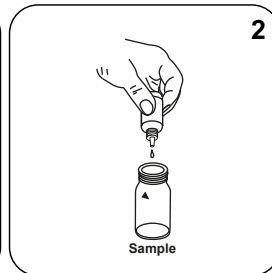
Het spoelbakje ledigen.



De druppelflessen verticaal houden en even grote druppels toevoegen door langzaam te drukken.



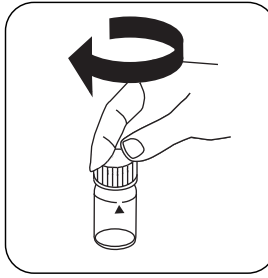
6 druppels DPD
1 bufferoplossing in het staalspoelbakje doen.



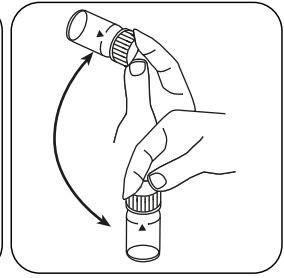
2 druppels DPD
1 reagensoplossing in het staalspoelbakje doen.



Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.

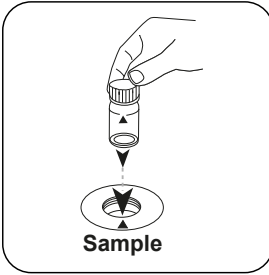


De spoelbakjes afsluiten.

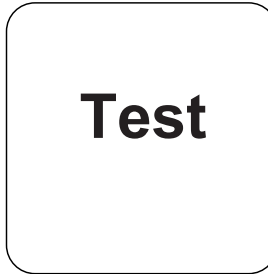


De inhoud mengen door om te draaien.

NL



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letteren.



De toets **TEST** (XD: **START**) indrukken.

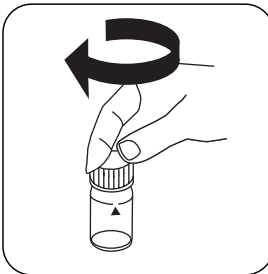
De display toont het resultaat in mg/L vrij chloor.

Uitvoering van de bepaling totaal chloor met vloeibaar reagens

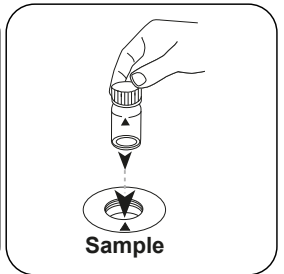
De methode in het apparaat selecteren.



Spoelbakje van 24 mm met **10 mL staal** vullen.



De spoelbakjes afsluiten.

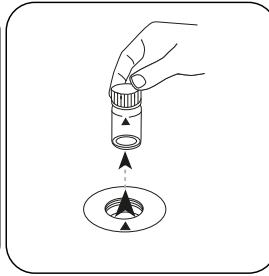


Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letteren.

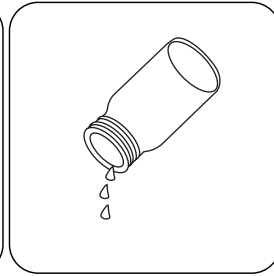


Zero

De toets **NUL** indrukken.



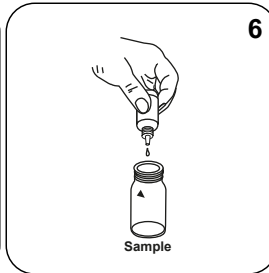
Het spoelbakje uit de meetschacht nemen.



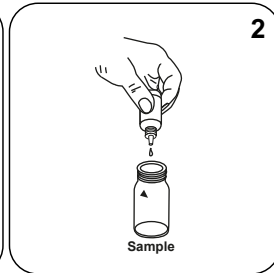
Het spoelbakje ledigen.



De druppelflessen verticaal houden en even grote druppels toevoegen door langzaam te drukken.



**6 druppels DPD
1 bufferoplossing** in het staal spoelbakje doen.



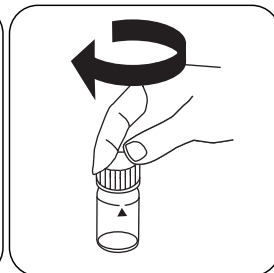
**2 druppels DPD
1 reagensoplossing** in het staal spoelbakje doen.



**3 druppels DPD
3 oplossing** in het staal spoelbakje doen.



Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.



De spoelbakjes afsluiten.



De inhoud mengen door om te draaien.



Het **staalspoelbakje** in de meetschacht plaats. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

NL



De reactietijd van **2 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L Totaal chloor.



Chemische methode

DPD

Aanhangsel

NL

Verstoringen

Permanente verstoringen

- Alle oxidatiemiddelen in de monsters reageren als chloor, wat tot extra resultaten leidt.

Uit te sluiten verstoringen

- Storingen veroorzaakt door koper en ijzer(III) worden door EDTA geëlimineerd.
- Concentraties van meer dan 4 mg/L chloor, bij gebruik van vloeibare reagentia, kunnen leiden tot resultaten binnen het meetbereik tot 0 mg/L. In dit geval moet het monster worden verdund met chloorvrij water. Voeg reagens toe aan 10 ml van het verdunde monster en herhaal de meting (plausibiliteitstest).

Verstoringen	verstoort vanaf
CrO_4^{2-}	0,01
MnO_2	0,01

Conform

EN ISO 7393-2

^{a)} bepaling van de vrije, gebonden, totaal mogelijke



Chloor HR T

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
DPD Nr. 1 HR	Tablet / 100	511500BT
DPD Nr. 1 HR	Tablet / 250	511501BT
DPD Nr. 1 HR	Tablet / 500	511502BT
DPD Nr. 3 HR	Tablet / 100	511590BT
DPD Nr. 3 HR	Tablet / 250	511591BT
DPD Nr. 3 HR	Tablet / 500	511592BT
Set DPD nr. 1 HR/nr. 3 HR #	per 100	517791BT
Set DPD nr. 1 HR/nr. 3 HR #	per 250	517792BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 100	515740BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 250	515741BT
DPD Nr. 1 hoog calcium ^{e)}	Tablet / 500	515742BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 100	515730BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 250	515731BT
DPD Nr. 3 hoog calcium ^{e)}	Tablet / 500	515732BT
DPD Nr.3 HR Evo	Tablet / 100	511920BT
DPD Nr. 3 HR Evo	Tablet / 250	511921BT
DPD Nr. 3 HR Evo	Tablet / 500	511922BT

Bemonstering

1. Tijdens de monstervoorbereiding moet worden vermeden dat het chloor wordt uitgestoten, bijvoorbeeld door pipetteren en schudden.
2. De analyse moet onmiddellijk na de bemonstering worden uitgevoerd.

Vorbereiding

1. Het schoonmaken van de spoelbakjes:
Aangezien veel huishoudelijke reinigingsmiddelen (bijv. afwasmiddelen) minder schadelijke stoffen bevatten, kan de bepaling van chloor leiden tot minder goede resultaten. Om deze meetfout uit te sluiten, moeten de glasapparaten chloorvrij zijn. Hiertoe wordt het glaswerk gedurende één uur onder natriumhypochlorietoplossing (0,1 g/L) bewaard en vervolgens grondig gespoeld met gedeïoniseerd water.
2. Voor de individuele bepaling van vrij chloor en totaal chloor is het zinvol om een aparte set spoelbakjes te gebruiken (zie EN ISO 7393-2, paragraaf 5.3).
3. De DPD-kleurontwikkeling vindt plaats bij een pH-waarde van 6,2 tot 6,5. De reagentia bevatten daarom een buffer voor de aanpassing van de pH-waarde. Sterk alkalisch of zuur water moet echter vóór de analyse in een pH-gebied tussen 6 en 7 (met 0,5 mol/L-zwavelzuur of 1 mol/L-natriumhydroxideoplossing) worden geplaatst.

Aantekeningen

1. Evo-tabletten kunnen worden gebruikt als alternatief voor de overeenkomstige standaardtabletten (bv. DPD nr. 3 Evo in plaats van DPD nr. 3).

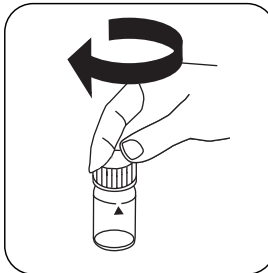


Uitvoering van de bepaling vrij chloor HR met tablet

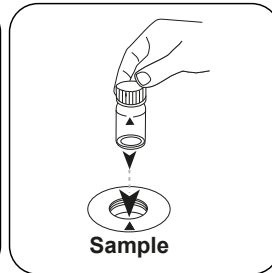
De methode in het apparaat selecteren.



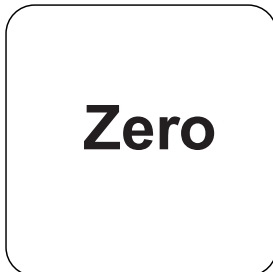
Spoelbakje van 24 mm met **10 mL staal** vullen.



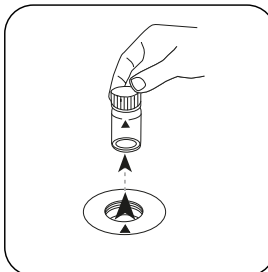
De spoelbakjes afsluiten.



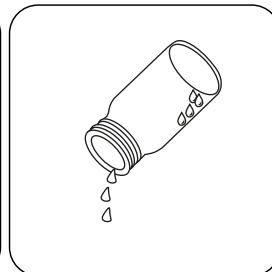
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



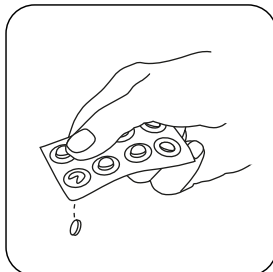
De toets **NUL** indrukken.



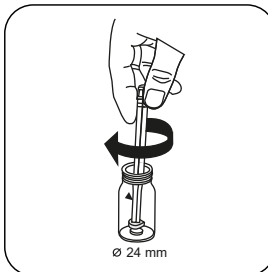
Het spoelbakje uit de meetschacht nemen.



Het spoelbakje tot op enkele druppels ledigen.



Een **DPD Nr. 1 HR tablet** toevoegen.



De tabletten onder lichte rotatie verpletteren.



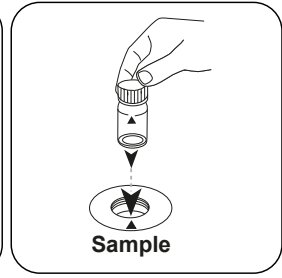
Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.



De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letteren.

NL

Test

De toets **TEST** (XD: **START**) indrukken.

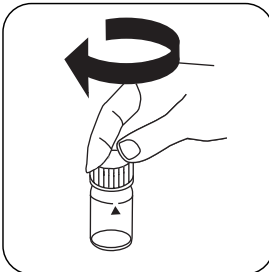
De display toont het resultaat in mg/L vrij chloor.

Uitvoering van de bepaling totaal chloor HR met tablet

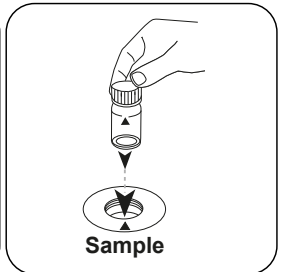
De methode in het apparaat selecteren.



Spoelbakje van 24 mm met **10 mL** staal vullen.



De spoelbakjes afsluiten.

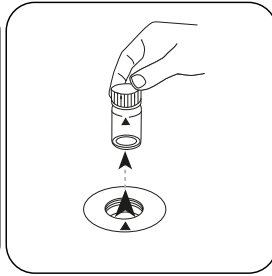


Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letteren.

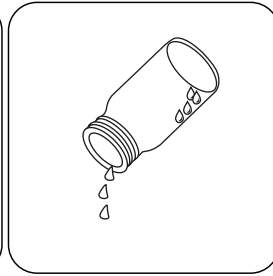


Zero

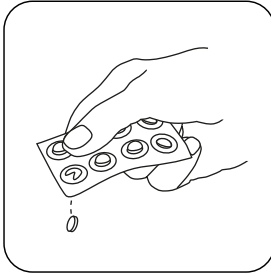
De toets **NUL** indrukken.



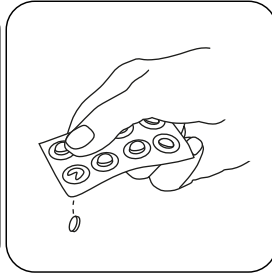
Het spoelbakje uit de meetschacht nemen.



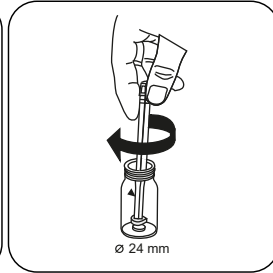
Het spoelbakje tot op enkele druppels ledigen.



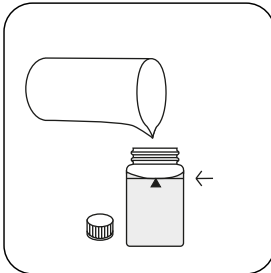
Een DPD Nr. 1 HR tablet toevoegen.



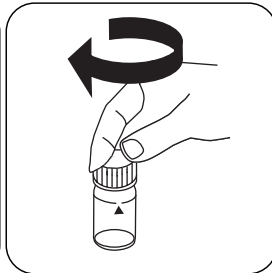
Een DPD Nr. 3 HR tablet toevoegen.



