

# Polyacrylate L

M338

1 - 30 mg/L Polyacryl

**POLY** 

**Turbidity** 

#### Instrument specific information

The test can be performed on the following devices. In addition, the required cuvette and the absorption range of the photometer are indicated.

Instrument Type	Cuvette	λ	Measuring Range
MD 100, MD 110	ø 24 mm	530 nm	1 - 30 mg/L Polyacryl
MD 600, MD 610, MD 640, XD 7000, XD 7500	ø 24 mm	660 nm	1 - 30 mg/L Polyacryl

#### **Material**

Required material (partly optional):

Reagents	Packaging Unit	Part Number
Cartouche C18	1 pc.	56A020101
KS173-P2-2,4 Dinitrophenol Indicator	65 mL	56L017365
QAC Buffer QA2	65 mL	56L018365
Polyacrylate L Reagent Set	1 pc.	56R019165
KS336-Propan-2-ol, 65 mL	65 mL	56L033665

The following accessories are required.

Accessories	Packaging Unit	Part Number
Pipette, 1000 μl	1 pc.	365045
Pipette tips, 0,1-1 ml (blue), 1000 pc.	1 pc.	419073

## **Application List**

- · Cooling Water
- · Boiler Water
- · Raw Water Treatment

### **Preparation**

· Preparing the cartridge:



- Remove the plunger from a suitable syringe. Attach the C18 cartridge to the syringe cylinder.
- 2. Add 5 ml of KS336 (propane-2-ol) to the syringe cylinder.
- 3. Using the plunger, press the solvent by drop through the cartridge.
- 4. Remove the solvent that has passed through.
- 5. Remove the plunger again. Fill the syringe cylinder with 20 ml of deionised water.
- With the help of the plunger, press the contents through the cartridge drop by drop.
- 7. Discard the deionised water that has flowed through.
- 8. The cartridge is now ready for use.

#### **Notes**

- If little or no turbidity is present at correct dose concentrations, the sample will need a pre-concentration step in order to detect this level of polyacrylate/polymer.
- Anomalous results occur when interferences are present as part of the sample components or from sample contaminants. In this case, the interference will need to be eliminated.
- 3. This test has been calibrated using polyacrylic acid 2'100 sodium salt in the range 1-30 mg/L. Other polyacrylates/polymers will give differing responses and therefore the test range will vary.

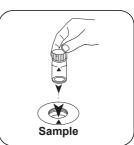


### **Determination of Polyacrylate with liquid reagent**

Select the method on the device.

For this method, a ZERO measurement does not have to be carried out every time on the following devices: XD 7000, XD 7500





Fill 24 mm vial with 10 mL Close vial(s). sample.

Place sample vial in the sample chamber. Pay attention to the positioning.





Press the **ZERO** button.

Remove the vial from the sample chamber.

For devices that require no ZERO measurement, start here.



Place 1 mL (25 drops) Polyacrylate Buffer A1 solution in the sample cuvette.



Close vial(s).



Invert several times to mix the contents.





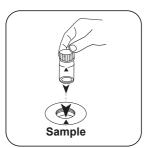
Place 1 mL (25 drops)
Polyacrylate Precipitant
A2 solution in the sample
cuvette.



Close vial(s).



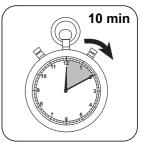
Invert several times to mix the contents.



Place **sample vial** in the sample chamber. Pay attention to the positioning.



Press the **TEST** (XD: **START**)button.



Wait for 10 minute(s) reaction time.

Once the reaction period is finished, the measurement takes place automatically.

The result in mg/L Polyacryl acid 2100 sodium salt appears on the display.



#### **Chemical Method**

Turbidity

### **Appendix**

### Calibration function for 3rd-party photometers

Conc. =  $a + b \cdot Abs + c \cdot Abs^2 + d \cdot Abs^3 + e \cdot Abs^4 + f \cdot Abs^5$ 

	ø 24 mm	□ 10 mm
а	5.21463 • 10 <sup>-1</sup>	5.21463 • 10 <sup>-1</sup>
b	3.45852 • 10 <sup>+1</sup>	7.43583 • 10+1
С	-2.38855 • 10 <sup>+1</sup>	-1.10411 • 10 <sup>+2</sup>
d	1.52167 • 10 <sup>+1</sup>	1.51229 • 10+2
е		
f		

#### **Bibliography**

W.B. Crummett, R.A. Hummel (1963), The Determination of Polyacrylamides in Water, American Water Works Association, 55 (2), pp. 209-219