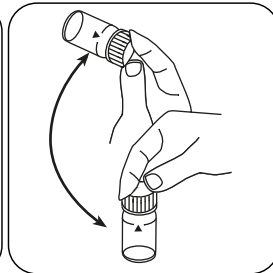
De tabletten onder lichte rotatie verpletteren.



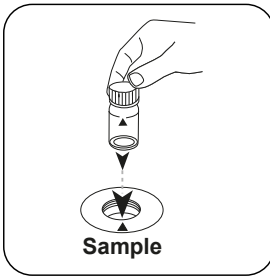
Het spoelbakje tot aan de **markering van 10 mL** met het **staal** vullen.



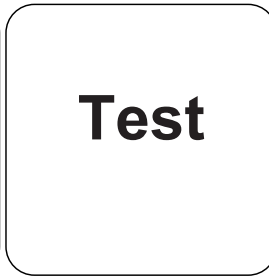
De spoelbakjes afsluiten.



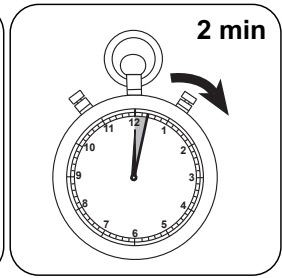
Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.



De reactietijd van **2 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L Totaal chloor.



Chemische methode

DPD

Aanhangsel

NL

Verstoringen

Permanente verstoringen

- Alle oxidatiemiddelen in de monsters reageren als chloor, wat tot extra resultaten leidt.

Uit te sluiten verstoringen

- Storingen veroorzaakt door koper en ijzer(III) worden door EDTA geëlimineerd.
- Als de reagenstabletten worden gebruikt voor monsters met een hoog calciumgehalte* en/of een hoge geleidbaarheid*, kan het monster troebel worden en kan de meting onjuist zijn. In dit geval is het DPD-nummer een alternatief. 1 High Calcium en het reagenstablet DPD-nr. 3 High Calcium te gebruiken.

*exacte waarden kunnen niet worden gegeven omdat de troebelheidsvorming afhankelijk is van de aard en samenstelling van het monsterwater.

Conform

EN ISO 7393-2

^{a)} bepaling van de vrije, gebonden, totaal mogelijke | ^{o)} hulpreagens, alternatief voor DPD-nr. 1 / nr. 3 in geval van troebelheid van het monster als gevolg van een hoog calciumionengehalte en/of een hoge geleidbaarheid | * met inbegrip van de mengstaaf



Koper T

M150

0.05 - 5 mg/L Cu^{a)}

Cu

Biquinoline

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
Koper Nr. 1	Tablet / 100	513550BT
Koper Nr. 1	Tablet / 250	513551BT
Koper Nr. 2	Tablet / 100	513560BT
Koper Nr. 2	Tablet / 250	513561BT
Set koper nr. 1/Nr. 2 [#]	per 100	517691BT
Set koper nr. 1/Nr. 2 [#]	per 250	517692BT
ValidCheck koper 2 mg/l	1 St.	48141525

Vorbereiding

1. Sterk alkalisch of zuur water moet vóór de analyse op een pH-waarde van 4 tot 6 worden ingesteld.

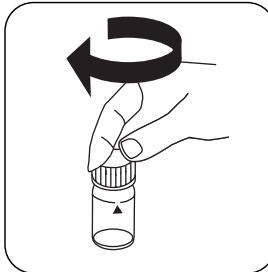
Uitvoering van de bepaling Koper, vrij met tablet

De methode in het apparaat selecteren.

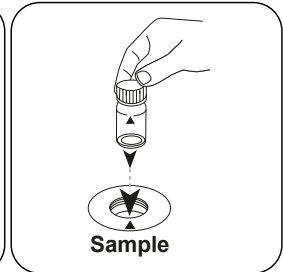
Selecteer bovendien de bepaling: vrij



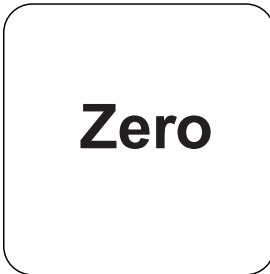
Spoelbakje van 24 mm met **10 mL staal** vullen.



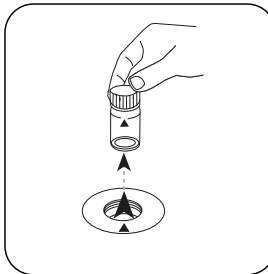
De spoelbakjes afsluiten.



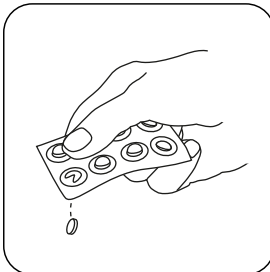
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letteren.



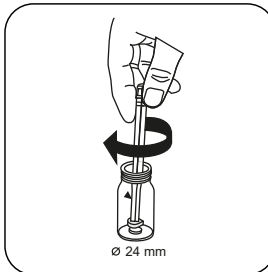
De toets **NUL** indrukken.



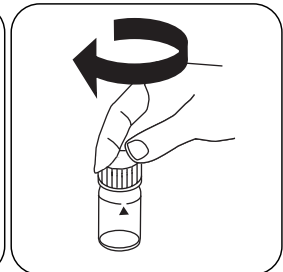
Het spoelbakje uit de meetschacht nemen.



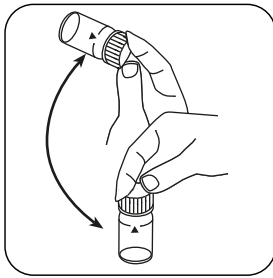
Een **COPPER Nr. 1** tablet toevoegen.



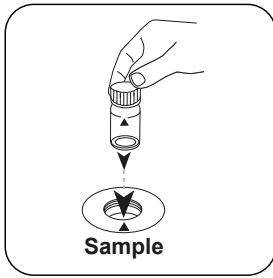
De tabletten onder lichte rotatie verpletteren.



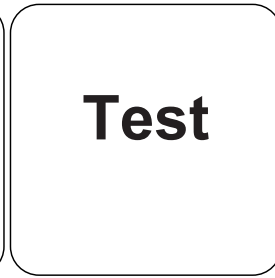
De spoelbakjes afsluiten.



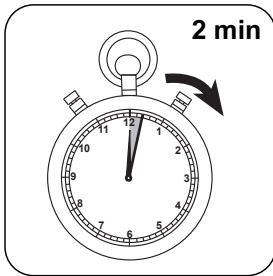
Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.



De reactietijd van **2 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L vrij koper.

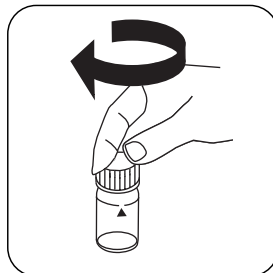
Uitvoering van de bepaling Koper, totaal met tablet

De methode in het apparaat selecteren.

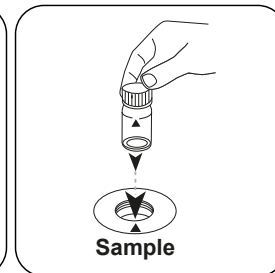
Selecteer bovendien de bepaling: totaal



Spoelbakje van 24 mm met **10 mL staal** vullen.



De spoelbakjes afsluiten.

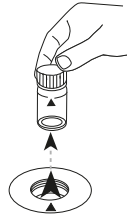


Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.

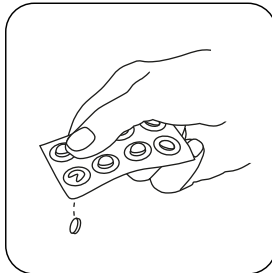


Zero

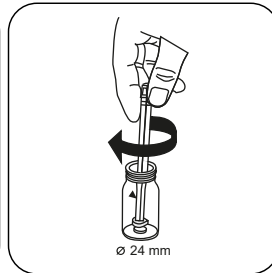
De toets **NUL** indrukken.



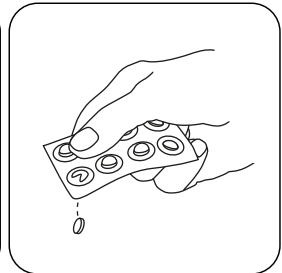
Het spoelbakje uit de meetschacht nemen.



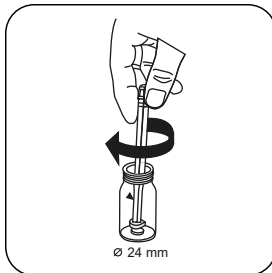
Een **COPPER Nr. 1** tablet toevoegen.



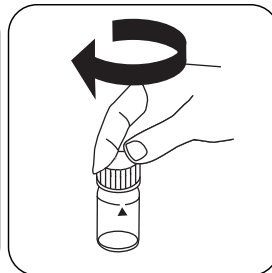
De tabletten onder lichte rotatie verpletteren en oplossen.



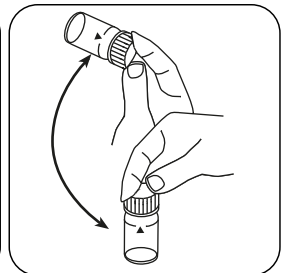
Een **COPPER Nr. 2** tablet toevoegen.



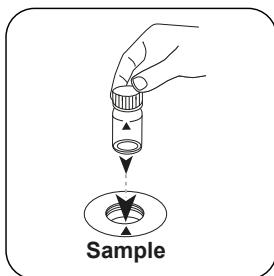
De tabletten onder lichte rotatie verpletteren.



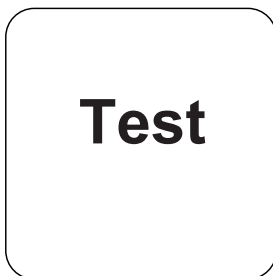
De spoelbakjes afsluiten.



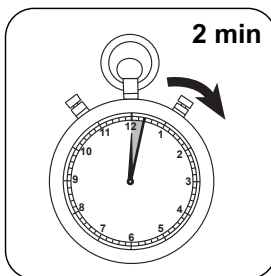
Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.



De reactietijd van **2 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L totaal koper.

Chemische methode

Biquinoline

Aanhangsel

Verstoringsen

Permanente verstoringen

1. Cyanide CN^- en Zilver Ag^+ beïnvloeden de bepaling.

Validatie van de methodes

Aantoonbaarheidsgrens	0.05 mg/L
Bepaalbaarheidsgrens	0.15 mg/L
Einde meetbereik	5 mg/L
Gevoeligheid	3.8 mg/L / Abs
Betrouwbaarheidsgrenzen	0.026 mg/L
Standaardafwijking procedure	0.011 mg/L
Variatiecoëfficiënt procedure	0.42 %

Literatuurverwijzing

Photometrische Analyse, Lange/Vedjelek, Verlag Chemie 1980

^{a)} bepaling van de vrije, gebonden, totaal mogelijke | ^{*} met inbegrip van de mengstaaf



CyA T

M160

10 - 160 mg/L CyA

CyA

Melamine

NL

Reagentia

Benodigd materiaal (deels optioneel):

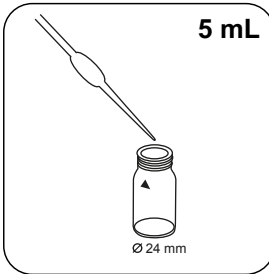
Reagentia	Verpakkingseenheid	Bestelnr.
CyA-test	Tablet / 100	511370BT
CyA-test	Tablet / 250	511371BT
VE-water	100 mL	461275
VE-water	250 mL	457022

Aantekeningen

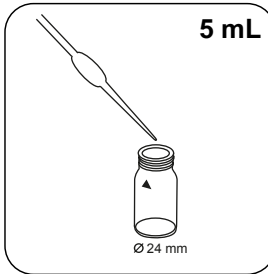
1. Cyanuurzuur veroorzaakt een zeer fijn verdeelde troebelheid met een melkachtig uiterlijk. Individuele deeltjes zijn niet te wijten aan de aanwezigheid van cyanuurzuur.

Uitvoering van de bepaling Cyanuurzuurtest met tablet

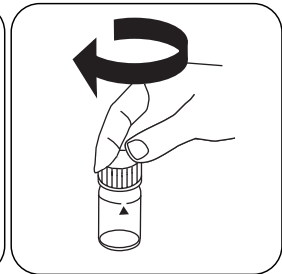
De methode in het apparaat selecteren.



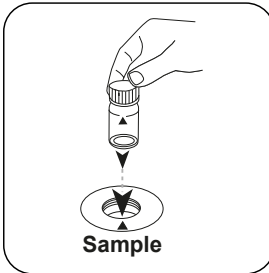
Spoelbakje van 24 mm met **5 mL gedeïoniseerd water** vullen.



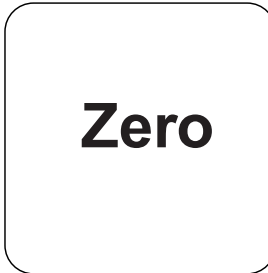
5 mL staal aan het spoelbakje toevoegen.



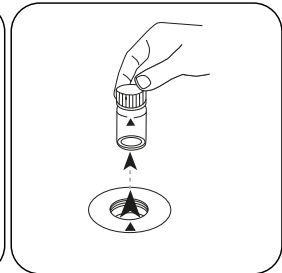
De spoelbakjes afsluiten.



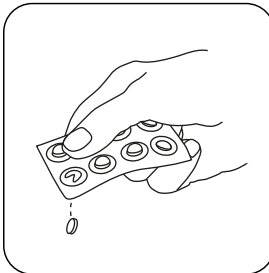
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



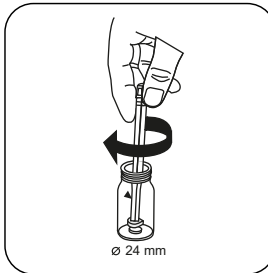
De toets **NUL** indrukken.



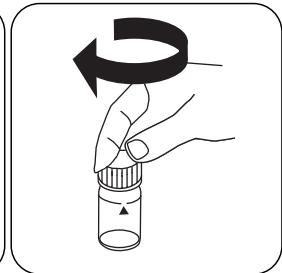
Het spoelbakje uit de meetschacht nemen.



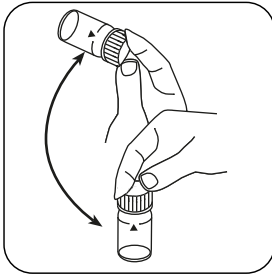
Een **CyA-test tablet** toevoegen.



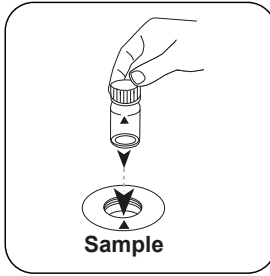
De tabletten onder lichte rotatie verpletteren.



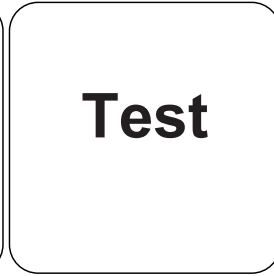
De spoelbakjes afsluiten.



De inhoud mengen door om te draaien (gedurende minstens 60 s tot de tablet volledig is opgelost).



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

De display toont het resultaat in mg/L Cyanuurzuur.

NL



Chemische methode

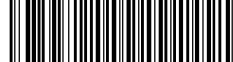
Melamine

Verstoringen

Permanente verstoringen

1. Onopgeloste deeltjes kunnen tot meerdere resultaten leiden. Daarom is het belangrijk om de tabletten volledig op te lossen.

NL



IJzer T

M220

0.02 - 1 mg/L Fe

FE

Ferrozine / Thioglycolaat

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
IJzer II LR (Fe ²⁺)	Tablet / 100	515420BT
IJzer II LR (Fe ²⁺)	Tablet / 250	515421BT
IJzer LR (Fe ²⁺ und Fe ³⁺)	Tablet / 100	515370BT
IJzer LR (Fe ²⁺ und Fe ³⁺)	Tablet / 250	515371BT

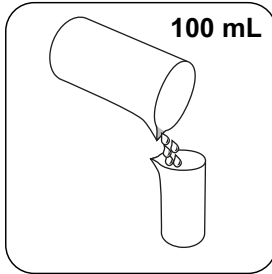
Voorbereiding

1. Water dat is behandeld met organische verbindingen als corrosiebescherming e.d. moet mogelijk worden geoxideerd worden om de ijzercomplexen te vernietigen. Hiertoe wordt een monster van 100 ml gemengd met 1 ml geconcentreerd zwavelzuur ($\geq 95\%$) en 1 ml geconcentreerd salpeterzuur ($\geq 65\%$) en op de helft ingedampd. Na afkoeling wordt de vertering uitgevoerd.

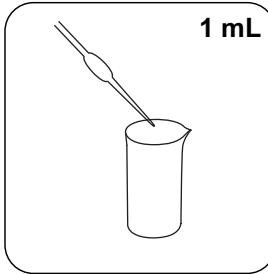
Aantekeningen

1. Deze methode bepaalt het totaal opgeloste Fe²⁺ en Fe³⁺.
2. Voor de bepaling van Fe²⁺ wordt het IRON (II) LR-tablet gebruikt in plaats van het IRON LR-tablet.

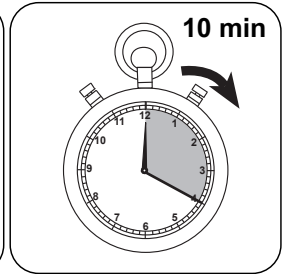
Ontsluiting



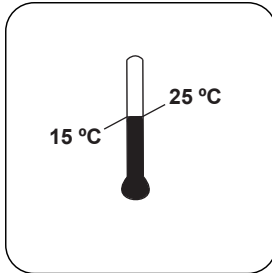
Een geschikte staalbeker met **100 mL** staal vullen.



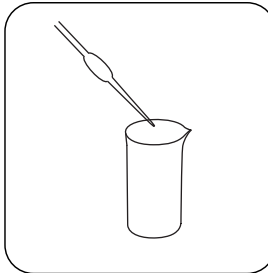
1 mL geconcentreerd zwavelzuur ($\geq 95\%$) toevoegen.



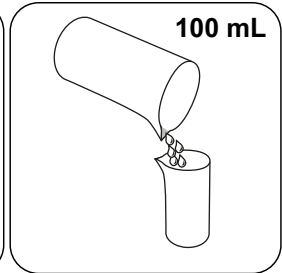
Het staal gedurende **10 minuten verwarmen**, of zolang tot alles volledig is opgelost.



Het staal laten afkoelen tot **kamertemperatuur**.



De **pH-waarde** van het staal met **Ammoniakoplossing (10-25 %)** afstellen op 3-5.



Het staal met **gedeïoniseerd water tot 100 mL** vullen.

Dit staal gebruiken voor de analyse van totaal opgelost en niet-opgelost ijzer.

Uitvoering van de bepaling IJzer (II,III), opgelost met tablet

De methode in het apparaat selecteren.

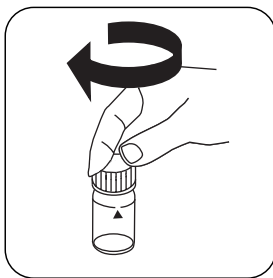
Voor de bepaling van **opgelost en niet-opgelost ijzer** de beschreven **ontsluiting** uitvoeren.



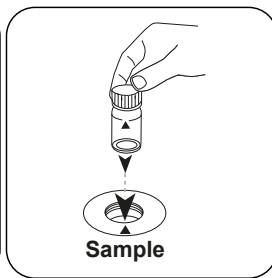
NL



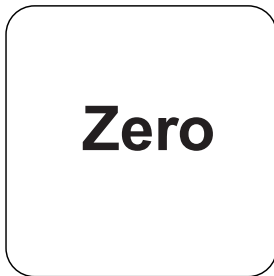
Spoelbakje van 24 mm met **10 mL staal** vullen.



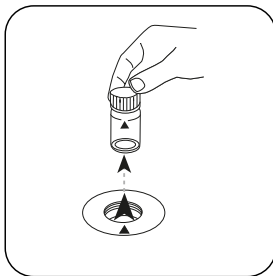
De spoelbakjes afsluiten.



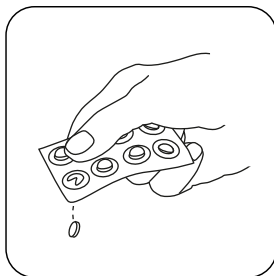
Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



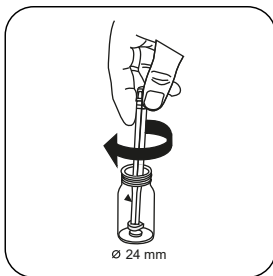
De toets **NUL** indrukken.



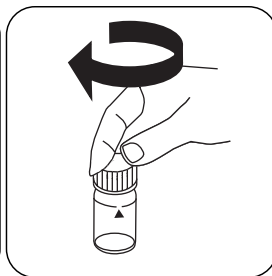
Het spoelbakje uit de meetschacht nemen.



Een **IRON LR tablet** toevoegen.



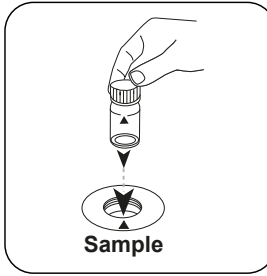
De tabletten onder lichte rotatie verpletteren.



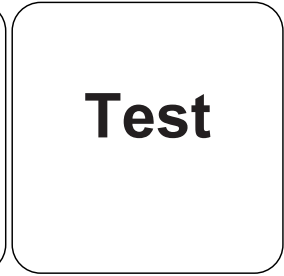
De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien

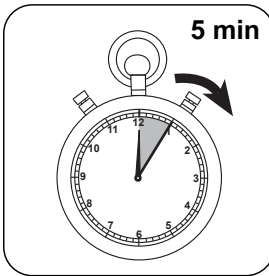


Het **staalspoelbakje** in de meetschacht plaats. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

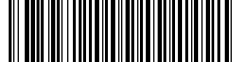
NL



De reactietijd van **5 minuten** afwachten.

Na afloop van de reactietijd wordt de meting automatisch uitgevoerd.

De display toont het resultaat in mg/L IJzer.



Chemische methode

Ferrozine / Thioglycolaat

Aanhangsel

NL

Verstoringsen

Uit te sluiten verstoringen

1. De aanwezigheid van koper verhoogt het meetresultaat met 10 %. Bij een concentratie van 10 mg/L koper in het monster wordt het resultaat verhoogd met 1 mg/L ijzer.
De verstoring kan worden geëlimineerd door toevoeging van thiourem

Validatie van de methodes

Aantoonbaarheidsgrens	0.01 mg/L
Bepaalbaarheidsgrens	0.016 mg/L
Einde meetbereik	1 mg/L
Gevoeligheid	0.92 mg/L / Abs
Betrouwbaarheidsgrenzen	0.013 mg/L
Standaardafwijking procedure	0.005 mg/L
Variatiecoëfficiënt procedure	1.23 %

Literatuurverwijzing

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



pH-waarde T

M330

6.5 - 8.4 pH

PH

Fenolrood

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
Fenolrood fotometer	Tablet / 100	511770BT
Fenolrood fotometer	Tablet / 250	511771BT
Fenolrood fotometer	Tablet / 500	511772BT

Aantekeningen

1. Voor de fotometrische pH-bepaling mogen alleen PHENOL RED-tabletten met een zwarte foliedruk en de term PHOTOMETER worden gebruikt.

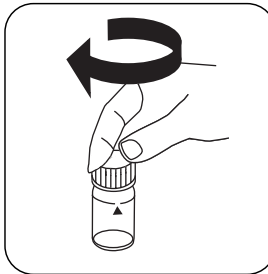


Uitvoering van de bepaling pH-waarde met tablet

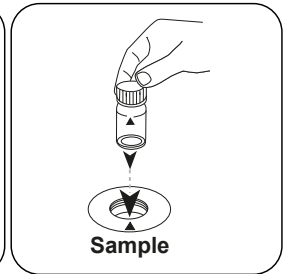
De methode in het apparaat selecteren.



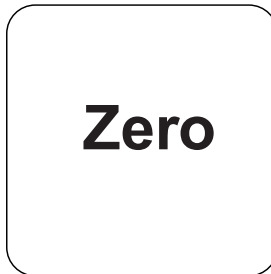
Spoelbakje van 24 mm met 10 mL staal vullen.



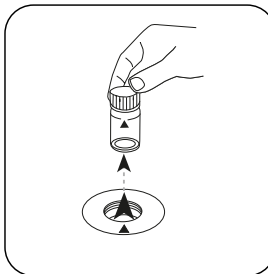
De spoelbakjes afsluiten.



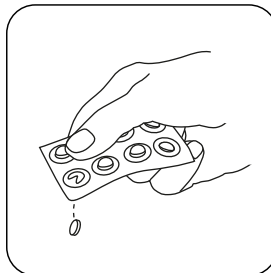
Het **staal**spoelbakje in de meetschacht plaatsen. Op de positionering letten.



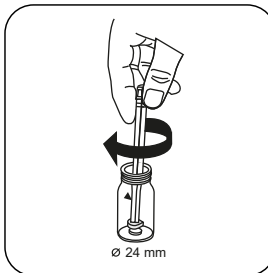
De toets **NUL** indrukken.



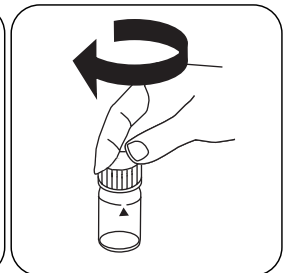
Het spoelbakje uit de meetschacht nemen.



Een **FENOLROOD FOTOMETER** tablet toevoegen.



De tabletten onder lichte rotatie verpletteren.



De spoelbakjes afsluiten.



Tabletten oplossen door om te draaien



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

De display toont het resultaat als pH-waarde.

NL

Chemische methode

Fenolrood

Aanhangsel

Verstoringsen

NL

Permanente verstoringen

1. Watermonsters met een lage carbonaathardheid* kunnen leiden tot onjuiste pH-waarden.

* $_{\text{K}84,3} < 0,7 \text{ mmol/l} \triangleq \text{Totale alkaliteit} < 35 \text{ mg/L CaCO}_3$.

Uit te sluiten verstoringen

1. pH-waarden onder 6,5 en boven 8,4 kunnen leiden tot resultaten binnen het meetbereik. Een plausibiliteitstest (pH-meter) wordt aanbevolen.
2. Zoutgebrek:
Voor zoutgehalten tot 2 g/L kan geen significante zoutfout worden verwacht als gevolg van het zoutgehalte van het reagenstablet. Indien het zoutgehalte hoger is, worden de gemeten waarden als volgt gecorrigeerd:

zoutgehalte van het monster (in g/L)	30 (zeewater)	60	120	180
Correctie	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾ na Kolthoff (1922)

²⁾ na Parson en Douglas (1926)

Literatuurverwijzing

Colorimetric Chemical Analytical Methods, 9th Edition, London



pH-waarde L

M331

6.5 - 8.4 pH

PH

Fenolrood

NL

Reagentia

Benodigd materiaal (deels optioneel):

Reagentia	Verpakkingseenheid	Bestelnr.
Fenolrood oplossing	15 mL	471040
Fenolrood oplossing	100 mL	471041
Fenolrood oplossing in verpakking van 6 stuks	1 St.	471046

Vorbereiding

- Door de verschillende druppelgroottes kan het meetresultaat grotere afwijkingen vertonen dan bij gebruik van tabletten.
Bij gebruik van een pipet (0,18 ml komt overeen met 6 druppels) kan deze afwijking worden geminimaliseerd.

Aantekeningen

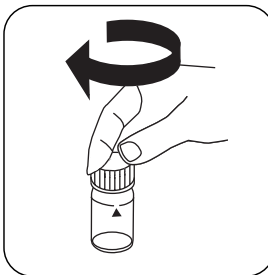
- Na gebruik moet de druppelfles meteen onmiddellijk worden gesloten met de schroefdop van dezelfde kleur.
- Bewaar het reagens bij +6 °C tot +10 °C op een koele plaats.

Uitvoering van de bepaling pH-waarde met vloeibaar reagens

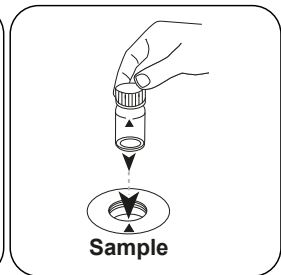
De methode in het apparaat selecteren.



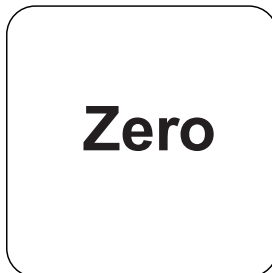
Spoelbakje van 24 mm met **10 mL staal** vullen.



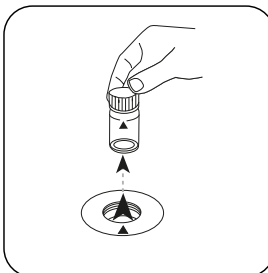
De spoelbakjes afsluiten.



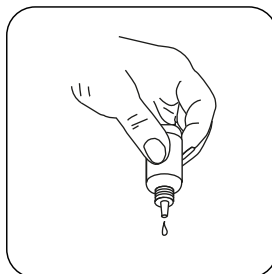
Het **staal spoelbakje** in de meetschacht plaatsen. Op de positionering letten.



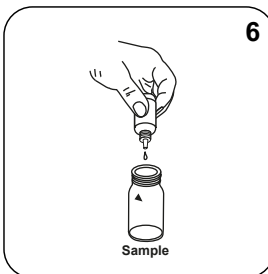
De toets **NUL** indrukken.



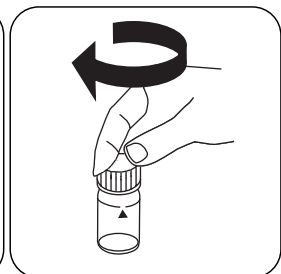
Het spoelbakje uit de meetschacht nemen.



De druppelflessen verticaal houden en even grote druppels toevoegen door langzaam te drukken.



6 druppels FENOLROOD-oplossing in het staal spoelbakje doen.



De spoelbakjes afsluiten.



De inhoud mengen door om te draaien.



Het **staalspoelbakje** in de meetschacht plaatsen. Op de positionering letten.



De toets **TEST** (XD: **START**) indrukken.

De display toont het resultaat als pH-waarde.

NL

Chemische methode

Fenolrood

Aanhangsel

Verstoringen

NL

Uit te sluiten verstoringen


1. Zoutgebrek: correctie van de gemeten waarde (gemiddelde waarden) voor monsters met een zoutgehalte van:

2. Zoutgehalte van het monster	Correctie
30 g/L (zeewater)	-0,15 ¹⁾
60 g/L	-0,21 ²⁾
120 g/L	-0,26 ²⁾
180 g/L	-0,29 ²⁾
¹⁾ na Kolthoff (1922)	²⁾ na Parson en Douglas (1926)

3. Bij het testen van gechloreerd water kan het aanwezige chloorgehalte de kleurreactie van het vloeibare reagens beïnvloeden. Dit wordt voorkomen door een klein kristal natriumthiosulfaat ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) aan de monsteroplossing toe te voegen voordat de PHENOL RED-oplossing wordt toegevoegd.

Literatuurverwijzing

Colorimetric Chemical Analytical Methods, 9th Edition, London

KS4.3 T / 20


方法名称

方法号

用于方法检测的条形码

测量范围

酸性 / 指示剂

屏幕显示: MD 100 / MD 110 / MD 200

化学方法

语言代码ISO 639-1

修订状态

仪器的具體信息

測試可以在以下設備上執行。此外還指出了所需的比色杯和光度計的吸收範圍。

儀器類型	比色皿	λ	測量範圍
MD 200, MD 600, MD 610, MD 640, MultiDirect, PM 620, PM 630	\varnothing 24 mm	610 nm	0.1 - 4 mmol/l $K_{S4.3}$
SpectroDirect, XD 7000, XD 7500	\varnothing 24 mm	615 nm	0.1 - 4 mmol/l $K_{S4.3}$

材料

所需材料 (部分可選) :

標題	包裝單位	貨號
Alka-M-Photometer	片劑 / 100	513210BT
Alka-M-Photometer	片劑 / 250	513211BT

應用列表

- 污水處理
- 飲用水處理
- 原水處理

備註

1. 術語總度-m、m-值、總碱度和酸容量 $K_{S4.3}$ 是相同的。
2. 準確地遵守 10 ml 的樣本體積對分析結果的準確度至關重要。

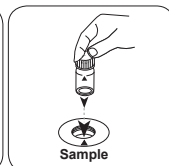
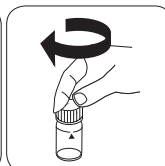
CN 方法手冊 01/20

开始测量

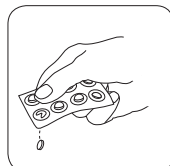
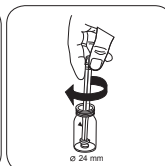
进行测定 $K_{s4.3}$ 片剂酸容量

选择设备中的方法。

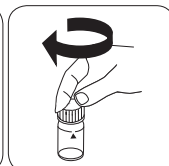
对于这种方法，在以下设备上不能进行 ZERO 测量：XD 7000, XD 7500

用 10 ml 样本填充 24 mm 比密封比色杯。
色杯。将样本比色杯放入测量轴
中。注意定位。

• • •

加入 ALKA-M-PHOTOME-
TER 片剂。

用轻微的扭转压碎片剂。



密封比色杯。

CN 方法手册 01/20

ZH



T 碱度-m

M30

5 - 200 mg/L CaCO₃

tA

酸性 / 指示剂

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
碱度 M 光度计	片剂 / 100	513210BT
碱度 M 光度计	片剂 / 250	513211BT

备注

1. 术语碱度-m、m-值、总碱度和酸容量 $K_{S4.3}$ 是相同的。
2. 准确地遵守 10 ml 的样本体积对分析结果的准确度至关重要。

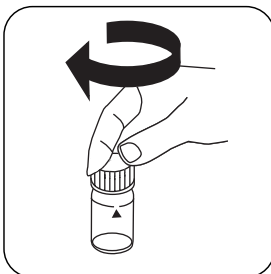


进行测定 总碱度 = 碱度 M = 片剂的 m-值

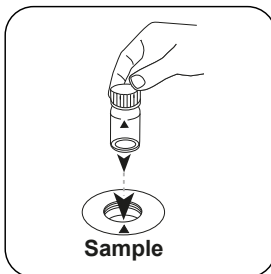
选择设备中的方法。



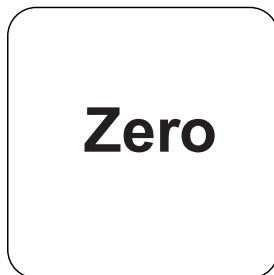
用 10 mL 样本填充 24 mm 比色杯。



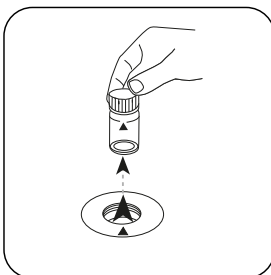
密封比色杯。



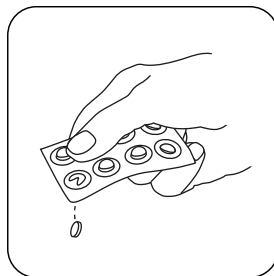
将样本比色杯放入测量轴中。注意定位。



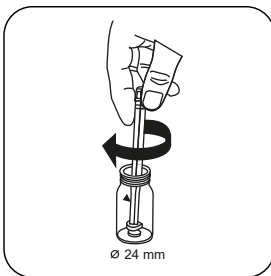
按下 ZERO 按钮。



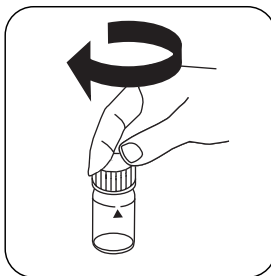
从测量轴上取下比色杯。



加入 ALKA-M-PHOTOMETER 片剂。

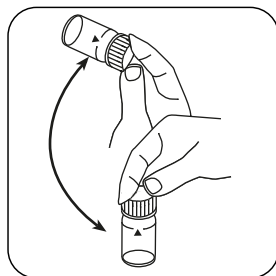


用轻微的扭转压碎片剂。

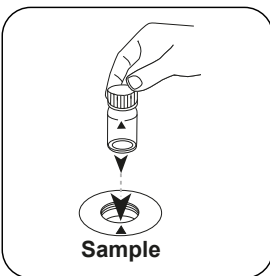


密封比色杯。

ZH



通过旋转溶解试剂。



将样本比色杯放入测量轴中。注意定位。



按下 **TEST (XD: START)** 按钮。

结果在显示屏上显示为 碱度-m。

ZH

分析

下表中输出数据也可转换为其他格式表示.

单位	参考表格	因素
mg/l	CaCO ₃	1
	°dH	0.056
	°eH	0.07
	°fH	0.1
	°aH	0.058
	K _{S4.3}	0.02

ZH

化学方法

酸性 / 指示剂

附录

源于

EN ISO 9963-1



T 氯

M100

0.01 - 6.0 mg/L Cl₂^{a)}

CL6

DPD

材料

所需材料 (部分可選) :

ZH

试剂	包装单位	货号
DPD No.1	片剂 / 100	511050BT
DPD No.1	片剂 / 250	511051BT
DPD No.1	片剂 / 500	511052BT
DPD No.3	片剂 / 100	511080BT
DPD No.3	片剂 / 250	511081BT
DPD No.3	片剂 / 500	511082BT
DPD No.1 高钙 ^{e)}	片剂 / 100	515740BT
DPD No.1 高钙 ^{e)}	片剂 / 250	515741BT
DPD No.1 高钙 ^{e)}	片剂 / 500	515742BT
DPD No.3 高钙 ^{e)}	片剂 / 100	515730BT
DPD No.3 高钙 ^{e)}	片剂 / 250	515731BT
DPD No.3 高钙 ^{e)}	片剂 / 500	515732BT
DPD No.4	片剂 / 100	511220BT
DPD No.4	片剂 / 250	511221BT
DPD No.4	片剂 / 500	511222BT
DPD No.3 Evo	片剂 / 100	511420BT
DPD No.3 Evo	片剂 / 250	511421BT
DPD No.3 Evo	片剂 / 500	511422BT
DPD No.4 Evo	片剂 / 100	511970BT
DPD No.4 Evo	片剂 / 250	511971BT
DPD No.4 Evo	片剂 / 500	511972BT

現有標準

标题	包装单位	货号
ValidCheck 氯 1.5 mg/l	1 片	48105510

取样

1. 在样本制备中，通过移液和摇动来避免氯的排气。
2. 取样后必须立即进行分析。

准备

1. 清洗比色杯：
由于许多家用清洁剂（例如洗碗用洗涤剂）含有还原剂，所以测定的氯结果可能会不足。为了排除这种测量误差，玻璃器皿应无氯。为此，将玻璃器皿在次氯酸钠溶液（0.1 g/L）下存放 1 小时，然后用去离子水（软化水）彻底冲洗。
2. 对于游离氯和总氯的单独测定，使用一套相应单独的比色杯是有意义的（参见 EN ISO 7393-2，第 5.3 段）。
3. DPD 显色发生在 pH 值在 6.2 至 6.5 时。因此该试剂含有用于调节 pH 值的缓冲液。但在分析前（用 0.5 mol/L 硫酸或 1 mol/L 氢氧化钠溶液）必须将强碱性或酸性水的 pH 范围调节到 6 和 7 之间。

备注

1. Evo 片剂可以作为相应标准片剂的替代品（如 DPD No.3 Evo 代替 DPD No.3）。



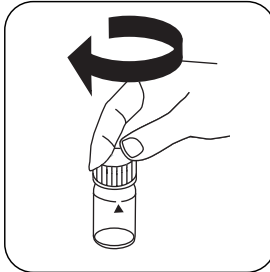
进行测定 余氯 片剂法

选择设备中的方法。

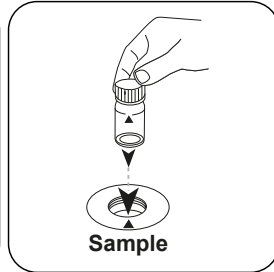
ZH



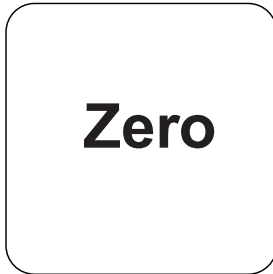
用 **10 mL** 样本填充 24 mm 比色杯。



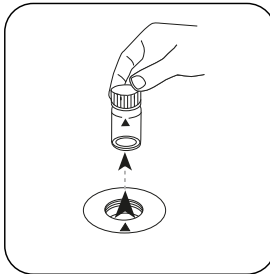
密封比色杯。



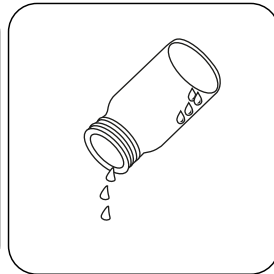
将样本比色杯放入测量轴中。
注意定位。



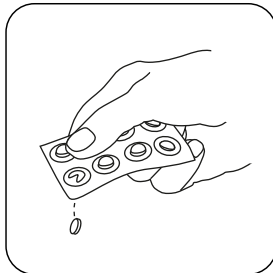
按下 **ZERO** 按钮。



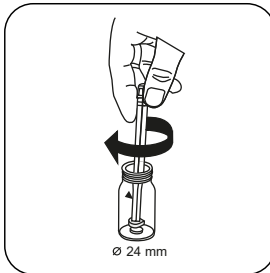
从测量轴上取下比色杯。



将比色杯倒空。



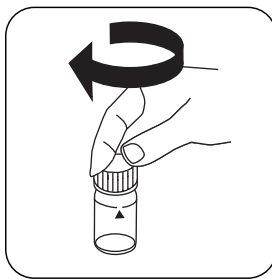
加入 **DPD No. 1** 片剂。



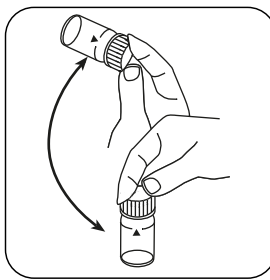
用轻微的扭转压碎片剂。



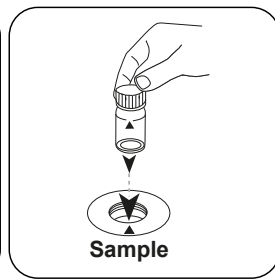
用样本将比色杯填充至
10 mL 刻度处。



密封比色杯。



通过旋转溶解片剂。

将样本比色杯放入测量轴中。
注意定位。

ZH

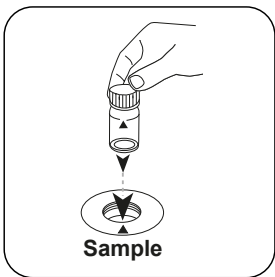
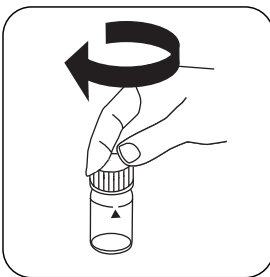
Test

按下 **TEST (XD: START)** 按钮。

结果在显示屏上显示为 mg / l 余氯。

进行测定 总氯 片剂法

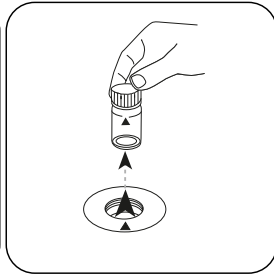
选择设备中的方法。

用 **10 mL** 样本填充 24 mm 比色杯。
密封比色杯。将样本比色杯放入测量轴中。
注意定位。



Zero

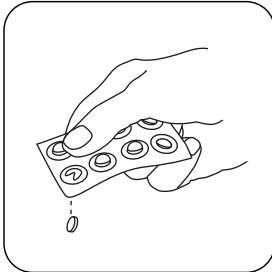
按下 **ZERO** 按钮。



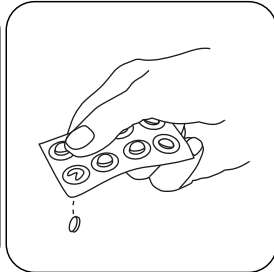
从测量轴上取下比色杯。



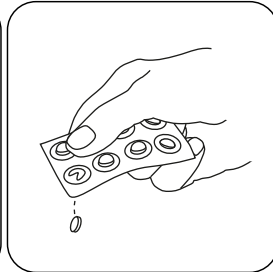
将比色杯倒空。



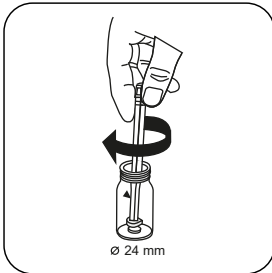
加入 **DPD No. 1** 片剂。



加入 **DPD No. 3** 片剂。



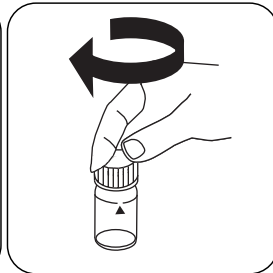
作为 DPD 1号和3号片剂的替代品，可以添加1个DPD 4号片剂。



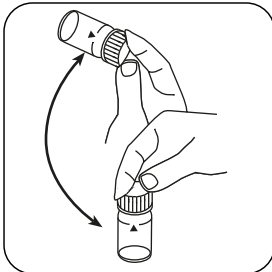
用轻微的扭转压碎片剂。



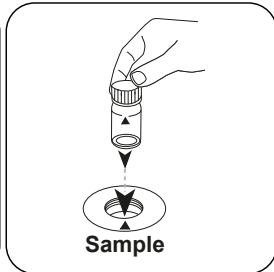
用样本将比色杯填充至 **10 mL** 刻度处。



密封比色杯。



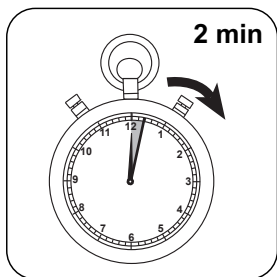
通过旋转溶解片剂。



将样本比色杯放入测量轴中。注意定位。

Test

按下 **TEST (XD: START)** 按钮。



等待 **2 分钟** 反应时间。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 mg / l 总氯。



化学方法

DPD

附录

ZH

干扰说明

持续干扰

- 存在于样本中的所有氧化剂都像氯一样反应，导致多重结果。

可消除干扰

- 铜和铁 (III) 的干扰必须通过 EDTA 消除。
- 对于高钙含量*和/或高电导率*的样本，使用试剂片可能会导致样本浑浊和相关的测量误差。在这种情况下，可选用试剂片 DPD 编号1 高钙和试剂片 DPD 编号3 高钙。
*不能给出精确值，因为浑浊的形成取决于样本水的类型和组成。
- 在使用片剂时，高于 10 mg/L 氯的浓度可导致测量范围内的结果高达 0 mg/L。氯浓度过高时应用无氯水稀释样本。将 10 mL 稀释的样本与试剂混合并重复测量 (可信度测试) 。

干扰	限/ [mg/l]
CrO ₄ ²⁻	0.01
MnO ₂	0.01

方法验证

检出限	0.02 mg/L
测定下限	0.06 mg/L
测量上限	6 mg/L
灵敏度	2.05 mg/L / Abs
置信范围	0.04 mg/L
标准偏差	0.019 mg/L
变异系数	0.87 %

一致性

EN ISO 7393-2

* 测定余氯，总氯和结合氯 | * 替代试剂，取代 DPD No. 1/No. 3 试剂，用于由高浓度钙离子和/或高电导率引起的浑浊水样分析



L 氯

M101

0.02 - 4.0 mg/L Cl₂^{a)}

CL6

DPD

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
DPD 1 缓冲溶液, 蓝瓶	15 mL	471010
DPD 1 缓冲溶液	100 mL	471011
DPD 1 缓冲溶液, 6 件装	1 片	471016
DPD 1 试剂溶液, 绿瓶	15 mL	471020
DPD 1 试剂溶液	100 mL	471021
DPD 1 试剂溶液, 6 件装	1 片	471026
DPD 3 溶液, 红瓶	15 mL	471030
DPD 3 溶液	100 mL	471031
DPD 3 溶液, 6 件装	1 片	471036
DPD 试剂套件	1 片	471056

現有標準

标题	包装单位	货号
ValidCheck 氯 1.5 mg/l	1 片	48105510

取样

1. 在样本制备中, 通过移液和摇动来避免氯的排气。
2. 取样后必须立即进行分析。

准备

1. 清洗比色杯 :
由于许多家用清洁剂 (例如洗碗用洗涤剂) 含有还原剂, 所以测定的氯结果可能会不足。为了排除这种测量误差, 玻璃器皿应无氯。为此, 将玻璃器皿在次氯酸钠溶液 (0.1 g/L) 下存放 1 小时, 然后用去离子水 (软化水) 彻底冲洗。
2. 对于游离氯和总氯的单独测定, 使用一套相应单独的比色杯是有意义的 (参见 EN ISO 7393-2, 第 5.3 段)。
3. DPD 显色发生在 pH 值在 6.2 至 6.5 时。因此该试剂含有用于调节 pH 值的缓冲液。但在分析前 (用 0.5 mol/l 硫酸或 1 mol/l 氢氧化钠溶液) 必须将强碱性或酸性水的 pH 范围调节到 6 和 7 之间。



备注

1. 使用后滴瓶必须立即用相同颜色的瓶盖重新密封。
2. 将试剂盒冷藏在 +6 °C至 + 10 °C。



进行测定 余氯 水剂法

选择设备中的方法。

ZH



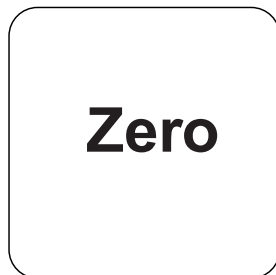
用 **10 mL** 样本填充 24 mm 比色杯。



密封比色杯。



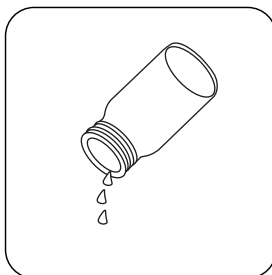
将样本比色杯放入测量轴中。
注意定位。



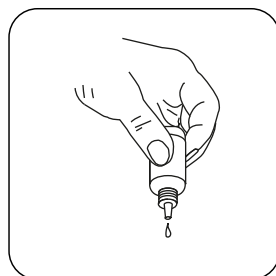
按下 **ZERO** 按钮。



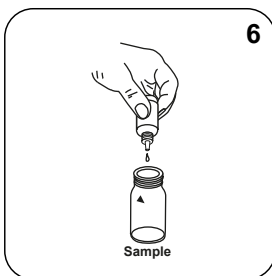
从测量轴上取下比色杯。



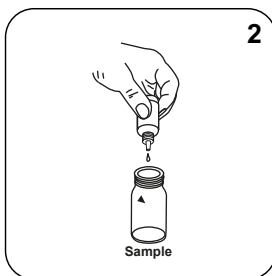
倒空比色杯。



垂直握住滴瓶，慢慢加入相同大小的滴剂。



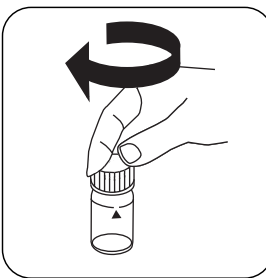
将 **6 滴 DPD 1 Buffer Solution** 添加到样本比色杯中。



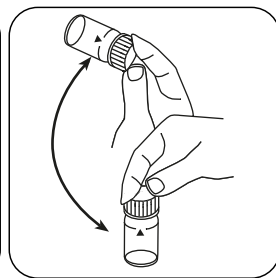
将 **2 滴 DPD 1 Reagent Solution** 添加到样本比色杯中。



用样本将比色杯填充至
10 mL 刻度处。

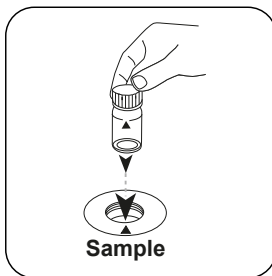


密封比色杯。

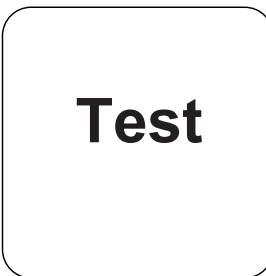


通过旋转混合内容物。

ZH



将样本比色杯放入测量轴
中。注意定位。



按下 **TEST (XD: START)** 按钮。

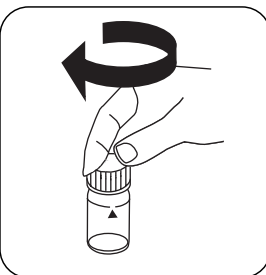
结果在显示屏上显示为 mg / l 余氯。

进行测定 总氯 水剂法

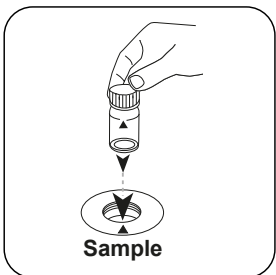
选择设备中的方法。



用 **10 mL** 样本填充 24 mm
比色杯。



密封比色杯。



将样本比色杯放入测量轴
中。注意定位。



Zero

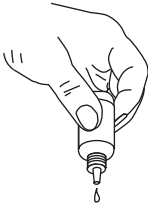
按下 **ZERO** 按钮。



从测量轴上取下比色杯。



倒空比色杯。



垂直握住滴瓶，慢慢加入相同大小的滴剂。



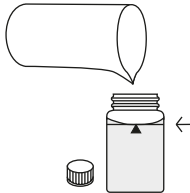
将 **6 滴 DPD 1 Buffer Solution** 添加到样本比色杯中。



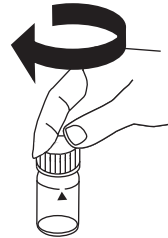
将 **2 滴 DPD 1 Reagent Solution** 添加到样本比色杯中。



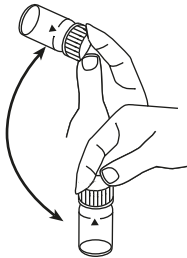
将 **3 滴 DPD 3 Solution** 添加到样本比色杯中。



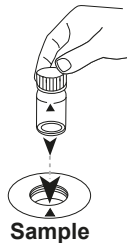
用样本将比色杯填充至 **10 mL 刻度处**。



密封比色杯。



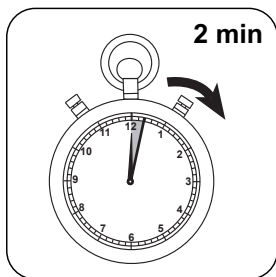
通过旋转混合内容物。



将样本比色杯放入测量轴中。注意定位。

Test

按下 **TEST (XD: START)** 按钮。



等待 2 分钟反应时间。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 mg / l 总氯。



化学方法

DPD

附錄

ZH

干扰说明

持续干扰

- 存在于样本中的所有氧化剂都像氯一样反应，导致多重结果。

可消除干扰

- 铜和铁 (III) 的干扰必须通过 EDTA 消除。
- 在使用液剂时，高于 4 mg/L 氯的浓度可导致测量范围内的结果高达 0 mg/L。在这种情况下应用无氯水稀释样本。将 10 ml 稀释的样本与试剂混合并重复测量 (可信度测试)。

干擾	從/ [mg/l]
CrO_4^{2-}	0,01
MnO_2	0,01

一致性

EN ISO 7393-2

^{a)} 测定余氯，总氯和结合氯



HR T 氯

M103

0.1 - 10 mg/L Cl₂^{a)}

CL10

DPD

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
DPD No.1 HR	片剂 / 100	511500BT
DPD No.1 HR	片剂 / 250	511501BT
DPD No.1 HR	片剂 / 500	511502BT
DPD No.3 HR	片剂 / 100	511590BT
DPD No.3 HR	片剂 / 250	511591BT
DPD No.3 HR	片剂 / 500	511592BT
套件 DPD No.1 HR/No.3 HR [#]	各100次	517791BT
套件 DPD No.1 HR/No.3 HR [#]	各250次	517792BT
DPD No.1 高钙 ^{e)}	片剂 / 100	515740BT
DPD No.1 高钙 ^{e)}	片剂 / 250	515741BT
DPD No.1 高钙 ^{e)}	片剂 / 500	515742BT
DPD No.3 高钙 ^{e)}	片剂 / 100	515730BT
DPD No.3 高钙 ^{e)}	片剂 / 250	515731BT
DPD No.3 高钙 ^{e)}	片剂 / 500	515732BT
DPD No.3 HR Evo	片剂 / 100	511920BT
DPD No.3 HR Evo	片剂 / 250	511921BT
DPD No.3 HR Evo	片剂 / 500	511922BT

取样

1. 在样本制备中, 通过移液和摇动来避免氯的排气。
2. 取样后必须立即进行分析。

准备

1. 清洗比色杯：
由于许多家用清洁剂（例如洗碗用洗涤剂）含有还原剂，所以测定的氯结果可能会不足。为了排除这种测量误差，玻璃器皿应无氯。为此，将玻璃器皿在次氯酸钠溶液（0.1 g/L）下存放 1 小时，然后用去离子水（软化水）彻底冲洗。
2. 对于游离氯和总氯的单独测定，使用一套相应单独的比色杯是有意义的（参见 EN ISO 7393-2，第 5.3 段）。
3. DPD 显色发生在 pH 值在 6.2 至 6.5 时。因此该试剂含有用于调节 pH 值的缓冲液。但在分析前（用 0.5 mol/L 硫酸或 1 mol/L 氢氧化钠溶液）必须将强碱性或酸性水的 pH 范围调节到 6 和 7 之间。

ZH

备注

1. Evo 片剂可以作为相应标准片剂的替代品（如 DPD No.3 Evo 代替 DPD No.3）。



进行测定 余氯 HR 片剂法

选择设备中的方法。

ZH



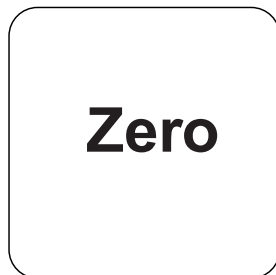
用 **10 mL** 样本填充 24 mm 比色杯。



密封比色杯。



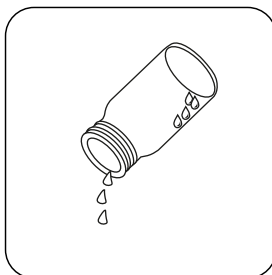
将样本比色杯放入测量轴中。
注意定位。



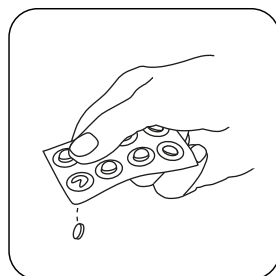
按下 **ZERO** 按钮。



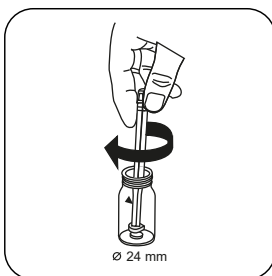
从测量轴上取下比色杯。



将比色杯倒空。



加入 **DPD No. 1 HR** 片剂。



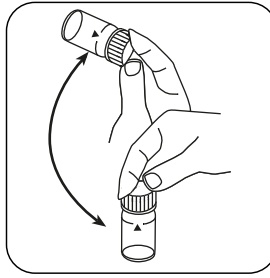
用轻微的扭转压碎片剂。



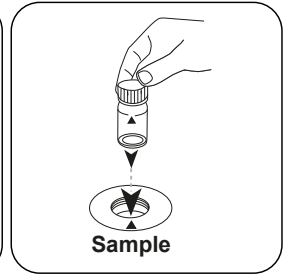
用样本将比色杯填充至 **10 mL** 刻度处。



密封比色杯。



通过旋转溶解片剂。

将样本比色杯放入测量轴中。
注意定位。

ZH

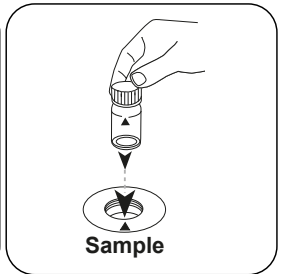
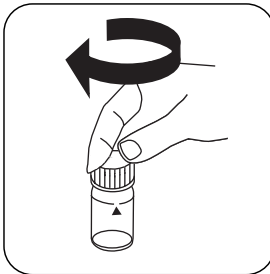
Test

按下 **TEST (XD: START)** 按钮。

结果在显示屏上显示为 mg / l 余氯。

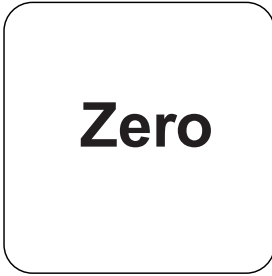
进行测定 总氯 HR 片剂法

选择设备中的方法。

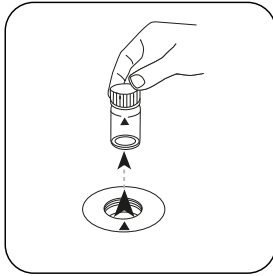
用 **10 mL** 样本填充 24 mm 比色杯。
密封比色杯。将样本比色杯放入测量轴中。
注意定位。



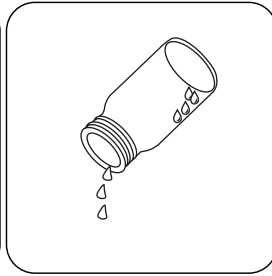
ZH



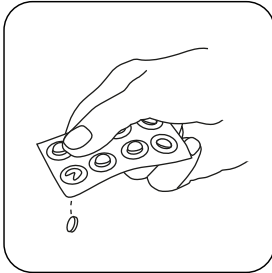
按下 **ZERO** 按钮。



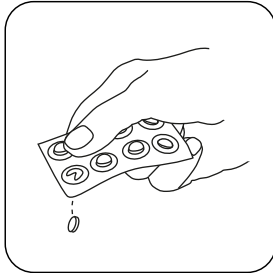
从测量轴上取下比色杯。



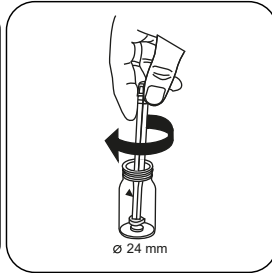
将比色杯倒空。



加入 **DPD No. 1 HR** 片剂。



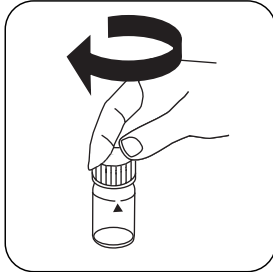
加入 **DPD No. 3 HR** 片剂。



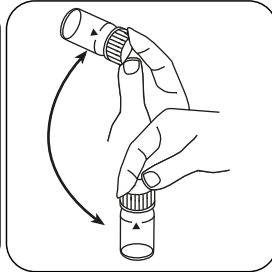
用轻微的扭转压碎片剂。



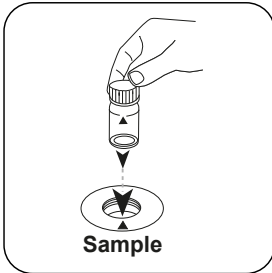
用样本将比色杯填充至 **10 mL** 刻度处。



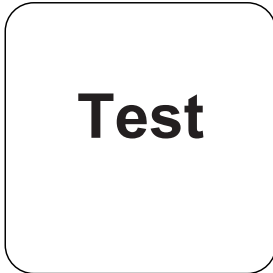
密封比色杯。



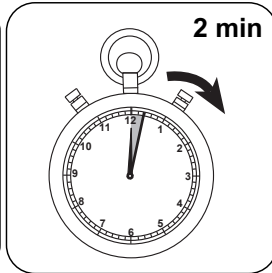
通过旋转溶解片剂。



将样本比色杯放入测量轴中。注意定位。




按下 **TEST (XD: START)** 按钮。



等待 **2 分钟** 反应时间。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 **mg / l 总氯**。



化学方法

DPD

附錄

干扰说明

持续干扰

- 存在于样本中的所有氧化剂都像氯一样反应，导致多重结果。

可消除干扰

- 铜和铁 (III) 的干扰必须通过 EDTA 消除。
- 对于高钙含量*和/或高电导率*的样本，使用试剂片可能会导致样本浑浊和相关的测量误差。在这种情况下，可选用试剂片 DPD 编号1 高钙和试剂片 DPD 编号3 高钙。
*不能给出精确值，因为浑浊的形成取决于样本水的类型和组成。

一致性

EN ISO 7393-2

^{a)} 测定余氯，总氯和结合氯 | ^{b)} 替代试剂，取代 DPD No. 1/No. 3 试剂，用于由高浓度钙离子和/或高电导率引起的浑浊水样分析 | * 含搅拌棒，10cm

ZH



T 铜

M150

0.05 - 5 mg/L Cu^{a)}

Cu

双喹啉

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
铜 No.1	片剂 / 100	513550BT
铜 No.1	片剂 / 250	513551BT
铜 No.2	片剂 / 100	513560BT
铜 No.2	片剂 / 250	513561BT
套件铜 No.1/No.2 [#]	各100次	517691BT
套件铜 No.1/No.2 [#]	各250次	517692BT
ValidCheck 铜 2 mg/l	1 片	48141525

准备

1. 在分析前应将强碱性或酸性水的 pH 从4到6 左右。

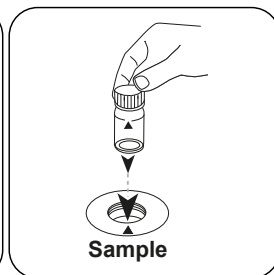
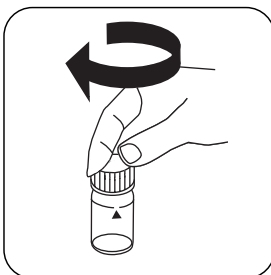
进行测定 余铜 片剂法

选择设备中的方法。

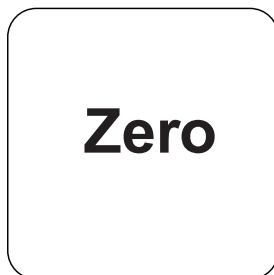
另外选择测定：余铜



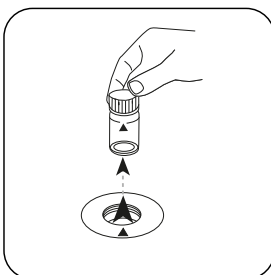
用 10 mL 样本填充 24 mm 比色杯。
密封比色杯。



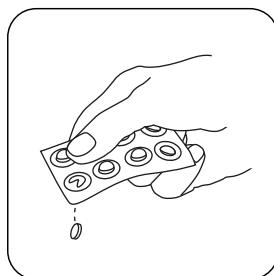
将样本比色杯放入测量轴中。注意定位。



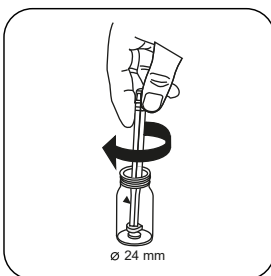
按下 ZERO 按钮。



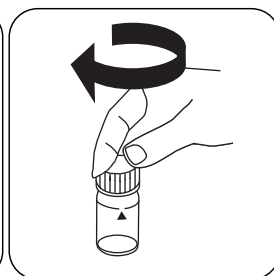
从测量轴上取下比色杯。



加入 **COPPER No. 1** 片剂。



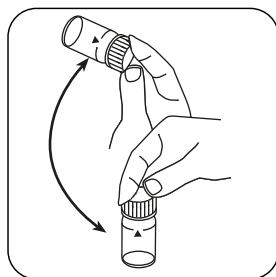
用轻微的扭转压碎片剂。



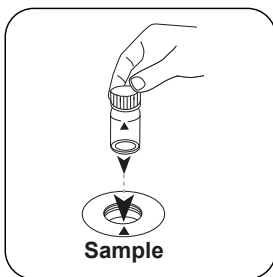
密封比色杯。



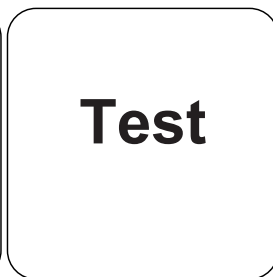
ZH



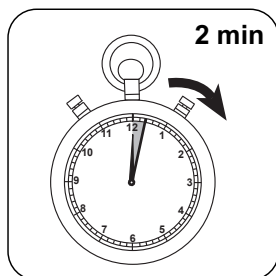
通过旋转溶解片剂。



将样本比色杯放入测量轴中。注意定位。



按下 **TEST (XD: START)** 按钮。



等待 2 分钟反应时间。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 mg/l 余铜。

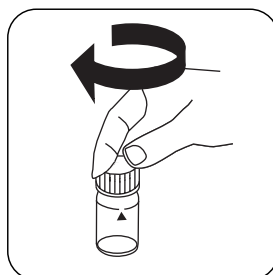
进行测定 总铜 片剂法

选择设备中的方法。

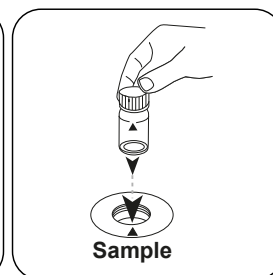
另外选择测定：总铜



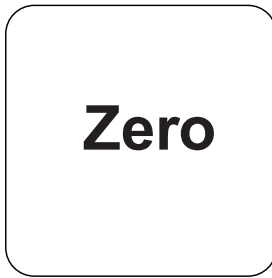
用 10 mL 样本填充 24 mm 比色杯。



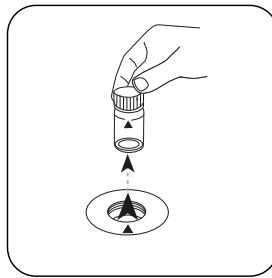
密封比色杯。



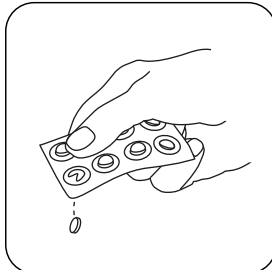
将样本比色杯放入测量轴中。注意定位。



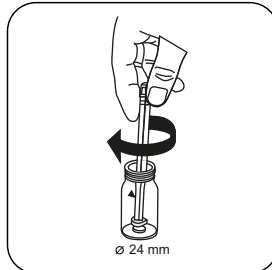
按下 **ZERO** 按钮。



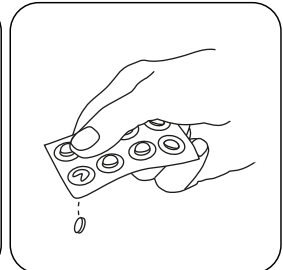
从测量轴上取下比色杯。



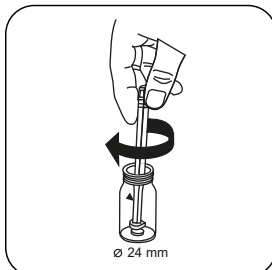
加入 **COPPER No. 1** 片剂。



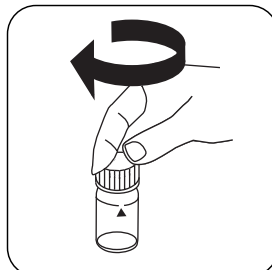
用轻微的扭转压碎片剂并溶解。



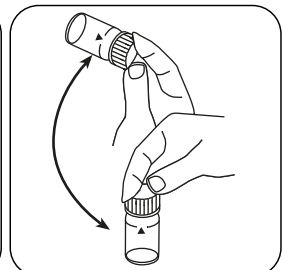
加入 **COPPER No. 2** 片剂。



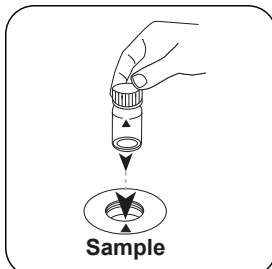
用轻微的扭转压碎片剂。



密封比色杯。



通过旋转溶解片剂。



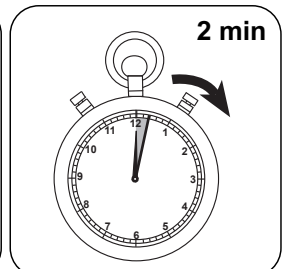
将样本比色杯放入测量轴中。注意定位。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 mg / l 总铜。



按下 **TEST (XD: START)** 按钮。



等待 **2 分钟** 反应时间。



ZH

化学方法

双喹啉

附錄

干扰说明

持续干扰

1. 氰化物CN⁻和银Ag⁺会干扰测定。

方法验证

检出限	0.05 mg/L
测定下限	0.15 mg/L
测量上限	5 mg/L
灵敏度	3.8 mg/L / Abs
置信范围	0.026 mg/L
标准偏差	0.011 mg/L
变异系数	0.42 %

参考文献

Photometrische Analyse, Lange/Vedjelek, Verlag Chemie 1980

^{a)} 测定余氯，总氯和结合氯 | * i含搅拌棒, 10cm



T 氰

M160

10 - 160 mg/L CyA

CyA

三聚氰胺

材料

所需材料 (部分可選) :

ZH

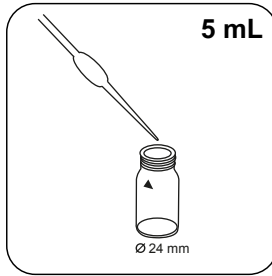
试剂	包装单位	货号
CYA 测试	片剂 / 100	511370BT
CYA 测试	片剂 / 250	511371BT
软化水	100 mL	461275
软化水	250 mL	457022

备注

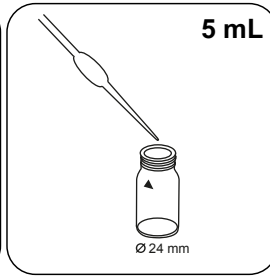
1. 氰尿酸引起非常细微的分散浑浊和乳白色的外观。单个颗粒不是由于氰尿酸的存在造成的。

进行测定 用片剂进行氰尿酸测试

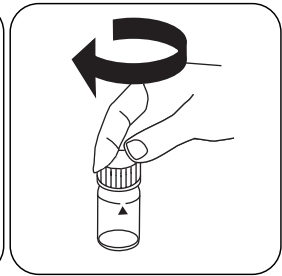
选择设备中的方法。



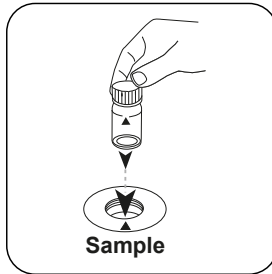
用 5 mL 去离子水填充 24 mm 比色杯。



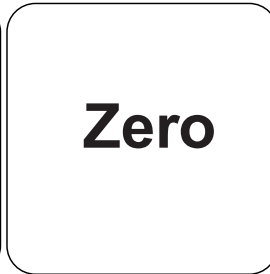
添加 5 mL 样本到比色杯中。



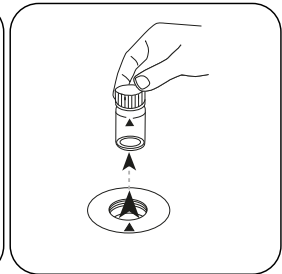
密封比色杯。



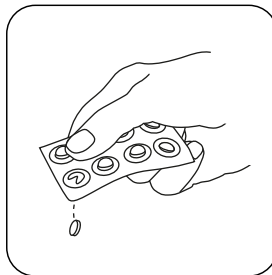
将样本比色杯放入测量轴中。注意定位。



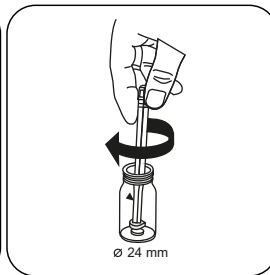
按下 **ZERO** 按钮。



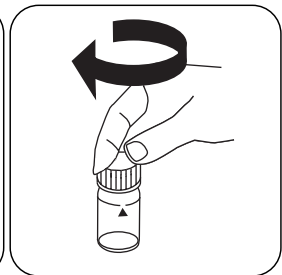
从测量轴上取下比色杯。



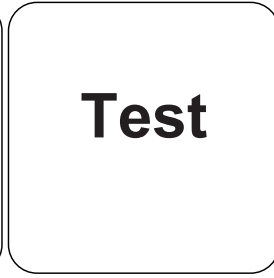
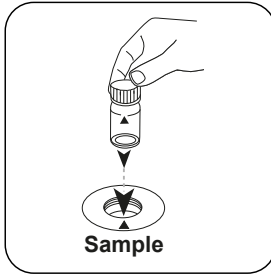
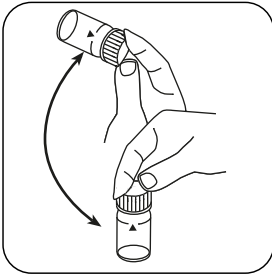
加入 **CyA-Test** 片剂。



用轻微的扭转压碎片剂。



密封比色杯。



ZH

通过旋转混合内容物（至少 60 秒，直到药片完全溶解）。将样本比色杯放入测量轴中。注意定位。

结果在显示屏上显示为 mg / l 三聚氰酸。

按下 **TEST (XD: START)** 按钮。



化学方法

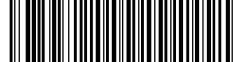
三聚氰胺

干扰说明

持续干扰

1. 未溶解的颗粒可能会导致结果过高。因此，完全溶解剂很重要。

ZH



T 铁

M220

0.02 - 1 mg/L Fe

FE

Ferrozine/巯乙酸盐

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
铁 II LR (Fe ²⁺)	片剂 / 100	515420BT
铁 II LR (Fe ²⁺)	片剂 / 250	515421BT
铁 LR (Fe ²⁺ und Fe ³⁺)	片剂 / 100	515370BT
铁 LR (Fe ²⁺ und Fe ³⁺)	片剂 / 250	515371BT

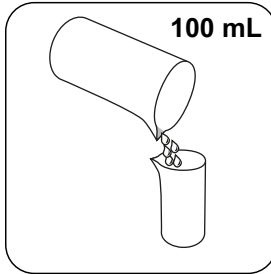
准备

1. 在必要情况下, 已用有机化合物作为腐蚀抑制剂处理的水须被氧化, 从而破坏铁复合物。为此, 将 100 ml 样本与 1 ml 浓硫酸和 1 ml 浓硝酸混合并蒸发至一半。冷却后进行消解。

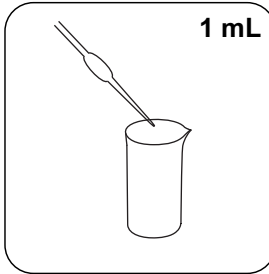
备注

1. 用这种方法测定总溶解的 Fe²⁺ 和 Fe³⁺。
2. 为了测定 Fe²⁺, 使用 IRON (II) LR 片剂代替 IRON LR 片剂。

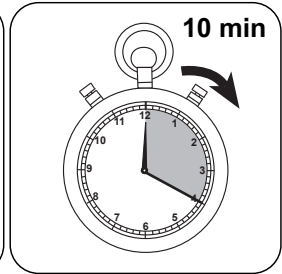
消解



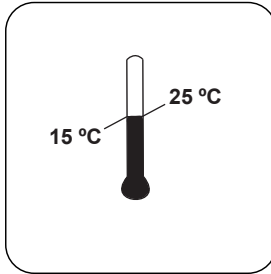
用 100 mL 样本填充合适的样本容器。



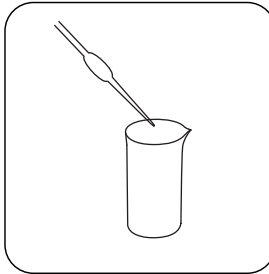
加入 1 mL 浓硫酸 ($\geq 95\%$)。



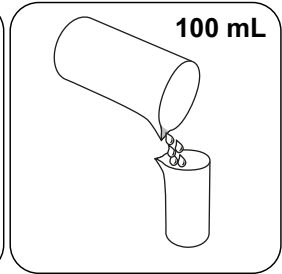
将样本加热 10 分钟，或一直加热直到其完全溶解。



将样本冷却到室温。



将样本的 pH 值从氨溶液 (10-25%) 调节到 3-5。



将样本用去离子水填充至 100 mL。

使用该样本来分析 总溶解铁。

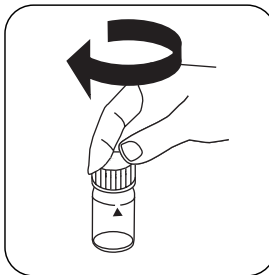
进行测定 铁 (II,III) ，用片剂溶解

选择设备中的方法。

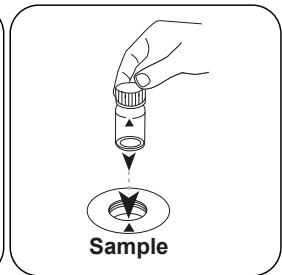
为了测定 溶解和未溶解铁，进行 中所述的消解。



用 10 mL 样本填充 24 mm 比色杯。



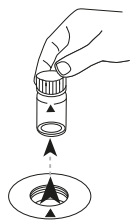
密封比色杯。



将样本比色杯放入测量轴中。注意定位。

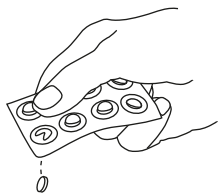


Zero

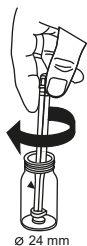


按下 **ZERO** 按钮。

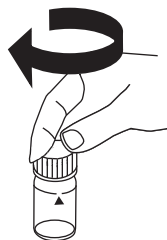
从测量轴上取下比色杯。



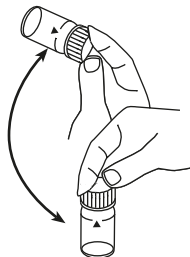
加入 **IRON LR** 片剂。



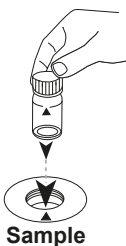
用轻微的扭转压碎片剂。



密封比色杯。



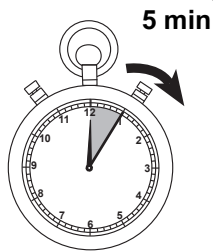
通过旋转溶解片剂。



将样本比色杯放入测量轴中。注意定位。

Test

按下 **TEST (XD: START)** 按钮。



等待 **5 分钟** 反应时间。

反应时间结束后，自动进行测量。

结果在显示屏上显示为 **mg / l 铁**。

化学方法

Ferrozine/巯乙酸盐

附錄

干扰说明

可消除干扰

1. 铜的存在使测量结果增加了 10 %。样本中铜浓度为 10 mg/L 时，测量结果增加 1 mg/L 铁。
干扰可以通过添加硫脲来消除

方法验证

检出限	0.01 mg/L
测定下限	0.016 mg/L
测量上限	1 mg/L
灵敏度	0.92 mg/L / Abs
置信范围	0.013 mg/L
标准偏差	0.005 mg/L
变异系数	1.23 %

参考文献

Photometrische Analyse, Lange/ Vjedelek, Verlag Chemie 1980, S. 102



T pH 值

M330

6.5 - 8.4 pH

PH

苯酚红

材料

所需材料 (部分可選) :

ZH

试剂	包装单位	货号
酚红光度计	片剂 / 100	511770BT
酚红光度计	片剂 / 250	511771BT
酚红光度计	片剂 / 500	511772BT

备注

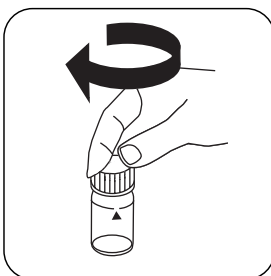
1. 对于光度 pH 值测定，只应使用标有 PHOTOMETER 的带有黑色烫印的 PHENOL RED 片剂。

进行测定 pH 值片剂

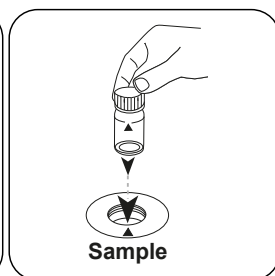
选择设备中的方法。



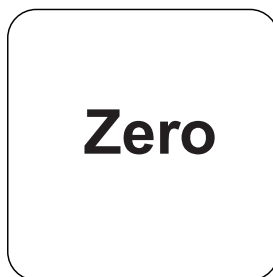
用 **10 mL** 样本填充 24 mm 比色杯。



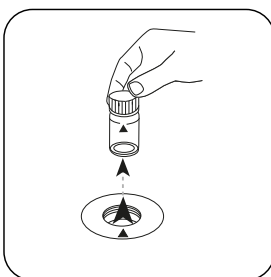
密封比色杯。



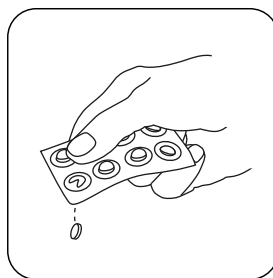
将样本比色杯放入测量轴中。注意定位。



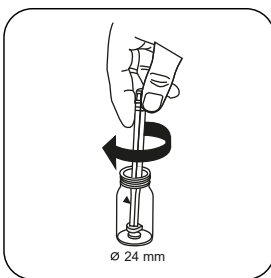
按下 **ZERO** 按钮。



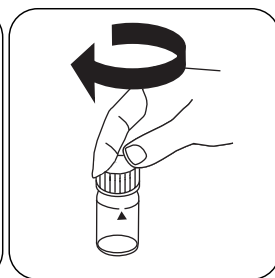
从测量轴上取下比色杯。



加入 **PHENOL RED PHOTOMETER** 片剂。



用轻微的扭转压碎片剂。



密封比色杯。

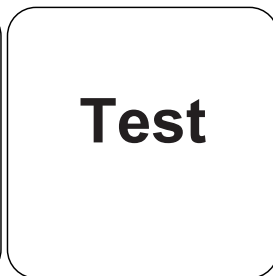
ZH



通过旋转溶解试剂。



将样本比色杯放入测量轴中。注意定位。



按下 **TEST (XD: START)** 按钮。

结果在显示屏上显示为 pH 值。

ZH

化学方法

苯酚红

附录

干扰说明

持续干扰

1. 碳酸盐硬度*低的水样可能会得出错误的 pH 值。
* $K_{S_{4,3}} < 0,7 \text{ mmol/l} \triangleq \text{总碱度} < 35 \text{ mg/L CaCO}_3$.

可消除干扰

1. pH 值低于 6.5 和高于 8.4 可导致测量范围内的结果。建议使用可信度测试 (pH 计)。
2. 盐误差：
对于盐含量高达 2 g/L，试剂片的盐含量不会引起明显的盐误差。对于较高的盐含量，测量值应进行如下校正：

样本的盐 含量以 g/ L 为单位	30 (海 水)	60	120	180
校正	-0,15 ¹⁾	-0,21 ²⁾	-0,26 ²⁾	-0,29 ²⁾

¹⁾ 根据 Kolthoff (1922)

²⁾ 根据 Parson 和 Douglas (1926)

参考文献

Colorimetric Chemical Analytical Methods, 9th Edition, London



L pH 值

M331

6.5 - 8.4 pH

PH

苯酚红

材料

所需材料 (部分可选) :

ZH

试剂	包装单位	货号
酚红溶液	15 mL	471040
酚红溶液	100 mL	471041
酚红溶液 6 件装	1 片	471046

准备

1. 由于液滴大小不同, 测量结果可能会比使用片剂时有更大的偏差。使用移液管 (0.18 ml 相当于 6 滴) 时这种偏差可以最小化。

备注

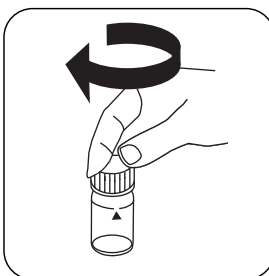
1. 使用后滴瓶必须立即用相同颜色的瓶盖重新密封。
2. 将试剂冷藏在 +6 °C 至 +10 °C。

进行测定 pH 值液剂

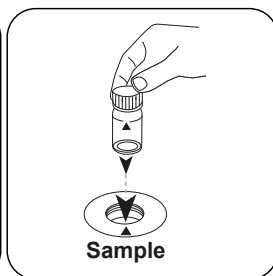
选择设备中的方法。



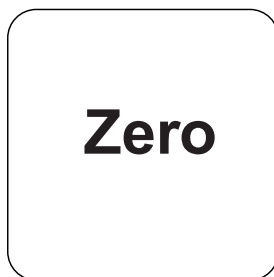
用 **10 mL** 样本填充 24 mm 比色杯。



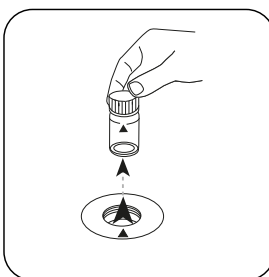
密封比色杯。



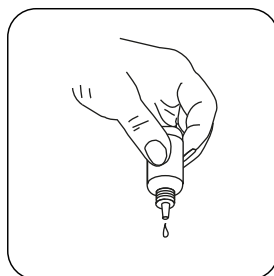
将样本比色杯放入测量轴中。注意定位。



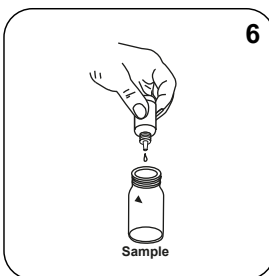
按下 **ZERO** 按钮。



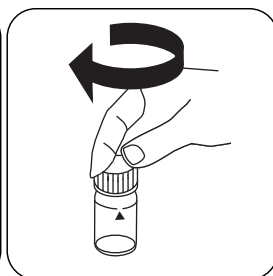
从测量轴上取下比色杯。



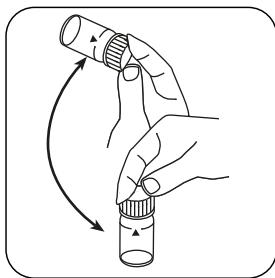
垂直握住滴瓶，慢慢加入相同大小的滴剂。



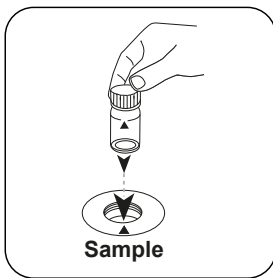
将 **6 滴 PHENOL Red-Lösung** 添加到样本比色杯中。



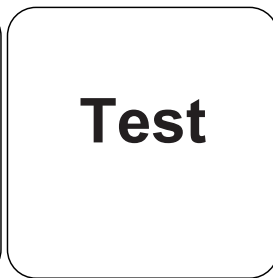
密封比色杯。



通过旋转混合内容物。



将样本比色杯放入测量轴中。注意定位。



按下 **TEST (XD: START)** 按钮。

结果在显示屏上显示为 pH 值。

ZH

化学方法

苯酚红

附录

干扰说明

可消除干扰

1. 盐误差：通过盐含量校正样本的测量值（平均值）：

样本盐含量	校正
30 g/L (海水)	-0,15 ¹⁾
60 g/L	-0,21 ²⁾
120 g/L	-0,26 ²⁾
180 g/L	-0,29 ²⁾

¹⁾根据 Kolthoff (1922)

²⁾根据 Parson 和 Douglas (1926)

3. 分析氯化水时存在的残余氯含量会影响液体试剂的显色反应。在添加 PHENOL RED 溶液之前，向样本溶液中加入一小块硫代硫酸钠晶体 ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5 \text{H}_2\text{O}$) 来防止这种情况。

参考文献

Colorimetric Chemical Analytical Methods, 9th Edition, London

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